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Preparation of Hollow Fe₃O₄/Pd@C NCs to stabilize

subminiature Pd nanoparticles for reduction of 4-nitrophenol

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Fig. S1 TEM image of the PVP-Fe₃O_{4.}



Fig. S2 Elemental mapping images of Fe (green), Pd(yellow), N (pink), C (blue) and Zn (red) of (A) Fe₃O₄/Pd²⁺@ZIF-8, (B) Fe₃O₄/Pd²⁺@PDA NCs.



Fig. S3 M-H curve of OA-stabilized Fe₃O₄ NPs.



Fig. S4 XRD patterns of Fe₃O₄/Pd²⁺@ZIF-8 (a) and Fe₃O₄/Pd²⁺@PDA NCs (b).



Fig. S5 Raman spectrum of Fe₃O₄/Pd@C NCs.



Fig. S6 FT-IR spectrums of Fe₃O₄/Pd²⁺@ZIF-8 (black), Fe₃O₄/Pd²⁺@PDA NCs (blue), Fe₃O₄/Pd@C NCs and pure PDA (green).

Table S1 Elemental analysis results in the XPS spectrum of $Fe_3O_4/Pd@C$ NCs catalysts.

Elements	At. /%	
С	88.80	
Ν	7.01	
Fe	3.89	
Pd	0.30	



Fig. S7 UV–vis absorption spectra of 4-NP (a); 4-NP + NaBH₄ (b) and 4-AP (c).



Fig. S8 Conversion of 4-NP in 5 successive cycles of reduction with 5.0 mg of $Fe_3O_4/Pd@C$ NCs catalysts.



Fig. S9 TEM image of Fe₃O₄/Pd@C NCs after 5 cycles.

catalyst	$k(s^{-1})$	Ref.
Fe ₃ O ₄ /Pd@C NCs	3.26×10 ⁻³	This work
p(AMPS)–Co composite	2.00×10 ⁻³	46
rGO/Pd-Fe ₃ O ₄ /PPy	3.20×10-3	47
Ni NPs	2.67×10 ⁻³	48
Pd–graphene nanohybrid	2.67×10 ⁻³	49
Au-Pd bimetallic NPs/graphene	1.45×10-2	50
Dumbbell-like Au-Fe ₃ O ₄	1.45×10 ⁻²	51

 Table S2 Comparison of rate constant values for the 4-NP reduction to 4-AP using various catalysts