

<Supporting Information>

Three novel polyoxometalate-based inorganic-organic hybrid materials based on 2,6-bis(1,2,4-triazol-1-yl)pyridine

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Table S1. Crystal data and refinement for the compounds **1–3**

Compound	1	2	3
Empirical formula	CoW ₅ C ₁₈ H ₁₆ N ₁₄ O ₁₇	Cd ₃ P ₂ W ₂₄ C ₅₄ H ₆₆ N ₄₂ O ₉₂	Ag ₃ PMo ₁₂ C ₁₈ H ₁₇ N ₁₄ O _{41.5}
Formula weight	1678.63	7587.02	2599.32
Crystal size (mm ³)	0.17×0.15×0.11	0.20×0.20×0.20	0.20×0.20×0.20
Wavelength (Å)	0.71073	0.71073	0.71073
Crystal system	Orthorhombic	Cubic	Monoclinic
Space group	<i>Cccm</i>	<i>Ia-3</i>	<i>P2₁/n</i>
<i>a</i> (Å)	13.0024(8)	30.375(4)	11.861(3)
<i>b</i> (Å)	21.9573(13)	30.375(4)	20.437(6)
<i>c</i> (Å)	22.2526(14)	30.375(4)	22.536(6)
β (°)	90.00	90.00	93.471(4)
Volume (Å ³)	6353.1(7)	28026(6)	5453(2)
Temperature (K)	296(2)	293(2)	296(2)
<i>Z</i>	8	8	4
<i>D</i> _{calcd} (g cm ⁻³)	3.510	3.591	3.166
μ (mm ⁻¹)	18.643	20.188	3.853
<i>F</i> (000)	6024	26864	4860
θ range for data collection (°)	1.82 to 25.00	2.51 to 24.99	2.07 to 25.00
<i>R</i> _{int}	0.0547	0.0675	0.0306
Index ranges, <i>hkl</i>	-15 ≤ <i>h</i> ≤ 15, -20 ≤ <i>k</i> ≤ 26, -26 ≤ <i>l</i> ≤ 26	-36 ≤ <i>h</i> ≤ 34, -36 ≤ <i>k</i> ≤ 36, -36 ≤ <i>l</i> ≤ 35	-14 ≤ <i>h</i> ≤ 11, -23 ≤ <i>k</i> ≤ 26, -26 ≤ <i>l</i> ≤ 26
Reflections collected	2897	4114	9456
Independent reflections	2359	4113	8979
<i>R</i> ₁ , <i>wR</i> ₂ (<i>I</i> > 2σ(<i>I</i>))	<i>R</i> ₁ =0.0305, <i>wR</i> ₂ =0.0621	<i>R</i> ₁ =0.0920, <i>wR</i> ₂ =0.1887	<i>R</i> ₁ =0.0329, <i>wR</i> ₂ =0.0757
<i>R</i> ₁ , <i>wR</i> ₂ (all data)	<i>R</i> ₁ =0.0428,	<i>R</i> ₁ =0.0920,	<i>R</i> ₁ =0.0384,

	$wR_2=0.0651$	$wR_2=0.1887$	$wR_2=0.0778$
<i>GOOF</i> on F^2	1.055	1.232	1.148
Largest diff. peak and hole ($e \text{ \AA}^{-3}$)	1.46/-1.822	4.667/-3.057	1.345/-1.009

$R = \sum(|F_o| - |F_c|) / \sum |F_o|$, $wR = \{ \sum w[(F_o^2 - F_c^2)^2] / \sum w[(F_o^2)^2] \}^{1/2}$, $w = 1 / [\sigma^2(F_o^2) + (ap)^2 + bP]$, $P = (F_o^2 + 2F_c^2) / 3$. **1:** $a = 0.0277$, $b = 31.3472$; **2:** $a = 0.0328$, $b = 5173.3159$; **3:** $a = 0.0284$, $b = 21.5563$.

