

ELECTRONIC SUPPORTING INFORMATION

First example of asymmetrical μ -oxo bridged dinuclear iron complex with terpyridine ligand: synthesis, crystal structure and electrochemical properties

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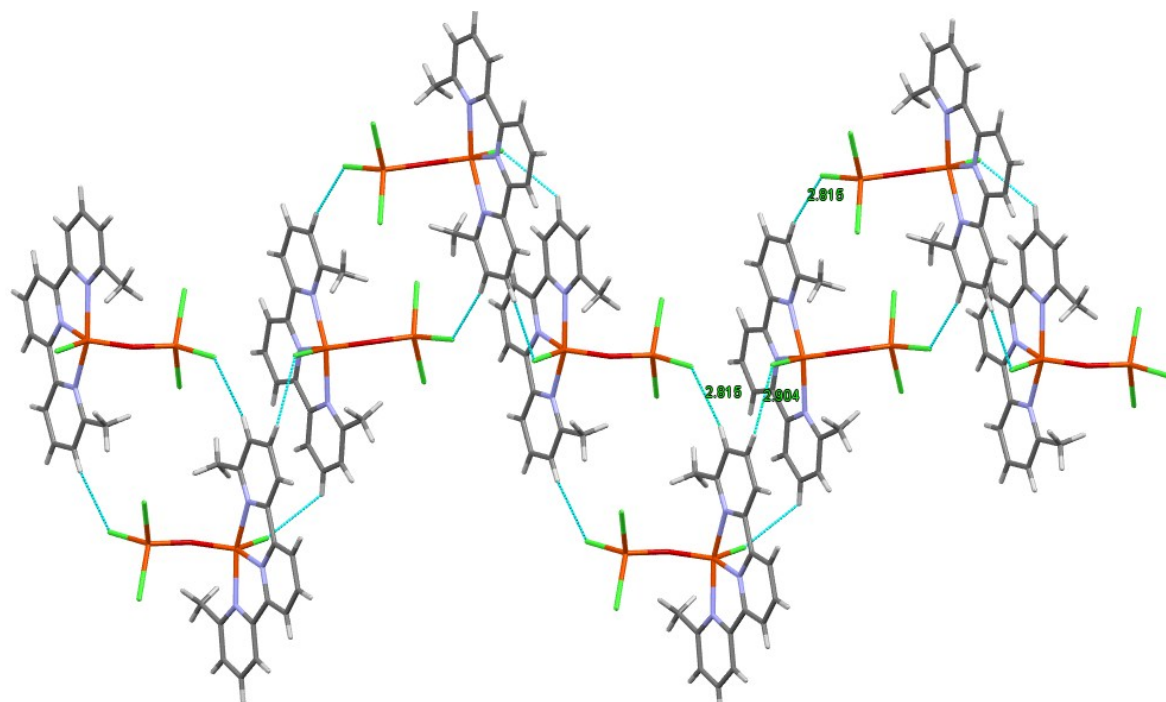


Figure S1. Short contacts in complex **1** that are vital for organization into 1-dimensional zig-zag chain structure.

