

Supporting Information For

Polyaniline Nanorods Randomly Assembly on the Sugarcane Bagasse Pith-based Carbon sheets with Promising Capacitive Performance

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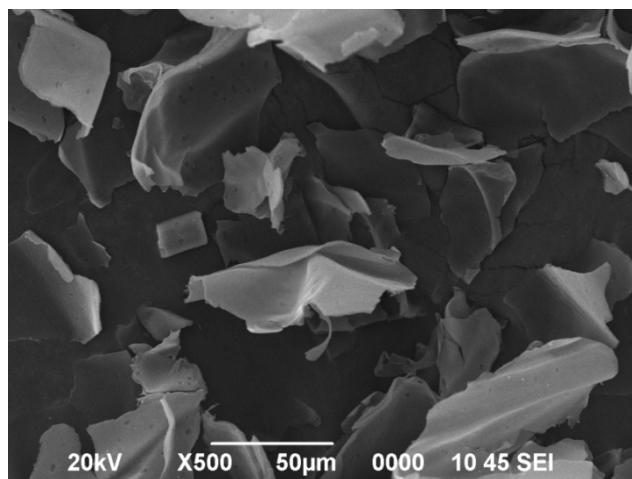


Fig. S1 SEM of the screened sugarcane pith

Table S1 The relative percentages of carbon species

Sample	C-C/C=C (%)	C-O (%)	C=O (%)	O-C=O (%)
SPCN	40.32	34.60	15.38	9.70
oxidized SPCN	49.30	19.62	6.79	24.29