Table S2. Selected bond lengths (Å) and bond angles (°) for compound 1

Bond	(Å)	Bond	(Å)	Bond	(Å)
W2–O7 #6	1.898(5)	W4–O10 #6	1.895(8)	W3–O7#7	1.913(6)
W1–O4#7	1.899(5)	W4–O8#7	1.924(6)	W2–O8#7	1.970(6)
Co1–N1#1	2.066(7)	Co1–N1#4	2.066(7)	Co1–O1W	2.320(10)
Co1–N1#2	2.066(7)	Co1–N1	2.066(7)	Co1–O1W #4	2.320(10)
O1–Co2	2.127(7)	W2–O4#7	1.938(5)	W3–O2#6	2.370(7)
N8–C8 #5	1.337(10)	O6–W2#3	1.8993(5)	O7–W2#6	1.898(5)
O8–W2#7	1.970(6)	O10–W4 #6	1.895(8)	Co2–N2#7	2.099(7)
Co2–N2#5	2.099(7)	Co2–N2#8	2.099(7)	Co2–O1#8	2.127(7)
Angles	(°)	Angles	(°)	Angles	(°)
N1#1–Co1–N1#4	176.7(4)	N1#1–Co1–N1#2	88.6(4)	N1#4–Co1–N1	88.6(4)
N1#4–Co1–N1#2	91.5(4)	N1#1–Co1–N1	91.5(4)	N1#1–Co1–O1W	91.6(2)
N1#2–Co1–O1W	88.4(2)	N1#2–Co1–N1	176.7(4)	N1–Co1–O1W	88.4(2)
N1#1–Co1–O1W#4	88.4(2)	N1#4–Co1–O1W	91.6(2)	N1#4–Co1–O1W#4	88.4(2)
N1#2–Co1–O1W#4	91.6(2)	N1–Co1–O1W#4	91.6(2)	O1W–Co1–O1W#4	180.000(1)
C4–N2–Co2	135.4(6)	C3–N2–Co2	121.7(6)	N2#7–Co2–N2#8	88.0(4)
C9#5–C10–H10A	121.2	N2#7–Co2–N2#5	180.0(5)	N2#5–Co2–N2#8	92.0(4)
N2#8–Co2–N2	180.0(3)	N2#7–Co2–N2	92.0(4)	N2#5–Co2–N2	88.0(4)
N2#5–Co2–O1	88.6(2)	N2#8–Co2–O1	88.6(2)	N2#7–Co2–O1	91.4(2)
N2–Co2–O1	91.4(2)	N2#7–Co2–O1#8	88.6(2)	N2#5–Co2–O1#8	91.4(2)
N2–Co2–O1#8	88.6(2)	N2#8–Co2–O1#8	91.4(2)		

Symmetry code for 1: #1: $x, -y+1/2, -z+1/2$; #2: $-x, -y+1, +z$; #3: $-x, -y, z$; #4: $-x, y, -z+1/2$; #5: $-x+1/2, -y+1/2, z$; #6: $-x, -y, -z$; #7: $x, y, -z$; #8: $-x+1/2, -y+1/2$.

Table S3. Selected bond lengths (Å) and bond angles (°) for compound 2

Bond	(Å)	Bond	(Å)	Bond	(Å)
Cd1–N2#3	2.30(2)	Cd1–N2	2.30(2)	Cd1–O1W	2.44(3)
Cd1–N7#7	2.35(3)	Cd1–N7#12	2.35(3)	Cd1–O1W#3	2.44(3)
W2–O13#5	2.48(3)	W1–O14#4	2.52(3)	W1–O4#2	1.938(18)
W1–O8#8	1.88(2)	W2–O7#8	1.95(2)	W3–O11#10	1.87(4)
W4–O1#2	2.043(11)	W3–O10#10	1.91(2)	O16–O2#10	1.54(4)
O1–W4#1	2.043(11)	O2–O16#9	1.54(4)	O7–W2#5	1.95(2)
O10–W3#11	1.91(2)	O4–W1#1	1.938(18)	O11–W3#11	1.87(4)
O16–O2#9	1.54(4)	O14–W1#4	2.52(3)	O13–O14#4	1.67(4)
N7–Cd1#6	2.35(3)	O14–W1#5	2.52(3)	O16–O2#11	1.54(4)
Angles	(°)	Angles	(°)	Angles	(°)
N2#3–Cd1–N2	96.1(11)	N2#3–Cd1–N7#7	98.2(8)	N2–Cd1–N7#12	98.2(8)
N2–Cd1–N7#7	94.8(8)	N2#3–Cd1–N7#12	94.8(8)	N2#3–Cd1–O1W	174.7(10)
N7#7–Cd1–O1W	84.1(10)	N7#7–Cd1–N7#12	160.5(13)	N7#12–Cd1–O1W	81.8(10)

