

Electronic Supplementary Information

Unprecedented surface stabilized InP quantum dots with bidentate ligands

Haewoon Seo^{*a}, Meehee Bang^b, Yongjin Kim^a, ChaeYeon Son^a, Heung Bae Jeon^{*b}
and Sang-Wook Kim^{*a}

^aDepartment of Molecular Science and Technology, Ajou University, Suwon 443-749, Korea

^bDepartment of Chemistry, Kwangwoon University, Seoul, 01897, Republic of Korea

Corresponding Author

*Author to whom correspondence should be addressed.

*Email: swkim@ajou.ac.kr

¹H and ¹³C NMR of hexadecane-1,2-dithiol

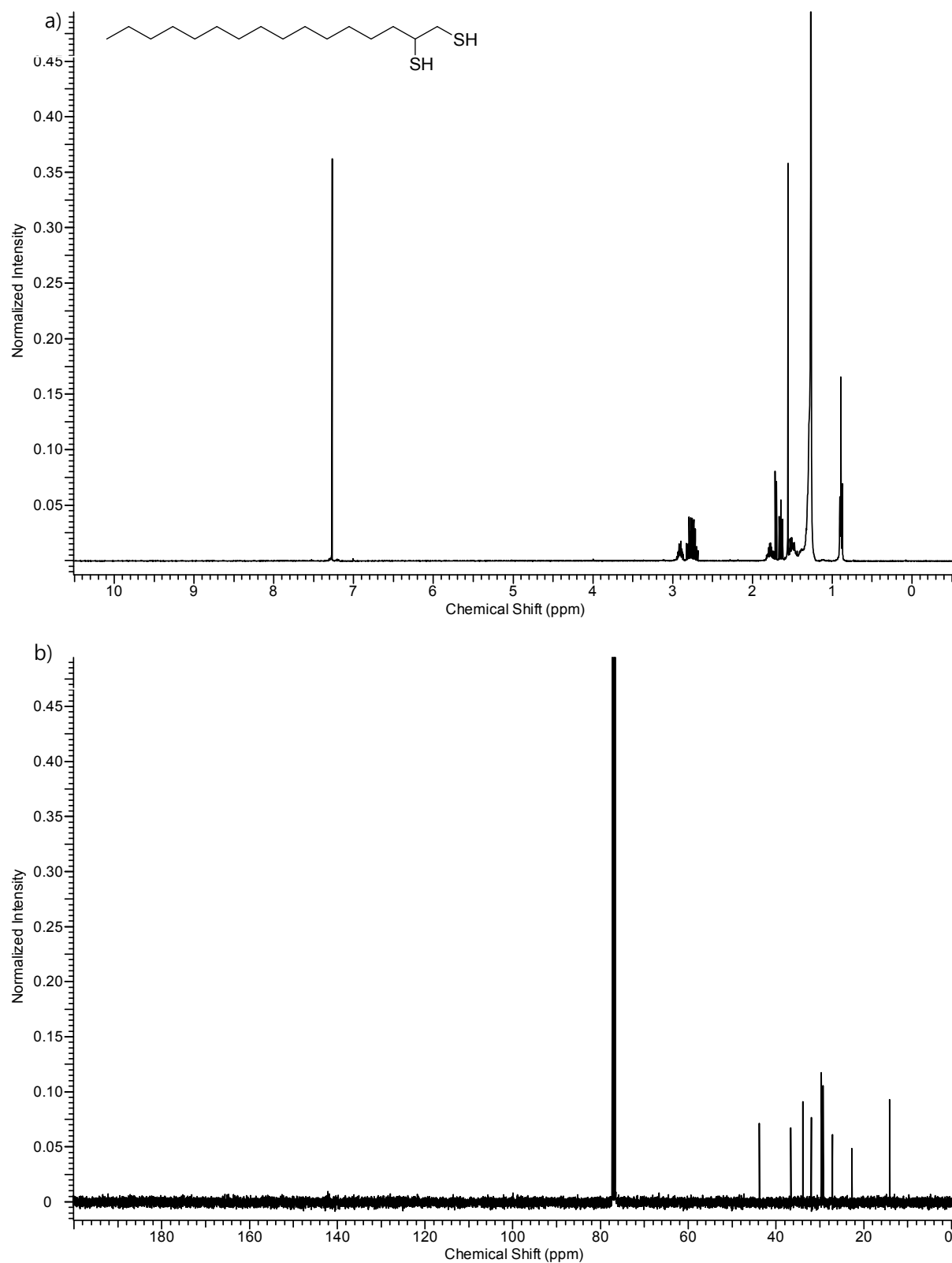


Figure S1. (a) ¹H NMR data of 1,2-hexadecanedithiol, (b) ¹³C NMR data of 1,2-hexadecanedithiol.