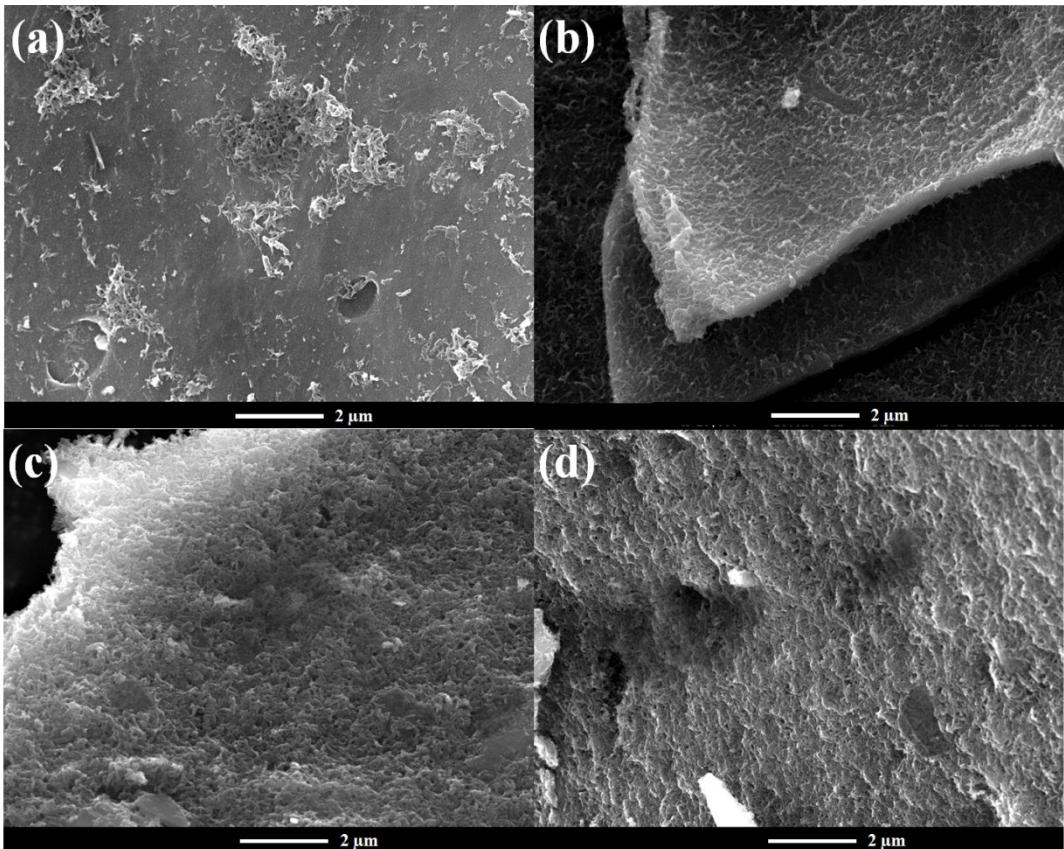


Fig. S2 FE-SEM of the oxidized SPCN/PANI-0.005 (a), the oxidized SPCN/PANI-0.01 (b), the oxidized SPCN/PANI-0.02 (c) and the oxidized SPCN/PANI-0.04 (d).

Table S2 Chemical composition of Oxidized SPCN/PANI-X by combustion analysis

Sample	C(wt%)	N(wt %)	PANI(wt %)
Oxidized SPCN/PANI-0.005	73.2	1.9	12.8
Oxidized SPCN/PANI-0.01	70.6	5.1	33.9
Oxidized SPCN/PANI-0.02	62.4	8.1	54.1
Oxidized SPCN/PANI-0.04	62.8	9.8	65.3

Calculation method: PANI(wt %)=N(wt %)/14×93.14.

Table S3 The pore structure parameters of the the SPCN, oxidized SPCN and oxidized SPCN/PANI-0.02

Sample	S _{BET} (m ² g ⁻¹)	V _{total} (cm ³ g ⁻¹)	D _{average} (nm)
SPCN	417	0.22	2.2
Oxidized SPCN	127	0.11	3.7
Oxidized SPCN/PANI-0.02	51	0.31	24.5
Oxidized SPCN/PANI-0.04	53	0.10	7.3

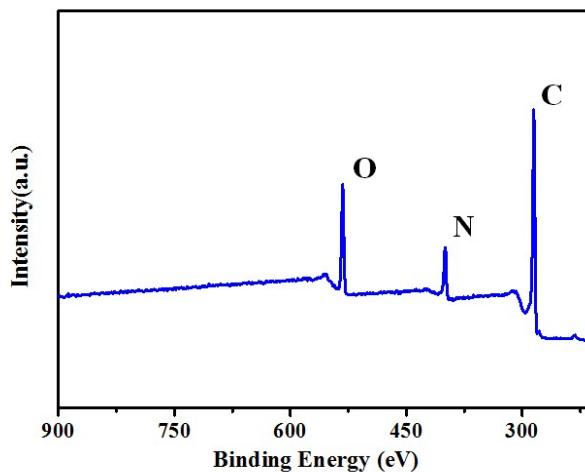


Fig. S3 XPS survey spectrum of the Oxidized SPCN/PANI-0.02.

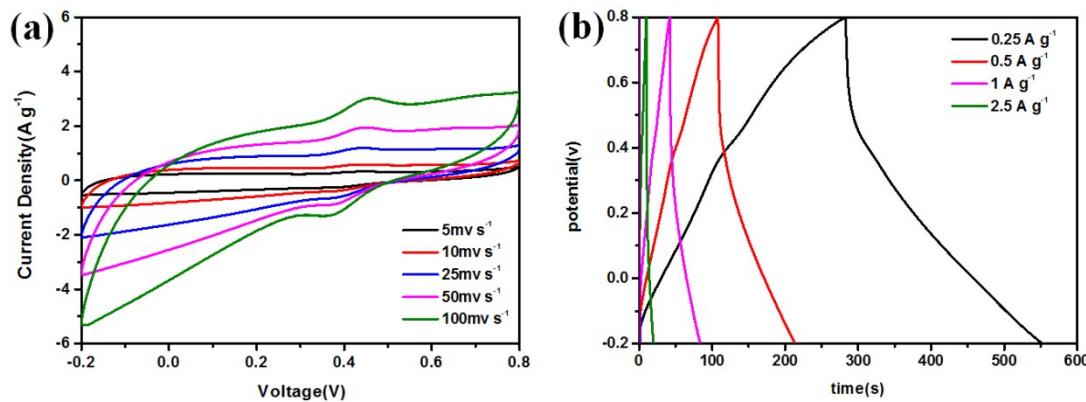


Fig. S4 CV curves of SPCN sample at different scan rates; (b) GCD curves of SPCN sample at different current densities ($1 \text{ M H}_2\text{SO}_4$).