N2#3–Cd1–O1W#3	88.4(10)	N2–Cd1–O1W	88.4(10)	N2–Cd1–O1W#3	174.7(10)
N7#7–Cd1–O1W#3	81.8(10)	N7#12–Cd1–O1W#3	84.1(10)	C5–N7–Cd1#6	123(2)
O8#8–W1–O14#4	91.8(13)	W3–O10–W3#11	140.9(13)	O6–W2–O13#5	95.3(12)
O3–W3–O10	100.9(12)	O15–W4–O1#2	106.5(11)	O12–W3–O11#10	153.9(17)
O11#10–W3–O10	86.9(17)	O1–W4–O1#2	113.2(6)	O16#9–O2–P2	48(2)
W1–O4–W1#1	139.3(11)	O13#8–O14–O13#5	97(3)	P2–O2–W4	133(2)

Symmetry code for 1: #1: $-z-1/2, -x, y-1/2$; #2: $-y, z+1/2, -x-1/2$; #3: $x+1, -y-1/2, z$; #4: $-x, -y, -z-1$; #5: $z+1/2, x, -y-1/2$; #6: $z+3/2, x-1, -y-3/2$; #7: $y+1, -z-3/2, x-3/2$; #8: $y, -z-1/2, x-1/2$; #9: $-x+1/2, -y-1/2, -z-3/2$; #10: $z+1, x-1/2, -y-1$; #11: $y+1/2, -z-1, x-1$; #12: $-y, z+1, x-3/2$.

Table S4. Selected bond lengths (Å) and bond angles (°) for compound **3**

Bond	(Å)	Bond	(Å)	Bond	(Å)
Ag1–N11	2.161(5)	Ag1–N11#4	2.161(5)	Ag2–N3	2.528(5)
Ag2–N14#1	2.222(5)	Ag2–N2	2.315(5)	Ag2–N4	2.564(6)
Ag3–N1#3	2.143(5)	Ag4–N5	2.251(5)	Ag4–N9	2.305(5)
Ag4–N8	2.514(5)	N14–Ag2#2	2.222(5)	Mo1–O33	1.858(4)
Angles	(°)	Angles	(°)	Angles	(°)
N11–Ag1–N11#4	180.0	N14#1–Ag2–N2	129.96(19)	N14#1–Ag2–N4	90.3(2)
N11#1–Ag2–N3	134.44(17)	N2–Ag2–N3	67.25(17)	N3–Ag2–N4	62.42(16)
N5–Ag4–N9	129.62(19)	N5–Ag4–N8	130.75(18)	N2–Ag2–N4	129.53(17)
N9–Ag4–N8	67.47(16)	C1–N4–Ag2	135.1(5)	N1#3–Ag3–N1	179.998(1)
C2–N1–Ag3	138.7(4)	N7–N4–Ag2	118.6(4)	C9–N2–Ag2	137.1(4)
N6–N2–Ag2	120.0(4)	C1–N1–Ag3	118.2(4)	C8–N5–Ag4	130.6(4)
C3–N3–Ag2	124.9(4)	C7–N3–Ag2	117.4(4)	C8–N6–N2	108.9(5)
C10–N8–Ag4	125.5(4)	C9–N5–Ag4	125.8(4)	C14–N8–Ag4	117.2(4)
C18–N9–Ag4	137.4(5)	N13–N9–Ag4	119.1(4)	C18–N14–Ag2#2	126.5(4)
C16–N11–Ag1	135.4(4)	C17–N11–Ag1	121.6(4)	C15–N14–Ag2#2	129.7(4)

Symmetry code for 1: #1: $x-1, y, z$; #2: $x+1, y, z$; #3: $-x-1, -y+2, -z-1$; #4: $-x, -y+3, -z-1$.

