

## Electronic Supplementary Information

### Alkylammonium thiostannate inorganic/organic hybrids as high-performance photocatalysts with a decoupled adsorption-photodegradation mechanism

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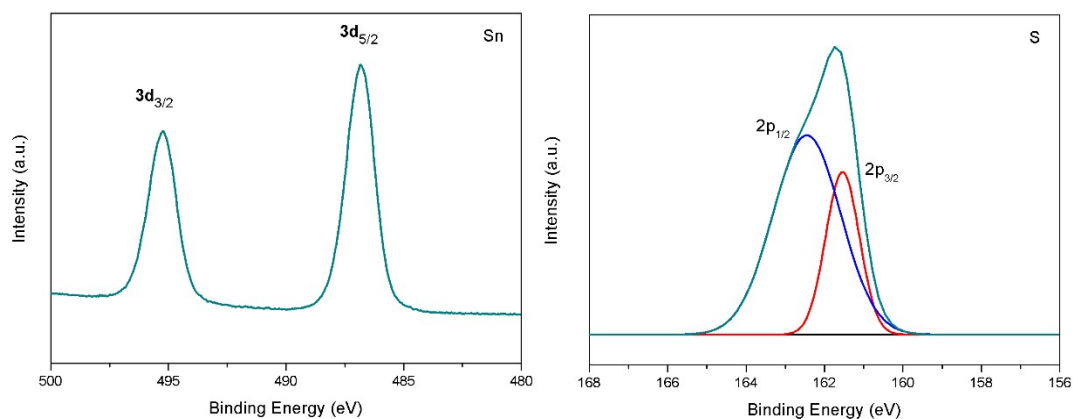
**Table S1.** The full elements analyses of three compounds

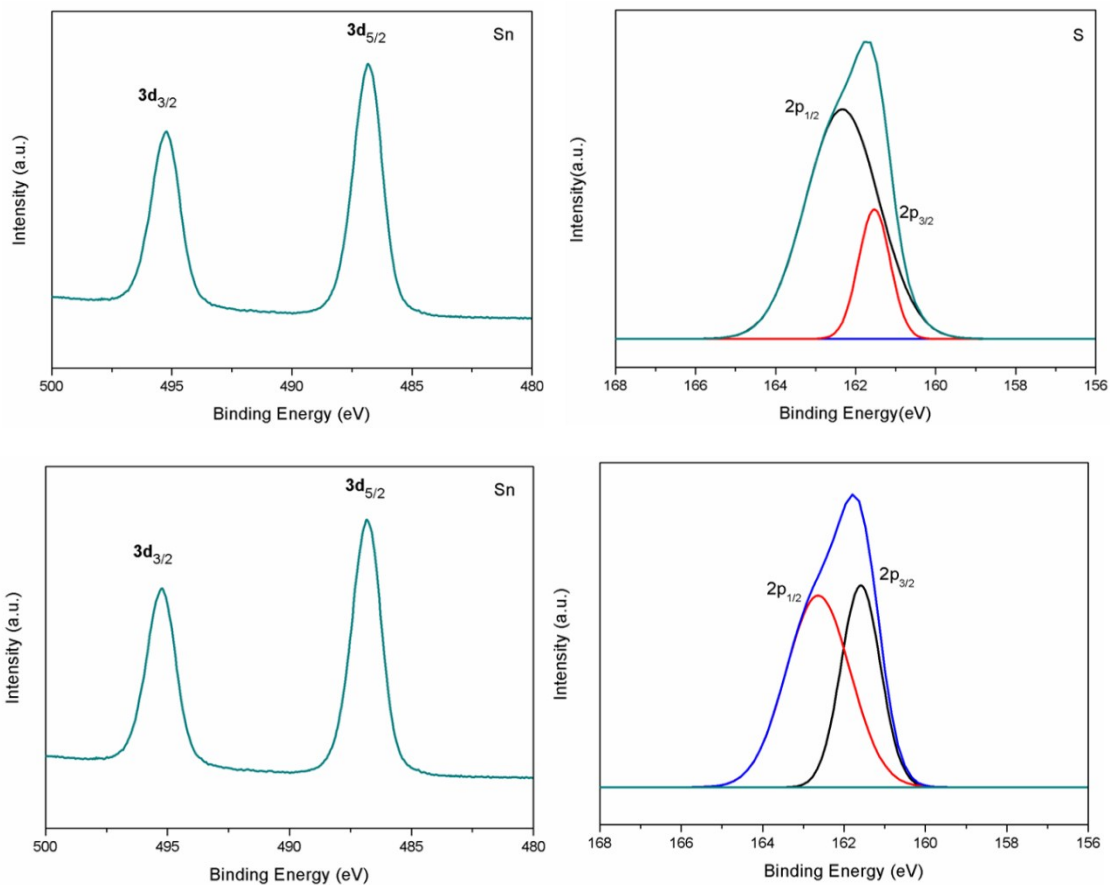
	H	C	N	Sn
$(C_4H_9NH_3)_2Sn_3S_7$	12.90 (13.18)	3.77 (3.84)*	3.276 (3.332)	48.38 (48.64)
$(C_6H_{13}NH_3)_2Sn_3S_7$	19.47 (18.35)	3.61 (3.57)	4.490 (4.08)	44.86 (45.26)
$(C_8H_{17}NH_3)_2Sn_3S_7$	20.94 (22.84)	2.91 (3.33)	4.599 (4.76)	41.91 (42.24)

