

Fluoroborophosphates: a family of potential deep ultraviolet

NLO materials

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Table S1. Crystal data and structure refinements for RbBPO₄F, CsBPO₄F and (NH₄)₂BPO₄F₂.

Formula	RbBPO ₄ F	CsBPO ₄ F	(NH ₄) ₂ BPO ₄ F ₂
Formula weight	210.25	257.69	179.86
Temperature	293(2) K	293(2) K	293
Wavelength	0.71073	0.71073	1.54178
Space group	<i>P</i> 2 ₁ 3	<i>P</i> 2 ₁ 3	<i>P</i> 2 ₁
a (Å), α (deg.)	7.6147, 90	7.7275, 90	4.5122, 90
b (Å), β (deg.)	7.6147, 90	7.7275, 90	11.5395, 89.957
c (Å), γ (deg.)	7.6147, 90	7.7275, 90	12.7360, 90
Volume (Å ³)	441.53(7)	461.44(6)	663.14(8)
Z, Calculated density	4, 3.163 Mg/m ³	4, 3.709 Mg/m ³	4, 1.802 Mg/m ³
Absorption coefficient	11.504 mm ⁻¹	8.303 mm ⁻¹	3.912 mm ⁻¹
F(000)	392	464	368
Theta range for data collection	3.78 to 28.72 deg	3.73 to 28.84 deg	3.47 to 74.53 deg
Limiting indices	-9 ≤ h ≤ 9, -10 ≤ k ≤ 9, -10 ≤ l ≤ 9	-10 ≤ h ≤ 8, -9 ≤ k ≤ 9, -10 ≤ l ≤ 10	-5 ≤ h ≤ 5, -9 ≤ k ≤ 14, -15 ≤ l ≤ 14
Reflections collected/unique	3167 / 369 [R(int) = 0.0814]	3389 / 400 [R(int) = 0.1683]	4078 / 2012 [R(int) = 0.0262]
Completeness to theta	96.2 %	96.8 %	97.2 %
Refinement method	Full-matrix least-squares on F _o ²	Full-matrix least-squares on F _o ²	Full-matrix least-squares on F _o ²
Final R indices [I > 2σ(I)]	R1 = 0.0272, wR2 = 0.0609	R1 = 0.0293, wR2 = 0.0662	R1 = 0.0383, wR2 = 0.0992
R indices (all data) ^a	R1 = 0.0297, wR2 = 0.0637	R1 = 0.0325, wR2 = 0.0693	R1 = 0.0386, wR2 = 0.0995
Largest diff. peak and hole	0.395 and -0.701 e.Å ⁻³	0.669 and -1.003 e.Å ⁻³	0.517 and -0.412 e.Å ⁻³

^aR₁ = Σ||F_o| - |F_c||/Σ|F_o| and wR₂ = [Σw(F_o² - F_c²)²/ΣwF_o⁴] 1/2 for F_o² > 2σ(F_o²).

Table S2. Selected bond lengths (Å) for RbBPO₄F, CsBPO₄F and (NH₄)₂BPO₄F₂.

