

Support information for

Two AIEE-active α -cyanostilbene derivatives containing BF_2 unit for detecting explosive picric acid in aqueous medium

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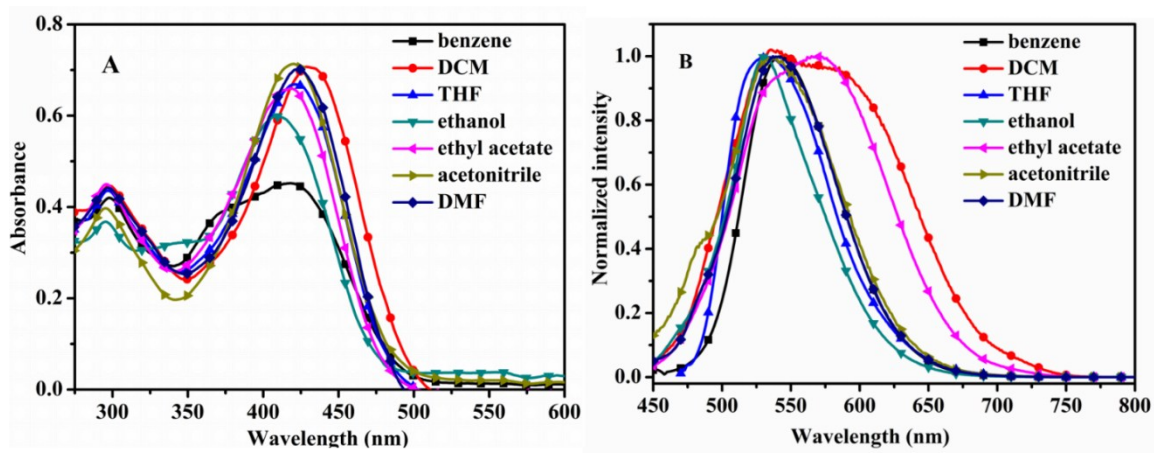


Fig. S1. UV-Vis spectra (A) and fluorescence spectra (B) of the compound **TPA-B** in different solvents.

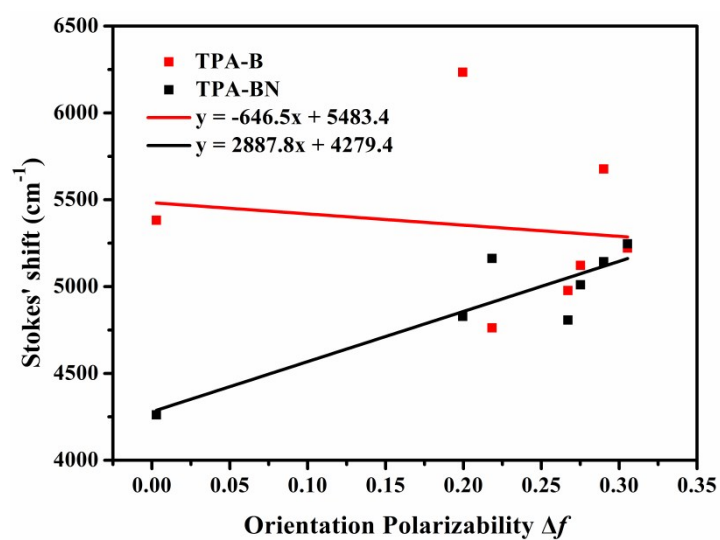


Fig. S2. Lippert-Mataga diagram of compounds **TPA-B** and **TPA-BN**.

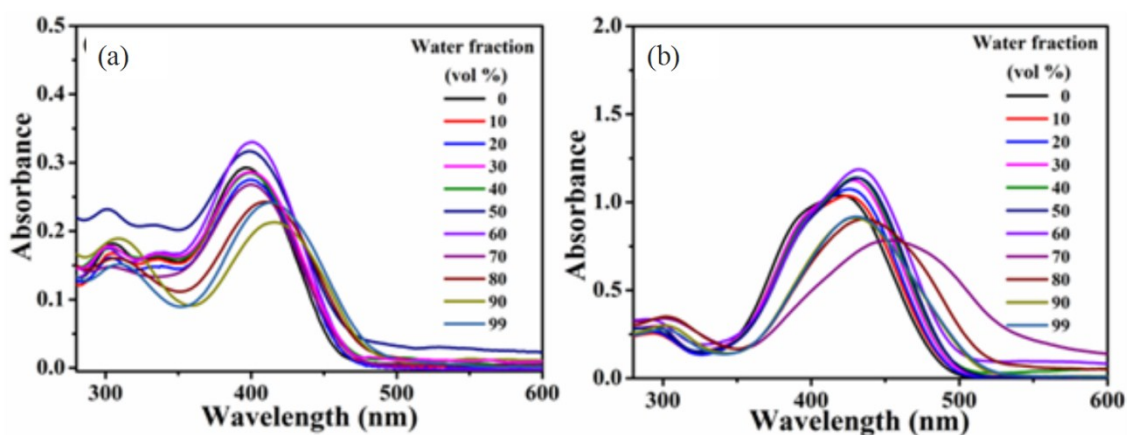


Fig. S3 UV-Vis spectra of **TPA-B** (a) and **TPA-BN** (b) in THF/water mixtures with different water volume fractions.

