Supporting information for

Chromenoquinoline-based two-photon fluorescent probe for highly specific and ultrafast visualizing sulfur dioxide derivatives in living cells and zebrafish

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Table of Contents

Pages

Materials and instruments
Determination of the fluorescence quantum yield
References
Measurement of two-photon cross-sections4
HeLa cell culture and imaging using CQ-SO ₂ 4
Cytotoxicity assays4
Fluorescence imaging in living zebrafish5
Synthesis 5-6
Table S17
Figure S18
Table S29
Figure S2-S4
Scheme S110
Figure S511
Figure S6-16 12-16

Materials and instruments. Unless otherwise stated, all reagents were purchased from commercial suppliers and used without further purification. Solvents used were purified by standard methods prior to use. Twice-distilled water was used throughout all experiments. Mass spectra were performed using an LCQ Advantage ion trap mass spectrometer from Thermo Finnigan or Agilent 1100 HPLC/MSD spectrometer. NMR spectra were recorded on an INOVA-400 spectrometer, using TMS as an internal standard. Electronic absorption spectra were obtained on a Labtech UV Power PC spectrometer. Photoluminescent spectra were recorded at room temperature with a HITACHI F4600 fluorescence spectrophotometer with the excitation and emission slit widths at 5.0 and 5.0 nm respectively. The fluorescence imaging of cells was performed with OLYMPUS FV1000 (TY1318) confocal microscopy. The pH measurements were carried out on a Mettler-Toledo Delta 320 pH meter. TLC analysis was performed on silica gel plates and column chromatography was conducted over silica gel (mesh 200-300), both of which were obtained from the Qingdao Ocean Chemicals.

Determination of the fluorescence quantum yield¹⁻³: Fluorescence quantum yields for chromenoquinoline **1** and **CQ-SO**₂ were determined by using rhodamine 6G ($\Phi_f = 0.95$ in H₂O) as a fluorescence standard. The quantum yield was calculated using the following equation:

$$\Phi_{\mathrm{F}(\mathrm{X})} = \Phi_{\mathrm{F}(\mathrm{S})} \left(A_{\mathrm{S}} F_{\mathrm{X}} / A_{\mathrm{X}} F_{\mathrm{S}} \right) \left(n_{\mathrm{X}} / n_{\mathrm{S}} \right)^2$$

Where Φ_F is the fluorescence quantum yield, *A* is the absorbance at the excitation wavelength, *F* is the area under the corrected emission curve, and *n* is the refractive index of the solvents used. Subscripts _S and _X refer to the standard and to the unknown, respectively.

References

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Measurement of two-photon cross-sections. The two-photon cross-section (σ) was determined by using a femtosecond (fs) fluorescence measurement technique. Chromenoquinoline **1** was dissolved in EtOH and PBS, at a concentration of 5.0× 10⁻⁵ M, and then the two-photon fluorescence was excited at 700-900 nm by using fluorescein in pH = 11 aqueous solution (σ = 32 GM in 810 nm) as the standard, whose two-photon property has been well characterized in the literature. The two-photon cross-section was calculated by using $\sigma = \sigma_r (F_t n_t^2 \Phi_r C_r)/(F_r n_r^2 \Phi_t C_s)$, where the subscripts t and r stand for the sample and reference molecules. *F* is the average fluorescence intensity integrated from two-photon emission spectrum, n is the refractive index of the solvent, *C* is the concentration, Φ is the quantum yield, and σ_r is the two-photon cross-section of the reference molecule.

HeLa cell culture and imaging using CQ-SO₂. HeLa cells were seeded in Dulbecco's modified Eagle'smedium (DMEM) supplemented with 10% fetal bovine serum for 24 h. Hella cells were incubated with 5.0 μ M CQ-SO₂ for 30 min, and then treated with exogenous HSO₃⁻ at 20 mM in the culture medium for 30 min at 37 °C. After washing with PBS three times to remove the remaining HSO₃⁻. the cells were imaged using OLYMPUS FV1000 (TY1318) confocal microscopy with an excitation filter of 405 nm or 800 nm.

Cytotoxicity assays. The toxicity of **CQ-SO**₂ towards living HeLa cells was determined by MTT (3-(4,5-dimethyl-2-thiazolyl)-2,5-diphenyl-2-H-tetrazolium bromide) assays. HeLa cells were grown in the modified Eagle's medium (MEM) supplemented with 10% FBS (fetal bovine serum) in an atmosphere of 5% CO₂ and 95% air at 37 °C. Immediately before the experiments, the cells were placed in a 96-well plate, followed by addition of increasing concentrations of probe **CQ-SO**₂ (99% MEM and 1% DMSO). The final concentrations of **CQ-SO**₂ was 0, 5, 10, 20, 50, 100 μ M (n = 5), respectively. The cells were then incubated at 37 °C in an atmosphere of 5% CO₂ and 95% air at 37 °C for 24 h, followed by MTT assays. Untreated assay with MEM (n = 5) was also conducted under the same conditions.

Fluorescence imaging in living zebrafish.