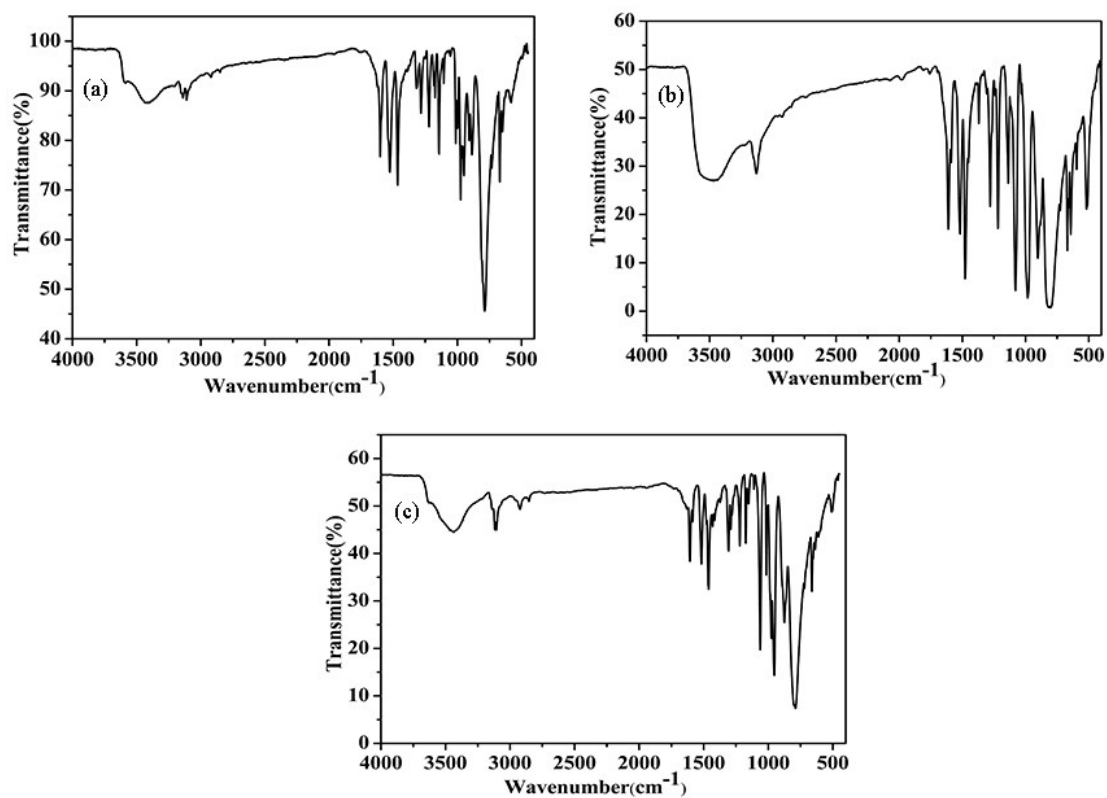


Fig. S1 IR spectra of compounds **1-3** (a-c).

