

Electronic Supplementary Material (ESI) for Journal of Materials Chemistry A
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Supplementary Information

New Reflowing Strategy Based on Lithiophilic Substrate towards Smooth and Stable Lithium Metal Anode

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Figures

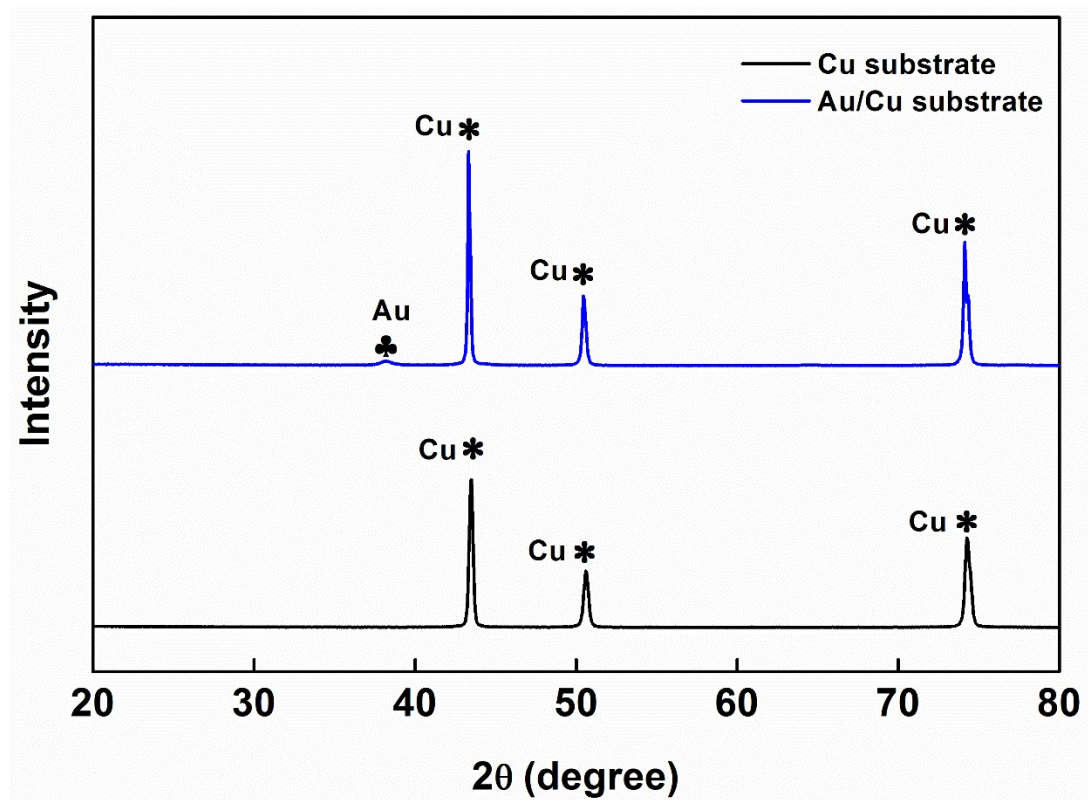


Fig. S1† XRD patterns of Cu and Au/Cu substrates.

