

Dibenzothiophene-S,S-dioxide – Fluorene Co-oligomers. Stable, Highly-Efficient Blue Emitters with Improved Electron Affinity

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Abstract: Incorporation of dibenzothiophene-S,S-dioxide units into conjugated fluorene oligomers changes the frontier orbital energy level and presents an effective way to increase the electron affinity of these materials, which are highly fluorescent with bright blue emission in both solution and solid state.

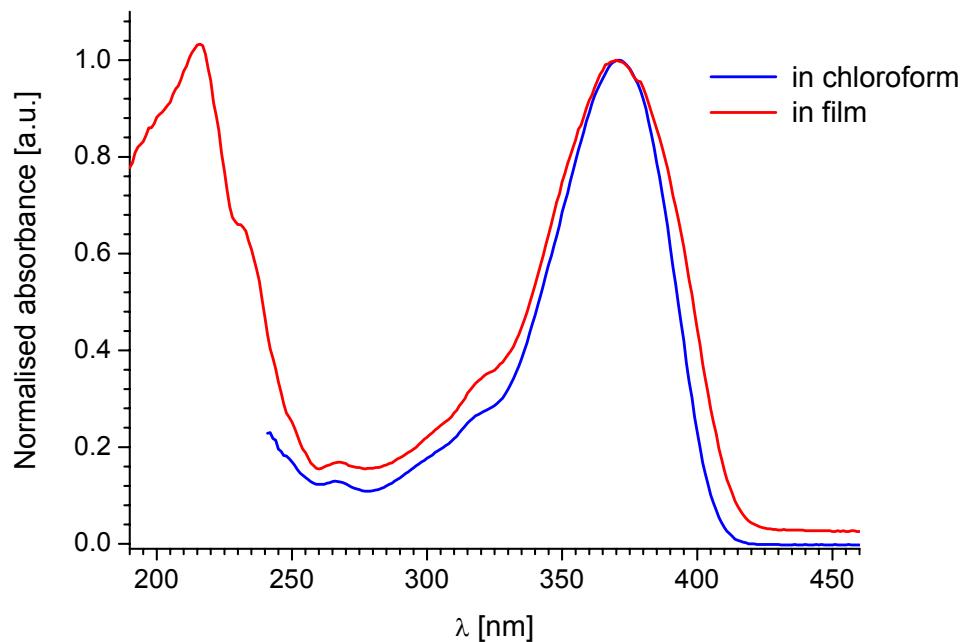


Figure S1. UV-Vis absorption spectra of pentafluorene FFFFF in chloroform ($\lambda_{\text{max}} = 371$ nm) and in film ($\lambda_{\text{max}} = 370$ nm).

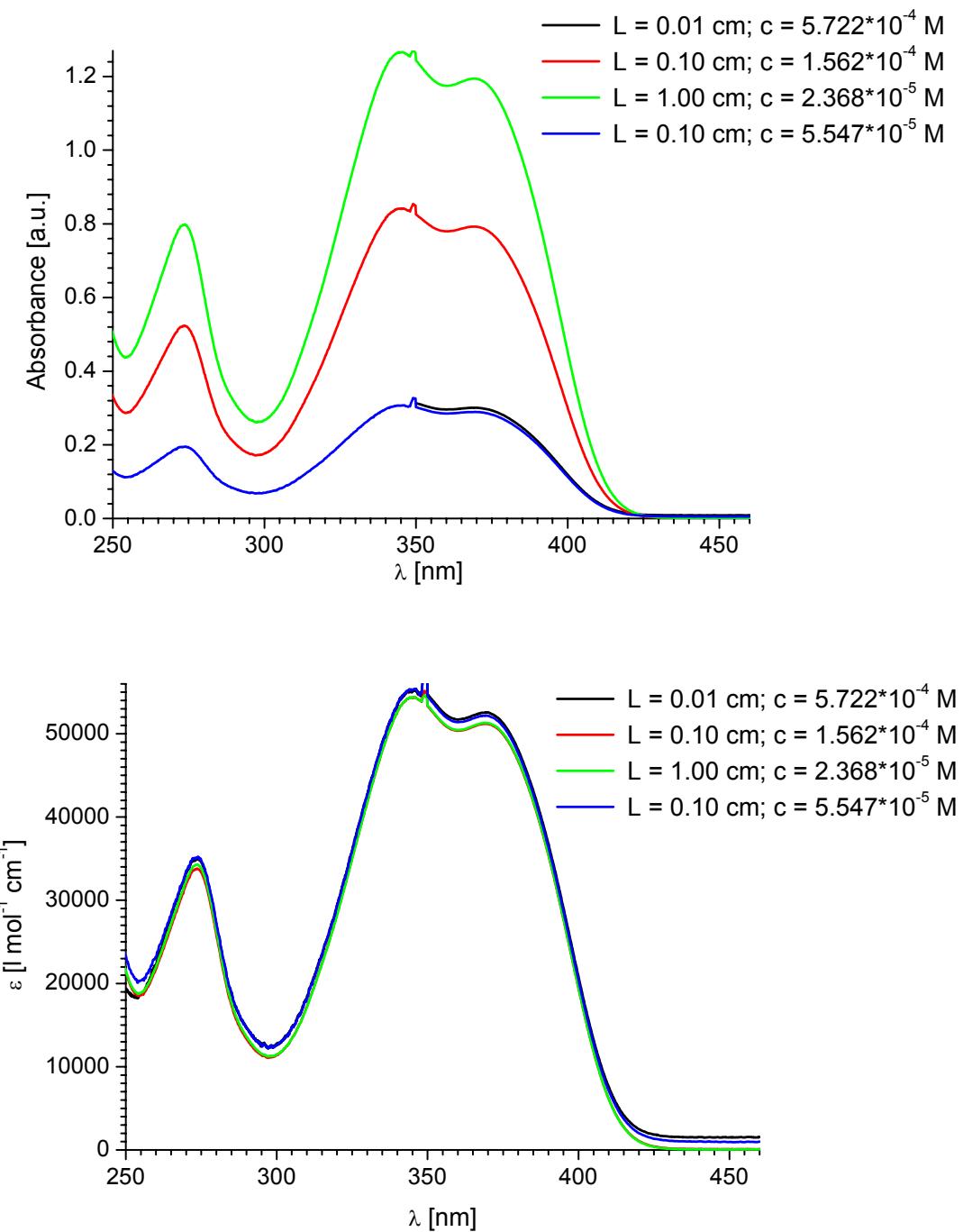


Figure S2. UV-Vis absorption spectra of FSF in chloroform.

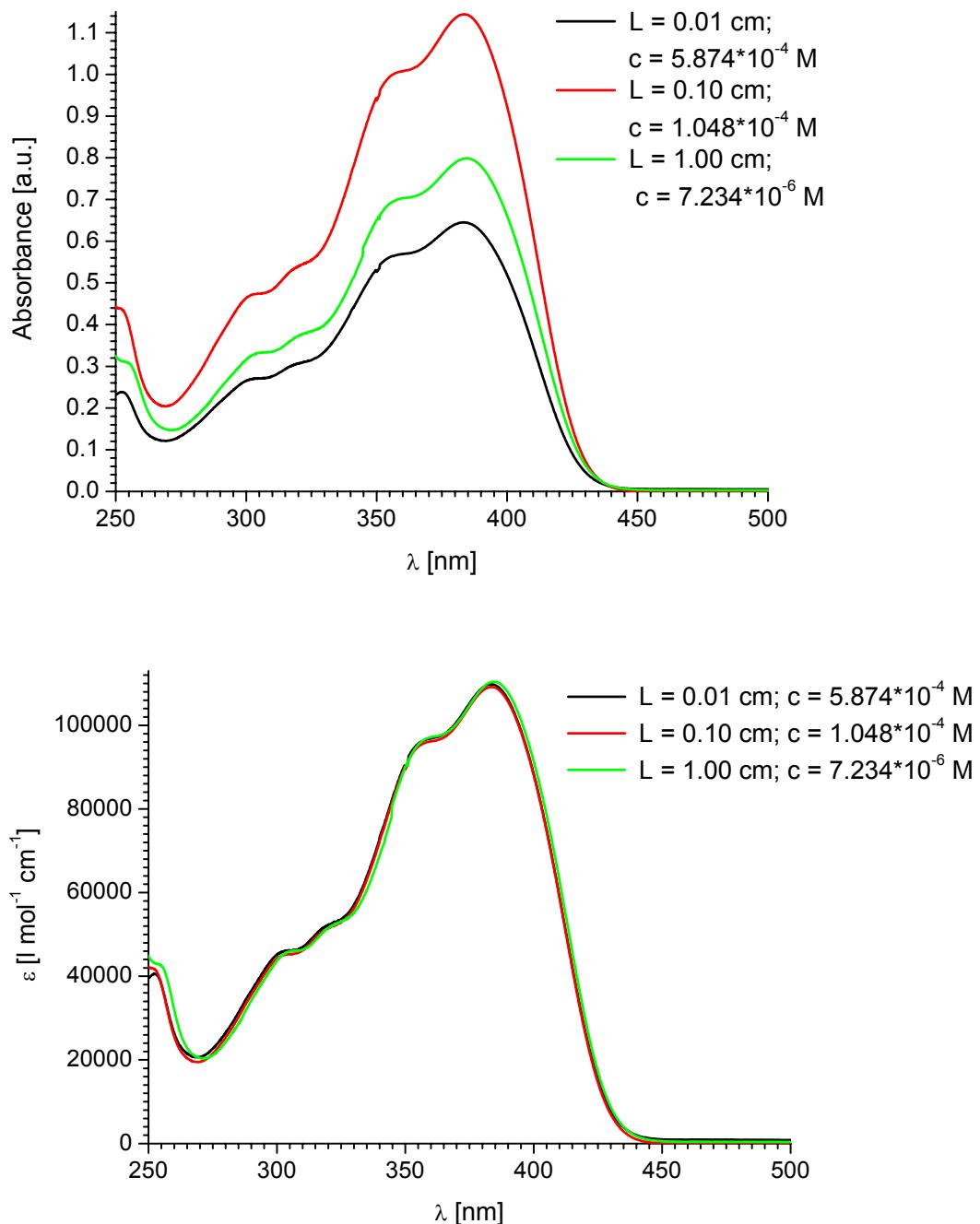


Figure S3. UV-Vis absorption spectra of FFSFF in chloroform.

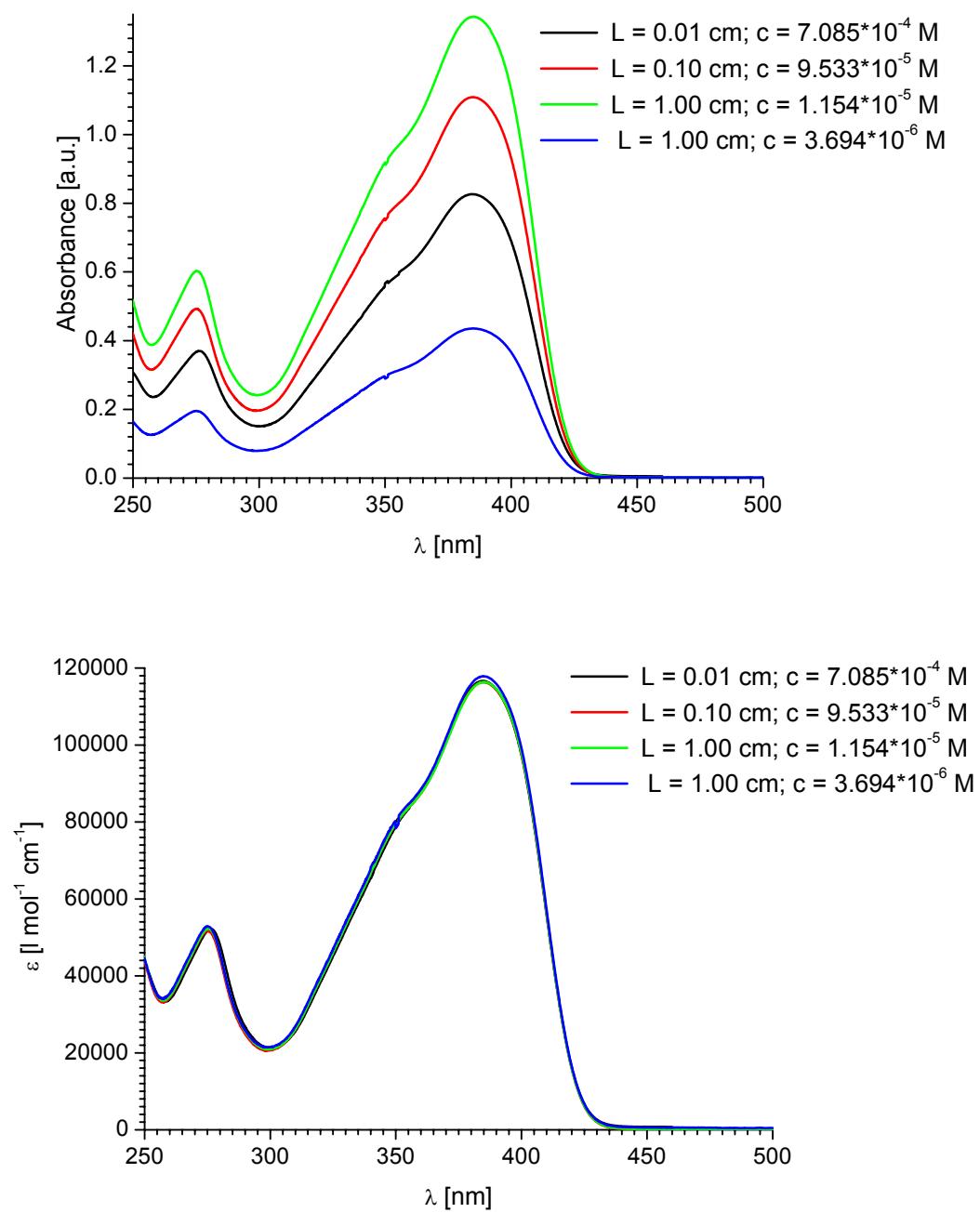


Figure S4. UV-Vis absorption spectra of FSFSF in chloroform.