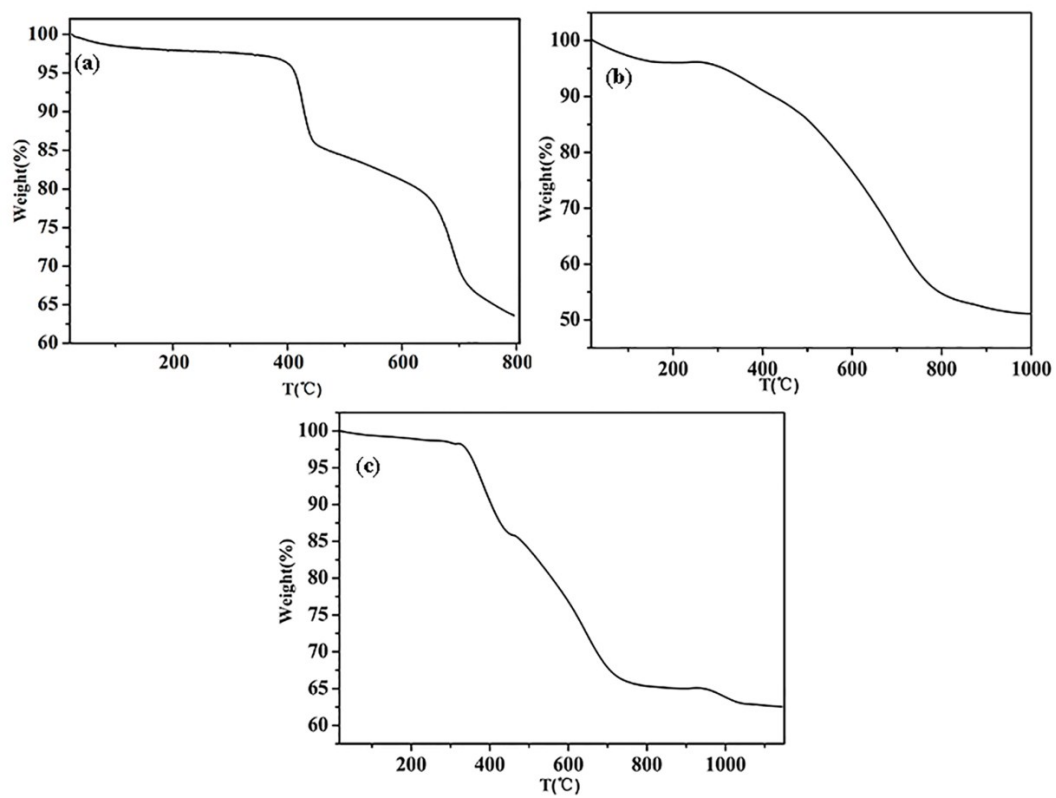


Fig. S2 TGA curves of compounds **1-3** (a-c).

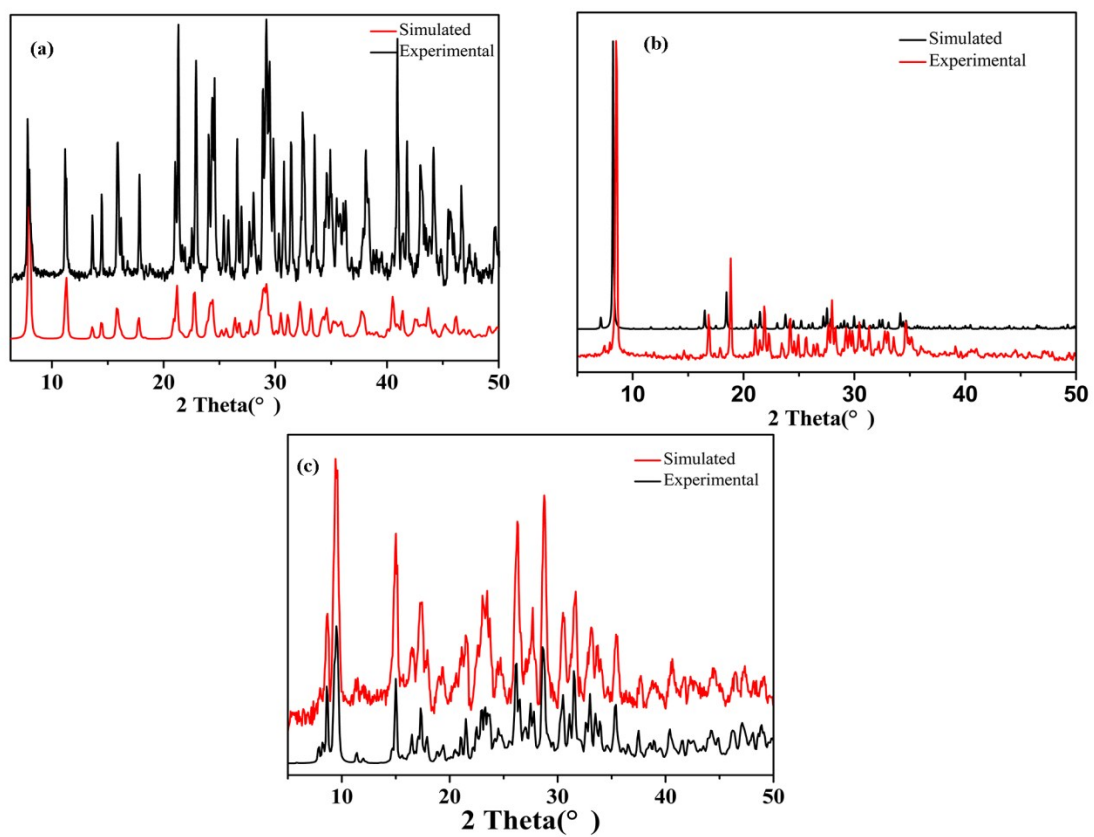


Fig. S3 PXRD patterns of compounds 1–3 (a–c).