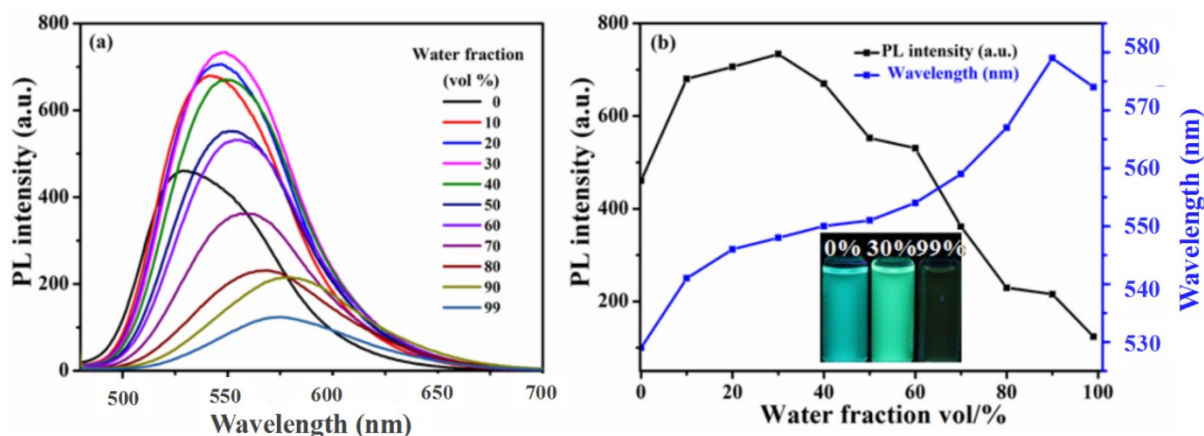


Fig. S4. PL emission (a) spectra changes of **TPA-B** (5.0×10^{-5} M) in THF/ H₂O mixtures with different water volume fractions; Plots of fluorescence intensity determined in THF-H₂O solutions versus water fractions (b). Insets: photos of **TPA-B** in THF-H₂O mixtures ($f_w = 0\%$, 30% and 99%) taken under 365 nm UV lamp.

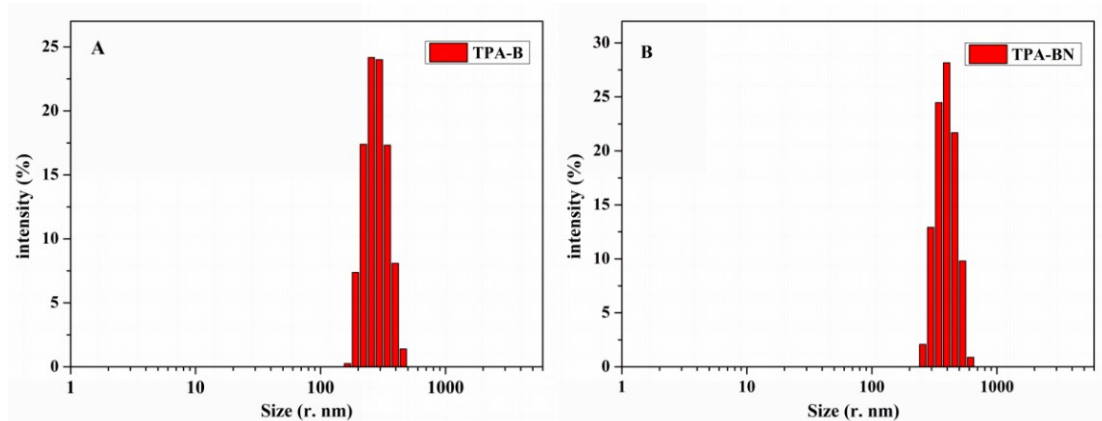


Fig. S5 The particle size analysis of **TPA-B** (A) in THF- H₂O mixtures ($f_w = 30\%$) and **TPA-BN** (B) in THF- H₂O mixtures ($f_w = 80\%$).

