

1 **Supporting information**

2 **Amorphous calcium organophosphates nanoshells as**
3 **potential carriers for drug delivery to Ca²⁺-enriched**
4 **surfaces.**

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8 ***Characterization methods***

9 *Wide angle X-ray scattering* (WAXS) experiments were performed using a Siemens D500 diffractometer
10 equipped with a conventional X-ray tube operating at a wavelength of $\lambda = 0.15418$ nm (Cu K α).

11 Alternatively, the CaPLiLX sample was left in contact with 0.05M CaCl₂ solution for 24 hs before analysis.

12 *X-Ray Diffraction analysis* (XRD) was performed with a Philips PW 1011/00 diffractometer using CuK α
13 radiation operated at 30 mA and 35 kV. XRD patterns were scanned in steps of 0.02°/2s in the range from 5°
14 to 60°. The CaPLiLX sample was deposited on glass slide with and without previously deposited Ca²⁺ and
15 was left to dry on a dessicator.

16 *Transmission electron microscopy* images of the CaP covered, and uncovered liposomes samples were taken
17 with a JEOL JEM 1200 EX II microscope. Samples were prepared by dripping on carbon-coated 300-mesh
18 copper grid and water evaporated in air. Alternatively, the samples were stained with 2% phosphotungstic
19 acid.

20 *High-resolution transmission electron microscopy* images of CaPLiLX were obtained with a FEI Talos 1162
21 microscope equipped with an energy-dispersive X-ray spectrometer (EDS, Si(Li) Jeol detector). The
22 microscope was used in scanning TEM mode (STEM). The sample was prepared by deposition of the
23 suspension on carbon-coated 400 mesh copper grid and evaporated under air. The images were analyzed by
24 Image J software.

25 *Surface composition of nanoshells* was obtained by Attenuated Total Reflectance – Fourier Transformed
26 Infrared spectroscopy (ATR-FTIR) in an Agilent Cary Series 630 spectrometer equipped with ZnSe accessory.
27 In all cases, samples were dropped on potassium bromide solids and was left to dry on air. Fifty scans over
28 the range of 600–4000 cm^{-1} were performed at a resolution of 2 cm^{-1} .

29 *Nanoshell size distribution and surface charge* were determined by dynamic light scattering (DLS) and
30 electrophoretic mobility (μ_e) measurements, respectively. Aqueous 0.1 M KCl samples of LiLX, and
31 CaPLiLX in the absence and in the presence of 0.05M of Ca^{2+} and Mg^{2+} were measured without filtration at
32 25 °C within a week with a Zetasizer Nano (NanoZSizerZEN3600, Malvern, U.K.) equipped with a He-Ne
33 633nm laser with a cell drive voltage of 30 V using a monomodal analysis model.

34 *Photoluminescence measurements* were performed using a Jobin-Yvon Spex Fluorolog FL3-11 spectrometer
35 equipped with a Xe lamp as the excitation source, a monochromator with 1 nm bandpass gap for selecting the
36 excitation and emission wavelengths, and a red sensitive R928 PM detector. The spectra were corrected for
37 the wavelength-dependent sensitivity of the detector and the source and the emission spectra were corrected
38 for Raman scattering by using the solvent emission spectrum. Fluorescence lifetime and time-resolved
39 anisotropy measurements were completed with a TCSPC (time-correlated single-photon counting) with LED
40 excitation at 341 nm The anisotropy curves were fitted according to $r(t) = r_\infty + r_0 x e^{-t/\theta}$, where $r(t)$ is the
41 anisotropy, r_∞ is the residual anisotropy; r_0 is the fundamental anisotropy and θ is the rotational correlation
42 time.

43 Time resolved emission spectroscopy (TRES) was performed, and results were fitted to an exponential model
44 until optimal values χ^2 and residuals were obtained. The emission spectrum associated with each fluorescence
45 lifetime may be obtained taking the contribution of each decay lifetime to the overall emission at a given
46 wavelength, weighted by the emission intensity at the emission maximum.

