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## **Electronic Supplementary Information**

Microbial Synthesis of Hollow Porous Prussian blue@yeast

Microspheres and their synergistic enhancement of organic pollutant
removal performance

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## **Figures**

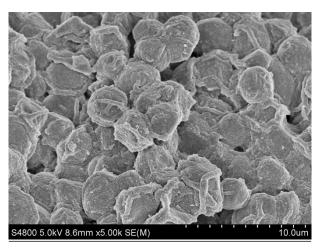


Fig.S1 FE-SEM image of PB@yeast synthesized by chemical precipitation.

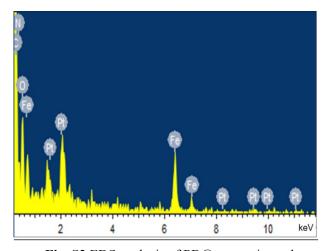


Fig. S2 EDS analysis of PB@yeast microsphere

(a) 
$$CH_3$$
  $CH_3$   $CH_$ 

Fig.S3 Basic molecular structures of (a) fluorescent brightener-CXT; (b) Methylene Blue(MB); (c)

Methyl Orange (MO).

**Tables** 

<u>Table S1. Comparison of the literature results about the photo-Fenton performance of PB-based materials in dyes containing wastewater treatment.</u>

catalysts	Reaction conditions					Major findings			Ref.
	dyes	$\underline{C}_{\underline{0}}$	<u>H<sub>2</sub>O<sub>2</sub></u>	Catal.	<u>рН</u>	<u>Time</u>	Removal	<u>k</u>	
		$(mg \cdot L^{-1})$	(mmol·L <sup>-1</sup> )	(g L-1)		(min)	<u>rate (%)</u>	(min <sup>-1</sup> )	
PB NPs	RhB	<u>7.5</u>	1.0	0.1	<u>5.0</u>	<u>120</u>	<u>80</u>	0.0048	[1]
PB/TiO <sub>2</sub>	<u>RhB</u>	<u>12</u>	$0.4*10^{3}$	<u>1.0</u>	<u>5.0</u>	<u>30</u>	<u>95</u>	0.1020	[2]
CNT/PB	MO	0.08	1.0	Ξ	6.0	<u>160</u>	<u>37</u>	0.0024	[3]
$PB/\gamma$ - $Al_2O_3$	<u>OG</u>	<u>90</u>	9.0	<u>6.5</u>	3.0	<u>60</u>	<u>100</u>	<u>0.035</u>	<u>[4]</u>
PB@yeast	CXT	<u>25</u>	0.2*103	0.5	5.0	<u>40</u>	<u>99</u>	<u>0.1467</u>	<u>This</u>
									<u>work</u>