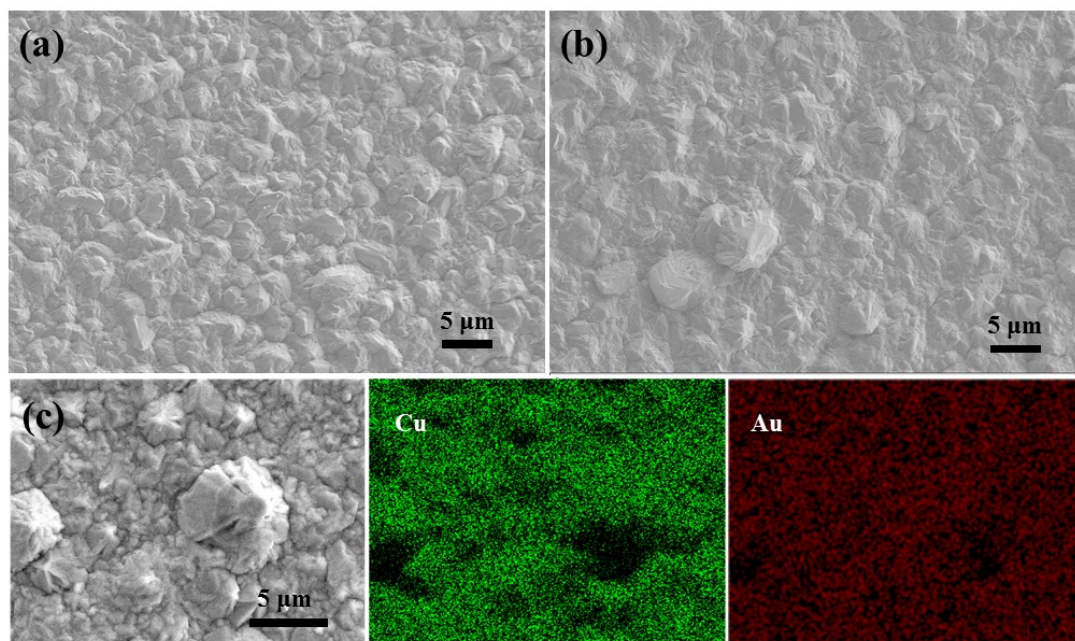


Fig. S2† SEM images of (a) Cu and (b) Au/Cu substrates. (c) The energy dispersive X-ray spectrometer (EDS) mapping for the distribution of carbon and golden elements on the surface of Au/Cu substrate.

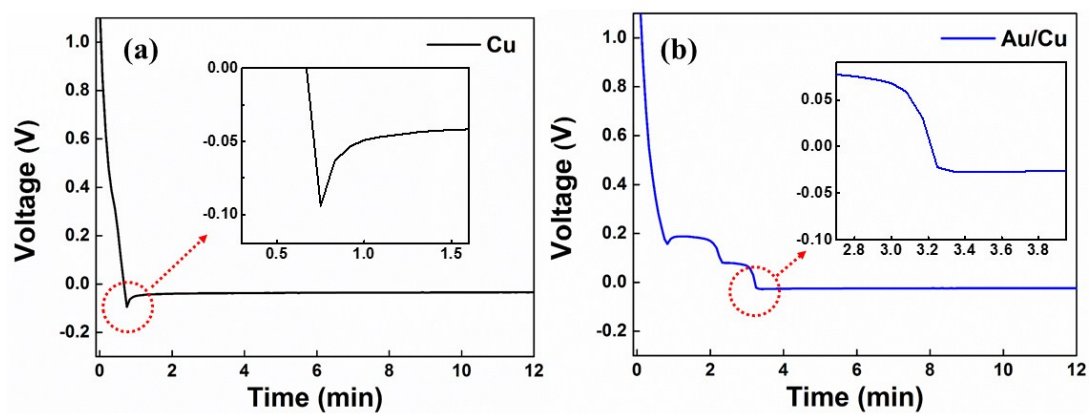


Fig. S3† Voltage profiles during initial Li deposition at 0.5 mA cm⁻² onto a) Cu and b) Au/Cu substrates. Insets in (a) and (b) are the corresponding magnified images.

