Supporting information

C₃N₄-digested 3D construction of hierarchical metallic phase MoS₂ nanostructures

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Figure S1. The selected range of XRD pattern of m-MS, m-CNMS-800 and bulk MoS₂.



Figure S2. SEM images of $(a, b) C_3N_4$ template and (c, d) m-MS.



Figure S3. The TEM images of m-CNMS at a low magnification.



Figure S4. High-resolution TEM images of m-CNMS and the corresponding lattice distance in selected regions.



Figure S5. The EDX spectrum of m-CNMS corresponding to the EDX mapping.



Figure S6. (a) Digital Photos of C_3N_4 after 160, 180, 190 and 200 °C hydrothermal reaction; (b) UV-Visible absorption spectra of C_3N_4 solution after different hydrothermal reaction.



Figure S7. SEM images of C_3N_4 templates with different morphologies and the corresponding prepared MoS₂ product. Note: bulk C_3N_4 was synthesized by calcining melamine at 550 °C in air; 1D C_3N_4 was synthesized by calcining nitric acid treated melamine at 550 °C in air; meso C_3N_4 was synthesized by calcining melamine-cyanuric acid hybrid at 550 °C in air.



Figure S8. XRD patterns of C_3N_4 templates with different morphologies and the corresponding prepared MoS_2 product.



Figure S9. CV curves for m-CNMS (a), m-MS (b), m-CNMS-800 (c) and bulk MoS₂ (d). The

capacitive current of all samples was collected at 0.05 V. Note: ECSA= Cdl / Cs. Cdl is calculated by CV at various scan rates in the non-Faradaic region, and Cdl is equal to the slope of the function of double layer charging current with scan rate. Cs is a specific electrochemical double layer capacitance of an atomically smooth surface, and the value of Cs = 40 μ F/cm² in this study.

| Element | Мо | S | С | Ν | S/Mo | C/N |
|---------|------------|------------|------------|-----------|------|------|
| EDX | 18.5 at. % | 42.6 at. % | 37.4 at. % | 1.5 at. % | 2.3 | 24.9 |
| XPS | 19.2 at. % | 36.9 at. % | 41.6 at. % | 2.3 at. % | 1.9 | 18.0 |

Table S1. Element content of m-CNMS from XPS and EDX.

| Catalysts | morphology | Current | Overpotential at | Tafel | ECSA | Ref |
|------------------------|--------------------------|---------|--------------------------|------------------------|------------------------------------|------|
| | | Density | corresponding j | slope | (cm ² _{ECSA}) | |
| | | | mA cm ⁻² (mV) | (mVdec ⁻¹) | | |
| 1T MoS ₂ | Flower-like | 10 | 215 | 50.2 | 165 | This |
| | sphere | 80 | 260 | | | work |
| 1T-2H MoS ₂ | nanosheet | 20 | 320 | 65 | 60.3 | 1. |
| $2 H MoS_2$ | Thin film with | 10 | 210 | 44 | / | 2. |
| | vertically | | | | | |
| | aligned MoS ₂ | | | | | |
| 1T' MoS ₂ | Nanofilm | 10 | 175 | 100 | / | 3. |
| 1T MoS ₂ | Nanosheet | 10 | 187 | 43 | 0.55 | 4. |
| 1T MoS ₂ | Mesoporous | 10 | 153 | 43 | 1577 | 5. |
| | Nanosheet | | | | | |
| 1T MoS ₂ | Nanosheet | 10 | 180 | 41 | / | 6. |
| $2 H MoS_2$ | Nanofilm | 250 | 400 | 50 | / | 7. |
| $2H MoS_2$ | Mesoporous | 10 | 210 | 74 | / | 8. |
| | foam | | | | | |
| Co_3S_4 $@MoS_2$ | Hollow ZIF-like | 10 | 210 | 88 | 202.5 | 9. |
| | structure | | | | | |
| 2H-1T MoS ₂ | vertically | 10 | 203 | 60 | / | 10. |
| | aligned flakelet | | | | | |
| | on nanosheet | | | | | |

Table S2. Comparison of HER performance of m-CNMS with other MoS₂-based electrocatalysts.

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