\*The data in the bracket was the calculated value from the formula  $A_2Sn_3S_7$

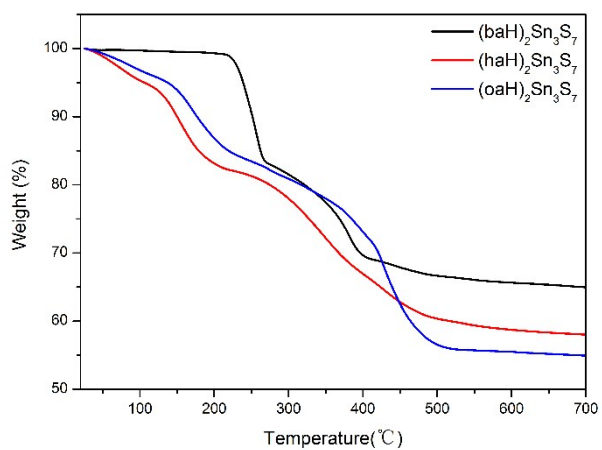
**Table S2** Binding Energies (in eV) of the Main Core Level Spectrum for the  $A_2Sn_3S_7$

	S2p		Sn3d	
	2p <sub>1/2</sub>	2p <sub>3/2</sub>	3d <sub>3/2</sub>	3d <sub>5/2</sub>
$(baH)_2Sn_3S_7$	162.55	161.54	495.32	486.82
$(haH)_2Sn_3S_7$	162.54	161.59	495.32	486.82
$(oaH)_2Sn_3S_7$	162.57	161.57	495.32	486.87