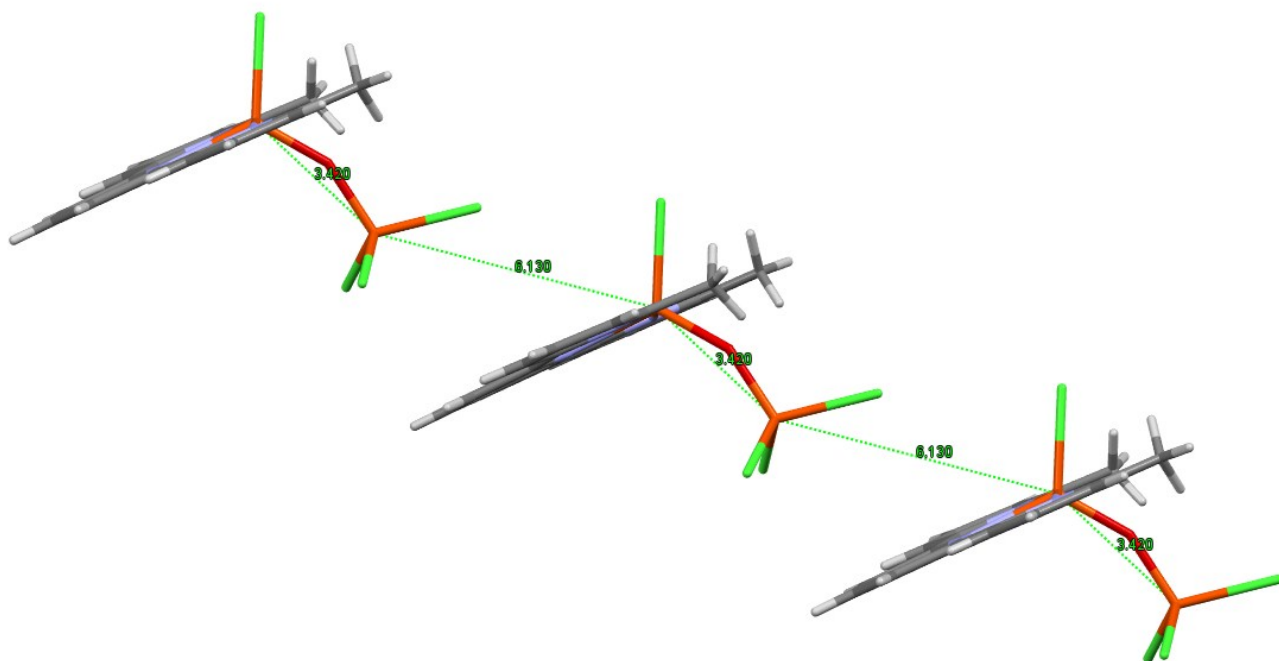


Figure S2. One-dimensional zig-zag chain structure in complex **1**. Short contacts to molecules responsible for organization were omitted for clarity, green lines represent intra- and intermetallic distances

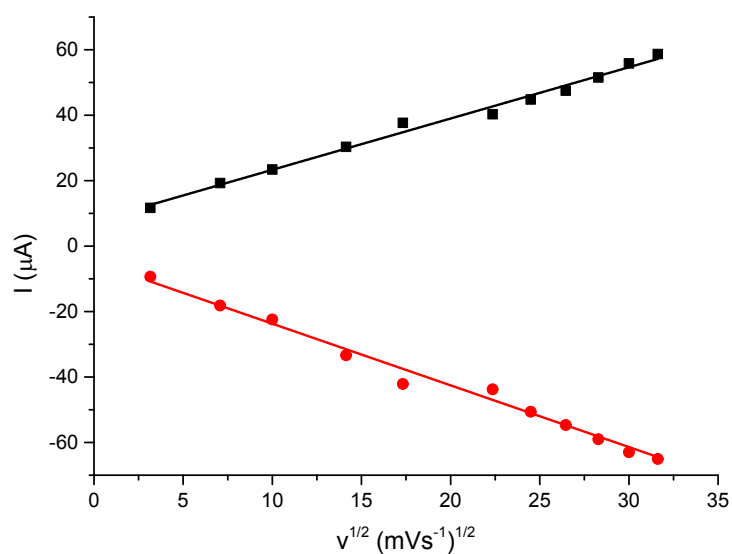


Figure S3. Linear relationship of the anode (■) and cathode (●) peak currents versus the square root of the scanning speed for the complex **2**. $R^2 = 0.989$ and 0.984 for ip_a and ip_c respectively.

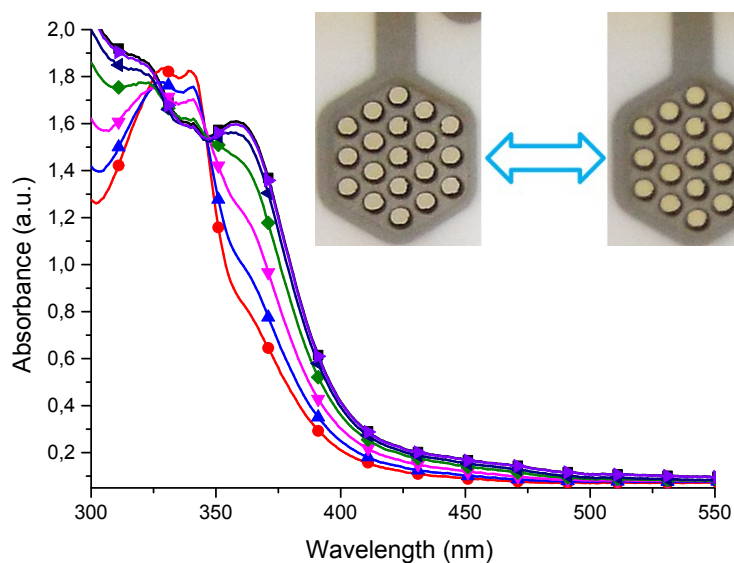


Figure S4. The spectroelectrochemical properties of complex **2** in 0.1M TBAClO₄ solution in acetonitrile as a supporting electrolyte by applying: no potential (■), +100 (●), +200 (▲), +300 (▼), +400 (◆), +500 (◄), and 600 (►) mV potentials versus Ag/AgCl gel reference electrode held for 30 s per potential. Insert: photographs of the Fe²⁺ (left) and Fe³⁺ (right) state of **2** by applying a potential for 1 min.

