Supporting Information

High-Performance Electrochromic Films with Fast Switching Times Using Transparent/Conductive Nanoparticle-Modulated Charge Transfer[†]

Junsang Yun,‡^a Yongkwon Song,‡^a Ikjun Cho,^a Yongmin Ko,^a Cheong Hoon Kwon,^a and Jinhan Cho*^a

^aDepartment of Chemical & Biological Engineering, Korea University, 145 Anam-ro, Seongbuk-gu, Seoul 02841, Republic of Korea. E-mail: jinhan71@korea.ac.kr

[‡]These authors contributed equally to this work.



Fig. S1 FTIR spectra of pristine OAm-WO_{2.72} NRs and TREN. In the case of OAm-WO_{2.72} NRs, the C–H stretching peaks (*ca.* 2,926 and 2,854 cm⁻¹) and the W–O stretching peaks (*ca.* 927 and 870 cm⁻¹) derived from the long alkyl chains of OAm ligands and WO_{2.72} NRs, respectively, are detected. However, in the case of TREN, there are no distinct absorption peaks at those ranges.



Fig. S2 Cross-sectional FE-SEM images and the corresponded thickness of $(WO_{2.72}$ NR/TREN)_n multilayers as a function of bilayer number (n).



Fig. S3 Enlarged planar FE-SEM image of $(WO_{2.72} NR/TREN)_{40}$ multilayers. As shown in the image, the continuous deposition of NR-type $WO_{2.72}$ forms a porous structure due to the random packing effects, which can facilitate an intercalation of Li⁺ ions and an incorporation of additional transparent/conductive ITO NPs.



Fig. S4 Optical transmittance spectra of $(WO_{2.72} \text{ NR/TREN})_n$ multilayers as a function of bilayer number (n) under applied potentials ranging from -1.0 V to -4.0 V. The optical modulations are obtained from the variations of transmittance between bleached state (+ 4.0 V) and colored state (- 4.0 V) at a wavelength of 633 nm. (a) n = 10, (b) n = 20, (c) n = 30, and (d) n = 40.



Fig. S5 Stability test of $(WO_{2.72} NR/TREN)_{20}$ multilayers under alternating potentials of -4.0 V and +4.0 V for 30 s interval.



Fig. S6 (a) Cyclic voltammograms of $(WO_{2.72} NR/PEI)_{20}$ multilayers in a scan rate ranging from 10 to 100 mV s⁻¹. (b) Square root of scan rates $(v^{1/2})$ -dependent redox peak current densities (I_p) from CV curves of $(WO_{2.72} NR/TREN)_{20}$ and $(WO_{2.72} NR/PEI)_{20}$ multilayers.



Fig. S7 Optical transmittance spectra of $(WO_{2.72} NR/PEI)_{20}$ multilayers at colored state (from -1.0 V to -4.0 V). In this case, the optical modulations between bleached state (+ 4.0 V) and colored state (- 4.0 V) at a wavelength of 633 nm is measured to be 35.1%.



Fig. S8 (a) FTIR spectra of pristine OAm-ITO NPs and TREN. (b) FTIR spectra and schematic representation of (ITO NP/TREN)_n multilayers as a function of bilayer number (n). The C–H stretching peaks originated from the long alkyl chains of OAm ligands at 2,926 and 2,854 cm⁻¹ appeared and disappeared repeatedly according to the alternating deposition of OAm-ITO NPs and TREN. These phenomena imply a formation of (ITO NP/TREN)_n multilayers through ligand-exchange reactions between OAm ligands and TREN.



Fig. S9 Magnified planar FE-SEM image of $(WO_{2.72} NR/TREN/ITO NP/TREN)_{20}$ multilayers. As shown in the image, the formed multilayers still exhibited nanoporous structure after the incorporation of ITO NPs, facilitating the diffusion of Li⁺ ions into the EC films.



Fig. S10 Cyclic voltammograms of $(WO_{2.72} NR/TREN/ITO NP/TREN)_{20}$ multilayers in a scan rate ranging from 10 to 100 mV s⁻¹.



Fig. S11 (a) Cyclic voltammograms at a scan rate of 100 mV s⁻¹ and (b) Nyquist plots of WO_{2.72} NR-based EC films with ITO NPs (m = 20), without ITO NPs (n = 20), and the pristine ITO NP films.