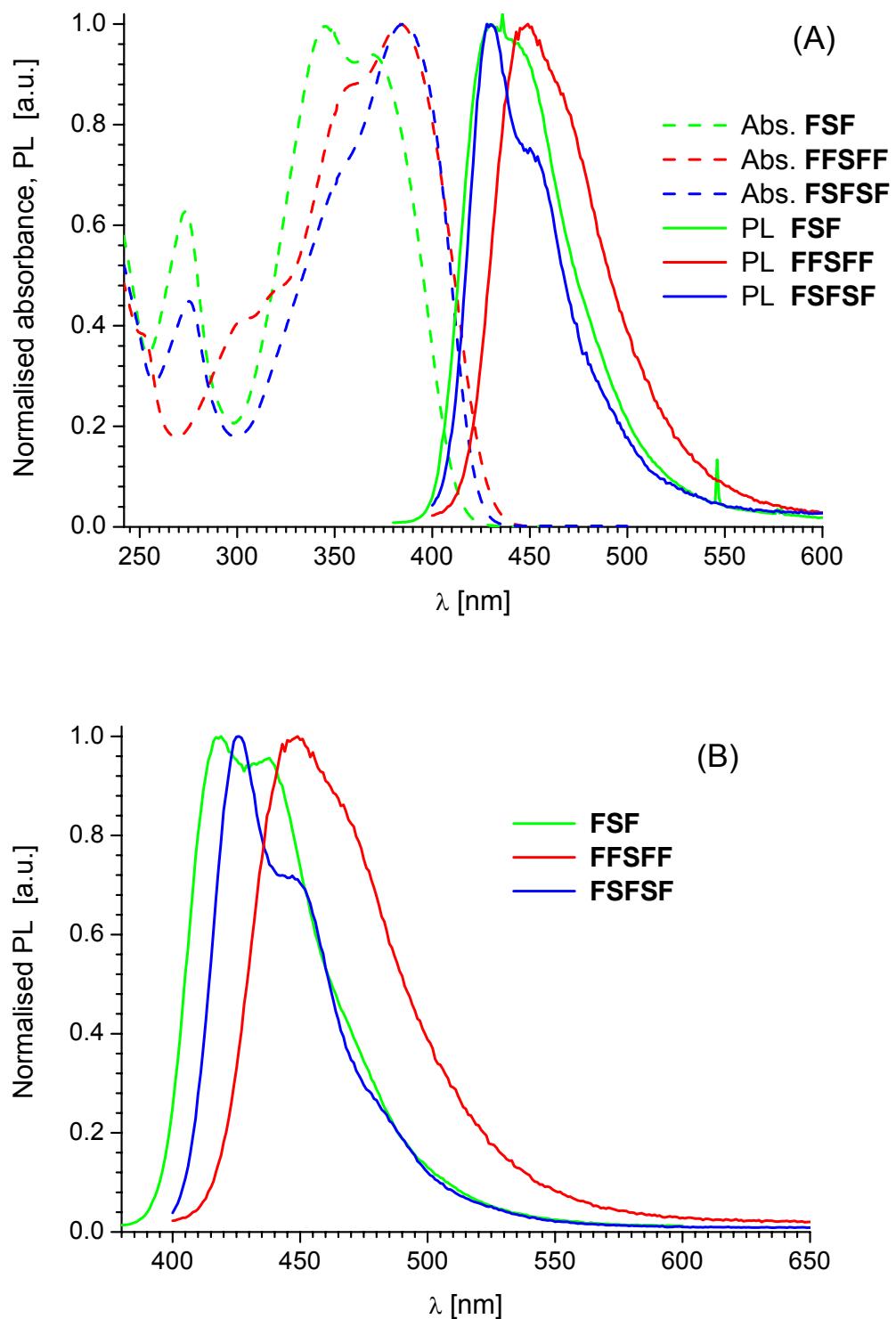


Figure S5. (A) UV-Vis absorption and photoluminescence spectra of **FSF**, **FFSFF** and **FSFSF** in chloroform.
(B) Photoluminescence spectra of **FSF**, **FFSFF** and **FSFSF** in tetrahydrofuran.

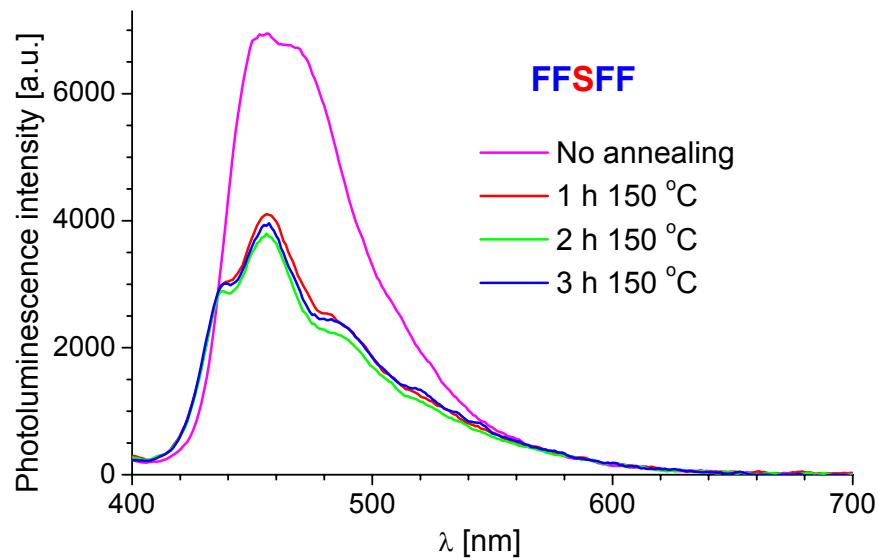


Figure S6. Evolution of the photoluminescence spectra of FFSFF films on annealing in an inert atmosphere.

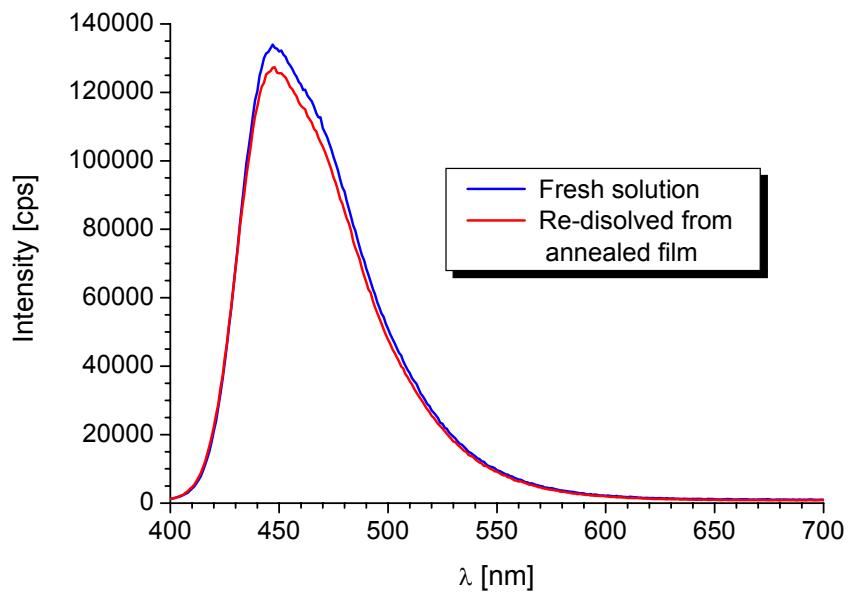


Figure S7. Photoluminescence spectra of FFSFF in chloroform solution: (—) fresh solution, $\lambda_{PL} = 449$ nm, $\Phi_{PL} = 67\%$; (—) solution of FFSFF prepared by dissolution of the film, annealed at 150 °C for 3 h (see Fig. S6) and then stored at ambient temperature in air for 6 months. The spectra are corrected for differences in absorption. Analysis shows that the PL intensity of the re-dissolved film is > 95% of the fresh solution, which is within the accuracy of the experiment.

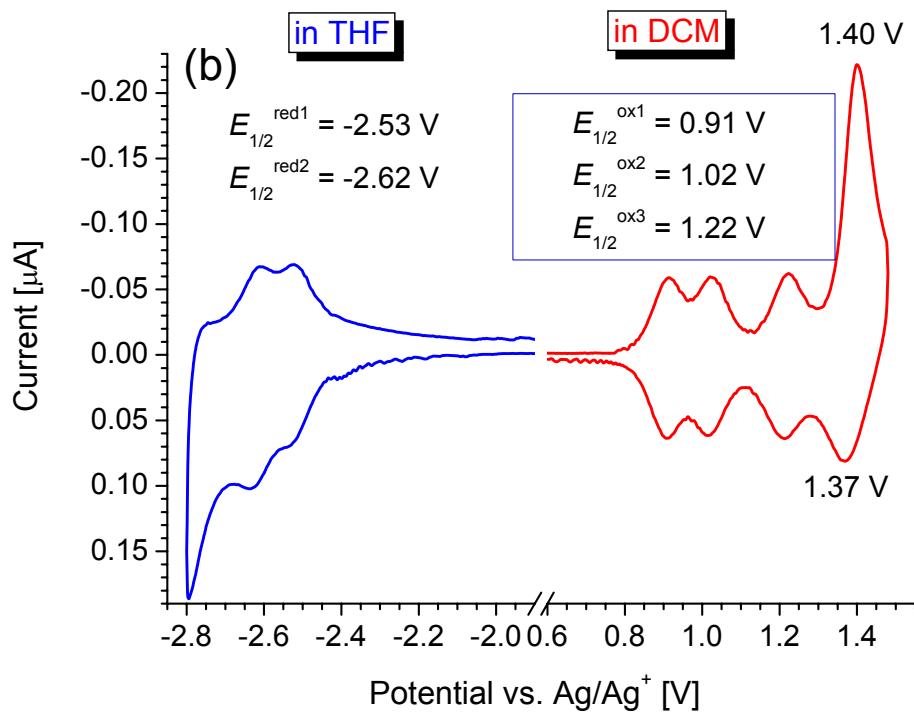
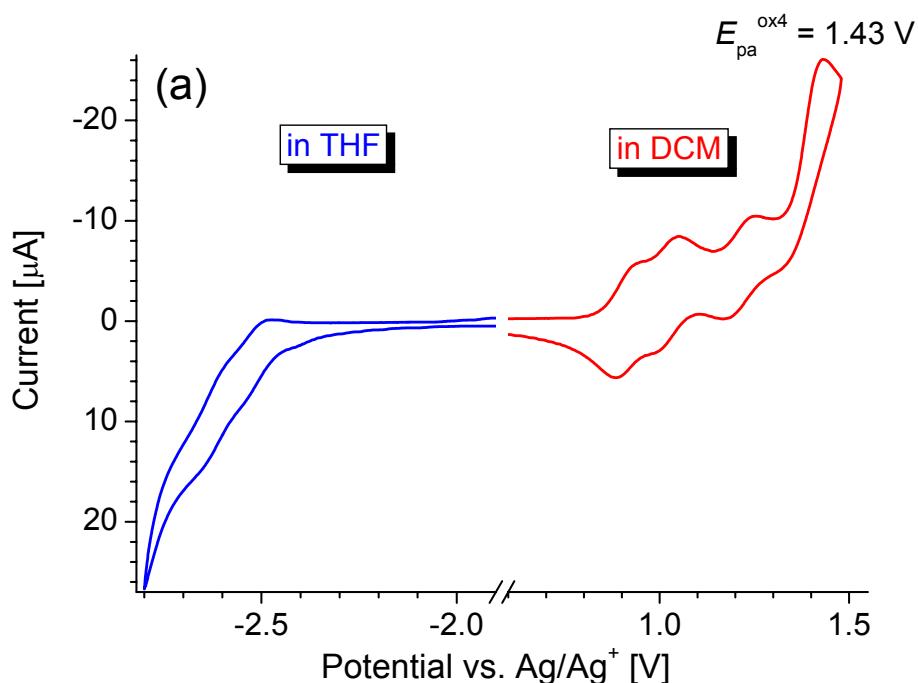


Figure S8. Cyclic voltammetry of FFFFF (a) and deconvoluted CV spectra (b); electrolyte 0.2 M Bu_4NPF_6 , scan rate 100 mV s⁻¹, 20 °C.

Experimental Part

General

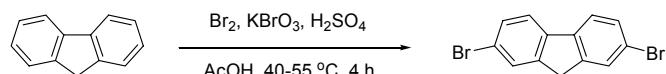
Elemental analyses were obtained on a Carlo-Erba Strumentazione instrument. Melting points were determined in open-end capillaries using a Stuart Scientific melting point apparatus SMP3 and were uncorrected. Solution ^1H NMR and ^{13}C NMR spectra were recorded on Varian Unity 300, Bruker Avance 400 and Varian Inova 500 spectrometers operating at (^1H) 299.91, 400.13, 499.99 and (^{13}C) 75.42, 100.62, 124.99 MHz, respectively. Chemical shifts are reported in ppm relative to TMS as internal standard. Mass spectra were obtained on a Micromass Autospec instrument operating in EI mode at 70 eV. MALDI-TOF spectra were obtained on an Applied Biosystems Voyager-DE STR operating in reflector mode.

UV-Vis absorption spectra were recorded on Perkin Elmer Lamda 19, Cary 5E and Genesis 10 spectrophotometers. Photoluminescence spectra were recorded on a Jobin Horiba Fluoromax 3, with an excitation at 390 nm, PL quantum yields (Φ_{PL}) in solution were measured using anthracene as a standard, $\Phi_{\text{PL}} = 27\%$ in ethanol solution.¹ Films of dibenzothiophene-S,S-dioxide/fluorene co-oligomers were spin-coated onto a quartz substrate from chloroform solution. Φ_{PL} of films were measured with an integrating sphere as described previously.²

Synthesis

2,7-Dibromofluorene, 2,7-dibromo-9,9-dihexylfluorene, 2-bromo-9,9-dihexylfluorene, 9,9-dihexylfluorene-2,7-diboronic acid (**2**) and 9,9-dihexylfluorene-2-boronic acid (**3**) have been described in the literature. We performed scale-up syntheses of these known compounds and we present our modified procedures for these widely used intermediates in oligo/polyfluorene chemistry.

2,7-Dibromofluorene.

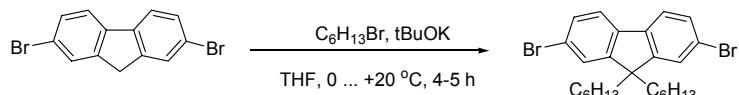


Fluorene (292 g, 1.76 mol) was dissolved in acetic acid (2600 cm³) at ~70 °C and H₂SO₄ (98%, 25 cm³) was added slowly to this solution. The reaction mixture was allowed to cool to ~50 °C with stirring, and a solution of bromine (150 cm³, 2.92 mol) in acetic acid (200 cm³) was added dropwise for 2–3 h, keeping the temperature at 40–55 °C to avoid crystallization of the fluorene. When ca. 1/3 – 1/2 of bromine was added, 2,7-dibromofluorene started to crystallise. Simultaneously with addition of a second half of bromine, KBrO₃ (100 g, 0.60 mol) was added in small portions (**CAUTION: add slowly, exothermic reaction!**) at 40–55 °C with vigorous stirring, which promotes the heavy precipitation of 2,7-dibromofluorene. The mixture was stirred for 3–4 h, then allowed to cool gradually to room temperature. After cooling the mixture to 10 °C, the solid was filtered off, washed with 70% AcOH (500 cm³) and water until pH 7, and dried affording the crude product as a cream-coloured solid (481 g, 85 %) of > 95% purity (by ^1H NMR). To further purify the product it was stirred in AcOH (~1000 cm³) at reflux (no full dissolution) for 4 h, cooled, filtered off, washed with AcOH and dried. Yield 455 g, 80%.

^1H NMR (400 MHz, CDCl₃): δ 7.66 (2H, d, $J_{1,3} = 1.8$ Hz, H-1,8), 7.59 (2H, d, $J_{3,4} = 8.0$ Hz, H-4,5), 7.50 (2H, dd, $J_{3,4} = 8.0$ Hz, $J_{1,3} = 1.8$ Hz, H-3,6), 3.89 (2H, s, CH₂).

^{13}C NMR (100 MHz, CDCl₃): δ 144.79, 139.69, 130.15, 128.31, 121.19, 120.94, 36.56.

2,7-Dibromo-9,9-dihexylfluorene (adapted from ref. ³).