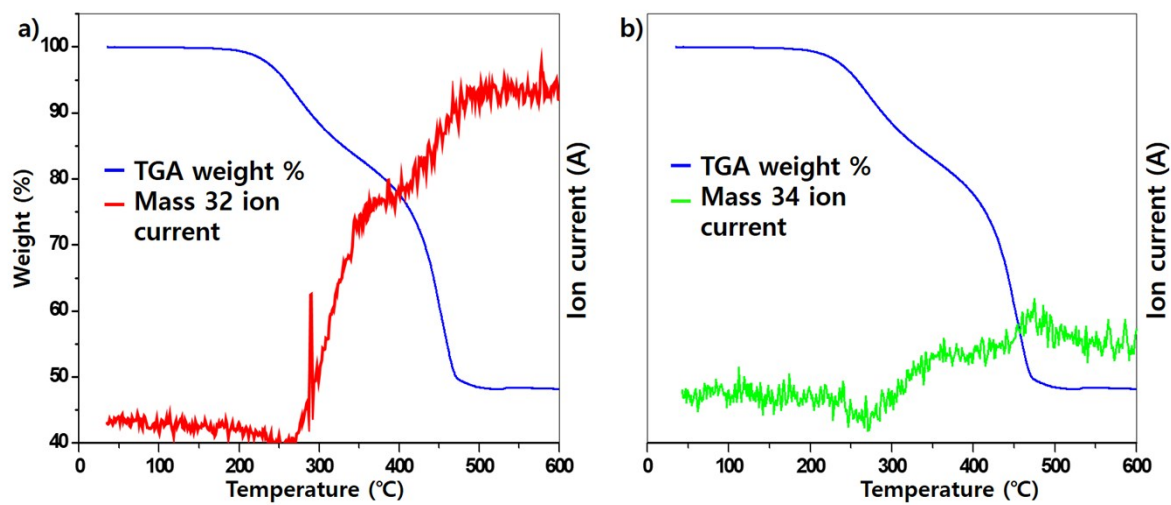


Figure S2. (a) TGA spectrum of dithiol exchanged QD and STA-Mass spectrum of 32 m/z ion, (b) TGA spectrum of dithiol exchanged QD and STA-Mass spectrum of 34 m/z ion.