Fig. S12 Optical transmittance spectra of $(WO_{2.72} NR/TREN/ITO NP/TREN)_m$ multilayers with increasing periodic number (m) under applied potentials ranging from -1.0 V to -4.0 V. The optical modulations between bleached state (+ 4.0 V) and colored state (- 4.0 V) at a wavelength of 633 nm are also shown. (a) m = 10, (b) m = 20, (c) m = 30, and (d) m = 40.



Fig. S13 Comparison of optical modulations at a wavelength of 633 nm between $(WO_{2.72}$ NR/TREN)_n and $(WO_{2.72}$ NR/TREN/ITO NP/TREN)_m multilayers as a function of bilayer (n) or periodic (m) number. The WO_{2.72} NR-based EC films with ITO NPs exhibit the higher optical modulations at the same layer number of WO_{2.72} NRs compared to the EC films without ITO NPs.



Fig. S14 Photographic images of electrochromic $(WO_{2.72} \text{ NR/TREN})_n$ and $(WO_{2.72} \text{ NR/TREN})_m$ multilayers under applied potentials of + 4.0 V (bleached state) and – 4.0 V (colored state). In this case, the $WO_{2.72}$ NR-based EC films with ITO NPs display a deeper color change than the EC films without ITO NPs.



Fig. S15 (a) CEs of $(WO_{2.72} NR/TREN/ITO NP/TREN)_m$ multilayers as a function of periodic number (m). (b) Comparison of CEs between $(WO_{2.72} NR/TREN)_n$ and $(WO_{2.72} NR/TREN/ITO NP/TREN)_m$ multilayers at the same layer number of $WO_{2.72} NR$ s.



Fig. S16 Cycling retention test of $(WO_{2.72} NR/TREN/ITO NP/TREN)_{20}$ multilayers under alternating potentials of -4.0 V and +4.0 V for 30 s interval.

Electrodes	Method	t _c (s)	$\mathbf{t}_{\mathbf{b}}\left(\mathbf{s} ight)$	ΔΤ (%)	CE (cm ² /C)	Reference
(WO _{2.72} NR/TREN /ITO NP/TREN) ₁₀	LbL assembly	4.1	1.5	32.0 at 633 nm	33.2	Our work
(WO _{2.72} NR/TREN /ITO NP/TREN) ₂₀	LbL assembly	5.0	3.0	41.6 at 633 nm	41.5	Our work
(WO _{2.72} NR/TREN /ITO NP/TREN) ₃₀	LbL assembly	8.2	11.4	52.9 at 633 nm	48.9	Our work
(WO _{2.72} NR/TREN /ITO NP/TREN) ₄₀	LbL assembly	10.9	15.2	55.8 at 633 nm	55.2	Our work
WO _{2.72} NW films	Langmuir-Blodgett	10	2	11.5 at 633 nm**	-	S1
WO _{2.72} NW films	Langmuir-Blodgett	30	16	49.2 at 633 nm**	-	S1
MoO ₃ -W _{0.71} Mo _{0.29} O ₃ hybrid films*	Drop casting	17.2	28.4	41.9 at 633 nm	19.0	S2
WO ₃ /Ag/WO ₃ films	Sputtering deposition	15.9	6.6	35.5 at 650 nm	28.3	S3
P ₈ W ₄₈ /W ₁₈ O ₄₉ nanocomposites*	LbL assembly	52	26	39.0 at 500 nm**	21.4	S4

Table S1. Comparison of EC performance of WO_x -based films in lithium-based electrolytes.

W _{0.71} Moo nar	0.29O3/PEDOT:PSS	Spray LbL assembly	17.9	10.5	65.1 at 633 nm	52.8	S5
Ну	perbranched WO ₃ films	Pulsed laser deposition	0.9	55	67.3 at 660 nm	65.4	S6
I a-W	Dual-phase O ₃ /WO ₃ films	Inkjet printing	5	5	12.8 at 633 nm**	3.12	S7
,	WO3 films	Langmuir-Blodgett	>3.6	>3.1	25.9 at 630 nm	71.3	S 8
[WO ₂ (0	D ₂)H ₂ O]•1.66H ₂ O films	Electrophoretic deposition	7.8	1.7	32.0 at 632 nm	11.5	S 9
PEI/V nar	VO ₃ nanosheets nocomposites	LbL assembly	660	11	37.5 at 633 nm**	32.0	S10

* Mo: Molybdenum, P₈W₄₈: K₂₈Li₅H₇P₈W₄₈O₁₈₄•92H₂O polyoxometalates.

** EC performance was evaluated from given data in the literature.

Supplementary references

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