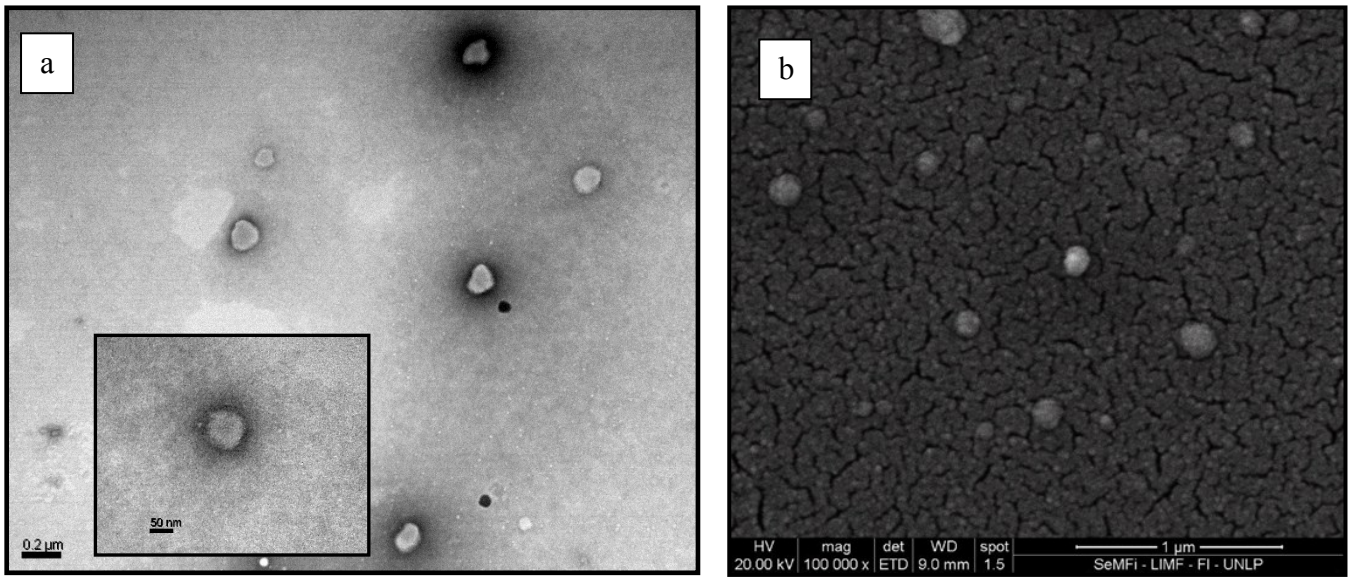
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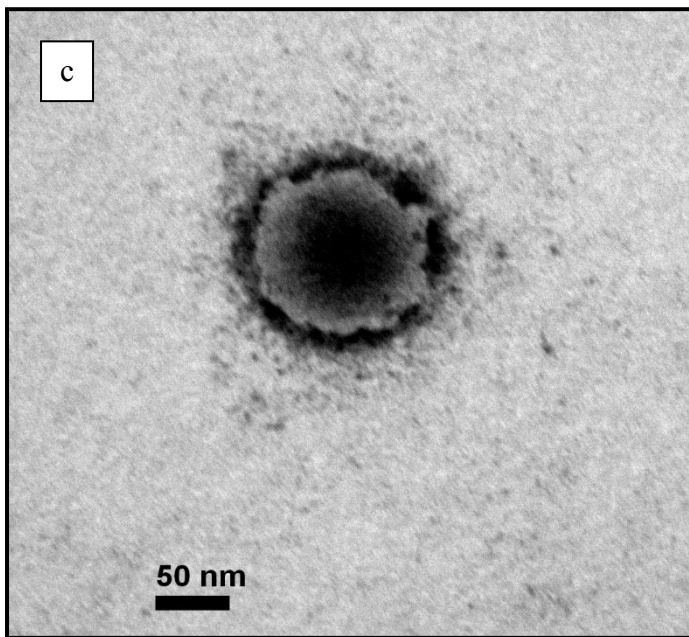
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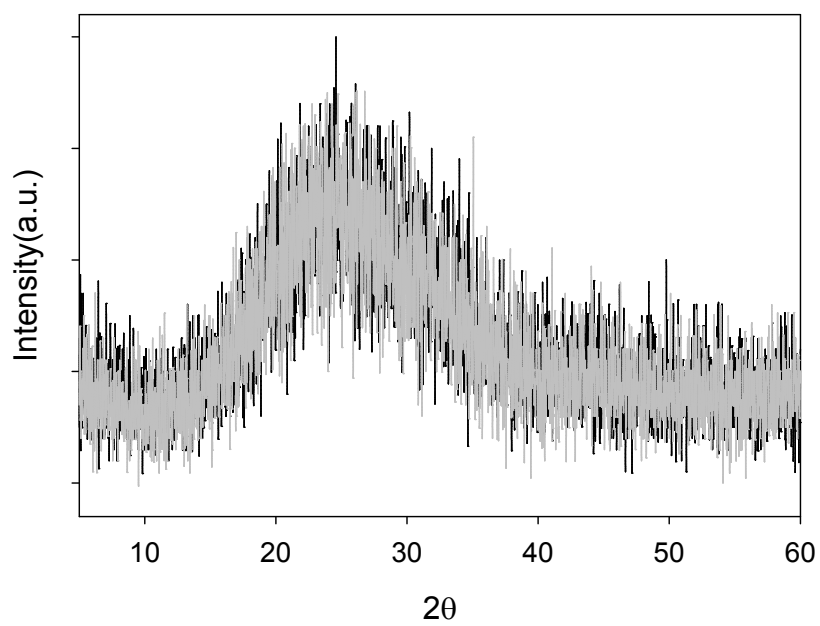