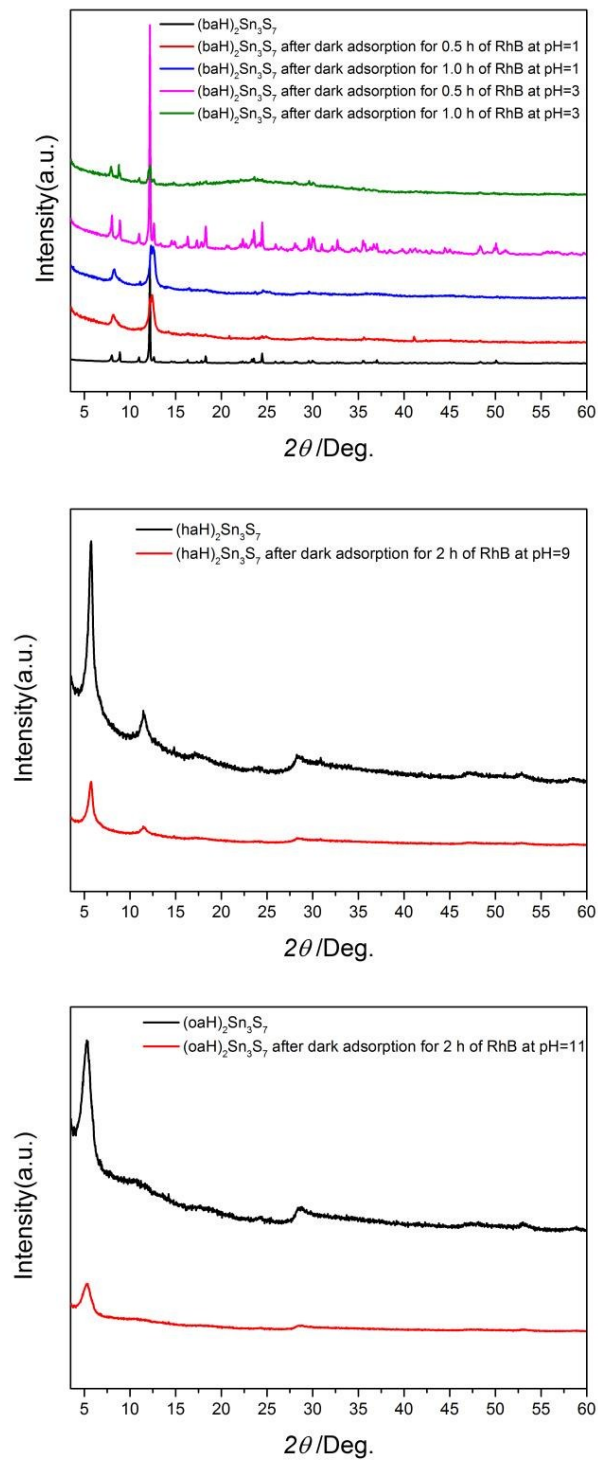
**Figure.S1** The Sn and S XPS spectra of  $A_2Sn_3S_7$ , A was baH, haH, and oaH, respectively from top to bottom.



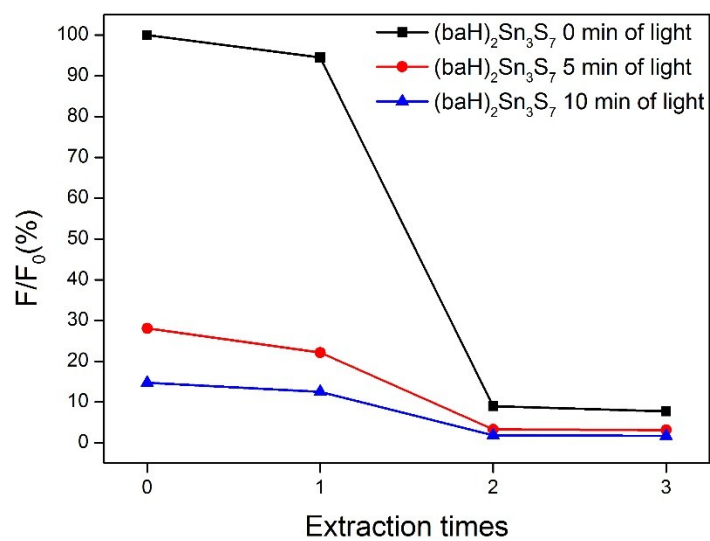
**Figure.S2** Thermogravimetric curves for  $A_2Sn_3S_7$