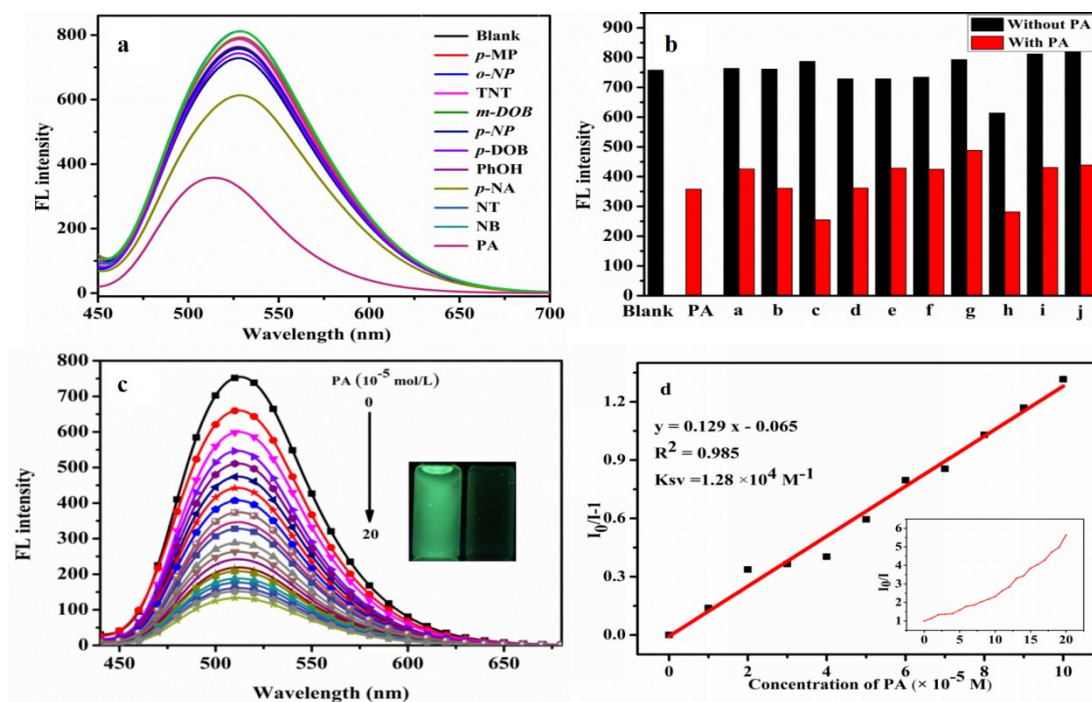


Fig. S6 (a) FL emission spectra obtained for different analytes (100 ppm); (b) quenching percentages of compound TPA-B (10 mM) with different analytes (100 ppm) in THF/water (v/v=2:8) mixtures before (black) and after (red) the addition of 100 ppm. a: *p*-MP, b: *o*-NP, c: TNT, d: *m*-DOB, e: *p*-NP, f: *p*-DOB, g: PhOH, h: *p*-NA, i: NT, j: NB; (c) FL spectra of TPA-B in THF/water (v/v = 7 : 3) containing different amounts of PA; (d) corresponding Stern-Volmer plot for PA detection. Inset: Stern-Volmer plot obtained at a lower concentration of PA.

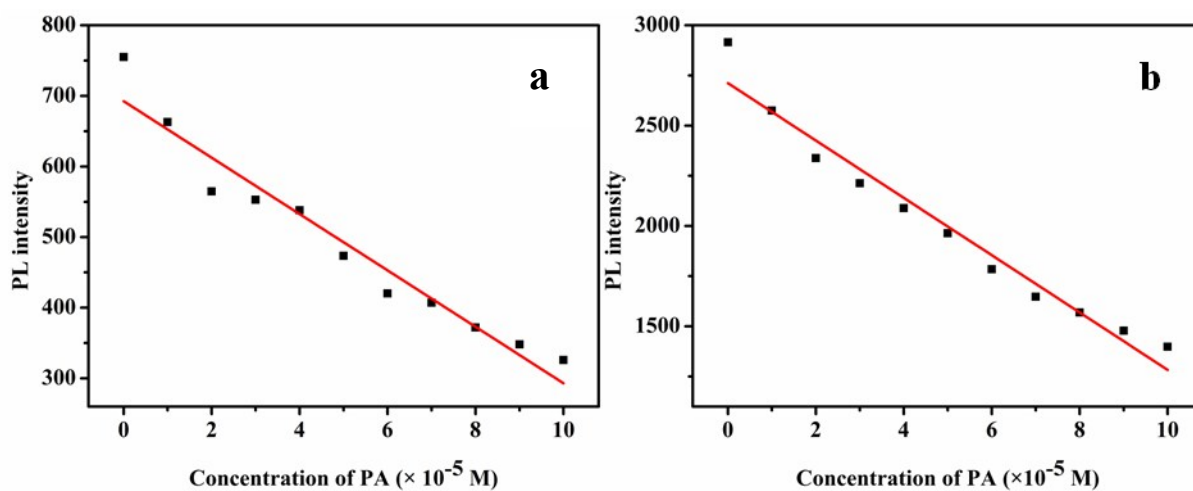
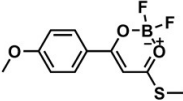
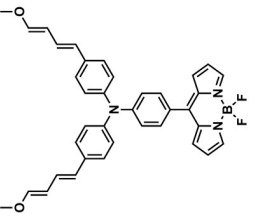
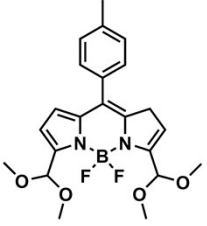
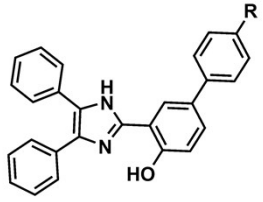