Wild type zebrafish were purchased from the Nanjing EzeRinka Biotechnology Co. Ltd. Animal handling procedures were approved by the Animal Ethics Committee of Guangxi Normal University (No. 20150325-XC). The zebrafish were kept at 28 °C and optimal breeding conditions. The zebrafish then incubated with NaHSO₃ (30 μ M) in the culture medium for 30 min at 28 °C. After washing with PBS three times to remove the remaining NaHSO₃. Then, the zebrafish were further incubated with probe **CQ-SO₂** (10 μ M) for 30 min at 28°C. The imaging experiments were recorded with a 10x objective lens using OLYMPUS FV1000 (TY1318) confocal microscopy. The fluorescence emission was collected at green channel when excitation at 800 nm.

Synthesis of intermediate 1

In a 50 ml round bottomed flask were added 3-bromoprop-1-yne (2.60 g, 22.0 mmol), 4-(diethylamino)-2-hydroxybenzaldehyde (3.86 g, 20.0 mmol), K₂CO₃ (3.04 g, 22.0 mmol) and then was dissolved in DMF (15.0 ml), the reaction mixture was stirred at room temperature for 10 h. After completion of the reaction, the reaction mixture was poured on cold water (50 ml) and extracted with DCM (15 ml x 3). The combined organic layer was washed with water (50 ml x 3). The solvent was removed under reduced pressure to obtain a brown residue which was purified by column chromatography over silica gel to obtain the pure **intermediate 1** (3.97 g, 85.8%) as a sandy beige solid. ¹H NMR (400 MHz, CDCl₃) 10.13 (s, 1H), 7.73 (d, J = 8.9 Hz, 1H), 6.33 (dd, J = 8.8Hz, 1H), 6.23 (d, J = 2.3 Hz, 1H), 4.80 (d, J = 2.4 Hz, 2H), 3.43 (q, J = 7.1 Hz, 4H), 2.58 (t, J = 2.4 Hz, 1H), 1.23 (t, J = 7.1 Hz, 6H). ¹³C NMR (100 MHz, CDCl₃) 185.92, 160.99, 152.53, 129.64, 113.62, 104.08, 93.37, 77.26, 75.15, 55.09, 43.94, 11.60. MS (ESI) m/z =232.1 [M+H]⁺; HRMS (ESI) Calcd for C₁₄H₁₈NO₂⁺ ([M+H)⁺: 232.1338, Found 232.1326.

Synthesis of chromenoquinoline 1

In a 25 ml round bottomed flask placed under nitrogen atmosphere were added **intermediate 1** (231 mg, 1.0 mmol), 4-aminoacetophenone (135 mg, 1.0 mmol), and CuCl (29 mg, 0.3 mmol) was dissolved in dry DMF (5.0 ml), the reaction mixture was stirred at 110 °C for 5 h. After completion of the reaction, the reaction mixture was poured on cold water (20 ml) and extracted with DCM (10 ml x 3). The combined organic layer was washed with water (20 ml x 3). The solvent was removed under reduced pressure to obtain a black residue which was purified by column chromatography over silica gel to obtain the **chromenoquinoline 1** (129.8 mg, 37.5 %) as a straw yellow solid. ¹H NMR (400 MHz, CDCl₃) δ 8.37-8.26 (m, 2H), 8.17 (dd, *J* = 8.8, 1.9 Hz, 1H), 8.12-8.04 (m, 1H), 7.82 (s, 1H), 6.51 (dd, *J* = 9.0, 2.5 Hz, 1H), 6.22 (d, *J* = 2.5 Hz, 1H), 5.28 (s, 2H), 3.49-3.36 (q, *J* = 7.1 Hz, 4H), 2.70 (s, 3H), 1.22 (t, *J* = 7.1 Hz, 6H). ¹³C NMR (100 MHz, CDCl₃) 197.40, 159.65, 151.82, 133.28, 131.62, 129.36, 128.74, 127.90, 127.35, 125.40, 107.20, 98.01, 68.41, 44.67, 26.70, 12.71. MS (ESI) m/z =347.1 [M+H]⁺. HRMS (ESI) Calcd for C₂₂H₂₃N₂O₂⁺ ([M+H]⁺): 347.1760, Found, 347.1746.

Synthesis of probe CQ-SO₂

Phosphorous oxychloride (POCl₃) (5 mL) was slowely added to dimethylformamide (DMF) (5 mL) at 0-5 °C under the stirring about half an hour. Then the **chromenoquinoline 1** (35 mg)by dissolving it into DMF (3 mL) was added to this cooled system under the stirring for 3 h, and the reaction mixture was subsequently heated at 75 °C for 6 h. The reaction mixture was cooled to room temperature and then poured into ice cold water (30 mL). Then the reaction system was neutralised with sodium carbonate and subsequently was filtered and washed with cold water, dried and crystallised from ethanol to give brick-red solid probe **CQ-SO₂** (21 mg, 53 %). ¹H NMR (400 MHz, CDCl₃) 10.20 (d, *J* = 6.8 Hz, 1H), 8.22 (d, *J* = 8.9 Hz, 1H), 8.11 (d, *J* = 2.1 Hz, 1H), 8.00 (d, *J* = 8.7 Hz, 1H), 7.84 (dd, *J* = 9.0, 2.2 Hz, 1H), 7.73 (s, 1H), 6.74 (d, *J* = 6.8 Hz, 1H), 6.45 (dd, *J* = 9.0, 2.4 Hz, 1H), 6.16 (d, *J* = 2.4 Hz, 1H), 5.22 (s, 2H), 3.35 (t, *J* = 7.1 Hz, 4H), 1.16 (q, *J* = 7.1, 5.8 Hz, 6H). ¹³C NMR (100 MHz, CDCl₃) 190.38, 158.84, 150.58, 130.24, 126.71, 125.66, 125.10, 124.74, 123.14, 106.19, 96.99, 67.38, 43.65, 11.68. MS (ESI) m/z =393.1 [M+H]⁺. HRMS (ESI) Calcd for C₂₃H₂₂ClN₂O₂⁺ ([M+H]⁺): 393.1370, Found, 393.1353.