RbBPO ₄ F		CsBPO ₄ F		(NH ₄) ₂ BPO ₄ F ₂	
Rb(1)-O(2)	3.082(2)	Cs(1)-O(2)	3.136(3)	B(1)-F(1)	1.398(7)
Rb(1)-O(2)#1	3.082(2)	Cs(1)-O(2)#1	3.136(3)	B(1)-O(1)	1.416(7)
Rb(1)-O(2)#2	3.082(2)	Cs(1)-O(2)#2	3.136(3)	B(1)-F(2)	1.434(6)
Rb(1)-O(1)#3	3.152(2)	Cs(1)-F(1)#3	3.263(3)	B(1)-O(4)#1	1.478(6)
Rb(1)-O(1)#4	3.152(2)	Cs(1)-F(1)#4	3.263(3)	B(2)-F(3)	1.395(6)
Rb(1)-O(1)#5	3.152(2)	Cs(1)-F(1)#2	3.263(3)	B(2)-F(4)	1.399(6)
Rb(1)-O(1)#6	3.232(2)	Cs(1)-O(1)#5	3.263(4)	B(2)-O(5)	1.410(7)
Rb(1)-O(1)#7	3.232(2)	Cs(1)-O(1)#6	3.263(4)	B(2)-O(6)#2	1.493(7)
Rb(1)-O(1)#8	3.232(2)	Cs(1)-O(1)#7	3.263(4)	P(1)-O(3)	1.505(3)
Rb(1)-F(1)#9	3.258(2)	Cs(1)-O(1)#8	3.328(4)	P(1)-O(2)	1.511(3)
Rb(1)-F(1)#10	3.258(2)	Cs(1)-O(1)#9	3.328(4)	P(1)-O(1)	1.557(3)
Rb(1)-F(1)	3.258(2)	Cs(1)-O(1)#10	3.328(4)	P(1)-O(4)	1.560(4)
B(1)-F(1)	1.408(7)	B(1)-F(1)	1.408(12)	P(2)-O(7)	1.495(3)
B(1)-O(1)#11	1.460(3)	B(1)-O(1)#10	1.461(5)	P(2)-O(8)	1.497(4)
B(1)-O(1)#12	1.460(3)	B(1)-O(1)#11	1.461(5)	P(2)-O(6)	1.559(4)
B(1)-O(1)	1.460(3)	B(1)-O(1)	1.461(5)	P(2)-O(5)	1.590(4)
P(1)-O(2)	1.484(4)	P(1)-O(2)	1.495(7)		
P(1)-O(1)	1.554(2)	P(1)-O(1)	1.560(3)		
P(1)-O(1)#13	1.554(2)	P(1)-O(1)#12	1.560(3)		
P(1)-O(1)#14	1.554(2)	P(1)-O(1)#13	1.560(3)		

Symmetry codes for the generated atoms:

For RbBPO₄F: #1 $x+1/2, -y+3/2, -z$; #2 $-x+3/2, -y+2, z-1/2$; #3 $-y+1, z+1/2, -x+1/2$;
 #4 $z+1/2, -x+3/2, -y$; #5 $-x+3/2, -y+1, z-1/2$; #6 $y+1/2, -z+3/2, -x+1$; #7 $-x+2, y+1/2, -z+1/2$;
 #8 $-z+3/2, -x+2, y-1/2$; #9 $-x+2, y+1/2, -z-1/2$; #10 $x-1/2, -y+3/2, -z$; #11 $-z+1, x-1/2, -y+1/2$;
 #12 $y+1/2, -z+1/2, -x+1$; #13 $-y+3/2, -z+1, x-1/2$; #14 $z+1/2, -x+3/2, -y+1$.

For CsBPO₄F: #1 $-x+1, y+1/2, -z+1/2$; #2 $-x+3/2, -y, z+1/2$; #3 $x, y, z+1$; #4 $x-1/2, -y+1/2, -z$; #5 y, z, x ;
 #6 $-x+1, y-1/2, -z+1/2$; #7 $z+1/2, -x+1/2, -y+1$; #8 $z+1, x, y$; #9 $x+1/2, -y+1/2, -z$; #10 $-y+1, z+1/2, -x+1/2$;
 #11 $-z+1/2, -x+1, y-1/2$; #12 $z+1/2, -x+1/2, -y$; #13 $-y+1/2, -z, x-1/2$.

For (NH₄)₂BPO₄F₂: #1 $x-1, y, z$; #2 $x+1, y, z$.

Table S3. Hydrogen bond lengths [\AA] for $(\text{NH}_4)_2\text{BPO}_4\text{F}_2$.

N(1)...O(3)	2.801(6)	N(1) ...F(2)#2	2.912(6)
N(1) ...O(3)#1	2.808(7)	N(2) ...O(2)#2	2.839(6)
N(1) ...O(8)#1	2.818(5)	N(2) ...F(4)#3	2.887(5)
N(2) ...O(4)#4	2.895(6)	N(4) ...O(7)	2.764(6)
N(2) ...F(1)	2.923(7)	N(4) ...O(7)#1	2.778(6)
N(3) ...O(8)#1	2.854(6)	N(4) ...F(2)#2	2.855(5)
N(3) ...O(2)	2.889(5)	N(4) ...O(6)#6	2.866(6)
N(3) ...F(3)#5	2.988(5)		

Symmetry codes for the generated atoms:

#1 $x+1, y, z$; #2 $-x, y+1/2, -z+1$; #3 $x, y, z-1$; #4 $-x+1, y+1/2, -z+1$; #5 $-x, y-1/2, -z+2$; #6 $-x, y+1/2, -z+2$.

Table S4. The calculated SHG tensors of RbBPO₄F, CsBPO₄F and (NH₄)₂BPO₄F₂.