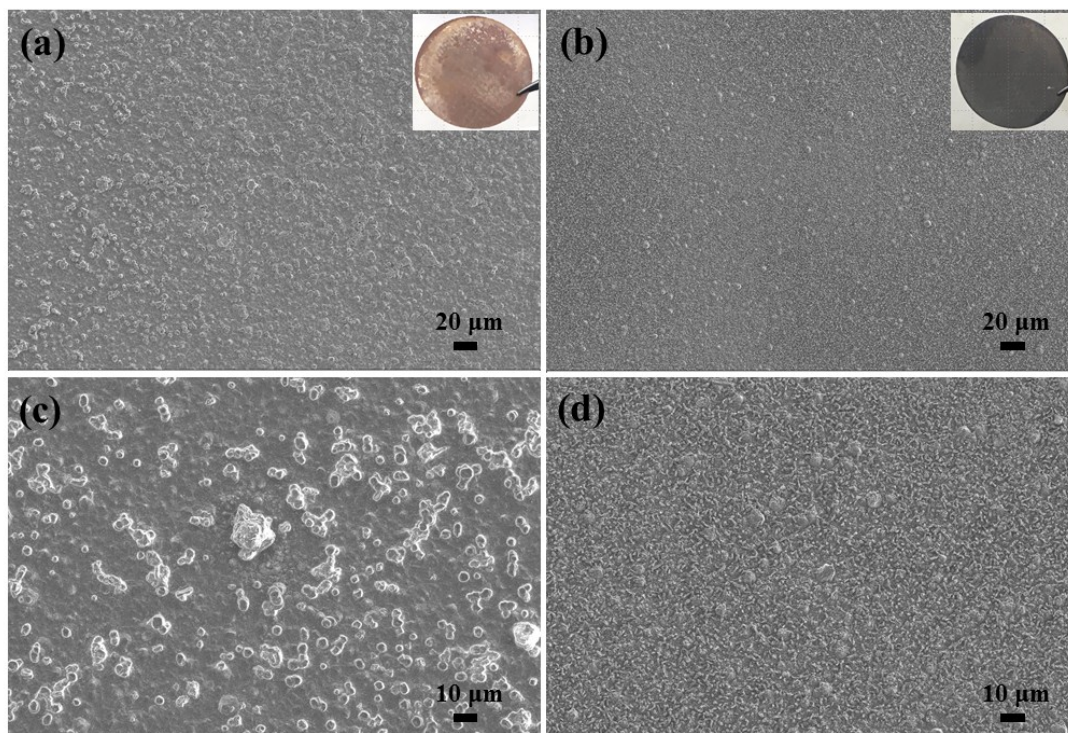


Fig. S4† SEM images of (a, c) Cu and (b, d) Au/Cu substrates with 0.1 mAh cm^{-2} Li deposition under the current density of 0.5 mA cm^{-2} . Insets are their corresponding digital photographs, with each electrode having the diameter of 14 mm.

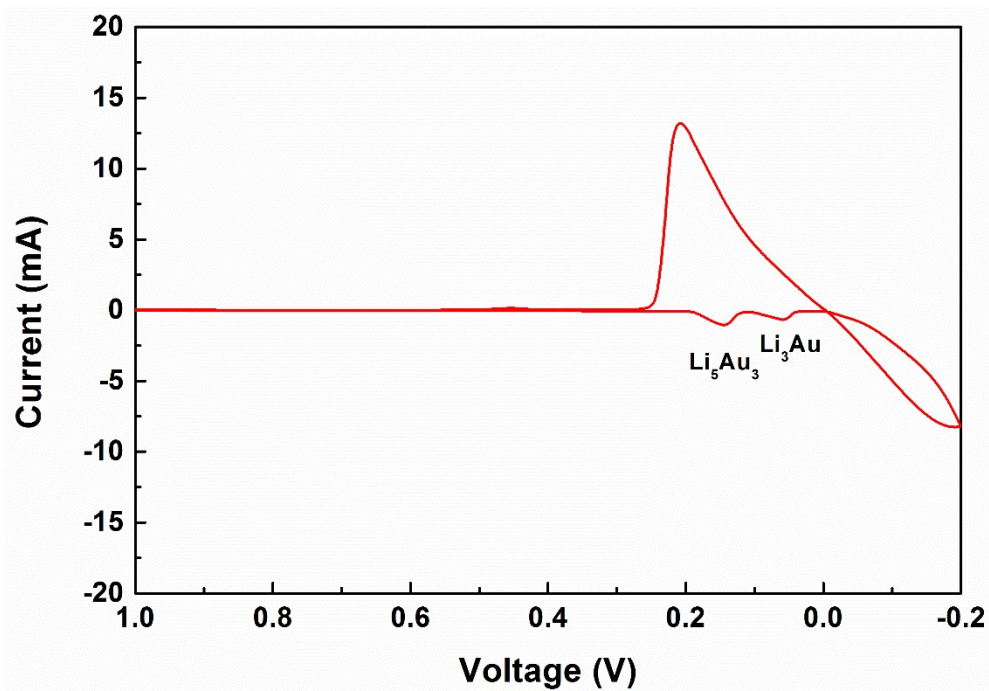


Fig. S5† The CV measurements of the half cell with Au/Cu electrode were conducted with a potential window from +1V to -0.2 V at a scan rate of 0.5 mV s⁻¹.

