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CIDNP as a tool to unveil the reaction mechanism: Interaction of mixed phosphonium-iodonium ylide with *p*-methoxyphenylacetylene

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1. Time-evolution of the ³¹P NMR spectra after mixing ylide 1 and acetylene 2



Fig. S1. ³¹P NMR spectrum of ylide 1.

Phosphonium salt 5 (δ 21.4 ppm) is the first product detected in 120 s after mixing the reagents. (Fig. S2)



Fig. S2. ³¹P NMR spectrum of the mixture 1 + 2 at 120 s after mixing

In 6 s two more products appear (δ 14.4 and 14.1 ppm), with the increasing signal of salt 5. The product with δ 14.4 ppm is unidentified, and the signal at 14.1 ppm belongs to 4. (Fig. S3)



Fig. S3. ³¹P NMR spectrum of the mixture 1 + 2 at 126 s after mixing.

The signal of **5** continues to grow and at 150 s new intermediates with δ 10.2 and -3.65 ppm appear, with the latter signal being emissive. (Fig. S4)



Fig. S4. ³¹P NMR spectrum of the mixture 1 + 2 at 150 s after mixing.

During the following 48 s the ylide concentration decreases, the signal of salt 5 drastically increases, the signals at 14.4 and 14.1 ppm pass their maxima and the signal of 4 becomes zero at 198 s. By this time several signals of low negative intensity appear at δ 22.3, 2.2 and – 3.9 ppm. (Fig. S5)



Fig. S5. ³¹P NMR spectrum of the mixture 1 + 2 at 198 s after mixing.

Further on the signal of **4** becomes emissive and attains a negative minimum and the signal of salt **5** attains an absorption maximum at 288 s. By this time the signal at 14.4 ppm decreases and low positive signals at 16.9, 15.4, -2.75 and -2.8 ppm appear.(Fig. S6)



Fig. S6. ³¹P NMR spectrum of the mixture 1 + 2 at 288 s after mixing.

The signal of 4 begins to grow and at 360 s crosses a zero point and continues to grow further, whereas the signal of salt 5 decreases. (Figs. S7, S8)



Fig. S7. ³¹P NMR spectrum of the mixture 1 + 2 at 360 s after mixing.



Fig. S8. ³¹P NMR spectrum of the mixture 1 + 2 at 390 s after mixing.

The ylide concentration decreases and at 420 s (7 min) **1** is consumed completely. By this time the signal of **4** almost attains its final value and stops increasing, whereas the signal of **5** decreases for 2 min more and attains its final value at 9 min. All minor emission signals

disappear by 360 s and the products with signals at 22.3, 2.2 and -3.9 ppm present as minor products in the final mixture. The signals at 14.4 and 10.2 ppm disappear by 378–390 s. The signals at 16.9, 15.4, -2.75 and -2.8 ppm attain their highest values by 7–8 min and then slowly decrease: signals at 15.4 and -2.75 ppm disappear in 30 min and the signals at 16.9 and -2.8 ppm in 150 min. (Figs. S9–S11)



Fig. S10. ³¹P NMR spectrum of the mixture 1 + 2 at 15 min after mixing.



Fig. S11. ³¹P NMR spectrum of the mixture 1 + 2 at 150 min after mixing.



Figure S12 Kinetics of normalized integral yields of the main products, numbers correspond to the numbers of reagents and products in Scheme 1, for Σ 3 see the text.

2. Time-evolution of the ¹H NMR spectra after mixing ylide 1 and acetylene 2 in DCM_{d2} [1]/[2] = 1/3



Fig. S13 Initial ¹H NMR spectrum after mixing the reagents



Fig. S14 ¹H NMR spectrum in 6 min after mixing the reagents.



Fig. S15 ¹H NMR spectrum in 15 min after mixing the reagents.

3. Time-evolution of the ¹H NMR spectra after mixing ylide 1 and acetylene 2 in DCM [1]/[2] = 1/3, C₆D₆ is added as a standard (δ (C₆H₆) =7.15 ppm)



Fig. S17 ¹H NMR spectrum in 5 min after mixing the reagents.



Fig. S18 ¹H NMR spectrum in 15 min after mixing the reagents.