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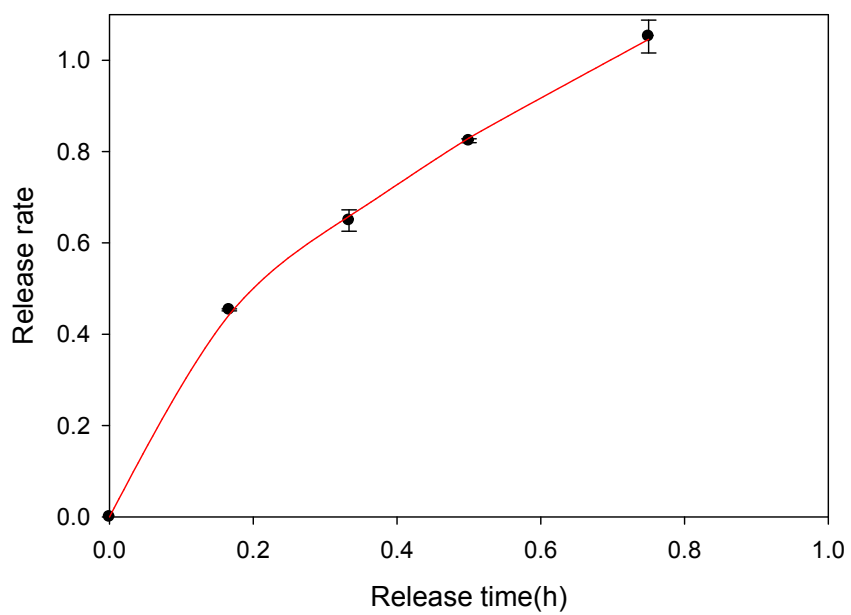
67 **Fig. S1.** TEM (a) and SEM (b) images of CaPLiX and LiLX (inset (a)). Core-shell structure of CaPLiX
68 observed by TEM (c).