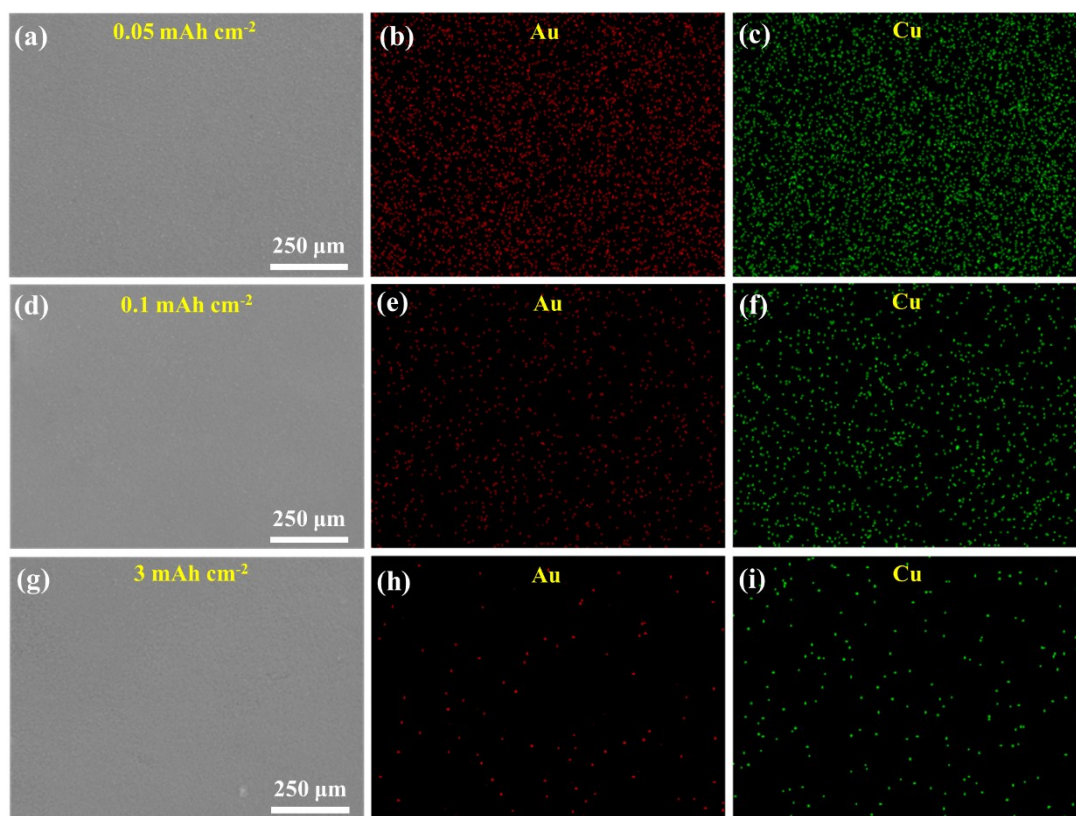


Fig. S6† The EDS mapping result of Au and Cu elements on the surface of Li/Au/Cu electrode with Li plating capacities of (a-c) 0.05, (d-f) 0.1, and (g-i) 3 mAh cm⁻², respectively.