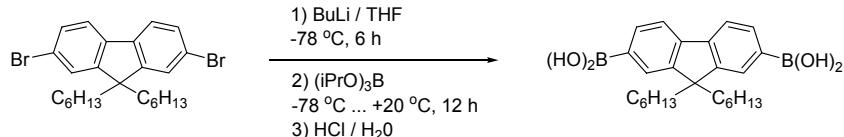


Under argon, a 3 L three-neck flask was charged with 2,7-dibromofluorene (130.0 g, 0.40 mol), 1-bromohexane (220 cm³, 1.57 mol) and dry THF (1000 cm³). After full dissolution the mixture was cooled to 0 °C and a solution of potassium *tert*-butoxide (100.6 g, 0.90 mol) in dry THF (1000 cm³) was added dropwise at 0 – +5 °C with vigorous stirring during 1.5 h. Upon adding the *tert*-butoxide solution the reaction mixture became orange (generation of fluorene anion) and then the colour changed to light pink (at the end of *tert*-butoxide addition no orange colour is produced, indicating that the alkylation reaction has been completed). The mixture was stirred at room temperature for 4 h, filtered from the KBr precipitate, and the solid was washed on the filter with DCM. The filtrate was evaporated on a rotavapor, the residue was dissolved in DCM (1500 cm³), washed with water, dried over MgSO₄, and the solvent was evaporated. Excess 1-bromohexane was removed *in vacuo* (80 °C, 1 mbar) yielding crude product (196.7 g, 99.6 %) as yellow

crystals. This was purified by column chromatography (7×17 cm column, silica gel, eluent – petrol ether, bp 40–60 °C) to afford 2,7-dibromo-9,9-dihexylfluorene (179.5 g, 91 %) as colourless plates. The material can also be additionally recrystallised from hexane or ethanol.

^1H NMR (500 MHz, CDCl_3): δ 7.51 (2H, d, $J = 7.8$ Hz, H-1,8), 7.45 (2H, dd, $J = 1.8$ Hz and 7.8 Hz, H-3,6), 7.44 (2H, d, $J = 1.8$ Hz, H-4,5), 1.96–1.87 (4H, m, $\text{CH}_2\text{C}_5\text{H}_{11}$), 1.16–1.08 (4H, m, $\text{CH}_2\text{CH}_2\text{CH}_2\text{C}_3\text{H}_7$), 1.08–0.98 [8H, m, $(\text{CH}_2)_3\text{CH}_2\text{CH}_2\text{CH}_3$], 0.78 (6H, t, $J = 7.4$ Hz, CH_3), 0.62–0.53 (4H, m, $\text{CH}_2\text{CH}_2\text{C}_4\text{H}_9$).

9,9-dihexylfluorene-2,7-diboronic acid (3).

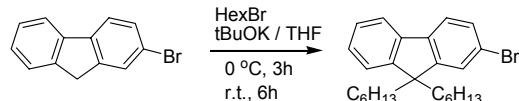


To a stirred solution of 2,7-dibromo-9,9-dihexylfluorene (30.0 g, 60.9 mmol) in dry THF (1000 cm^3) under argon, a solution of BuLi in hexane (2.5 M; 54 cm^3 , 135 mmol) was added dropwise at -78°C . The mixture was stirred at this temperature for 6 h to give a white suspension. Triisopropylborate (60 cm^3 , 258 mmol) was added quickly and the mixture was stirred overnight allowing the temperature to rise gradually to room temperature. Water (300 cm^3) was added and the mixture was stirred at r.t. for 4 h. Organic solvents were removed on a rotavapor (35°C , 40 mbar), water (1100 cm^3) was added and the mixture was acidified with concentrated HCl. The product was extracted into diethyl ether ($7 \times 300 \text{ cm}^3$), the organic layer was dried over MgSO_4 and solvent was removed on a rotavapor. The residue was dissolved in acetone (110 cm^3) and reprecipitated into a mixture of water (130 cm^3) and concentrated HCl (70 cm^3) affording product **3** (24.3 g, 95 %) as a white powder. The product can be additionally purified by dissolution in acetone (100 cm^3) and addition of hexane (200 cm^3) to this solution.

^1H NMR (400 MHz, acetone- d_6): δ 7.99 (2H, dd, H -1,8), 7.90 (2H, dd, $J_{3,4} = 7.6$ Hz, $J_{1,3} = 1.3$ Hz, H-3,6), 7.80 (2H, dd, $J_{3,4} = 7.6$ Hz, $J_{1,4} = 0.6$ Hz, H-4,5), 7.19 [4H, s, $\text{B}(\text{OH})_2$], 2.12–2.00 (4H, m, $\text{CH}_2\text{C}_5\text{H}_{11}$), 1.2–0.9 [12H, m, $(\text{CH}_2)_2(\text{CH}_2)_3\text{CH}_3$], 0.74 (6H, t, $J = 7.2$ Hz, CH_3), 0.64–0.54 (4H, m, $\text{CH}_2\text{CH}_2\text{C}_4\text{H}_9$).

^{13}C NMR (100 MHz, acetone- d_6): δ 150.87, 144.09, 133.87, 129.39, 119.92, 55.50, 41.09, 32.27, 30.39, 24.57, 23.16, 14.21

2-Bromo-9,9-dihexylfluorene

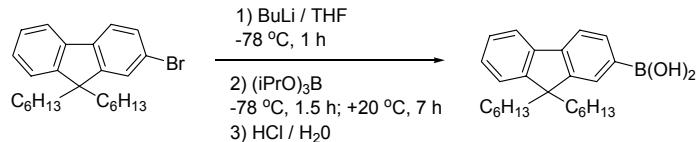


1-Bromohexane (230 cm^3) was added to a solution of 2-bromofluorene (106 g, 0.43 mol) in dry THF (1000 cm^3) under argon and cooled to 0°C . Potassium *tert*-butoxide (111 g, 0.99 mol) was dissolved in dry THF (1000 cm^3) and added dropwise at -5°C for 2 h to the above solution. The reaction was left stirring under argon at 0°C for 3 h and then it was allowed to warm up to room temperature and stirred for another 6 h. The obtained precipitate of KBr was filtered off and washed with DCM. The combined filtrates were evaporated and chromatographed on silica, eluted with petrol ether. 1-Bromohexane was removed under high vacuum (70°C , 0.9 mbar), followed by another purification by column chromatography on silica to afford 171 g (95%) of the title product as a light yellowish oil.

^1H NMR (400 MHz, CDCl_3): δ 7.67–7.63 (1H, m), 7.54 (1H, dd, $J = 0.8$ and 7.6 Hz, H-3), 7.46–7.42 (2H, m), 7.36–7.29 (3H, m), 2.00–1.86 (4H, m, $-\text{CH}_2\text{C}_5\text{H}_{11}$), 1.16–1.07 (4H, m, $\text{CH}_2\text{CH}_2\text{CH}_2\text{C}_3\text{H}_7$), 1.07–0.97 [8H, m, $(\text{CH}_2)_3\text{CH}_2\text{CH}_2\text{CH}_3$], 0.76 (6H, t, $J = 7.6$ Hz, CH_3), 0.64–0.55 (4H, m, $\text{CH}_2\text{CH}_2\text{C}_4\text{H}_9$).

^{13}C NMR (75 MHz, CDCl_3): δ 152.92, 150.26, 140.09, 139.98, 129.82, 127.41, 126.87, 126.61, 126.08, 122.83, 120.98, 119.70, 55.34, 40.27, 31.46, 29.63, 23.64, 22.56, 13.99.

9,9-Dihexylfluorene-2-boronic acid (2).



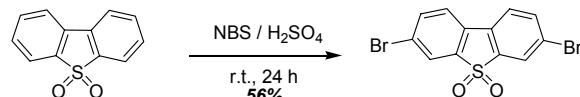
A 1.6 M solution of *n*-butyllithium in hexane (130 cm^3 , 0.208 mol) was added dropwise over 30 min to a solution of 2-bromo-9,9-dihexylfluorene (65.0 g, 0.157 mol) in dry THF (2300 cm^3) at -78°C under argon and the mixture was stirred for 1 h. Then triisopropylborate (113 cm^3 , 0.487 mol) was added dropwise for 60 min at -78°C and the reaction

mixture was vigorously stirred under argon at this temperature until the light pink colour of the solution disappeared (*ca.* 1.5 h), then the temperature was allowed to rise gradually to room temperature and the reaction mixture was left stirring under argon for 7 h at room temperature to give a colourless suspension. THF was removed up to *ca.* 500 cm³ and the solid was washed with water (800 cm³). 13% HCl (800 cm³) was added and the mixture was vigorously stirred for 3 h. The product was extracted into diethyl ether, washed with water, dried over MgSO₄ and purified by column chromatography on silica, eluting first with petrol ether and then with mixture of petrol ether and toluene (3:1, 1:1 v/v). Yield: 37.09 g (62 %) of a colourless solid.

¹H NMR (200 MHz, CDCl₃): δ 8.32 (1H, d, J = 7.6 Hz, H-4), 8.22 (1H, s, H-1), 7.90 (1H, d, J = 7.6 Hz, H-5), 7.86–7.77 (1H, m), 7.46–7.32 (3H, m), 2.22–1.96 (4H, m, –CH₂C₅H₁₁), 1.20–0.96 [12H, m, CH₂CH₂(CH₂)₃CH₃], 0.75 (6H, t, J = 6.4 Hz, CH₃), 0.78–0.58 (4H, m, (4H, m, CH₂CH₂C₄H₉).

¹³C NMR (75 MHz, CDCl₃) δ 151.64, 150.12, 145.57, 140.72, 134.61, 129.67, 128.94 (br.), 127.92, 126.83, 123.03, 120.40, 119.26, 55.05, 40.42, 31.53, 29.75, 23.80, 22.58, 14.01.

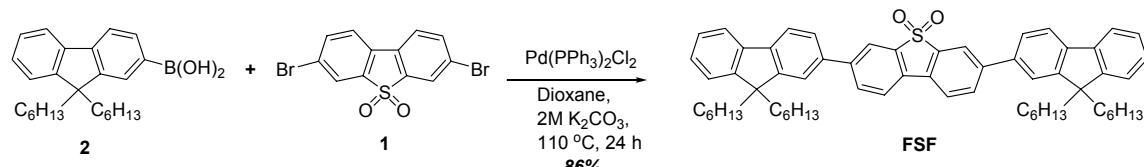
3,7-Dibromodibenzothiophene-S,S-dioxide (1)



Dibenzothiophene-S,S-dioxide (100.0 g, 0.46 mol) was dissolved in concentrated H₂SO₄ (3000 cm³). *N*-bromosuccinimide (NBS) (82.3 g, 0.46 mol) was added to this solution in several portions and the mixture was stirred at room temperature for 1 h. Additional NBS (82.3 g, 0.46 mol) was added to the mixture, which was then vigorously stirred at room temperature for 24 h. The precipitation started in 2–3 h and a lot of white solid was formed at the end of the process. The solid was filtered off, washed with H₂SO₄ (200 cm³), then with H₂O until neutral and recrystallised from chlorobenzene to obtain 3,6-dibromodibenzothiophene-S,S-dioxide **1** as colourless needles (96 g, 56%), m.p. 317–318 °C. Lit.⁴ m.p. 288–290 °C.

¹H NMR (300 MHz, CDCl₃): δ 7.93 (2H, d, J = 1.8 Hz, H-4,6), 7.77 (2H, dd, J = 1.8 and 8.4 Hz, H-2,8), 7.64 (2H, d, J = 8.1 Hz, H-1,9).

3,7-Bis(9,9-di-*n*-hexylfluorene-2-yl)dibenzothiophene-S,S-dioxide (FSF).



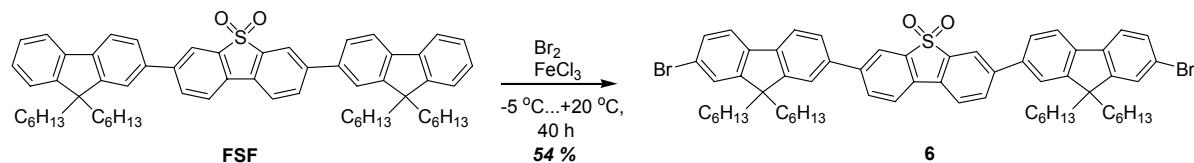
Under argon, to a mixture of 3,6-dibromodibenzothiophene-S,S-dioxide **1** (0.37 g, 1.00 mmol), 9,9-di-*n*-hexylfluorene-2-boronic acid **2** (0.77 g, 2.04 mmol) and dichlorobis(triphenylphosphine)palladium(II) (16 mg, 0.02 mmol), degassed 2 M potassium carbonate aqueous solution (4 cm³) and 1,4-dioxane (10 cm³) were added via a syringe. The reaction was stirred under argon with heating at 110 °C (oil bath) for 24 h with protection from the sunlight. The resulting slurry was poured into 5% NaCl-aqueous solution, the product was extracted with dichloromethane (50 cm³), the organic layer was washed with water until pH 7 and dried over anhydrous magnesium sulphate. After evaporation of the solvent, the residue was purified by column chromatography on silica gel eluting first with petroleum ether to remove by-products and then with petrol ether (PE) – dichloromethane mixture (PE:DCM, 1:1 v/v), to obtain the title product **FSF** (0.76 g, 86%) as a light yellow powder, m.p. 153 °C.

¹H NMR (500 MHz, CDCl₃): δ 8.16 (2H, d, J = 1.6 Hz, H-4,6 dibenzothiophene), 7.96 (2H, dd, J = 1.6 and 7.8 Hz, H-2,8 dibenzothiophene), 7.91 (2H, d, J = 7.8 Hz, H-1,9 dibenzothiophene), 7.81 (2H, d, J = 8.0 Hz, H-4 fluorene), 7.75 (2H, dd, J = 1.5 and 7.0 Hz, H-5 fluorene), 7.63 (2H, dd, J = 1.8 and 8.0 Hz, H-3 fluorene), 7.61 (2H, s, H-1 fluorene), 7.39–7.33 (6H, m, H-6,7,8 fluorene), 2.06–2.00 (8H, m, CH₂C₅H₁₁), 1.16–1.00 [24H, m, CH₂CH₂(CH₂)₃CH₃], 0.77 (12H, t, J = 7.3 Hz, CH₃), 0.70–0.60 (8H, m, CH₂CH₂C₄H₉).

¹³C NMR (100 MHz, CDCl₃): δ 151.89, 151.09, 144.16, 141.82, 140.28, 138.61, 137.44, 132.57, 129.95, 127.56, 126.92, 125.87, 122.96, 121.91, 121.24, 120.64, 120.30, 120.02, 55.34, 40.47, 31.51, 29.70, 23.79, 22.59, 14.01
m/z (ES+): 880 (M⁺, 100%). [Exact Mass (calcd): 880.5253]

Anal. Calcd for C₆₂H₇₂O₂S (M.W. 881.30): C, 84.50; H, 8.23; S, 3.64. Found: C, 84.45; H, 8.21; S, 3.58

3,7-Bis(2-bromo-9,9-di-*n*-hexylfluorene-7-yl)dibenzothiophene-*S,S*-dioxide (6).



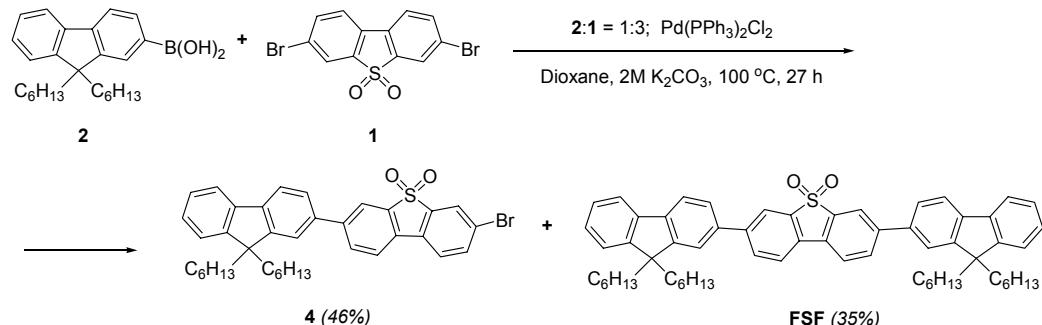
To a solution of **FSF** (0.50 g, 0.57 mmol) in chloroform (4.5 cm³) containing FeCl₃ (14 mg, 0.09 mmol; 0.15 equivalents) a solution of bromine (0.19 g, 1.21 mmol; 2.14 equivalents) in chloroform (1.3 cm³) was added at -5 °C (*it is important that the reaction proceeds in the dark to avoid bromination of the aliphatic part of the molecule*). The solution was allowed to warm to room temperature and stirred for 40 h. The resulting solution was washed with water and several times with a 0.3 M sodium carbonate aqueous solution. The aqueous layer was additionally extracted with chloroform and the combined organic solution was dried over anhydrous magnesium sulfate. Removing the solvent afforded crude product (0.49 g, 83%), which was recrystallised from acetone to yield compound **6** (0.32 g, 54%) as a light yellow powder, m.p. 241–242.5 °C.