	RbBPO₄F	CsBPO₄F	(NH₄)₂BPO₄F₂
calculated SHG coefficient tensors ($\times 10^{-9}$ esu)	$d_{14}=d_{25}=d_{36}=1.24$	$d_{14}=d_{25}=d_{36}=1.75$	$d_{14}=d_{25}=d_{36}=1.10$ $d_{16}=d_{21}=0.55$ $d_{22}=2.10$ $d_{23}=d_{34}=1.57$

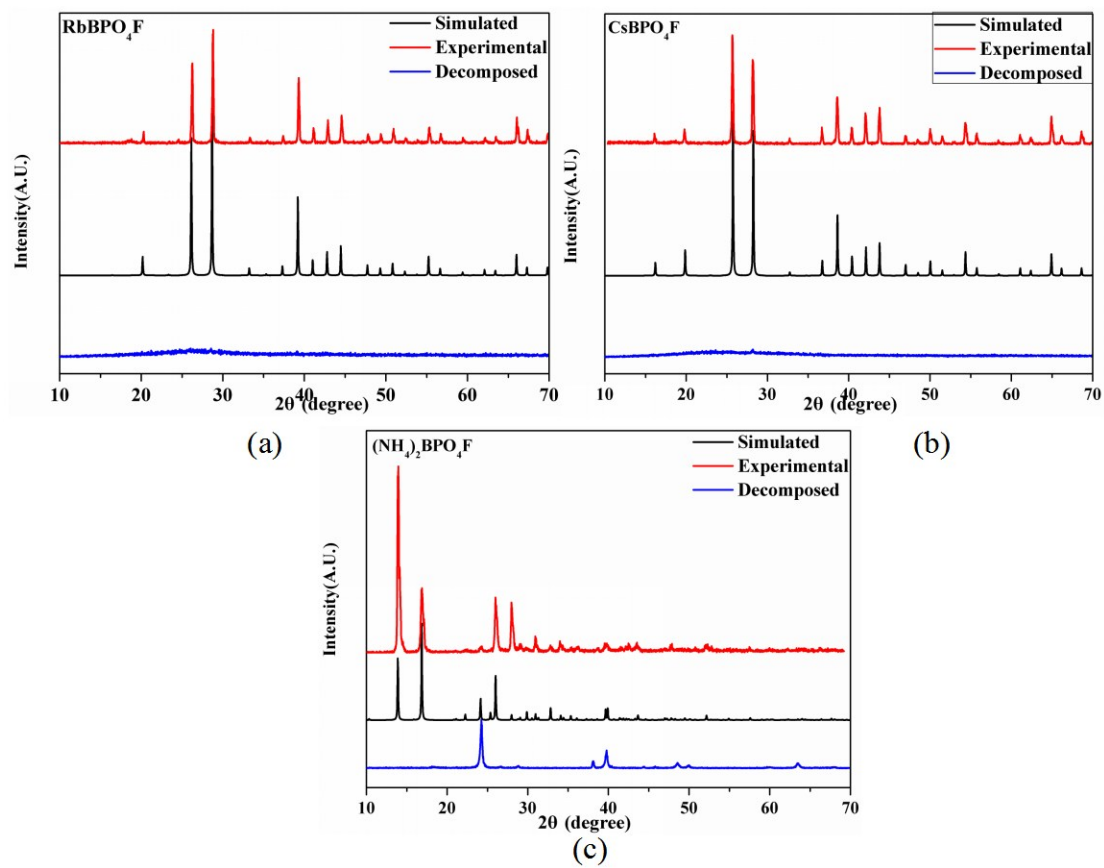


Figure S1. Powder X-ray diffraction patterns of RbBPO_4F , CsBPO_4F and $(\text{NH}_4)_2\text{BPO}_4\text{F}_2$.

