Supporting Information

Metallic Ti₃O₅ Hierarchical Porous Microspheres with Enhanced Photothermal Property

Ruifeng Du,^{a,b} Wei Liu,^b Hua Bai,^b Hongtao Wang,^a* and Guangcheng Xi^b*

^aSchool of Chemical and Material Engineering, Fuyang Normal University, Fuyang 236037, China

Email: hwang@fynu.edu.cn

^bInstitute of Industrial and Consumer Product Safety, Institution Chinese Academy of

Inspection and Quarantine, No. 11, Ronghua South Road, Beijing 100176, P. R. China.

E-mail: xiguangcheng@caiq.org.cn

1. Caculation of Photothermal Conversion Efficiency of γ-Ti₃O₅ HPMs

According to the Beer-Lambert Law, the mass extinction coefficient α of the γ -Ti₃O₅ HPMs can be calculated using equation (1):

$$A = \alpha LC \qquad (1)$$

In equation (1), A refers to the absorbance of γ -Ti₃O₅ HPMs at 808 nm, α is the mass extinction coefficient of γ -Ti₃O₅ (L g⁻¹ cm⁻¹), L is the optical length of the quartz cuvette (cm), and C is the mass concentration (g L⁻¹). According to Figure S5, the calculated mass extinction coefficient of Ti₃O₅ is 0.87 L g⁻¹ cm⁻¹.

The photothermal conversion efficiency (η) of the γ -Ti₃O₅ HPMs was determined by equation (2) as the previous reports^[31,42,43]. The temperature change of the γ -Ti₃O₅ aqueous solution (1 mg mL⁻¹) was recorded as a function of time under continuous irradiation of 808 nm NIR laser at a power of 1 W cm⁻², in which the irradiation lasted for 600 s, and then the laser was shut off (figure 4e,f).

$$\eta = \frac{hS(T_{max} - T_{surr}) - Q_{Dis}}{I(1 - 10^{-A308})}$$
(2)

Following the previous reports, the value of hS can be calculated by equation (3) - (5). The T, T_{max} and T_{sur} are random temperature, the maximum temperature after irradiation and the surrounding temperature. In this work, we measured the temperature change at concentration of 1mg/mL, the T_{max} - T_{surr} = 45.3 °C, and the τ_s was calculated to be 345.82 s by equation (3) and (4) as Figure 4 e and f. m and C_p in equation (5) are the mass and heat capacity of solvent (water). In this work m=1 g, C = 4.2 J g⁻¹ °C. The hS value was determined to be 11.93 mW / °C.

$$\theta = \frac{T - Tsur}{Tmax - Tsur}$$
(3)

$$t = \tau s \ln \theta$$
(4)
$$hS = \frac{\sum_{i} \text{miCp}, i}{\tau s}$$
(5)

Take the hS value into equation (2). At concentration of 1 mg/mL γ -Ti₃O₅ HPMs, A₈₀₈=0.865. Q_{Dis} expresses the heat dissipated due to light absorption by the pure water without γ -Ti₃O₅. It is measured to be 15.96 mW. *I* is the incident laser power (1000 mW). Substituting these values into equation (2), the 808 nm laser photothermal conversion efficiency (η) of γ -Ti₃O₅ HPMs can be calculated to be 65.29 %.

Supporting Figures



Figure S1. SEM images of the as-synthesized TiO₂ HPMs.



Figure S2. XRD pattern of the γ -Ti₃O₅ HPMs.



Figure S3. EDS spectrum of the γ -Ti₃O₅ HPMs.



Figure S4. Ti2p XPS spectrum of the γ -Ti₃O₅ nanosheet microspheres, which demonstrates that Ti⁴⁺ and Ti³⁺ ions coexist in the sample.



Figure S5. The linearly fitted plots of absorbance versus concentration of γ -Ti₃O₅ aqueous suspension at 808 nm.



Figure S6. The stability of γ -Ti₃O₅ HPMs in physiological medium PBS, FBS and RPMI1640. a. The aggration and precipation of γ -Ti₃O₅ HPMs keep in room temperature for 1, 3 and 5 hours. b. Photothermal photographs of γ -Ti₃O₅ HPMs dispersed in FBS for 1, 3 and 5 hours after 10 min laser irradiation at the wavelengths of 808 nm and 980nm (2 W cm-2), respectively. c. Absorption spectra of γ -Ti₃O₅ HPMs (200 µg mL⁻¹) dispersed in ethanol, RPMI1640 and FBS for 1, 7 and 28 days.



Figure S7. Infrared thermographic maps and temperature increase in wells of control and nanoparticles at different time points (Hela cells exposed to 808 nm laser at 2W).



Figure S8. Haematological data of the mice intravenously injected with the γ -Ti₃O₅ HPMs at 1, 7 and 28 days post-injection. The terms are following: white blood cells (WBC), neutrophils (NE), lymphocyte (LY), red blood cells (RBC), haemoglobin (HGB), haematocrit (HCT), mean corpuscular volume (MCV), mean corpuscular haemoglobin (MCH), mean corpuscular haemoglobin concentration (MCHC), red blood cell volume distribution width (RDW), platelets (PLT) and platelet distribution width (PDW).



Figure S9. Histological data (haematoxylin and eosin stained images) obtained from the liver, spleen, kidney, heart and lung of the γ -Ti₃O₅ HPM-treated mice at 1, 7 and 28 days post tail vein injection. Scale bar, 100 µm.