¹H NMR (300 MHz, CDCl₃): δ 8.14 (2H, s, br, H-4,6 dibenzothiophene), 7.96 (2H, dd, *J* = 1.5 and 8.1 Hz, H-2,8 dibenzothiophene), 7.91 (2H, d, *J* = 8.1 Hz, H-1,9 dibenzothiophene), 7.78 (2H, d, *J* = 8.1 Hz, H-4 fluorene), 7.65–7.58 (6H, m), 7.51–7.48 (4H, m), 2.07–1.96 (8H, m, CH₂C₅H₁₁), 1.20–1.00 [24H, m, CH₂CH₂(CH₂)₃CH₃], 0.78 (12H, t, *J* = 6.9 Hz, CH₃), 0.69–0.57 (8H, m, CH₂CH₂C₄H₉).

¹³C NMR (100 MHz, CDCl₃) δ 153.31, 151.54, 143.99, 140.72, 139.32, 138.65, 137.93, 132.62, 130.17, 130.05, 126.27, 126.10, 121.99, 121.64, 121.38, 121.27, 120.69, 120.45, 55.70, 40.37, 31.50, 29.63, 23.76, 22.60, 14.01. m/z (EI): 1036 (M⁺, 45%, ⁷⁹Br, ⁷⁹Br), 1038 (M⁺, 100%, , ⁷⁹Br, ⁸¹Br), 1040 (M⁺, 67%, ⁸¹Br, ⁸¹Br). [Exact Mass (calcd): 1036.3463].

Anal. Calcd for C₆₂H₇₀Br₂O₂S (M.W. 1039.09): C, 71.66; H, 6.79; Br, 15.38; S, 3.09. Found: C, 71.49; H, 6.87; Br, 15.13; S, 3.19

3-Bromo-7-(9,9-di-*n*-hexylfluorene-2-yl)dibenzothiophene-*S,S*-dioxide (4)



The flask with 3,6-dibromodibenzothiophene-*S,S*-dioxide **1** (11.92 g, 31.86 mmol) and 9,9-di-*n*-hexyl-2-fluoreneboronic acid **2** (4.02 g, 10.62 mmol) was flushed with argon. Anhydrous 1,4-dioxane (350 cm³), degassed 2.3 M potassium carbonate aqueous solution (30 cm³) and dichlorobis(triphenylphosphine)palladium(II) (0.09 g, 0.13 mmol) were added. The reaction mixture was stirred under argon with heating at 100 °C (oil bath) for 27 h with protection from the sunlight. The resulting solution was concentrated and the product was extracted with dichloromethane (250 cm³), washed with water until pH 7 and dried over anhydrous magnesium sulphate. After evaporation of the solvent, the residue was purified by column chromatography on silica gel eluting first with PE to remove by-products. Further elution with a mixture of PE:DCM (1:1 v/v) yielded first compound **FSF** (1.64 g, 35%) and then compound **4** (3.06 g, 46%) as a yellowish powder, m.p. 177.5–178.5 °C.

Compound 4:

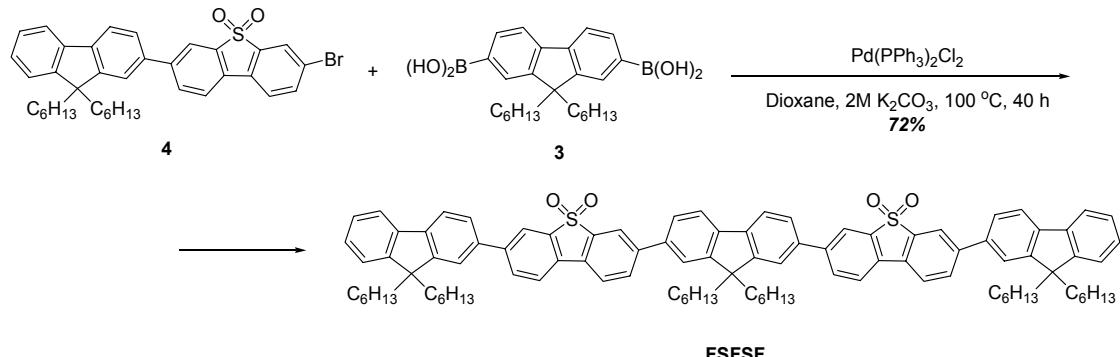
¹H NMR (400 MHz, CDCl₃): δ 8.10 (1H, d, *J* = 1.6 Hz, H-6 benzo thiophene), 7.97 (1H, d, *J* = 1.6 Hz, H-4 benzo thiophene), 7.93 (1H, dd, *J* = 1.6 and 8.0 Hz, H-8 benzo thiophene), 7.84 (1H, d, *J* = 8.0 Hz, H-9 benzo thiophene), 7.79 (1H, d, *J* = 8.0 Hz), 7.78 (1H, dd, *J* = 1.6 and 8.4 Hz, H-2 benzo thiophene), 7.76–7.72 (1H, m), 7.70 (1H, d, *J* = 8.4 Hz, H-1 benzo thiophene), 7.60 (1H, dd, *J* = 1.8 and 7.8 Hz), 7.58 (1H, m), 7.39–7.34 (3H, m), 2.06–1.99 (4H, m, CH₂C₅H₁₁), 1.16–0.99 [12H, m, CH₂CH₂(CH₂)₃CH₃], 0.76 (6H, t, *J* = 7.0 Hz, CH₃), 0.69–0.59 (4H, m, CH₂CH₂C₄H₉).
¹³C NMR (100 MHz, CDCl₃) δ 151.93, 151.08, 144.75, 141.98, 140.21, 139.36, 138.17, 137.20, 136.97, 132.67, 130.48, 129.01, 127.62, 126.94, 125.89, 125.51, 123.96, 122.97, 122.89, 121.92, 121.23, 120.65, 120.32, 120.05, 55.34, 40.43, 31.49, 29.68, 23.77, 22.57, 13.99.

m/z (EI): 626 (M^+ , 96%, ^{79}Br), 628 (M^+ , 100%, ^{81}Br). [Exact Mass (calcd): 626.1854].

Anal. Calcd for $C_{37}H_{39}BrO_2S$ (M.W. 627.67): C, 70.80; H, 6.26; Br, 12.73; S, 5.11. Found: C, 70.92; H, 6.33; Br, 12.60, S, 5.13.

In spite of the high excess of dibromide **1** over boronic acid **2** in this synthesis (**1:2** = 3 mol : 1 mol), a substantial amount of **FSF** was formed in the reaction, probably due to the high reactivity of **4** compared to that for **1**. When the ratio of **1:2** was increased to 1:7, this did not increase the yield of compound **4** due to the low solubility of 3,6-dibromodibenzothiophene-S,S-dioxide **1**, which does not completely dissolve in the mixture.

2,7-Bis[7-(9,9-di-n-hexylfluorene-2-yl)dibenzothiophene-S,S-dioxide-3-yl]-9,9-di-n-hexylfluorene (FSFSF).



To a flask containing compound **4** (0.51 g, 0.81 mmol), 9,9-di-n-hexylfluorene-2,7-diboronic acid (0.14 g, 0.33 mmol) and dichlorobis(triphenylphosphine)palladium(II) (11 mg, 0.02 mmol), degassed 2 M potassium carbonate aqueous solution (1.5 cm³) and 1,4-dioxane (4 cm³) were added via syringe. The reaction mixture was stirred under argon with heating on an oil bath at 100 °C for ca. 40 h under protection from light. Filtration, washing with dioxane (ca. 1 cm³), n-hexane (ca. 3.5 cm³) and water (to remove inorganic salts) afforded crude product (0.42 g), which was purified by column chromatography on silica gel eluting with a mixture PE:DCM, 1:1 to afford the title product **FSFSF** (0.33 g, 72%), as a light yellow powder, m.p. > 270 °C.

¹H NMR (500 MHz, CDCl₃): δ 8.17 (4H, dd, *J* = 1.5 and 6.0 Hz), 8.02–7.95 (4H, m), 7.93 (2H, d, *J* = 8.0 Hz), 7.92 (2H, d, *J* = 8.0 Hz), 7.86 (2H, d, *J* = 8.0 Hz), 7.81 (2H, d, *J* = 7.5 Hz), 7.76 (2H, dd, *J* = 1.5 and 6.0 Hz), 7.69–7.60 (8H, m), 7.41–7.33 (6H, m), 2.16–2.09 [4H, m, CH₂C₅H₁₁ (central fluorene)], 2.08–2.00 [8H, m, CH₂C₅H₁₁ (terminal fluorenes)], 1.18–1.00 [36H, m, CH₂CH₂(CH₂)₃CH₃], 0.782 [6H, t, *J* = 7.0 Hz, CH₃ (central fluorene)], 0.773 [12H, t, *J* = 7.0 Hz, CH₃ (terminal fluorenes)], 0.74–0.60 (12H, m, CH₂CH₂C₄H₉).

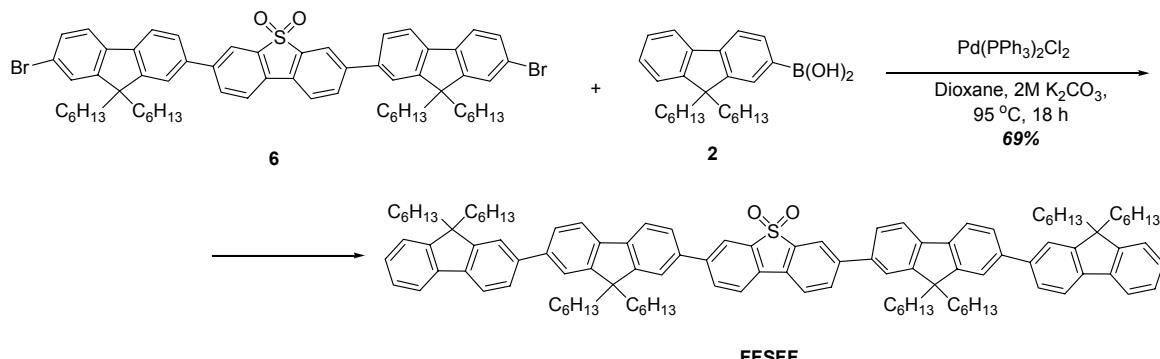
¹³C NMR (75 MHz, CDCl₃): δ 152.28, 151.89, 151.08, 144.20, 143.93, 141.83, 140.96, 140.26, 138.63, 138.60, 137.99, 137.39, 132.62, 132.61, 130.11, 129.89, 127.58, 126.93, 126.12, 125.87, 122.97, 121.97 (×2), 121.31, 121.22, 120.71, 120.66, 120.62, 120.31, 120.03, 55.69, 55.34, (40.47, 40.49), (31.54, 31.51), 29.71, (23.91, 23.80), (22.62, 22.60), 14.02.

m/z (MALDI-TOF, dithranol matrix/THF): 1426.78 (93%), 1427.78 (100%), 1428.78 (71%), 1429.78 (34%).

[Exact Mass (calcd): 1426.7846. Calculated MS: m/z 1427.79 (100.0%), 1426.78 (93.4%), 1428.79 (53.0%), 1429.79 (18.5%), 1429.78 (9.0%), 1428.78 (8.4%)].

Anal. Calcd for $C_{99}H_{110}O_4S_2$ (M.W. 1428.06): C, 83.26; H, 7.76; S, 4.49. Found: C, 83.07; H, 7.93; S, 4.39

3,7-Bis[7-(9,9-di-n-hexylfluorene-2-yl)-9,9-di-n-hexylfluorene-2-yl]dibenzothiophene-S,S-dioxide (FFSFF).



To a flask containing compound **6** (0.20 g, 0.19 mmol), 9,9-di-n-hexyl-2-fluoreneboronic acid (0.22 g, 0.58 mmol) and dichlorobis(triphenylphosphine)palladium(II) (9 mg, 0.01 mmol), degassed 2 M potassium carbonate aqueous solution (0.9 cm³) and 1,4-dioxane (5 cm³) were added via syringe. The reaction mixture was stirred under argon with heating on

an oil bath at 95 °C for 18 h with protection from light. After removing the solvent, the product was extracted with dichloromethane (50 cm³), washed with water and dried over anhydrous magnesium sulphate. After evaporation of the solvent, the residue was purified by column chromatography on silica gel eluting first with cyclohexane to remove by-products and then with mixture of cyclohexane and dichloromethane (cHex:DCM=3:1 v/v) to obtain the title product, **FFSFF** (0.21 g, 69%), as a light yellow powder, m.p. 225.6–226.3 °C.

¹H NMR (500 MHz, CDCl₃): δ 8.18 (2H, d, *J* = 1.5 Hz, H-4,6 dibenzothiophene), 7.99 (2H, dd, *J* = 1.5 and 8.0 Hz, H-2,8 dibenzothiophene), 7.93 (2H, d, *J* = 8.0 Hz, H-1,9 dibenzothiophene), 7.85 (2H, d, *J* = 7.0 Hz), 7.83 (2H, d, *J* = 8.0 Hz), 7.80 (2H, d, *J* = 7.5 Hz), 7.75 (2H, d, *J* = 7.0 Hz), 7.70–7.63 (12H, m), 7.39–7.31 (6H, m), 2.16–2.08 [8H, m, CH₂C₅H₁₁ (inner fluorene)], 2.08–1.98 [8H, m, CH₂C₅H₁₁ (terminal fluorene)], 1.18–1.02 [48H, m, CH₂CH₂(CH₂)₃CH₃], 0.778 [12H, t, *J* = 7.0 Hz, CH₃ (inner fluorene)], 0.773 [12H, t, *J* = 7.0 Hz, CH₃ (terminal fluorene)], 0.78–0.66 (16H, m, CH₂CH₂C₄H₉).

¹³C NMR (100 MHz, CDCl₃) δ 152.20, 151.87, 151.51, 151.00, 144.16, 141.53, 141.11, 140.73, 140.46, 140.31, 139.44, 138.65, 137.42, 132.58, 129.98, 127.05, 126.80, 126.28, 126.08, 125.99, 122.93, 121.94, 121.50, 121.43, 121.31, 120.65, 120.37, 120.27, 119.92, 119.75, 55.50, 55.17, (40.54, 40.37), (31.49, 31.47), 29.68, (23.85, 23.78), (22.59, 22.56), 14.02.

m/z (MALDI-TOF, dithranol matrix/THF): 1544.92 (95%), 1545.92 (100%), 1546.92 (88%), 1547.91 (62%), 1548.91 (27%).

[Exact Mass (calcd): 1545.0261. Calculated MS: m/z 1546.03 (100.0%), 1545.03 (82.6%), 1547.03 (60.0%), 1548.04 (23.8%), 1549.04 (7.0%), 1548.03 (4.5%), 1547.02 (3.7%)].