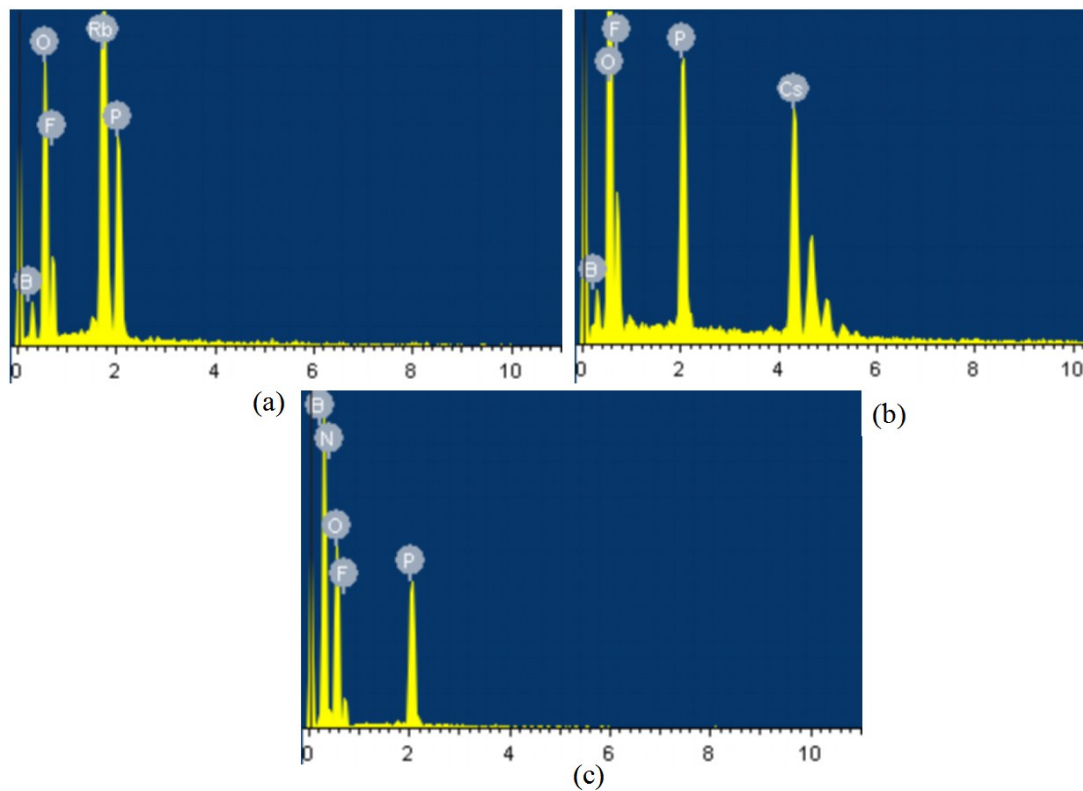


Figure S2. The Energy dispersive X-ray spectroscopy of RbBPO_4F , CsBPO_4F and $(\text{NH}_4)_2\text{BPO}_4\text{F}_2$.

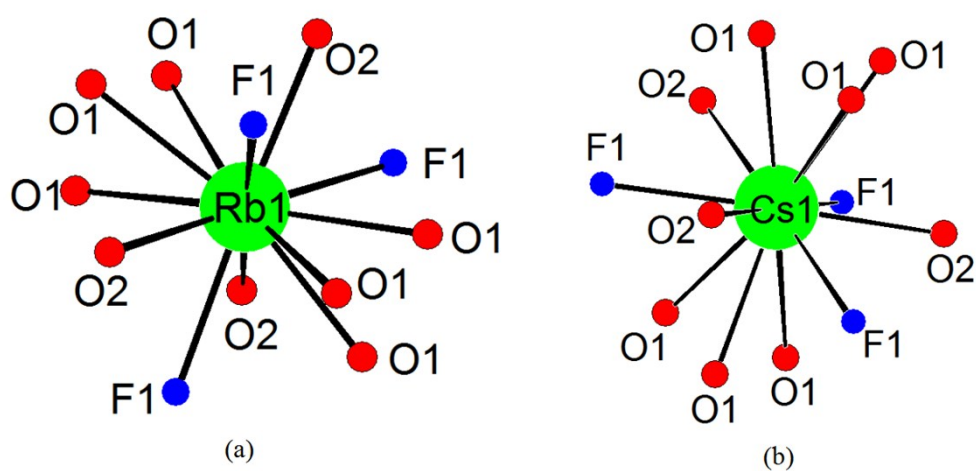


Figure S3. The coordination environments around Rb^+ and Cs^+ in RbBPO_4F and CsBPO_4F .