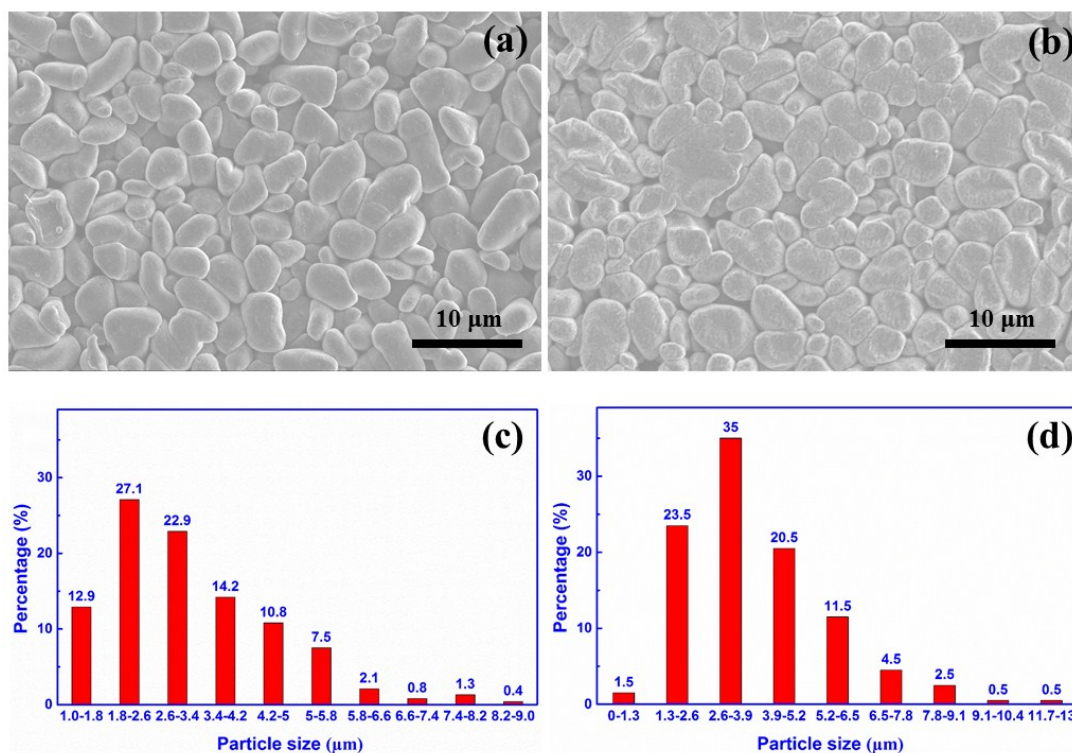


Fig. S7† SEM images of (a) untreated Li/Au/Cu and (b) RF-Li/Au/Cu electrodes with low magnification. The particle size distribution of (c) untreated Li/Au/Cu and (d) RF-Li/Au/Cu electrodes was measured by Nano Measure software. For untreated Li/Au/Cu electrode, there are 40% of particles whose size is below 2.6 nm, indicating there is a large number of small particles. In contrast, there is only 25% of particles for RF-Li/Au/Cu electrode whose size is below 2.6 nm. On the other hand, RF-Li/Au/Cu electrode also has more particles with diameters > 5 nm (more than 19.5%) than that of untreated Li/Au/Cu electrode (only 12.1%).

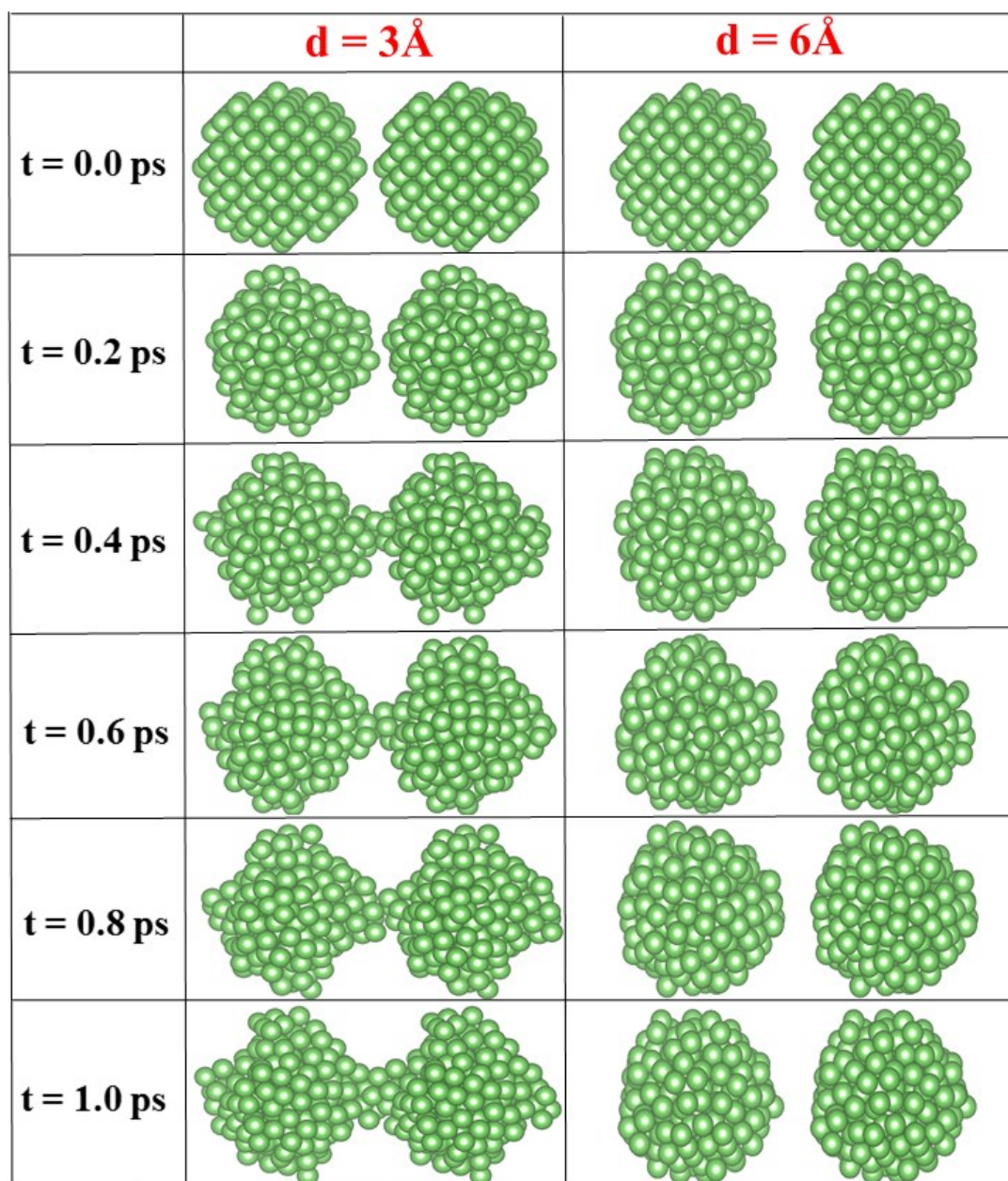


Fig. S8† The evolution processes of Li diffusion for $d=3\text{\AA}$ and $d=6\text{\AA}$ models using the Ab-initio molecular dynamics simulations. For $d=3\text{\AA}$ model, there is a tendency that Li particles diffuse into together after 0.4 ps, while those of $d=6\text{\AA}$ model still in a separate state after 1.0 ps.