Anal. Calcd for C₁₁₂H₁₃₆O₂S (M.W. 1546.34): C, 86.99; H, 8.86; S, 2.07. Found: C, 86.89; H, 8.91; S, 2.15

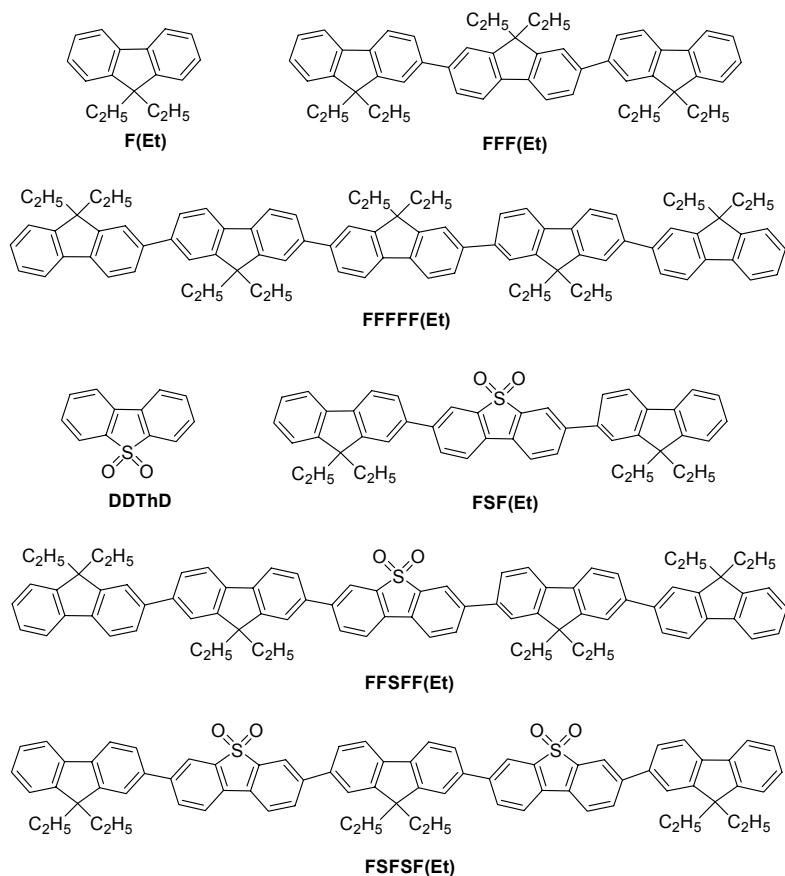
Cyclic voltammetry and Spectroelectrochemistry.

Cyclic voltammetry experiments were performed on a BAS-CV50W electrochemical workstation with *iR* compensation at 100 mV s⁻¹, using Ag/Ag⁺ (0.01 M AgNO₃ in dry acetonitrile), platinum disk (\varnothing = 1.6 mm or 2.0 mm) and platinum wire as the reference, working and counter electrodes, respectively. Oxidation potentials were measured in dichloromethane and reduction potentials were measured in dry tetrahydrofuran solution under argon at room temperature. The solutions contained the substrate in concentrations *ca.* 2–5 × 10⁻⁴ M, together with *n*-Bu₄NPF₆ (0.2 M) as the supporting electrolyte.

Spectroelectrochemical measurements were performed on a Varian Cary 5E spectrophotometer in a 1 mm quartz cell using a Pt grid as the working electrode and Pt wire as the counter and reference electrodes, in DCM solution with 0.1 M Bu₄NPF₆ as the supporting electrolyte.

Computational Procedures

The *ab initio* computations were performed for compounds with 9,9-diethyl substituents on the fluorene ring (instead of the 9,9-dihexylfluorene derivatives studied experimentally) to decrease the computation time. The optimisation of the geometries of dibenzothiophene-S,S-dioxide (**S**), 9,9-diethylfluorene **F(Et)**, its trimer **FFF(Et)** and pentamer **FFFFF(Et)**, as well as compounds **FSF(Et)**, **FFSFF(Et)**, and **FSFSF(Et)** were carried out with the Gaussian 98⁵ package of programs at density-functional theory (DFT) level using Pople's 6-31G split valence basis set supplemented by *d*-polarisation functions on heavy atoms. DFT calculations were carried out using Becke's three-parameter hybrid exchange functional⁶ with Lee–Yang–Parr gradient-corrected correlation functional (B3LYP).⁷ Thus, optimization of the geometries and calculation of electronic structures were performed at B3LYP/6-31G(d) level of theory. Contours of HOMO and LUMO orbitals were visualised using Molekel v.4.3 program.⁸ No constraints of bonds/angles/dihedral angles were applied in the calculations and all the atoms were free to optimise.



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**B3LYP/6-31(d) optimised geometries of
F(Et), FFF(Et), FFFF(Et), S, FSF(Et), FFSFF(Et) and FSFSF(Et)**

9,9-Diethylfluorene F(Et)*E* = -658.6761028 Hartree

Dipole moment: 0.3637 Debye.

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	3.460451	0.538079	0.000117
2	6	0	3.015416	1.863983	0.000018
3	6	0	1.648977	2.154080	-0.000015
4	6	0	0.734657	1.098842	0.000067
5	6	0	1.180882	-0.238250	0.000114
6	6	0	2.542993	-0.519937	0.000156
7	6	0	-0.734588	1.098852	-0.000050
8	6	0	-1.180862	-0.238225	-0.000138
9	6	0	0.000023	-1.208909	-0.000004
10	6	0	-1.648879	2.154116	0.000065
11	6	0	-3.015329	1.864058	0.000014
12	6	0	-3.460401	0.538169	-0.000140
13	6	0	-2.542971	-0.519878	-0.000208
14	6	0	0.000395	-2.123093	-1.260299
15	6	0	-0.000042	-1.403528	-2.612432
16	6	0	-0.000287	-2.123112	1.260285
17	6	0	-0.000360	-1.403593	2.612441
18	1	0	4.526572	0.327008	0.000170
19	1	0	3.738656	2.675247	-0.000048
20	1	0	1.307951	3.186443	-0.000127
21	1	0	2.899903	-1.547510	0.000243
22	1	0	-1.307822	3.186468	0.000201
23	1	0	-3.738547	2.675342	0.000098
24	1	0	-4.526528	0.327126	-0.000197
25	1	0	-2.899918	-1.547438	-0.000324
26	1	0	0.878037	-2.781219	-1.198775
27	1	0	-0.876501	-2.782218	-1.198805
28	1	0	0.000128	-2.133117	-3.430287
29	1	0	-0.885500	-0.770125	-2.729340
30	1	0	0.884924	-0.769485	-2.729589
31	1	0	-0.877610	-2.781644	1.198569
32	1	0	0.876917	-2.781849	1.198976
33	1	0	-0.000586	-2.133215	3.430269
34	1	0	0.884944	-0.770014	2.729583
35	1	0	-0.885473	-0.769731	2.729425

FFF(Et)

HF=-1973.6464711 Hartree

Dipole moment: 0.1697 Debye

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.729096	0.791568	-0.073076
2	6	0	0.729158	0.791531	0.072480
3	6	0	1.174266	-0.545209	0.121834
4	6	0	-0.000027	-1.516443	-0.000254
5	6	0	-1.174284	-0.545145	-0.122326
6	6	0	-2.525537	-0.831409	-0.252725
7	6	0	-3.468906	0.213647	-0.337062
8	6	0	-3.005068	1.542669	-0.287760
9	6	0	-1.649204	1.838507	-0.157346
10	6	0	1.649266	1.838460	0.156725
11	6	0	3.005123	1.542591	0.287242
12	6	0	3.468873	0.213559	0.336679
13	6	0	2.525507	-0.831497	0.252360
14	6	0	-0.130203	-2.430502	1.253577
15	6	0	0.130123	-2.430447	-1.254133
16	6	0	-4.916749	-0.082318	-0.467483
17	6	0	4.916756	-0.082408	0.467300
18	6	0	-5.363107	-1.162699	-1.253084
19	6	0	-6.718428	-1.463196	-1.378287
20	6	0	-7.655026	-0.674160	-0.708360

21	6	0	-7.228036	0.412774	0.080319
22	6	0	-5.876954	0.706370	0.199581
23	6	0	5.876989	0.706199	-0.199799
24	6	0	7.228079	0.412675	-0.080320
25	6	0	7.654978	-0.674097	0.708617
26	6	0	6.718338	-1.463062	1.378568
27	6	0	5.363015	-1.162655	1.253130
28	6	0	-9.118712	-0.753380	-0.642646
29	6	0	-9.583395	0.283034	0.193011
30	6	0	-8.419922	1.117194	0.728892
31	6	0	8.420026	1.116998	-0.728859
32	6	0	9.583459	0.283067	-0.192552
33	6	0	9.118694	-0.753238	0.643196
34	6	0	-10.015360	-1.639874	-1.242995
35	6	0	-11.382369	-1.483273	-1.003092
36	6	0	-11.845692	-0.455960	-0.174664
37	6	0	-10.945996	0.432632	0.427578
38	6	0	10.946096	0.432748	-0.426864
39	6	0	11.845745	-0.455660	0.175717
40	6	0	11.382338	-1.482879	1.004216
41	6	0	10.015293	-1.639558	1.243869
42	6	0	8.524254	2.607717	-0.291897
43	6	0	8.342500	1.073486	-2.283491
44	6	0	8.233841	-0.319624	-2.910835
45	6	0	8.612872	2.860792	1.216192
46	6	0	-8.342103	1.074158	2.283509
47	6	0	-8.233362	-0.318772	2.911239
48	6	0	-8.524299	2.607790	0.291490
49	6	0	-8.613348	2.860397	-1.216642
50	6	0	-0.267080	-1.709270	2.597882
51	6	0	0.266731	-1.709161	-2.598434
52	1	0	-2.872060	-1.861885	-0.261584
53	1	0	-3.720036	2.355268	-0.380393
54	1	0	-1.320262	2.874410	-0.130822
55	1	0	1.320316	2.874356	0.130108
56	1	0	3.720140	2.355151	0.379789
57	1	0	2.872039	-1.861965	0.261346
58	1	0	-0.998168	-3.086859	1.103044
59	1	0	0.747676	-3.090597	1.281089
60	1	0	0.998214	-3.086666	-1.103721
61	1	0	-0.747659	-3.090675	-1.281546
62	1	0	-4.633767	-1.759257	-1.793612
63	1	0	-7.033464	-2.299358	-1.997626
64	1	0	-5.544975	1.526410	0.831577
65	1	0	5.545033	1.526115	-0.831962
66	1	0	7.033360	-2.299075	1.998113
67	1	0	4.633605	-1.759201	1.793567
68	1	0	-9.660281	-2.440369	-1.887298
69	1	0	-12.091893	-2.165496	-1.463898
70	1	0	-12.912156	-0.346012	0.003278
71	1	0	-11.316929	1.228054	1.070183
72	1	0	11.317099	1.228098	-1.069518
73	1	0	12.912235	-0.345652	-0.002026
74	1	0	12.091823	-2.164972	1.465275
75	1	0	9.660160	-2.439977	1.888235
76	1	0	9.404468	3.041640	-0.785878
77	1	0	7.654691	3.140771	-0.700437
78	1	0	7.481341	1.680719	-2.594493
79	1	0	9.231399	1.583449	-2.679515
80	1	0	8.177773	-0.240631	-4.002473
81	1	0	9.101999	-0.939746	-2.664295
82	1	0	7.336550	-0.844347	-2.566797
83	1	0	8.690516	3.935414	1.416928
84	1	0	7.726678	2.486158	1.739115
85	1	0	9.491422	2.374307	1.652384
86	1	0	-7.480875	1.681463	2.594177
87	1	0	-9.230905	1.584272	2.679565
88	1	0	-8.177192	-0.239503	4.002852
89	1	0	-9.101533	-0.938964	2.664934
90	1	0	-7.336099	-0.843585	2.567256
91	1	0	-9.404379	3.041845	0.785597
92	1	0	-7.654617	3.140946	0.699640
93	1	0	-8.690921	3.934967	-1.417688
94	1	0	-7.727360	2.485504	-1.739732
95	1	0	-9.492102	2.373885	-1.652388
96	1	0	-0.353997	-2.437511	3.412232
97	1	0	0.603337	-1.077901	2.804906
98	1	0	-1.157835	-1.072963	2.621835
99	1	0	0.353774	-2.437375	-3.412795
100	1	0	-0.603857	-1.078009	-2.805395
101	1	0	1.157327	-1.072634	-2.622440

FFFFF(Et)

HF=-3288.6168836 Hartree
 Dipole moment: 0.0487 Debye

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.730910	0.052002	0.892746
2	6	0	0.730937	-0.052265	0.892730
3	6	0	1.177206	-0.089522	-0.444054
4	6	0	-0.000007	-0.000087	-1.415498
5	6	0	-1.177203	0.089293	-0.444027
6	6	0	-2.531432	0.183316	-0.730312
7	6	0	-3.476853	0.241446	0.314698
8	6	0	-3.012172	0.202815	1.643769
9	6	0	-1.653206	0.109562	1.939651
10	6	0	1.653254	-0.109850	1.939616
11	6	0	3.012214	-0.203089	1.643705
12	6	0	3.476870	-0.241691	0.314626
13	6	0	2.531430	-0.183543	-0.730367
14	6	0	-4.927267	0.336128	0.018241
15	6	0	4.927281	-0.336329	0.018154
16	6	0	-5.391561	1.113808	-1.060506
17	6	0	-6.749061	1.208368	-1.361268
18	6	0	-7.670091	0.514042	-0.574436
19	6	0	-7.225062	-0.268284	0.510588
20	6	0	-5.871790	-0.355908	0.804702
21	6	0	5.871778	0.355773	0.804587
22	6	0	7.225051	0.268202	0.510469
23	6	0	7.670110	-0.514137	-0.574534
24	6	0	6.749106	-1.208531	-1.361339
25	6	0	5.391603	-1.114022	-1.060573
26	6	0	-9.129763	0.414249	-0.653290
27	6	0	-9.576977	-0.439418	0.375516
28	6	0	-8.402082	-0.946891	1.212572
29	6	0	8.402038	0.946914	1.212406
30	6	0	9.576956	0.439434	0.375386
31	6	0	9.129777	-0.414284	-0.653395
32	6	0	-10.048660	0.997497	-1.527869
33	6	0	-11.405840	0.723593	-1.368574
34	6	0	-11.872173	-0.127019	-0.347400
35	6	0	-10.929982	-0.707194	0.527450
36	6	0	10.929952	0.707255	0.527322
37	6	0	11.872168	0.127084	-0.347505
38	6	0	11.405870	-0.723574	-1.368657
39	6	0	10.048700	-0.997530	-1.527950
40	6	0	8.520174	0.512621	2.702956
41	6	0	8.285933	2.498874	1.168510
42	6	0	8.158759	3.121999	-0.224913
43	6	0	8.646495	-0.992727	2.956034
44	6	0	-8.286049	-2.498862	1.168814
45	6	0	-8.158856	-3.122142	-0.224539
46	6	0	-8.520202	-0.512482	2.703086
47	6	0	-8.646427	0.992894	2.956055
48	6	0	-13.321484	-0.405836	-0.196370
49	6	0	13.321473	0.405930	-0.196451
50	6	0	-14.278186	0.606728	-0.416510
51	6	0	-15.631286	0.332946	-0.275830
52	6	0	-16.064316	-0.957584	0.088541
53	6	0	-15.130905	-1.970896	0.313444
54	6	0	-13.773467	-1.688606	0.169600
55	6	0	13.773421	1.688690	0.169598
56	6	0	15.130848	1.970991	0.313522
57	6	0	16.064284	0.957705	0.088614
58	6	0	15.631292	-0.332809	-0.275862
59	6	0	14.278202	-0.606605	-0.416614
60	6	0	17.529853	0.956352	0.160482
61	6	0	17.989859	-0.338522	-0.156051
62	6	0	16.820530	-1.276294	-0.456033
63	6	0	-16.820499	1.276462	-0.455998
64	6	0	-17.989859	0.338679	-0.156178
65	6	0	-17.529890	-0.956220	0.160307
66	6	0	-19.353224	0.612993	-0.163181
67	6	0	-20.258437	-0.409678	0.147166
68	6	0	-19.799838	-1.693216	0.461541
69	6	0	-18.432122	-1.976040	0.470575
70	6	0	18.432051	1.976161	0.470886

