Supporting Information

A turn-on fluorescence sensor for highly selective detection of Al^{3+} based on diarylethene and its application by test strips

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Figure S6. The binding constant of **10** with Al³⁺ was calculated to be $4.72 \times 10^4 \,\mathrm{L \cdot mol^{-1}}$.

Figure S7. The limit of detection (LOD) for Al^{3+} was 1.24×10^{-5} mol L⁻¹.

Figure S8. Fluorescence photos of Al³⁺ at different concentrations.

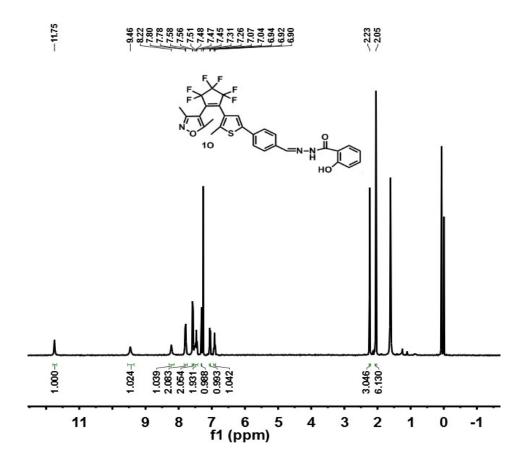
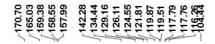


Figure S1. ¹H NMR spectrum of 1O.



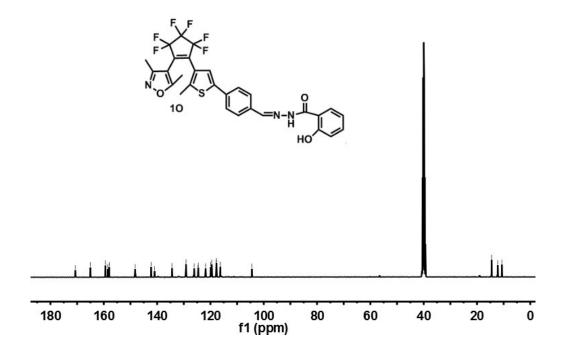


Figure S2. ¹³C NMR spectrum of **10**.

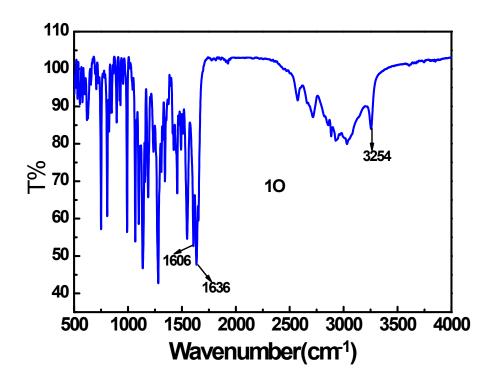


Figure S3. IR spectra of 10.

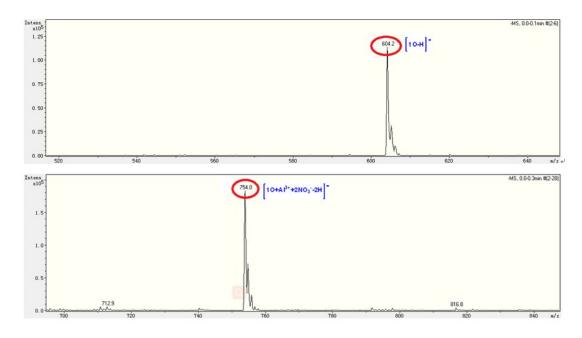


Figure S4. MS-ESI spectrum of 10 and 10-Al $^{3+}$.

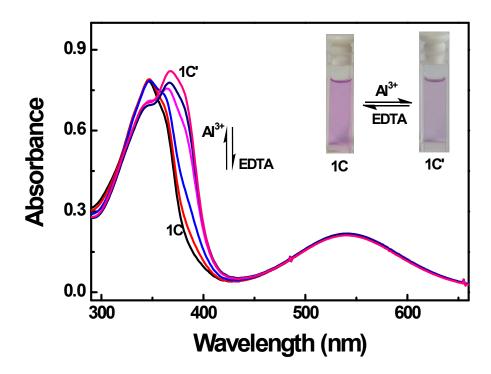


Figure S5. Absorption spectra changes of 1C induced by $Al^{3+}/EDTA$ in methanol solution (2.0 \times 10⁻⁵ mol L^{-1}).

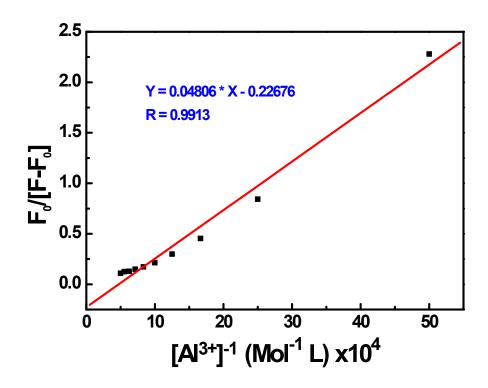


Figure S6. Hildebrand-Benesi plot based on the 1 : 1, the association constant of **10** with Al³⁺ was calculated to be $4.72 \times 10^4 \, \text{L} \cdot \text{mol}^{-1}$.

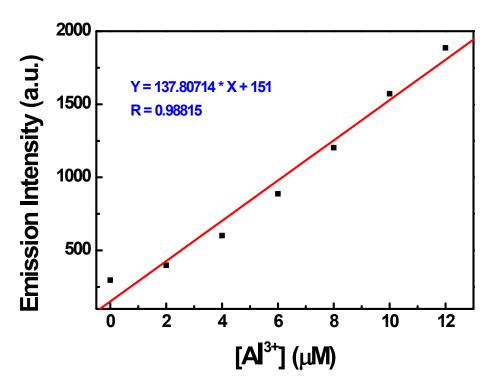


Figure S7.The limit of detection (LOD) for Al^{3+} was 1.24×10^{-5} mol L^{-1} .

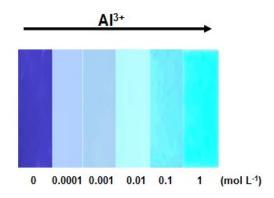


Figure S8. Fluorescence photos of Al^{3+} at different concentrations.