## Electronic Supporting Information (ESI) for

## A dinuclear cobalt cluster as electrocatalyst for oxygen reduction reaction

Yun-Wu Li, Wen-Jie Zhang, Chun-Xia Li, Lin Gu, Hong-Mei Du, Hui-Yan Ma,\* Su-Na Wang,\* and Jin-Sheng Zhao\* Shandong Provincial Key Laboratory / Collaborative Innovation Center of Chemical Energy Storage and Novel Cell Technology, and School of Chemistry and Chemical Engineering, Liaocheng University, Liaocheng 252000, PR China.

<sup>\*</sup>Corresponding authors.

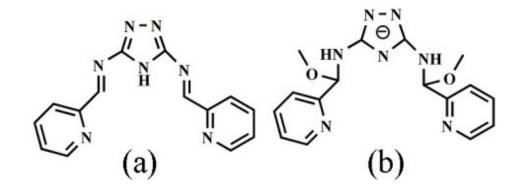
E-mail addresses: mahuiyanyan@163.com (H. Y. Ma), wangsuna@lcu.edu.cn (S. N. Wang), j.s.zhao@163.com (J. S. Zhao).

## **S1. Experimental Section**

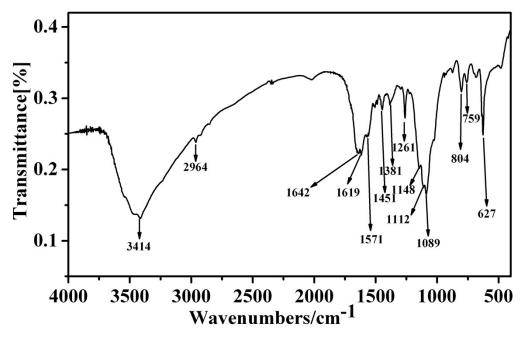
*Syntheses of catalyst*: Dinuclear  $\{Co^{II}_2\}$  cluster catalyst was synthesized according our reported experiment [reference 13 in main text] and as bellow:

A mixture of ligand Hpmat (27.7 mg, 0.1 mmol) in CHCl<sub>3</sub>/MeOH (10/2 mL) was stirred and dissolved under 60 °C silicone oil bath. Then a solution of  $Co(ClO_4)_2 \cdot 6H_2O$  (55.01 mg, 0.15 mmol) in 5 mL MeOH was added dropwise in the above solution. The mixture stirred in 60 °C bath for 8 hours and then cooled to room temperature and filtered. Purple needle (macrography) shaped crystals of {Co<sub>2</sub>} catalyst were collected. Yield: 13% based on Co. Elemental analysis (%) for {Co<sub>2</sub>} catalyst, C<sub>16</sub>H<sub>18</sub>ClCoN<sub>7</sub>O<sub>6</sub> (M = 498.74): *Calcd*.: C, 38.53; H, 3.64; N, 19.66; *Found*: C, 38.81; H, 3.32; N, 19.46. IR (KBr disk, cm<sup>-1</sup>) see **Fig. S2** in **ESI**.

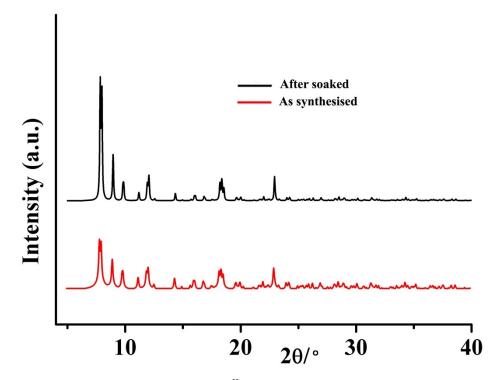
## **S2.** Figures in Supporting Information



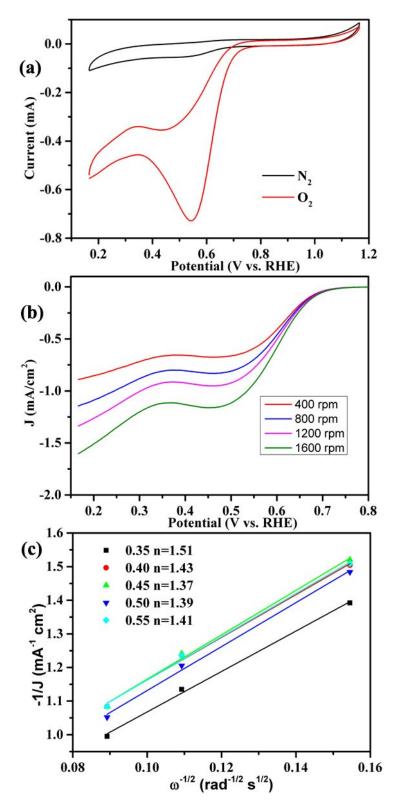
**Fig. S1** The initial Hpmat ligand (a) and final pmmat<sup>-</sup> ligand (b) of  $\{Co^{II}_2\}$  catalyst.



**Fig. S2** IR spectra of  $\{Co^{II}_2\}$  catalyst.



**Fig. S3** (a) The PXRD of as synthesised  $\{Co^{II}_2\}$  catalyst. (b) The PXRD of catalyst after soaked in mixture ink.



**Fig. S4** (a) CV curves of free ligand Hpmat. (b) LSV curves of free ligand Hpmat. (c) K-L plots of free ligand Hpmat at different potentials.

Compound	{Co <sup>II</sup> <sub>2</sub> } catalyst
Formula	C <sub>16</sub> H <sub>18</sub> ClCoN <sub>7</sub> O <sub>6</sub>
Fw	498.74
$\lambda/{ m \AA}$	0.71073
T/K	298(2)
Crystal system	Orthorhombic
Space group	Рсса
<i>a</i> [Å]	22.1561(18)
<i>b</i> [Å]	11.2746(7)
<i>c</i> [Å]	19.7428(17)
α[°]	90
β[°]	90
γ[°]	90
<i>V</i> (Å <sup>3</sup> )	4931.8(7)
Ζ	4
$D_c/\mathrm{Mg}\cdot\mathrm{m}^{-3}$	1.343
F(000)	2040
Reflections collected/unique	24089/4364
<i>R</i> <sub>int</sub>	0.2293
Data/Restraints/Parameters	4364/86/290
$R_1/wR_2 [I > 2\sigma(I)]^a$	0.2000/0.4782
$R_1/wR_2$ [(all data)] <sup><i>a</i></sup>	0.2977/0.5397
GOF on $F^2$	1.175
$^{a}R_{I} = \Sigma(  F_{0}  -  F_{C}  ) / \Sigma  F_{0}  \ wR_{2} =  $	$[\Sigma w( F_0 ^2 -  F_C ^2)^2 / (\Sigma w F_0 ^2)^2]^{1/2}.$

**S3. Table S1** Crystal Data and Structure Refinement Parameters for  $\{Co^{II}_2\}$  catalyst.