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70 **Fig. S2.** XRD patterns of CaP nanoshells deposited on glass slides in the presence (black line) and in the
71 absence (grey line) of Ca²⁺.

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74 **Fig. S3.** Release profile for free LX in acetate buffer solution. Line stands for the fitting to Kosmeyer-
75 Peppas model. Error bars stands for SD.

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77 **Table S1.** Parameters obtained from data fitted to Korsmeyer-Peppas and Weibull models drug release models
 78 for the release kinetics of LX from CaPLiX in Simulated body fluid (SBF), phosphate buffer (PBS), and
 79 acetate buffer.

	Korsmeyer-Peppas		Weibull	
	$RR = k_{KP} \times t^n$		$RR = 1 - \exp(-t^b/a)$	
	k_{KP}	n	a	b
SBF	0.55±0.01	0.14±0.02	1.23±0.06	0.13±0.02
Acetate	0.39±0.02	0.11±0.02	2.0±0.1	0.15±0.02
PBS	0.35±0.02	0.11±0.02	2.4±0.2	0.17±0.03
Free LX in PBS	0.59±0.05	0.6±0.1	--	--
Adsorbed LX on CaPLi	0.39±0.01	0.88±0.02	--	--

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81 The fitted parameters were obtained with a coefficient of determination $r^2 > 0.90$.

82 The complete set of data was also fitted to the Weibull model which considers the diffusion of drugs through
 83 porous materials.⁴⁴ In this case, the relation $RR = 1 - \exp(-t^b/a)$ was used, where the value of parameter b is
 84 indicative of the type of release mechanism and a is a scale factor. Fitting the release data to the Weibull
 85 model yields $r^2 > 0.91$ for all media. However, b values < 0.35 obtained for all the samples (Table S1) are not
 86 frequently found in the literature and are attributed to highly disordered materials.

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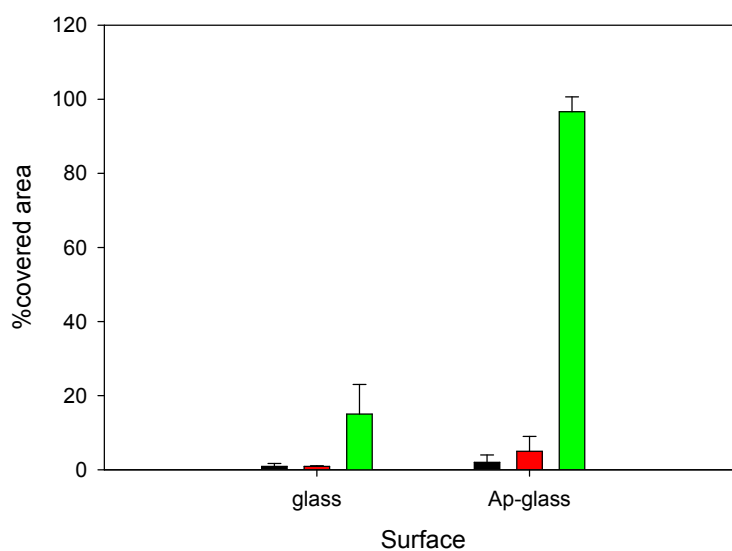
90 **Table S2.** Diameter and electrophoretic mobility of LiLX and CaPLiLX in the absence and the presence of
 91 Ca^{2+} and Mg^{2+} .

<i>Sample</i>	<i>Diameter(nm)(±SD)</i>	<i>Electrophoretic mobility(cm²V⁻¹s⁻¹)(±SD)</i>	<i>PDI</i>
<i>LiLX</i>	100.6 ± 0.6	-4.6 ± 0.2	0.259 ± 0.004
<i>CaPLiLX</i>	171.8 ± 0.6	-4.1 ± 0.1	0.204 ± 0.004
<i>CaPLiLX+Ca²⁺</i>	832 ± 86	-0.49 ± 0.06	0.75 ± 0.05
<i>CaPLiLX+Mg²⁺</i>	1182 ± 134 (64%) 228 ± 33 (36%)	--	0.79 ± 0.04

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93 CaPLiLX electrophoretic mobility of $-4.1 \pm 0.1 \text{ cm}^2\text{V}^{-1}\text{s}^{-1}$ indicates that the nanocarriers are negatively
 94 charged at pH 7.4. Moreover, the diminution of μ_e upon LiLX coating is indicative of CaP deposition over the
 95 liposomes.

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100 **Fig. S4.** Analysis of fluorescence covered area on unmodified glass and Ap-modified glass slides as a
 101 function of contact time: 0 min (black), 10 min (red) and 180 min (green). Error bars stands for SD.