The gravimetric capacitance of the SPCN sample is 67 F g^{-1} at current density of 0.25 A g^{-1} .

Table S4. Electrochemical performances of different PANI-based electrodes.

Electrode materials	Electrolyte	Capacitance (F g^{-1})	References
PANI hydrogels	$1 \text{ M H}_2\text{SO}_4$	$636 \text{ (at } 2 \text{ A g}^{-1}\text{)}$	S1
G/PANI nanorods/CNT	$1 \text{ M H}_2\text{SO}_4$	$638 \text{ (at } 0.5 \text{ A g}^{-1}\text{)}$	S2
3D RGO/S-PANI	$1 \text{ M H}_2\text{SO}_4$	$480 \text{ (at } 1 \text{ A g}^{-1}\text{)}$	S3
RGO/PANI/Vulcan carbon	$1 \text{ M H}_2\text{SO}_4$	$446 \text{ (at } 5 \text{ mV s}^{-1}\text{)}$	S4
CGR/CNT/PANI	$1 \text{ M H}_2\text{SO}_4$	$456 \text{ (at } 1 \text{ A g}^{-1}\text{)}$	S5
RGO/PANI nanoarrays	$1 \text{ M H}_2\text{SO}_4$	$727 \text{ (at } 1 \text{ A g}^{-1}\text{)}$	S6
RGO/PANI nanoarrays	$1 \text{ M H}_2\text{SO}_4$	$555 \text{ (at } 0.2 \text{ A g}^{-1}\text{)}$	S7

RGO/PANI	1 M H₂SO₄	385 (at 0.5 A g⁻¹)	S8
PANI /GrapheneNanoribbons	1 M H₂SO₄	340 (at 0.25 A g⁻¹)	S9
G/ PANI nanoarrays/CF	1 M H₂SO₄	1145 (at 1 A g⁻¹)	S10
G/PANI/CNT	PVA/H₂SO₄ gel	511 (at 1 A g⁻¹)	S11
RGO/PANI	1 M H₂SO₄	480 (at 1 A g⁻¹)	S12
PANI/CNT	1 M H₂SO₄	424 (at 0.1 A g⁻¹)	S13
PANI nanorods	1 M H₂SO₄	297(at 1 A g⁻¹)	S14
PANI hydrogel	1 M H₂SO₄	252(at 0.5 A g⁻¹)	S15
This work	1 M H₂SO₄	513(at 0.25 A g⁻¹)	

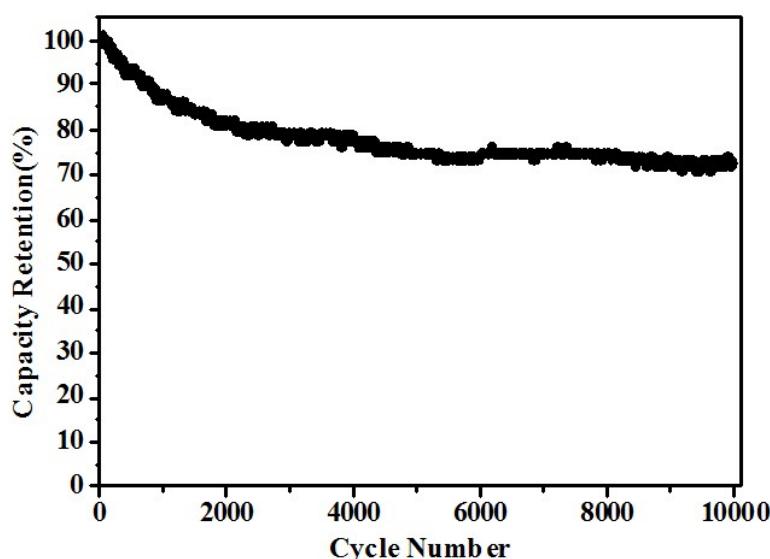


Fig. S5 The cycling stability of the Oxidized SPCN/PANI-0.02 at 5 A g⁻¹.

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