Probes	Detection limit	Living cell Imaging	Living Animals Imaging	Response time	Reference	
	0.11 μM	endogenous	no	5 min	Sensors and Actuators B 2018, 255, 1228- 1237	
Storoflag	50 nM	endogenous	no	3 min	Chem. Commun., 2016, 52, 10289- 10292; Anal. Chim. Acta, 2016, 937, 136- 142	
	0.3 µM	exogenous	yes	5 seconds	Anal.Chem., 2017, 89, 9388-9393	
N L C C C C C C C C C C C C C C C C C C	1.86 µM	exogenous	no	4.5 min	J. Mater. Chem. B, 2016, 4, 7888-7894	
, 0 0 1 1 1 0 0 1 1 0 0 1 0 0 0 0 0 0 0 0	3.75 μM	exogenous	yes	15 seconds	Sensors and Actuators B 2018, 268, 157-163	
	0.16 µM	endogenous	yes	35 seconds	Sensors and Actuators B 2018, 254,709-718	
C N OH	0.16 µM	exogenous	no	No data	Sensors and Actuators B 2016, 233, 1-6	
NO STA	2.21 nM	endogenous	no	10min	Dyes and Pigments 2016, 134, 297-305	
N CHO	16 nM	endogenous	yes	5 seconds	This work	

Table S1. Comparison of some two-photon fluorescent probes for selective detectionof SO_2 derivatives.



Figure S1. Two-photon action cross-sections of chromenoquinoline 1 in EtOH.

	DMSO	DMF	CH ₃ CN	EtOH	PBS
$\lambda_{\max(nm)}$	423	418	414	418	422
$\lambda_{em (nm)}$	536	526	525	541	556
Stokes Shifts (nm)	113	108	111	123	134
${\Phi}$	0.81	0.65	0.65	0.46	0.16
ε_{\max}	1.68×10^{4}	2.03×10 ⁴	2.23×10^{4}	2.00×10^{4}	1.12×10^{4}

 Table S2. Photophysical data of chromenoquinoline 1 in diverse solvents.



Figure S2. Absorption spectra of the probe **CQ-SO**₂ (5 μ M) in the absence (**•**) and presence (**•**) of SO₂ derivatives (25 μ M) in PBS buffer (pH 7.4, 25 mM, containing 20% DMSO as a cosolvent). Insets: images of aqueous probe **CQ-SO**₂ before (left) and after (right) treatment with NaHSO₃.



Figure S3. The pH effects of probe **CQ-SO**₂ in the absence (\bullet) or presence (\blacksquare) of SO₂ derivatives, excitation at 405 nm and emission at 520 nm.



Figure S4. HRMS (ESI) of the reaction mixture of CQ-SO₂ towards SO₂ derivatives.



Scheme S1. The two proposed response mechanism of probe $CQ-SO_2$ towards SO_2 derivatives.



Figure S5. ¹H-NMR comparison of the probe **CQ-SO**₂ with the product upon addition of HSO_3^- in DMSO-*d*6 and D₂O (V/V=7:3).



Figure S6. Cytotoxicity of the probe CQ-SO₂ on Hella cells determined by MTT.



Figure S7. One-photon fluorescence images of Hella cells. (a-b) The cells incubated with the probe CQ-SO₂ (5 μ M) for 30 min: (a) Bright field image and (b) fluorescent image generated from green channel; (c-d) The cells incubated with the probe CQ-SO₂ (5 μ M) for 30 min after pre-incubation with 20 μ M NaHSO₃ for 30 min: (c) Bright field image and (d) fluorescence image generated from green channel. Excitation at 405 nm. Scale bar =10 μ m.



Figure S8. The HRMS of intermediate 1.



Figure S9. The ¹H NMR of intermediate 1.



Figure S10. The ¹³C NMR of intermediate 1.



Figure S11. The HRMS of chromenoquinoline 1.



Figure S12. The ¹H NMR of chromenoquinoline 1.



Figure S13. The ¹³C NMR of chromenoquinoline 1.



Figure S14. The HRMS of probe CQ-SO₂.



Figure S15. The ¹H NMR of probe CQ-SO₂.



Figure S16. The ¹³C NMR of probe CQ-SO₂.