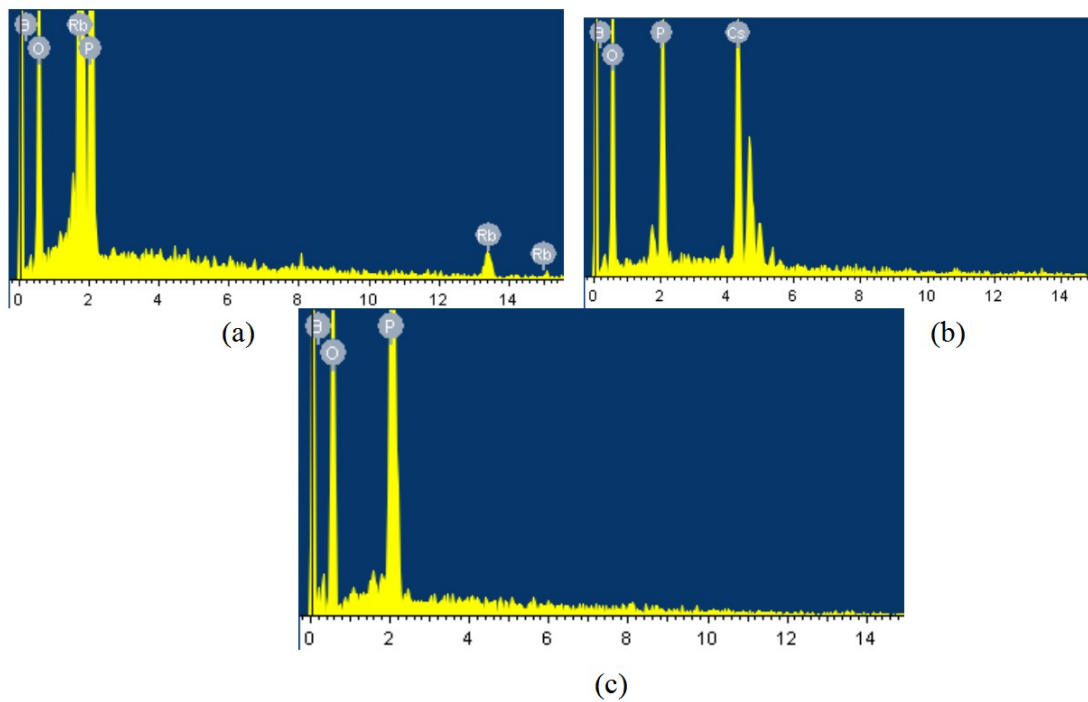


Figure S4. The Energy dispersive X-ray spectroscopy of the thermal decomposition residuals for RbBPO₄F, CsBPO₄F and (NH₄)₂BPO₄F₂.

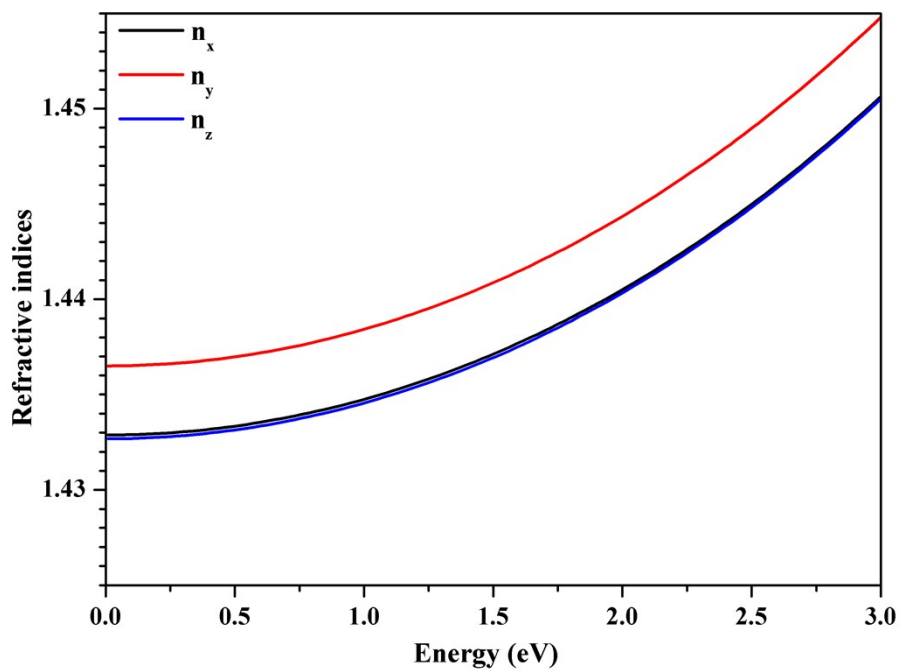


Figure S5. The calculated refractive indices of (NH₄)₂BPO₄F₂.