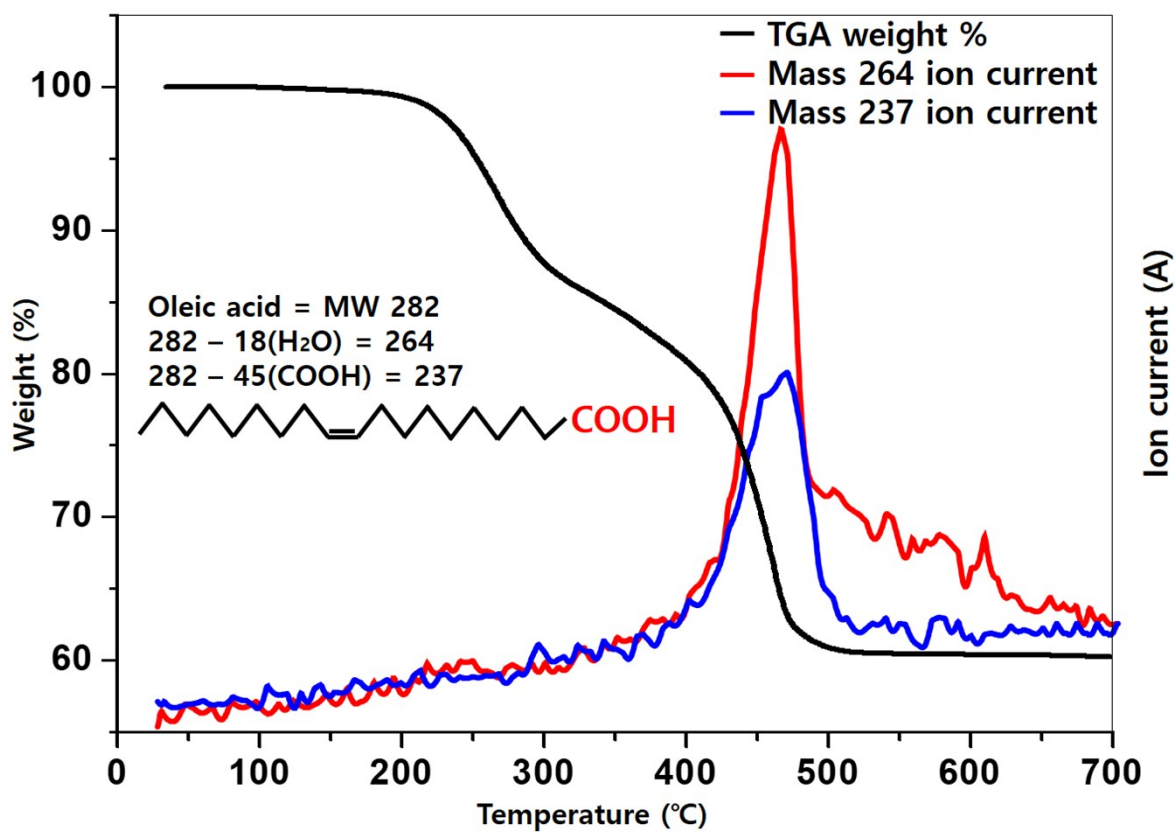


Figure S3. TGA spectrum of dithiol exchanged QD and STA-Mass spectrum of 264 m/z ion and 237 m/z ion.