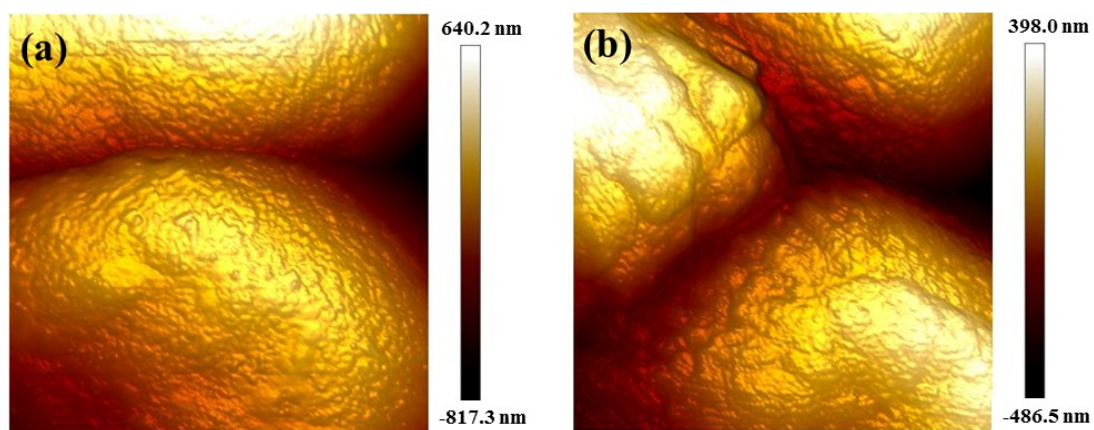


Fig. S9† 2D AFM height images of (a) untreated Li/Au/Cu and (b) RF-Li/Au/Cu electrodes.

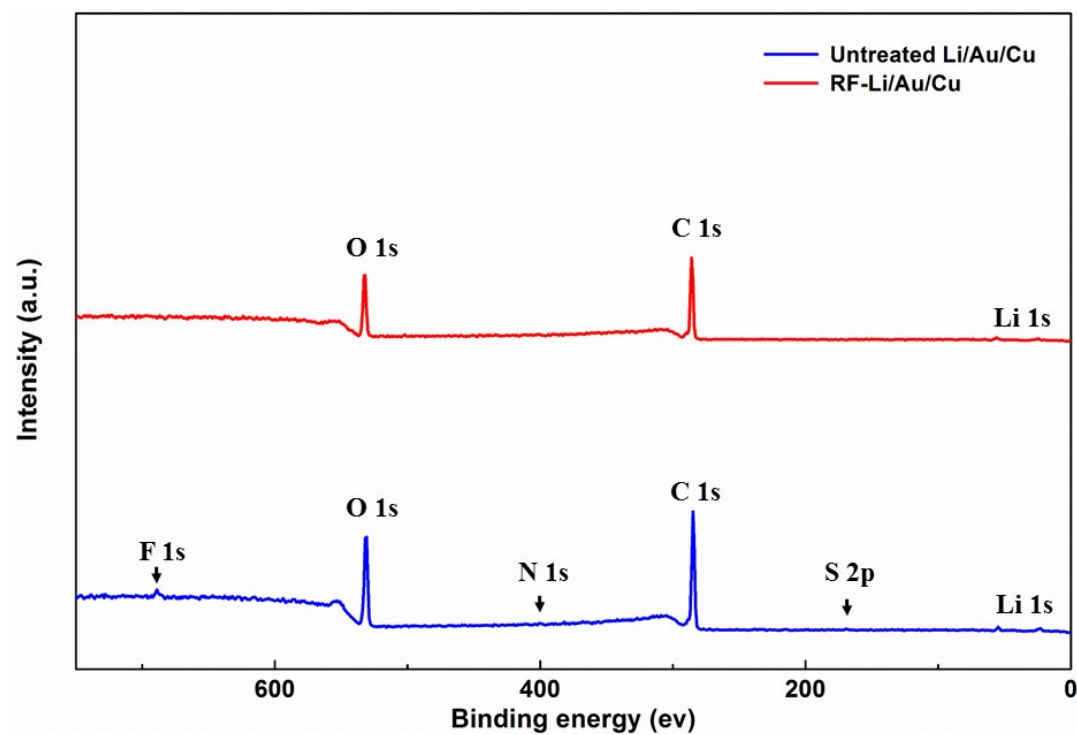


Fig. S10† XPS characterization of the untreated Li/Au/Cu and RF-Li/Au/Cu electrodes.