71	6	0	19.799771	1.693352	0.461934
72	6	0	20.258407	0.409841	0.147505
73	6	0	19.353228	-0.612819	-0.162977
74	6	0	16.767420	-2.473081	0.539372
75	6	0	16.891650	-1.851926	-1.900757
76	6	0	16.957215	-0.820630	-3.031243
77	6	0	16.688528	-2.106653	2.024404
78	6	0	-16.891510	1.852220	-1.900677
79	6	0	-16.767439	2.473149	0.539527
80	6	0	-16.956944	0.821019	-3.031259
81	6	0	-16.688648	2.106554	2.024523
82	6	0	-0.095018	-1.257007	-2.329400
83	6	0	0.094979	1.256885	-2.329328
84	6	0	0.193790	2.604707	-1.608480
85	6	0	-0.193799	-2.604874	-1.608632
86	1	0	-2.878148	0.184145	-1.760777
87	1	0	-3.729648	0.274748	2.456281
88	1	0	-1.323716	0.091100	2.975553
89	1	0	1.323784	-0.091412	2.975525
90	1	0	3.729705	-0.275035	2.456202
91	1	0	2.878127	-0.184341	-1.760839
92	1	0	-4.674740	1.672811	-1.655222
93	1	0	-7.078424	1.822659	-2.195614
94	1	0	-5.525101	-0.981604	1.623537
95	1	0	5.525065	0.981482	1.623404
96	1	0	7.078492	-1.822838	-2.195663
97	1	0	4.674803	-1.673080	-1.655262
98	1	0	-9.717694	1.650267	-2.331808
99	1	0	-12.120218	1.152883	-2.065179
100	1	0	-11.278062	-1.343607	1.337415
101	1	0	11.278005	1.343700	1.337273
102	1	0	12.120271	-1.152871	-2.065234
103	1	0	9.717759	-1.650345	-2.331862
104	1	0	9.388630	1.028169	3.135327
105	1	0	7.641601	0.899669	3.237250
106	1	0	7.418723	2.788762	1.777485
107	1	0	9.165966	2.916860	1.676339
108	1	0	8.078403	4.212208	-0.147394
109	1	0	9.030285	2.893908	-0.847420
110	1	0	7.267885	2.757004	-0.746370
111	1	0	8.728846	-1.191129	4.030699
112	1	0	7.773834	-1.537820	2.581171
113	1	0	9.536209	-1.406697	2.470436
114	1	0	-7.418872	-2.788731	1.777843
115	1	0	-9.166117	-2.916760	1.676656
116	1	0	-8.078524	-4.212343	-0.146891
117	1	0	-9.030359	-2.894112	-0.847101
118	1	0	-7.267958	-2.757227	-0.746010
119	1	0	-9.388690	-1.027947	3.135490
120	1	0	-7.641655	-0.899543	3.237415
121	1	0	-8.728753	1.191381	4.030705
122	1	0	-7.773734	1.537902	2.581139
123	1	0	-9.536118	1.406890	2.470436
124	1	0	-13.943062	1.609093	-0.671362
125	1	0	-15.449973	-2.973488	0.587254
126	1	0	-13.046885	-2.482932	0.314840
127	1	0	13.046820	2.482993	0.314862
128	1	0	15.449887	2.973570	0.587415
129	1	0	13.943103	-1.608962	-0.671527
130	1	0	-19.720393	1.607664	-0.406279
131	1	0	-21.325600	-0.204115	0.143569
132	1	0	-20.513779	-2.477047	0.700456
133	1	0	-18.080745	-2.975362	0.715225
134	1	0	18.080645	2.975462	0.715582
135	1	0	20.513686	2.477176	0.700956
136	1	0	21.325573	0.204290	0.143972
137	1	0	19.720426	-1.607468	-0.406123
138	1	0	15.902593	-3.095635	0.271708
139	1	0	17.655195	-3.095986	0.363953
140	1	0	17.768362	-2.511661	-1.957784
141	1	0	16.016118	-2.499054	-2.048395
142	1	0	16.998608	-1.324915	-4.003432
143	1	0	16.077947	-0.167958	-3.029983
144	1	0	17.846105	-0.187061	-2.945038
145	1	0	16.650543	-3.014071	2.637776
146	1	0	17.560510	-1.524862	2.340876
147	1	0	15.792504	-1.516822	2.243888
148	1	0	-17.768232	2.511941	-1.957721
149	1	0	-16.015978	2.499379	-2.048179
150	1	0	-15.902583	3.095715	0.271984
151	1	0	-17.655192	3.096086	0.364117

152	1	0	-16.998255	1.325387	-4.003409
153	1	0	-16.077658	0.168372	-3.029966
154	1	0	-17.845824	0.187417	-2.945195
155	1	0	-16.650678	3.013900	2.638002
156	1	0	-17.560664	1.524746	2.340874
157	1	0	-15.792653	1.516674	2.243993
158	1	0	-0.966810	-1.131089	-2.985878
159	1	0	0.783236	-1.259830	-2.989586
160	1	0	0.966750	1.131005	-2.985842
161	1	0	-0.783295	1.259752	-2.989488
162	1	0	0.255225	3.421011	-2.337129
163	1	0	-0.681119	2.786198	-0.975497
164	1	0	1.084669	2.655381	-0.973921
165	1	0	-0.255239	-3.421133	-2.337330
166	1	0	0.681125	-2.786394	-0.975676
167	1	0	-1.084665	-2.655599	-0.974059

Dibenzothiophene-S,S-dioxide **S**

E = -1010.6677122 Hartree

Dipole moment: 5.5106 Debye

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	0.739396	0.954552	0.000036
2	6	0	-0.739398	0.954551	0.000031
3	6	0	-1.291507	-0.334065	-0.000103
4	16	0	0.000001	-1.589902	0.000073
5	6	0	1.291506	-0.334064	-0.000101
6	6	0	2.657404	-0.569394	-0.000223
7	6	0	3.511578	0.538248	-0.000148
8	6	0	2.985258	1.833660	0.000011
9	6	0	1.604886	2.050675	0.000078
10	6	0	-1.604888	2.050674	0.000060
11	6	0	-2.985259	1.833659	-0.000014
12	6	0	-3.511579	0.538246	-0.000144
13	6	0	-2.657404	-0.569395	-0.000211
14	8	0	0.000021	-2.335388	-1.267246
15	8	0	-0.000018	-2.334828	1.267732
16	1	0	3.049302	-1.581898	-0.000366
17	1	0	4.587284	0.389568	-0.000229
18	1	0	3.658462	2.686251	0.000080
19	1	0	1.214004	3.064117	0.000183
20	1	0	-1.214006	3.064116	0.000153
21	1	0	-3.658464	2.686249	0.000031
22	1	0	-4.587285	0.389566	-0.000205
23	1	0	-3.049302	-1.581900	-0.000328

FSF(Et)

HF=-2325.6383698 Hartree

Dipole moment: 5.7212 Debye

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z
1	6	0	-0.734237	0.437726	-0.066638
2	6	0	0.734183	0.437728	0.063396
3	6	0	1.285960	-0.850136	0.120445
4	16	0	-0.000093	-2.105708	-0.000802
5	6	0	-1.286103	-0.850146	-0.122640
6	6	0	-2.642369	-1.089985	-0.241855
7	6	0	-3.525139	0.005847	-0.314661
8	6	0	-2.978714	1.302938	-0.258905
9	6	0	-1.607880	1.525619	-0.135873
10	6	0	1.607871	1.525620	0.132049
11	6	0	2.978653	1.302952	0.255678
12	6	0	3.524977	0.005868	0.312644
13	6	0	2.642171	-1.089965	0.240313
14	8	0	-0.118191	-2.851432	1.261589
15	8	0	0.117984	-2.851978	-1.262871
16	6	0	-4.986565	-0.205343	-0.444700
17	6	0	4.986326	-0.205281	0.443631
18	6	0	-5.490551	-1.249991	-1.242585
19	6	0	-6.861107	-1.463625	-1.377802

20	6	0	-7.750234	-0.624840	-0.703478	
21	6	0	-7.262491	0.423062	0.103322	
22	6	0	-5.897041	0.633544	0.230029	
23	6	0	5.897227	0.633046	-0.231217	
24	6	0	7.262601	0.422745	-0.103376	
25	6	0	7.749837	-0.624414	0.704695	
26	6	0	6.860281	-1.462705	1.379069	
27	6	0	5.489809	-1.249256	1.242717	
28	6	0	-9.216350	-0.611043	-0.650879	
29	6	0	-9.621232	0.440946	0.196040	
30	6	0	-8.412286	1.187840	0.758480	
31	6	0	8.412820	1.187065	-0.758324	
32	6	0	9.621418	0.440896	-0.194173	
33	6	0	9.215993	-0.610460	0.653273	
34	6	0	-10.161046	-1.429695	-1.273543	
35	6	0	-11.517261	-1.188586	-1.044331	
36	6	0	-11.921696	-0.145364	-0.204630	
37	6	0	-10.973879	0.674618	0.420074	
38	6	0	10.974213	0.674599	-0.417278	
39	6	0	11.921644	-0.144720	0.208881	
40	6	0	11.516672	-1.187329	1.049083	
41	6	0	10.160305	-1.428463	1.277370	
42	6	0	8.416805	2.690927	-0.357310	
43	6	0	8.352052	1.102291	-2.312653	
44	6	0	8.339568	-0.307760	-2.910610	
45	6	0	8.474952	2.986622	1.144565	
46	6	0	-8.350193	1.104643	2.312817	
47	6	0	-8.336992	-0.304813	2.912163	
48	6	0	-8.416848	2.691316	0.355931	
49	6	0	-8.476364	2.985488	-1.146184	
50	1	0	-3.024059	-2.106059	-0.252600	
51	1	0	-3.645146	2.156446	-0.339554	
52	1	0	-1.228004	2.542938	-0.104865	
53	1	0	1.228060	2.542938	0.100187	
54	1	0	3.645105	2.156482	0.335917	
55	1	0	3.023795	-2.106053	0.251981	
56	1	0	-4.797112	-1.885306	-1.786328	
57	1	0	-7.223403	-2.271781	-2.007843	
58	1	0	-5.519158	1.424228	0.873759	
59	1	0	5.519752	1.423166	-0.875874	
60	1	0	7.222181	-2.270314	2.010038	
61	1	0	4.796025	-1.884140	1.786523	
62	1	0	-9.851153	-2.241915	-1.926243	
63	1	0	-12.264599	-1.816628	-1.521779	
64	1	0	-12.980693	0.030331	-0.035363	
65	1	0	-11.300082	1.482457	1.071178	
66	1	0	11.300825	1.481961	-1.068770	
67	1	0	12.980753	0.031008	0.040355	
68	1	0	12.263708	-1.814876	1.527653	
69	1	0	9.850009	-2.240215	1.930462	
70	1	0	9.271184	3.169056	-0.855483	
71	1	0	7.518588	3.156497	-0.786036	
72	1	0	7.456831	1.646138	-2.644365	
73	1	0	9.209920	1.659937	-2.712626	
74	1	0	8.290323	-0.254286	-4.003995	
75	1	0	9.243130	-0.865679	-2.643709	
76	1	0	7.474205	-0.883175	-2.565645	
77	1	0	8.471674	4.068653	1.318386	
78	1	0	7.614873	2.560790	1.672146	
79	1	0	9.383412	2.578019	1.599056	
80	1	0	-7.454758	1.648936	2.643219	
81	1	0	-9.207792	1.662566	2.712977	
82	1	0	-8.286578	-0.250250	4.005441	
83	1	0	-9.240805	-0.863067	2.646803	
84	1	0	-7.471966	-0.880526	2.566848	
85	1	0	-9.270870	3.169810	0.854369	
86	1	0	-7.518326	3.157462	0.783392	
87	1	0	-8.473283	4.067344	-1.321098	
88	1	0	-7.616747	2.559161	-1.674118	
89	1	0	-9.385223	2.576390	-1.599425	

