Electronic Supporting Information (ESI) for

A dinuclear cobalt cluster as electrocatalyst for oxygen reduction reaction

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S1. Experimental Section

Syntheses of catalyst: Dinuclear $\{Co^{II}_2\}$ cluster catalyst was synthesized according our reported experiment [reference 13 in main text] and as bellow:

A mixture of ligand Hpmat (27.7 mg, 0.1 mmol) in CHCl₃/MeOH (10/2 mL) was stirred and dissolved under 60 °C silicone oil bath. Then a solution of $Co(ClO_4)_2 \cdot 6H_2O$ (55.01 mg, 0.15 mmol) in 5 mL MeOH was added dropwise in the above solution. The mixture stirred in 60 °C bath for 8 hours and then cooled to room temperature and filtered. Purple needle (macrography) shaped crystals of {Co₂} catalyst were collected. Yield: 13% based on Co. Elemental analysis (%) for {Co₂} catalyst, C₁₆H₁₈ClCoN₇O₆ (M = 498.74): *Calcd*.: C, 38.53; H, 3.64; N, 19.66; *Found*: C, 38.81; H, 3.32; N, 19.46. IR (KBr disk, cm⁻¹) see **Fig. S2** in **ESI**.

S2. Figures in Supporting Information

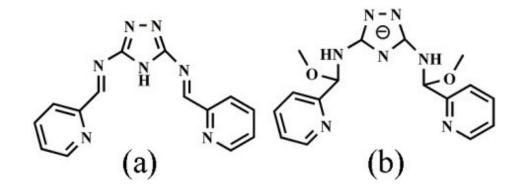


Fig. S1 The initial Hpmat ligand (a) and final pmmat⁻ ligand (b) of $\{Co^{II}_2\}$ catalyst.

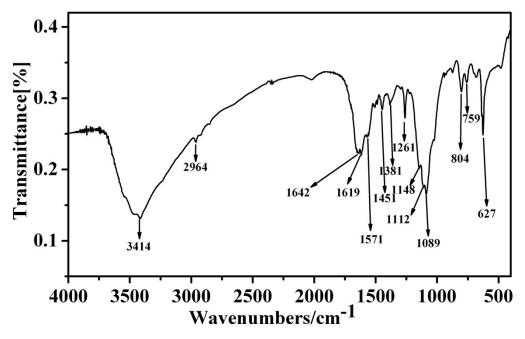


Fig. S2 IR spectra of $\{Co^{II}_2\}$ catalyst.

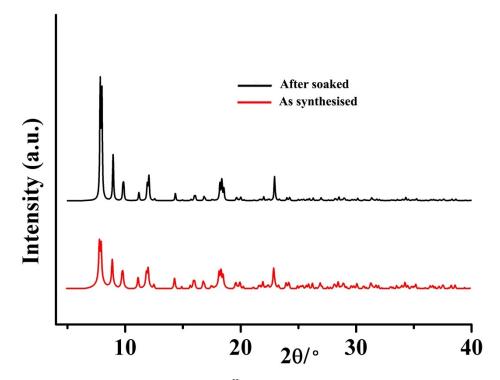


Fig. S3 (a) The PXRD of as synthesised $\{Co^{II}_2\}$ catalyst. (b) The PXRD of catalyst after soaked in mixture ink.

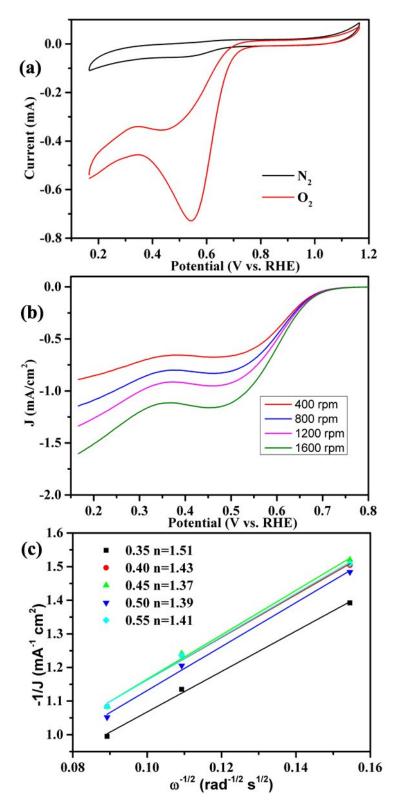


Fig. S4 (a) CV curves of free ligand Hpmat. (b) LSV curves of free ligand Hpmat. (c) K-L plots of free ligand Hpmat at different potentials.

Compound	{Co ^{II} ₂ } catalyst
Formula	C ₁₆ H ₁₈ ClCoN ₇ O ₆
Fw	498.74
$\lambda/{ m \AA}$	0.71073
T/K	298(2)
Crystal system	Orthorhombic
Space group	Рсса
<i>a</i> [Å]	22.1561(18)
<i>b</i> [Å]	11.2746(7)
<i>c</i> [Å]	19.7428(17)
α[°]	90
β[°]	90
γ[°]	90
<i>V</i> (Å ³)	4931.8(7)
Ζ	4
$D_c/\mathrm{Mg}\cdot\mathrm{m}^{-3}$	1.343
F(000)	2040
Reflections collected/unique	24089/4364
<i>R</i> _{int}	0.2293
Data/Restraints/Parameters	4364/86/290
$R_1/wR_2 [I > 2\sigma(I)]^a$	0.2000/0.4782
R_1/wR_2 [(all data)] ^{<i>a</i>}	0.2977/0.5397
GOF on F^2	1.175
$^{a}R_{I} = \Sigma(F_{0} - F_{C}) / \Sigma F_{0} \ wR_{2} = $	$[\Sigma w(F_0 ^2 - F_C ^2)^2 / (\Sigma w F_0 ^2)^2]^{1/2}.$

S3. Table S1 Crystal Data and Structure Refinement Parameters for $\{Co^{II}_2\}$ catalyst.