**Table S3.** Comparison of the adsorption capacities on various adsorbents for RhB.

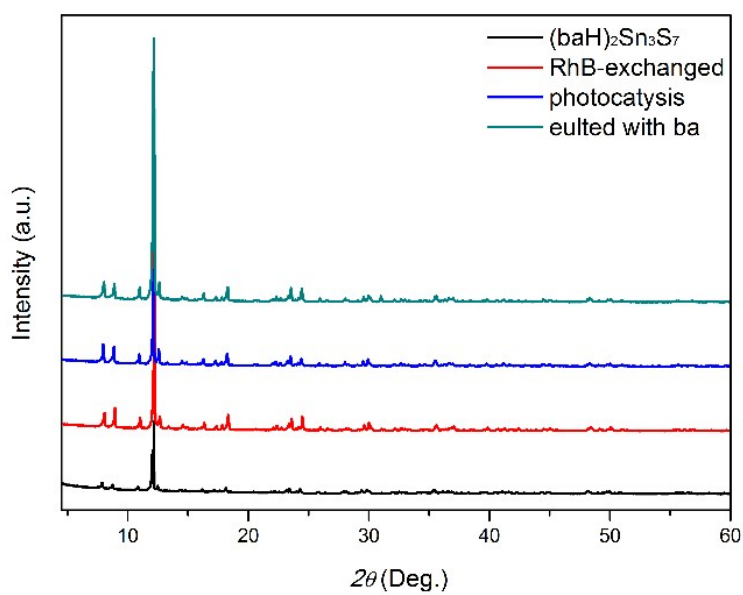
Adsorbent	Capacity(mg/g)	Ref.
(baH) <sub>2</sub> Sn <sub>3</sub> S <sub>7</sub>	1305	This work
(oaH) <sub>2</sub> Sn <sub>3</sub> S <sub>7</sub>	1149	This work
(haH) <sub>2</sub> Sn <sub>3</sub> S <sub>7</sub>	1047	This work
HP-TPPO-3	828.6	Ref <sup>33</sup>
In-MOF@GO-2	267	Ref <sup>34</sup>
Au-CN <sub>x</sub>	250	Ref <sup>35</sup>
SnS <sub>2</sub>	200.0	Ref <sup>36</sup>
HSA	185.61	Ref <sup>37</sup>
W <sub>18</sub> O <sub>49</sub> nanowire	120	Ref <sup>38</sup>
SnS <sub>2</sub> /rGO	94.07	Ref <sup>39</sup>
MoS <sub>2</sub>	49.2	Ref <sup>40</sup>



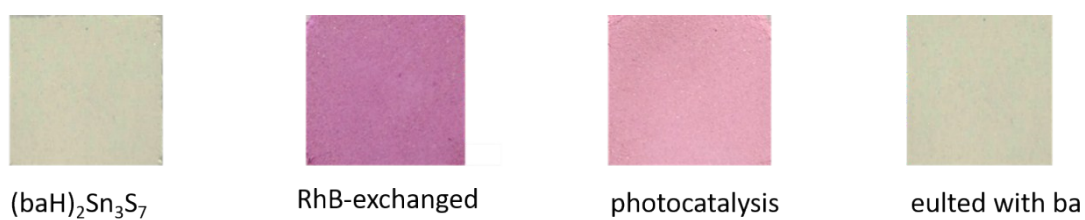
**Figure.S3** The PXRD of A<sub>2</sub>Sn<sub>3</sub>S<sub>7</sub> in solution with different pH at a specific time.