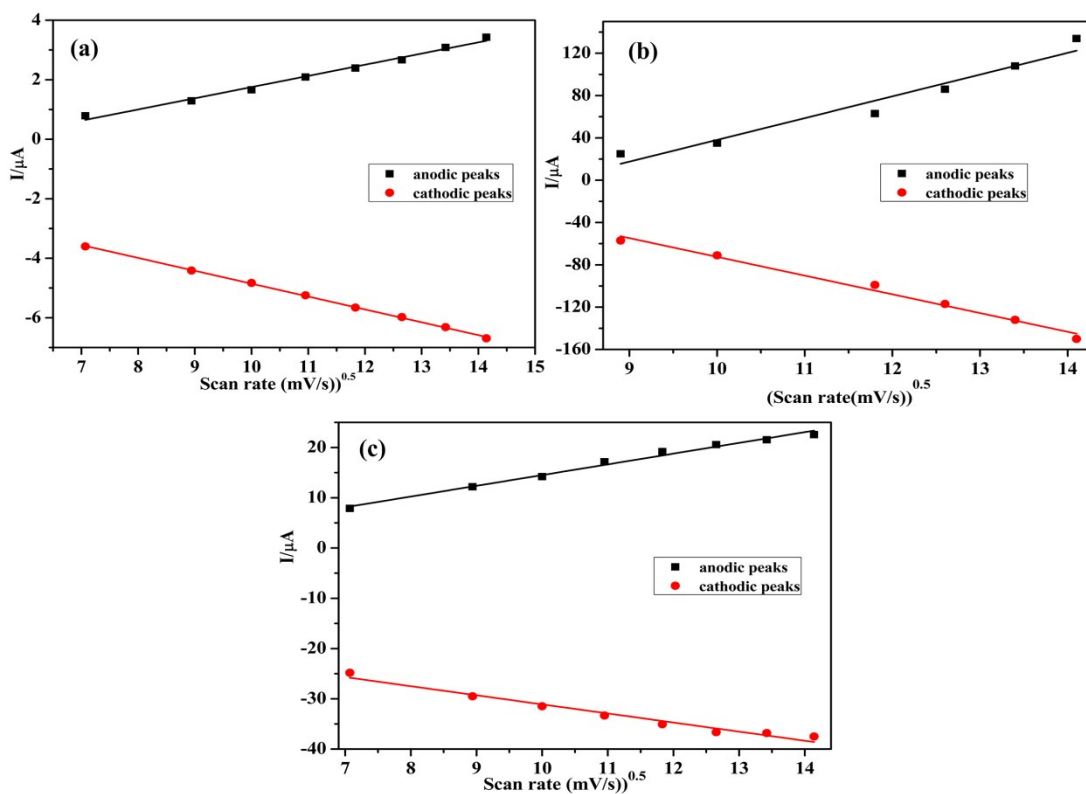


Fig. S4 (a), (b), (c) The plots of the anodic and the cathodic peak II-II' currents for 1-GCE, I-I' currents for 2-GCE, II-II' currents for 3-GCE against scan rates.

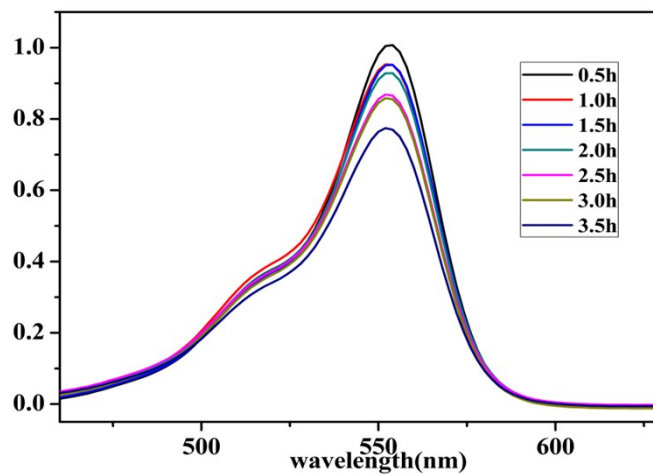


Fig. S5 Absorption spectra of the RhB solutions during the decomposition reaction without catalyst.