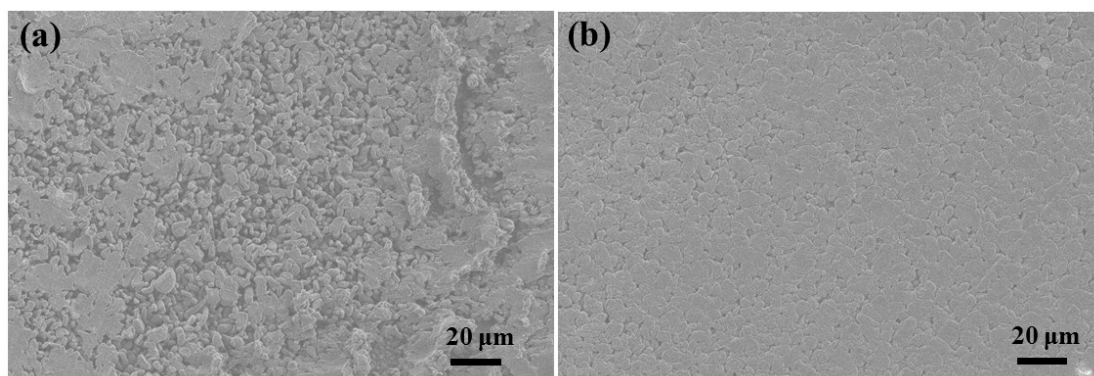


Fig. S11† The low-magnification SEM images of (a) untreated Li/Au/Cu and (b) RF-Li/Au/Cu electrodes after the 10th Li plating with a current density of 1 mA cm⁻².

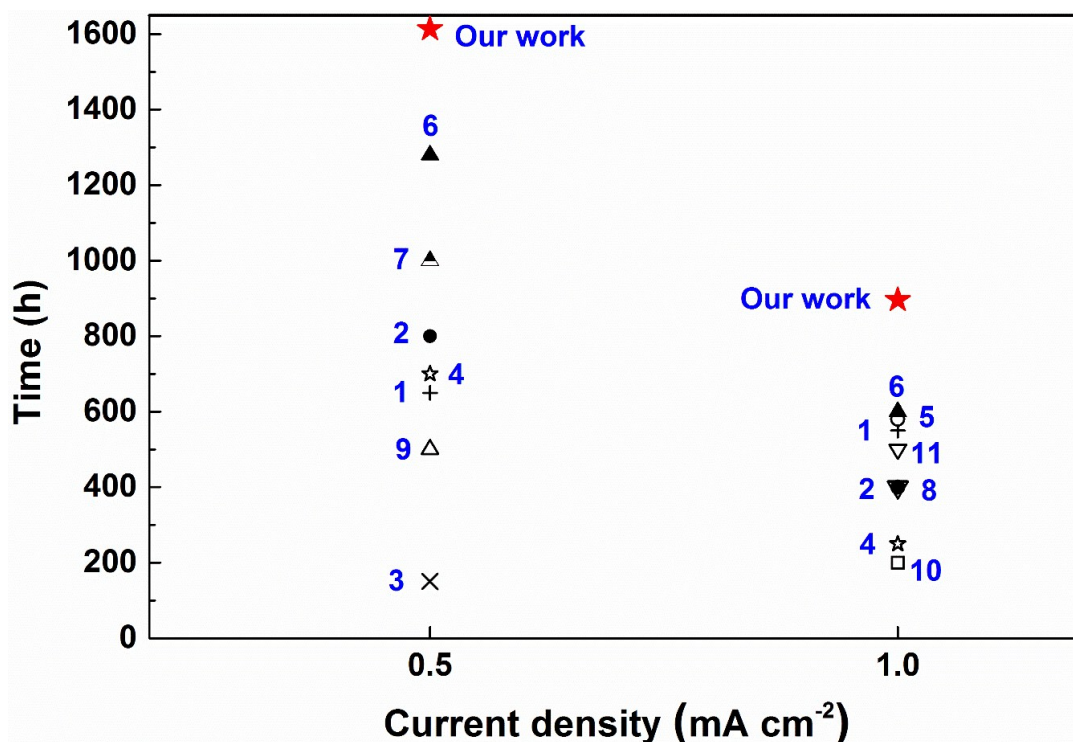


Fig. S12† Comparison of cycling performance of our data and the results reported by other groups under the current density of 0.5 mA cm⁻² or 1 mA cm⁻² with a capacity of 1 mAh cm⁻².