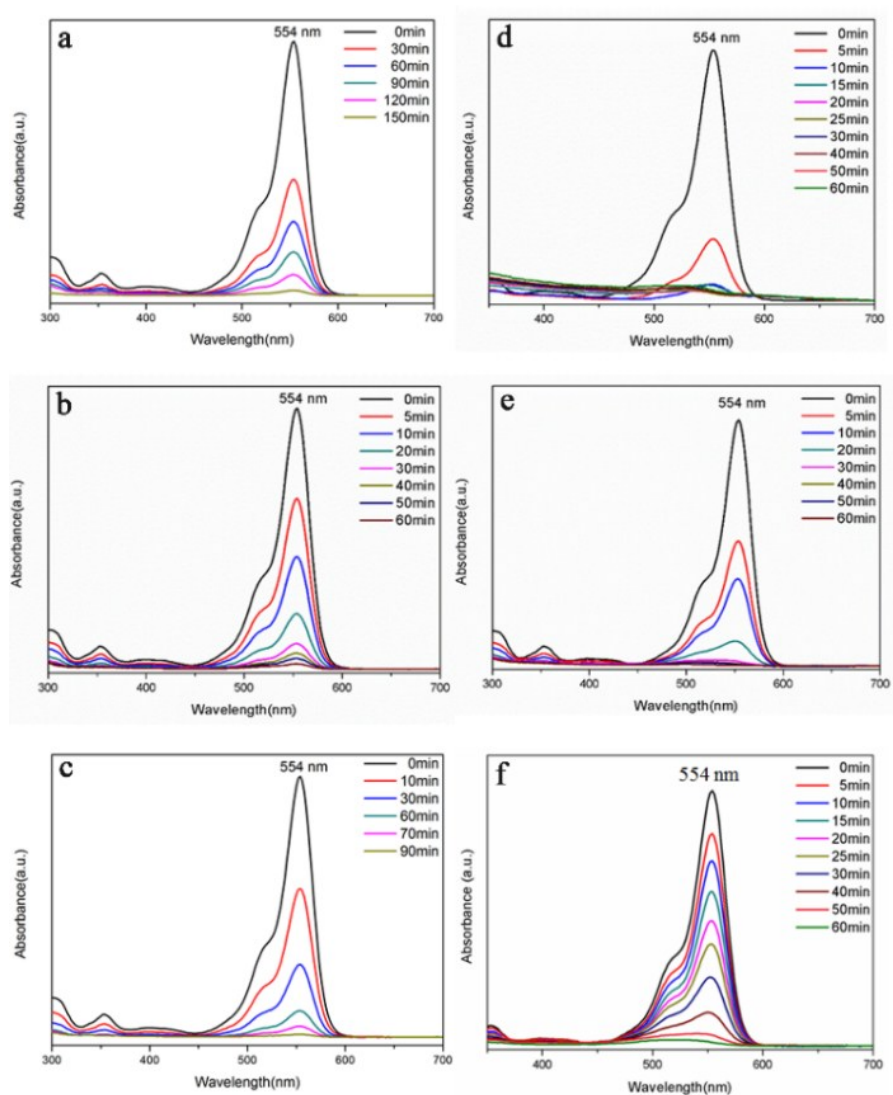
**Figure.S4** The Fluorescence intensity of undegraded RhB in photocatalytic process with multiple dispersive liquid-liquid microextraction for three extraction cycles



**Figure.S5** PXRD of (baH)<sub>2</sub>Sn<sub>3</sub>S<sub>7</sub> catalyst in cycle process.



**Figure.S6** The colored photos of (baH)<sub>2</sub>Sn<sub>3</sub>S<sub>7</sub> powder undergo ion-exchange, photocatalysis and elution in a cycle experiment



**Figure.S7** UV-Vis absorption spectra of RhB in blank experiments without illumination (a) (baH)<sub>2</sub>Sn<sub>3</sub>S<sub>7</sub>, (b) (haH)<sub>2</sub>Sn<sub>3</sub>S<sub>7</sub> and (c) (oaH)<sub>2</sub>Sn<sub>3</sub>S<sub>7</sub>; The UV-Vis absorption spectra of RhB in integrated adsorption-photocatalysis experiments (d) (baH)<sub>2</sub>Sn<sub>3</sub>S<sub>7</sub>, (e) (haH)<sub>2</sub>Sn<sub>3</sub>S<sub>7</sub> and (f) (oaH)<sub>2</sub>Sn<sub>3</sub>S<sub>7</sub>.