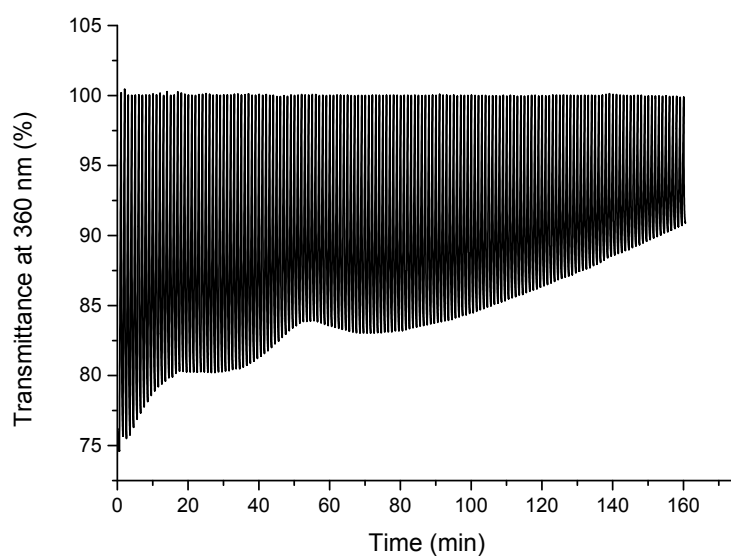


Figure S5. Changes in transmittance of complex **1** measured in anhydrous and deaerated acetonitrile with 0.1 M TBAClO₄ as a supporting electrolyte and monitored at 360 nm when switching between +100 mV and +600 mV potential at 60 s cycles.

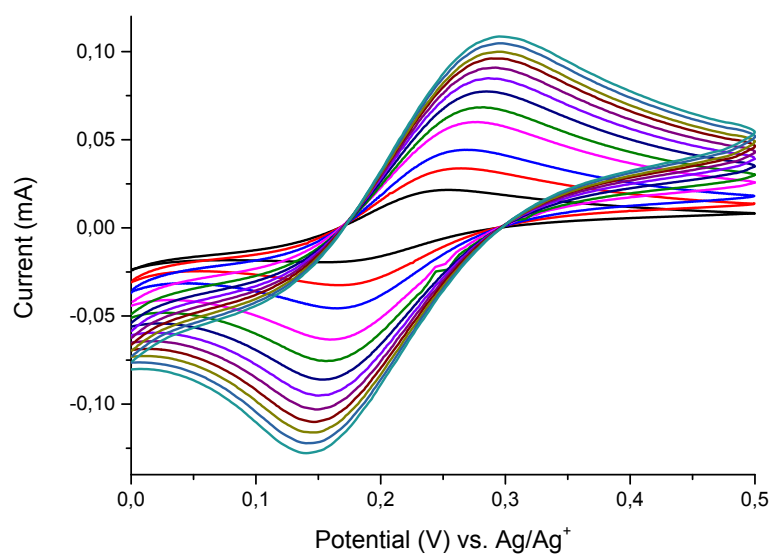


Figure S6. CVs of complex **1** measured in anhydrous and deaerated acetonitrile with 0.1 M TBAClO₄ as a supporting electrolyte at different scan rates.

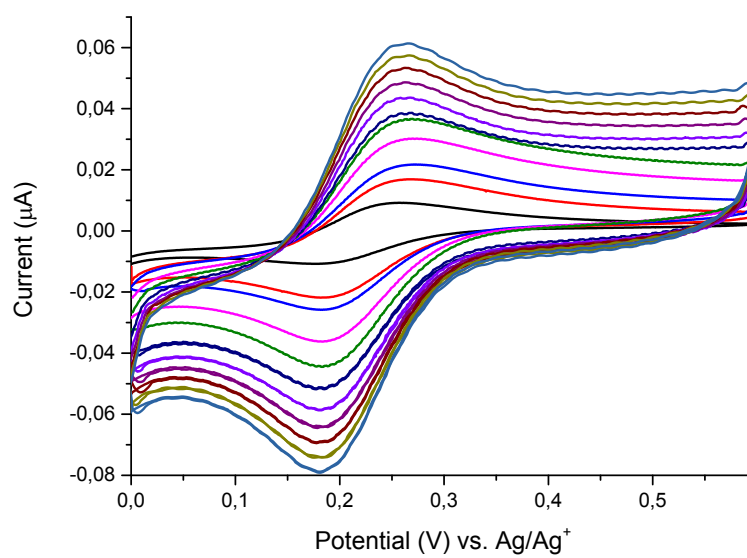


Figure S7. CVs of complex **2** measured in anhydrous and deaerated acetonitrile with 0.1 M TBAClO₄ as a supporting electrolyte at different scan rates.