Fig. S7 The linear relationship of TPA-B (a) and TPA-BN (b) between the fluorescence intensity and the PA concentration

Table S1 Performance comparison of organoboron chemosensors reported for PA

Compounds	Detecting system	K_{sv}/M^{-1}	Limit of detection (LOD)	Ref.
	Pure DCM solution	0.5×10^4	21.5×10^{-6}	[58]
	test paper (toluene solution)	2.1×10^6	49.8×10^{-9}	[56]
	In CH_3CN/H_2O solution	9.2×10^6	35.8×10^{-8}	[59]
 $R = NMe_2$ $= NPh_2$	Pure THF solution	$0.7 \times 10^4 /$ 1.0×10^4	/	[57]
TPA-BN/TPA-B	THF/ H_2O mixtures	$1.28 \times 10^4 /$ 1.07×10^4	$1.26 \times 10^{-6} /$ 1.51×10^{-6}	This work

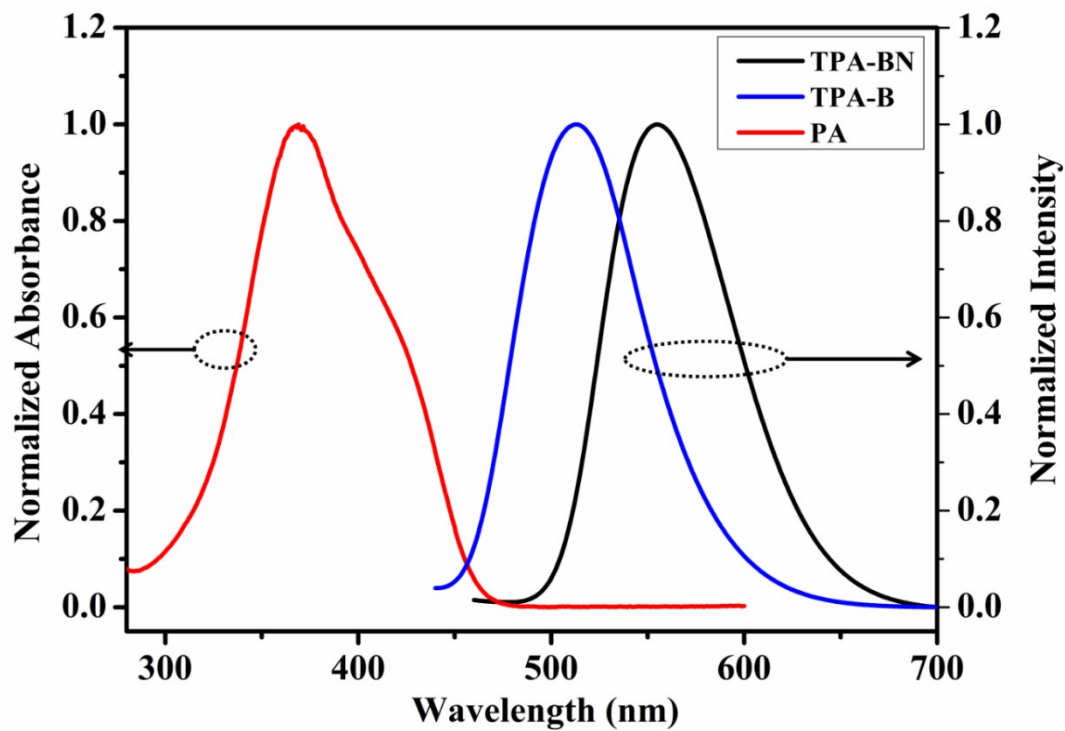


Fig. S8. UV-vis absorption spectra of PA; Normalized Fluorescence of TPA-B and TPA-BN.