FFSFF

HF=-3640.6088025 Hartree
 Dipole moment: 5.6270 Debye

Center Number	Atomic Number	Atomic Type	Coordinates (Angstroms)		
			X	Y	Z

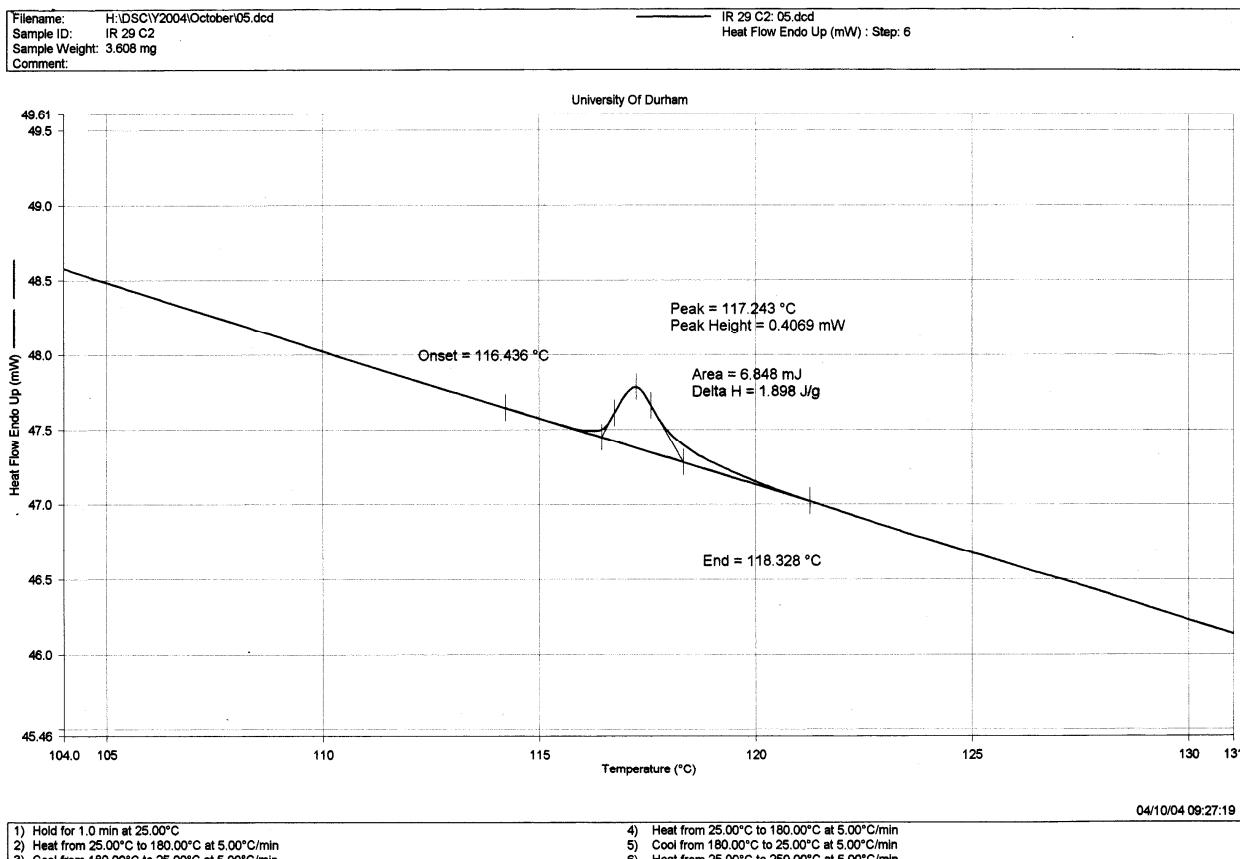
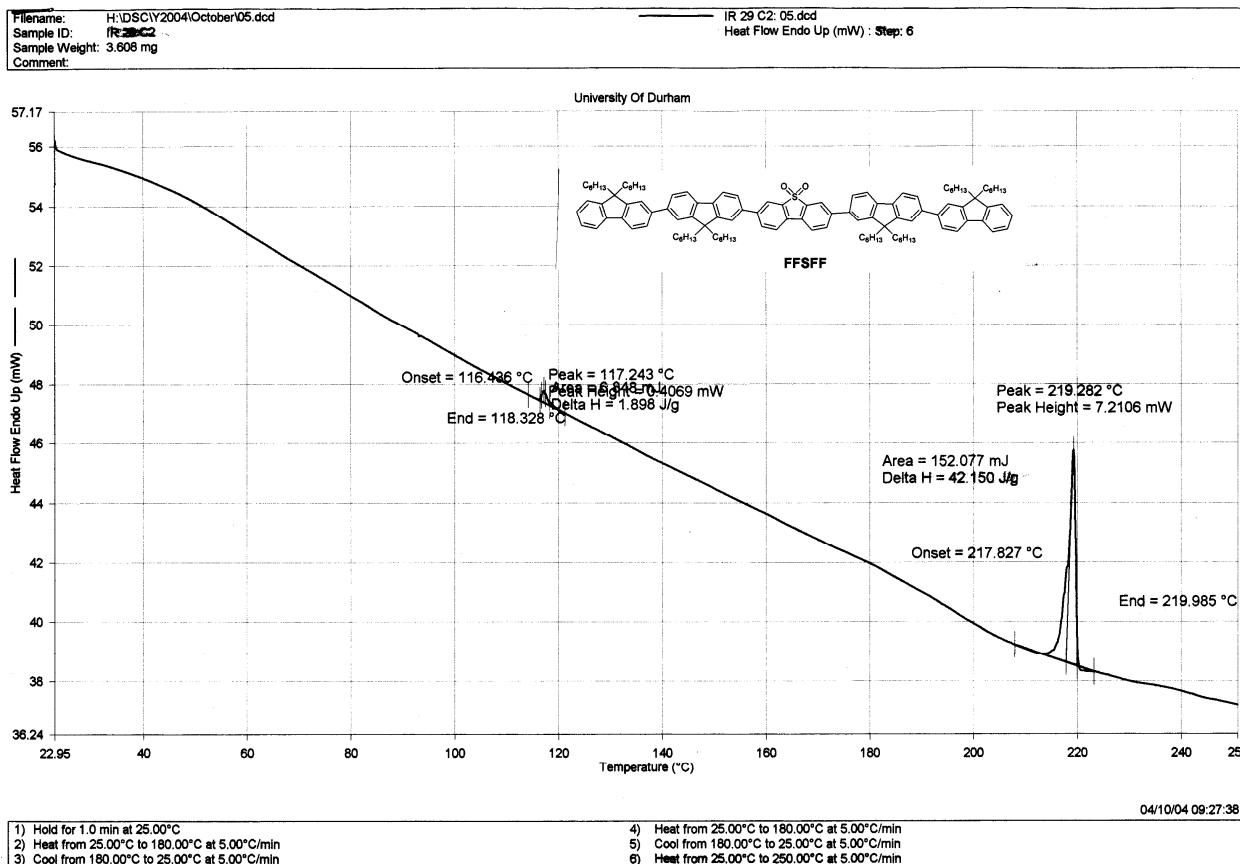
1	6	0	-0.734056	-0.174462	0.352944
2	6	0	0.737589	-0.182755	0.273078
3	6	0	1.283016	1.068583	-0.047751
4	16	0	-0.014456	2.302747	-0.245150
5	6	0	-1.296426	1.084274	0.097036
6	6	0	-2.658162	1.321805	0.099542
7	6	0	-3.537050	0.256296	0.379921
8	6	0	-2.978959	-1.009499	0.647288
9	6	0	-1.602817	-1.231922	0.633503
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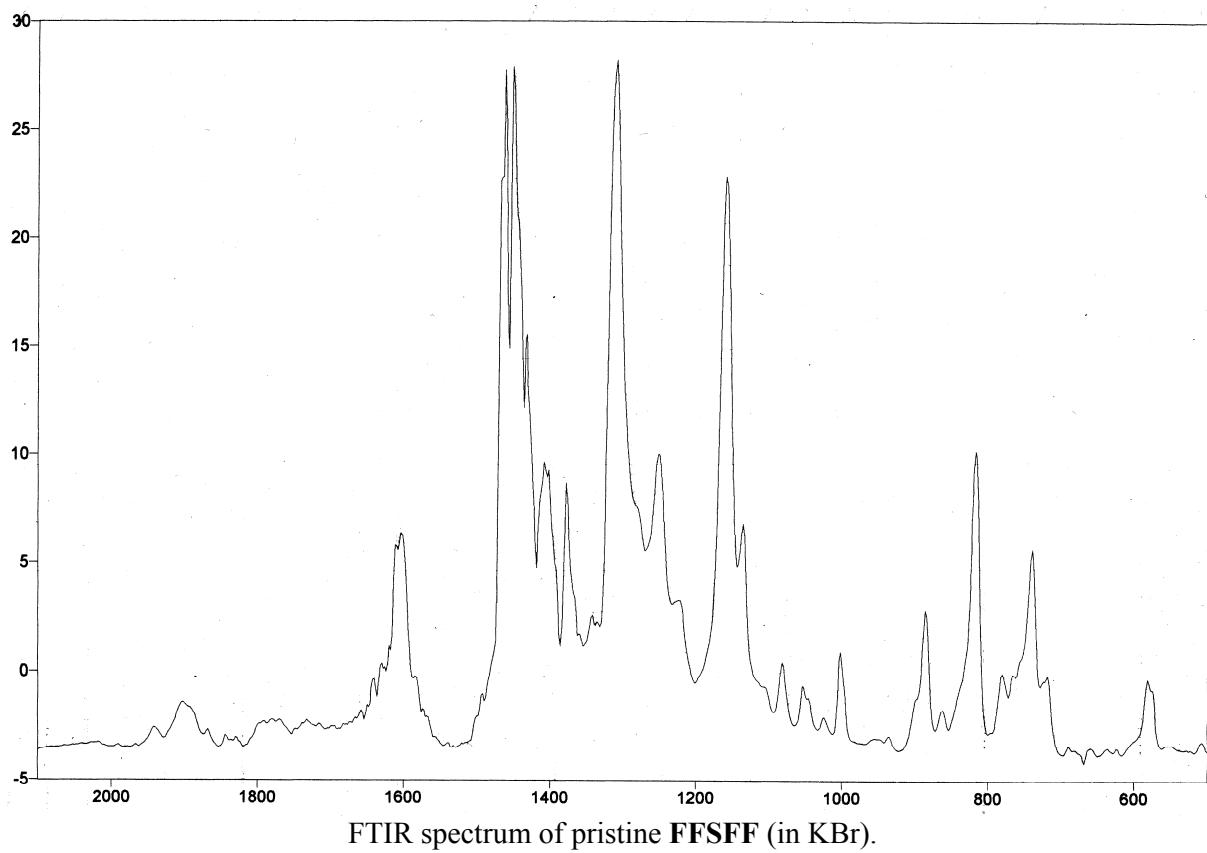
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Dipole moment: 8.7453 Debye

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8	6	0	3.014006	-1.927319	-0.080940
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13	6	0	-2.536501	0.451703	0.084250
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16	6	0	5.439073	0.765213	-0.907389
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23	6	0	-7.706736	0.263566	0.250043
24	6	0	-6.801493	1.061640	0.953846
25	6	0	-5.438776	0.771865	0.899733
26	6	0	9.173594	0.382253	-0.168545
27	6	0	9.762969	-0.586732	0.656156
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37	6	0	-11.974392	0.269186	-0.261118
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56	6	0	-18.059635	0.530773	0.459168
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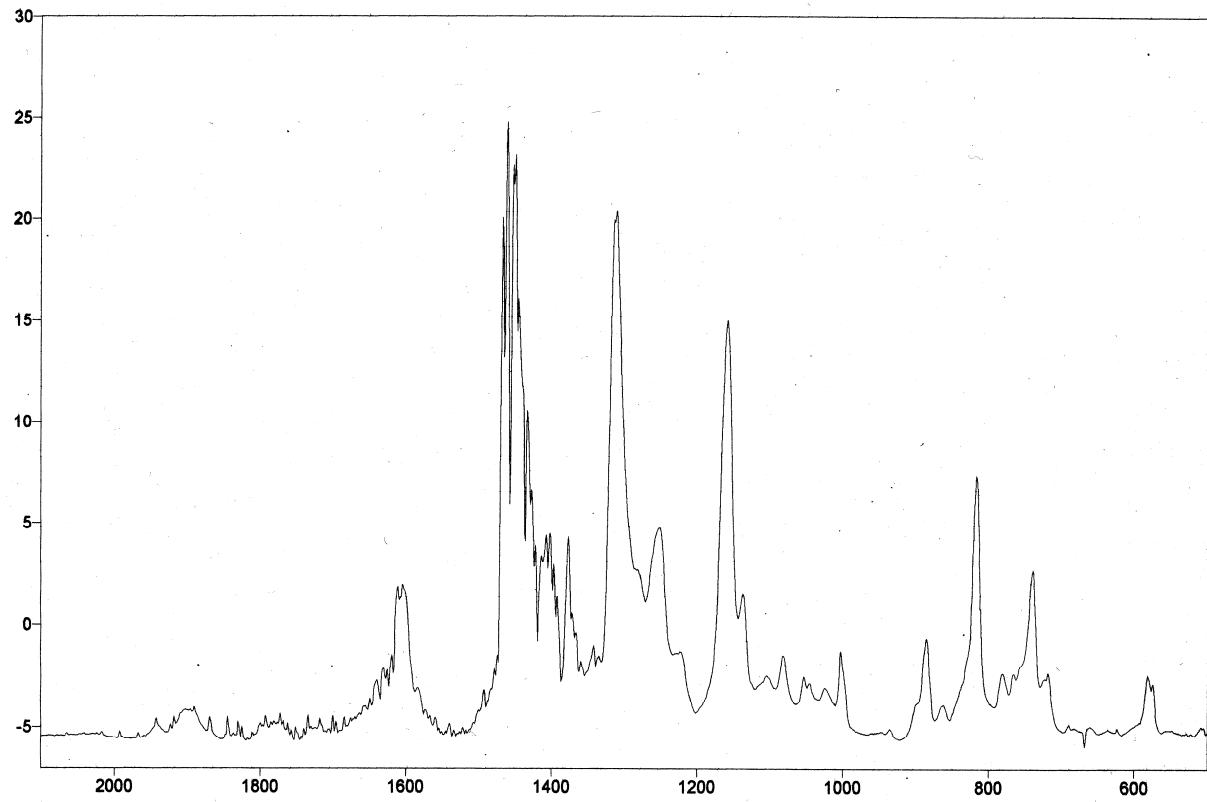
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96	1	0	11.535828	-1.451833	1.505967
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115	1	0	-17.644805	0.012547	3.209595
116	1	0	-17.621926	2.327263	2.602914
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118	1	0	-16.761573	4.304766	1.368577
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120	1	0	-17.721604	3.300040	0.272623
121	1	0	-16.823010	-2.332026	3.107615
122	1	0	-17.776136	-1.977427	1.658933
123	1	0	-16.008272	-2.019579	1.568358
124	1	0	17.624366	2.305062	-2.618625
125	1	0	15.872388	2.270120	-2.539565
126	1	0	15.893564	-0.058044	-3.125657
127	1	0	17.645881	-0.015144	-3.206242
128	1	0	16.761645	4.293158	-1.402725
129	1	0	15.952117	3.262792	-0.212742
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131	1	0	16.824368	-2.359058	-3.083005
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135	1	0	-0.830587	2.717156	1.217359
136	1	0	-0.921599	2.705871	-1.178491
137	1	0	0.831242	2.707608	-1.241046
138	1	0	-0.123630	2.050035	-3.437551
139	1	0	0.784036	0.687576	-2.765115
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Differential scanning calorimetry of FFSFF

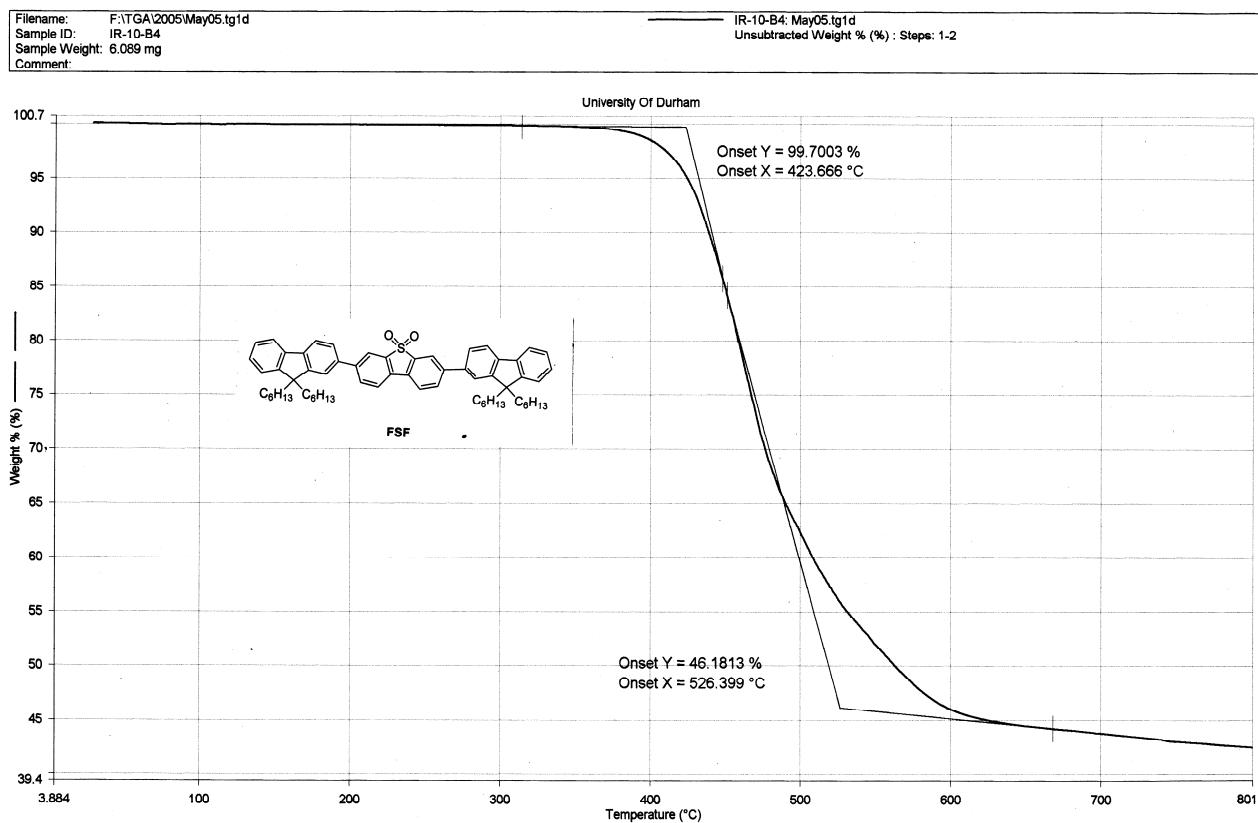


FTIR spectrum of pristine FFSFF (in KBr).