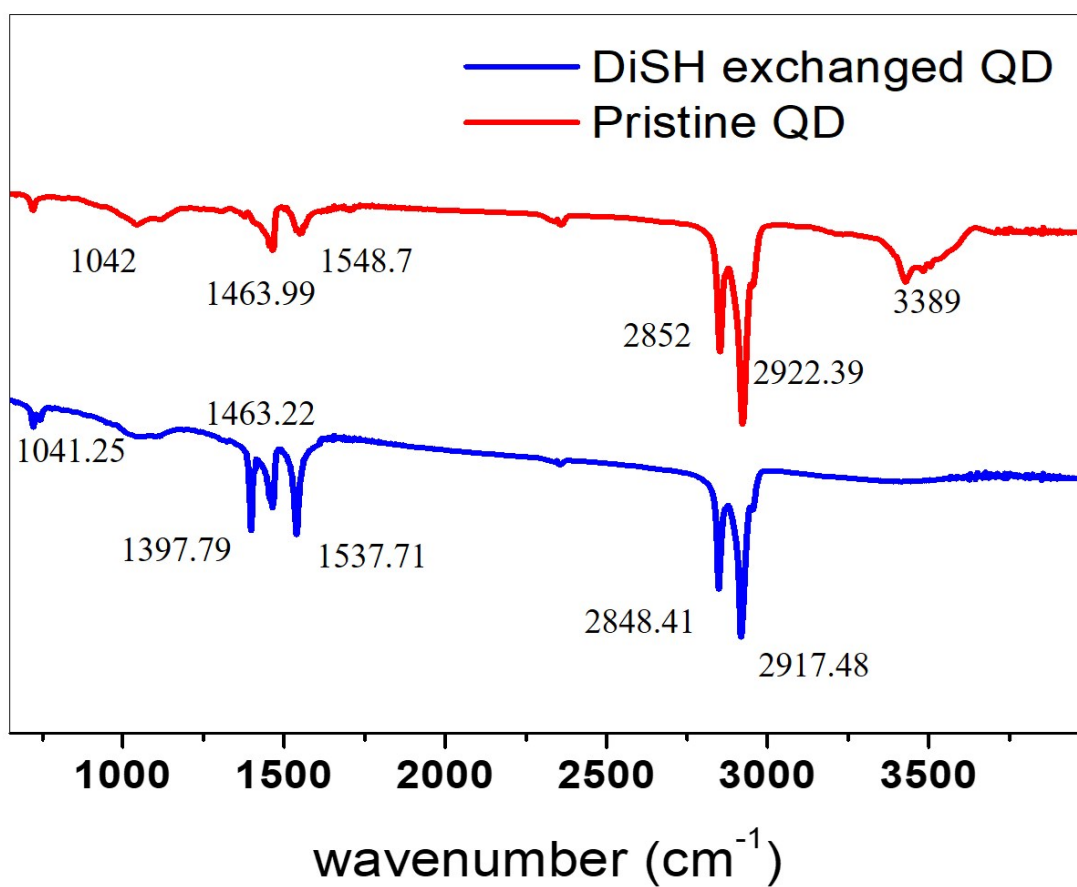


Figure S4. FT-IR spectrum of pristine and dithiol exchanged QD.

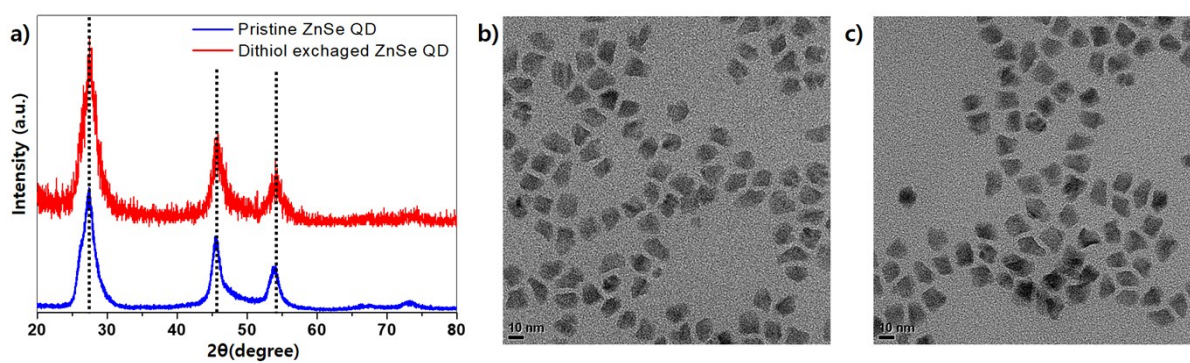


Figure S5. (a) XRD spectrum of pristine and dithiol exchanged ZnSe QD.
TEM image of (b) pristine ZnSe QD and (c) dithiol ligand exchanged ZnSe QD.

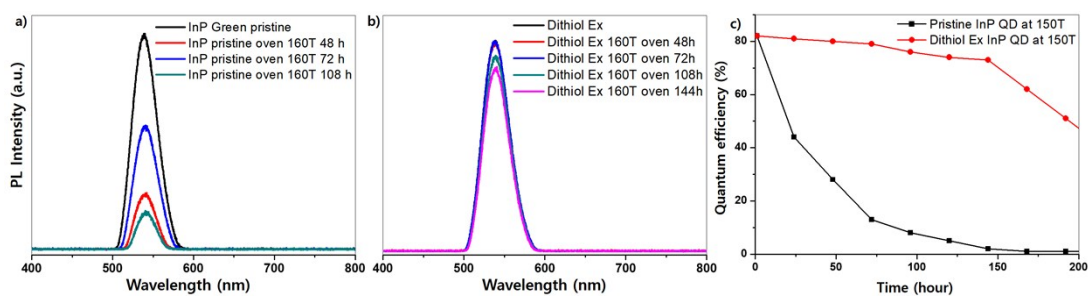


Figure S6. PL emission spectrum of 160°C tested QD (a) pristine InP QD, (b) dithiol ligand exchanged QD, (c) plot of PL QY change

	C(carbon) %	H(hydrogen) %	S(sulfur) %
Reference QD #1	72.314	11.843	1.8794
Reference QD #2	71.394	11.776	1.8847
Dithiol Ex QD #1	63.299	10.487	4.9778
Dithiol Ex QD #2	64.119	10.582	4.5275

Table S1. CHNS elemental analysis data of pristine InP QD and dithiol exchanged InP QD