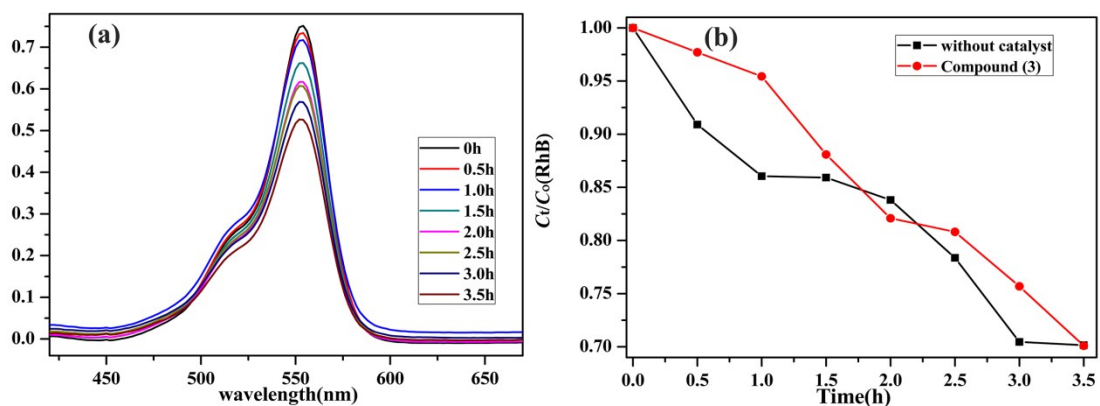


Fig. S6 (a) Absorption spectra of the RhB (10 mg L^{-1}) solution during the decomposition reaction under UV light irradiation with the use of compound **3**. (b) The comparison of the compound and no crystal decomposition rate of RhB in the same conditions.

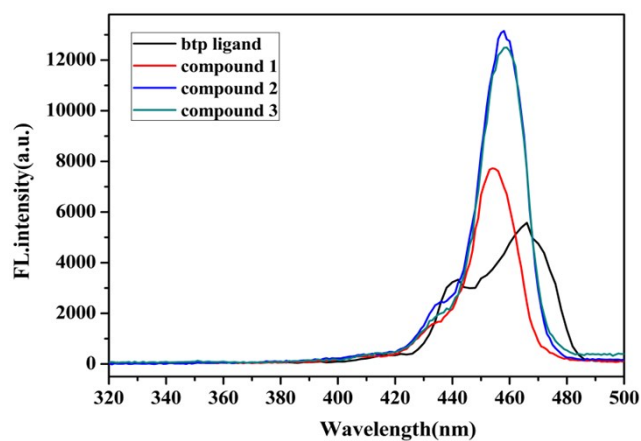


Fig. S7 Solid-state photoluminescence spectra of compounds **1–3** and the btp ligand at the room temperature.

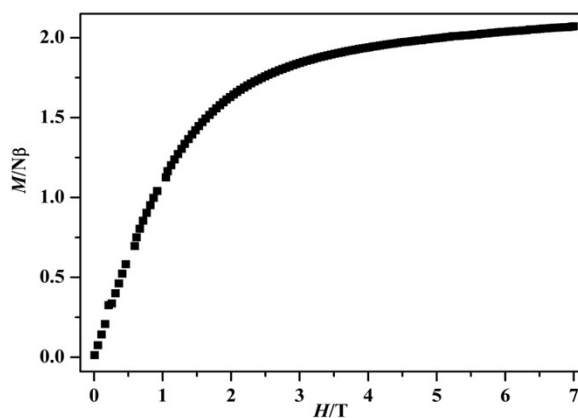


Fig. S8 The plot of M vs. H .