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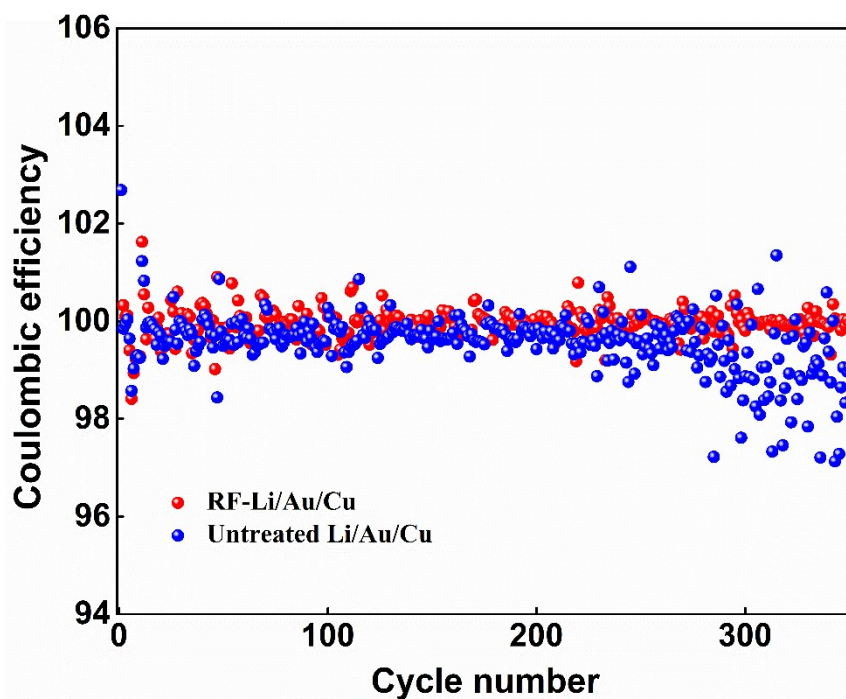


Fig. S13† Coulombic efficiency of LiFePO₄ full cells with RF-Li/Au/Cu and untreated Li/Au/Cu electrodes at 1 C rate.

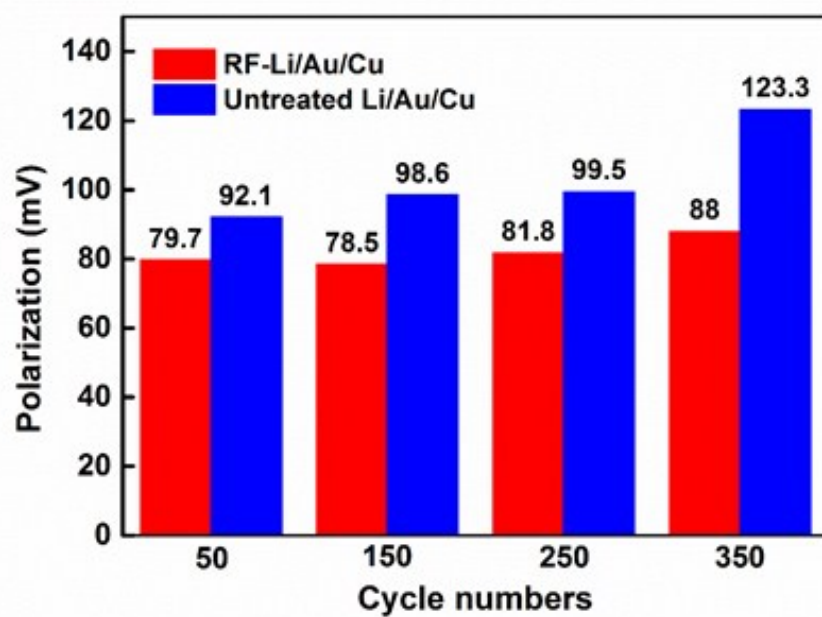


Fig. S14† Polarization comparison of LiFePO₄ full cells with untreated Li/Au/Cu and RF-Li/Au/Cu electrodes.