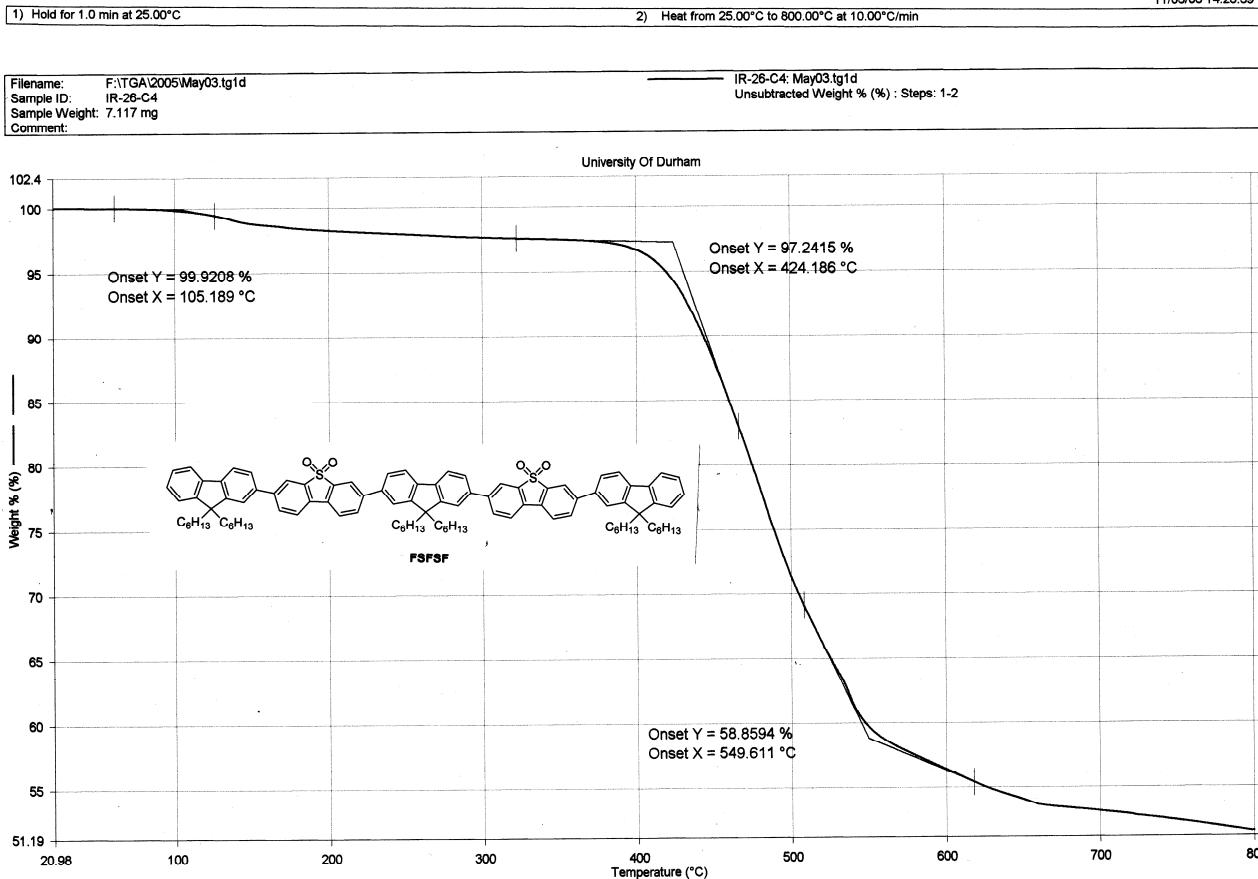


FTIR spectrum of FFSFF after annealing (in KBr).

[The film of FFSFF was drop-casted from chloroform on quartz plate and annealed at 180 °C for 24 h in air. Then it was redissolved in dichloromethane, the solvent was evaporated and the IR spectrum of FFSFF in KBr was recorded.]

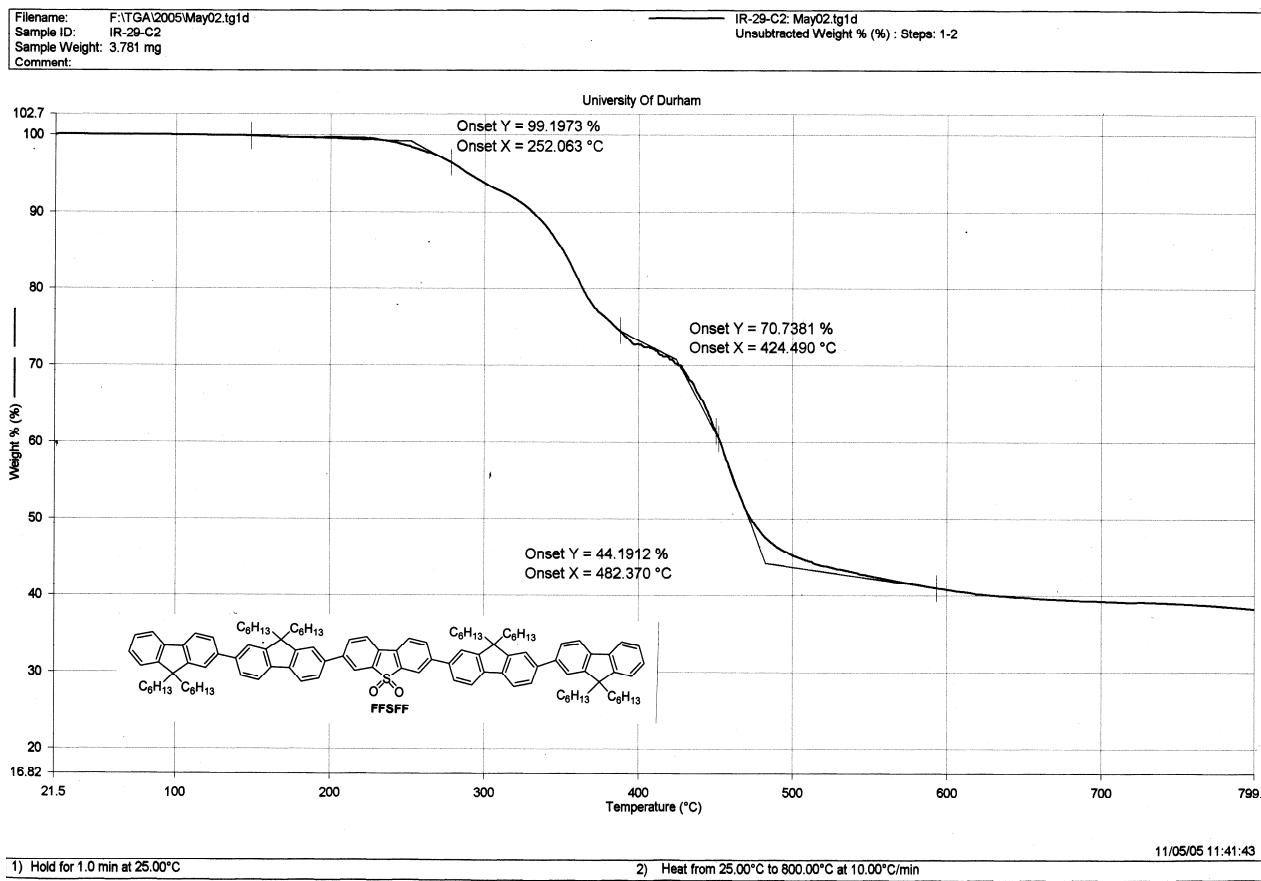


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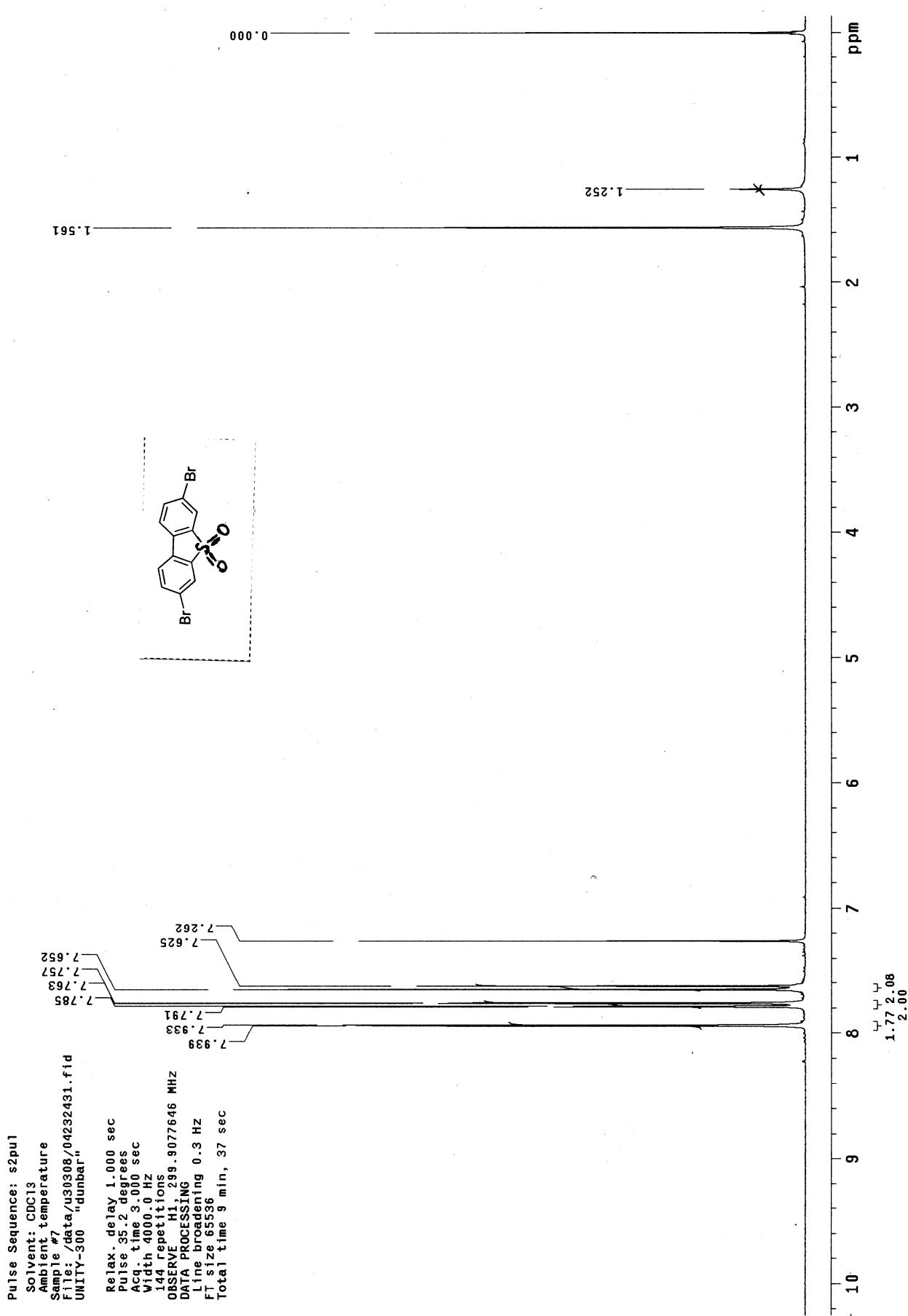


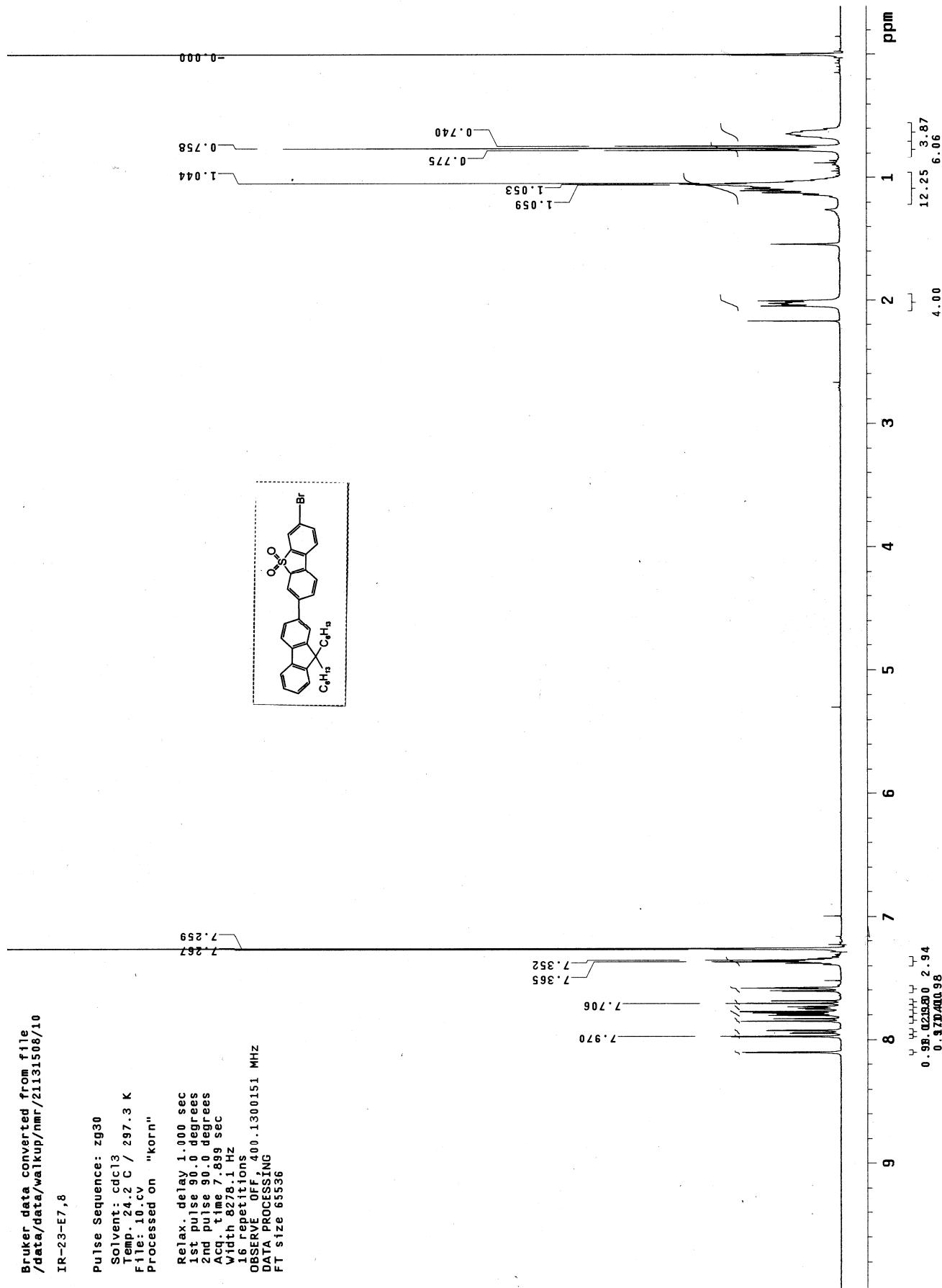
11/05/05 11:39:44

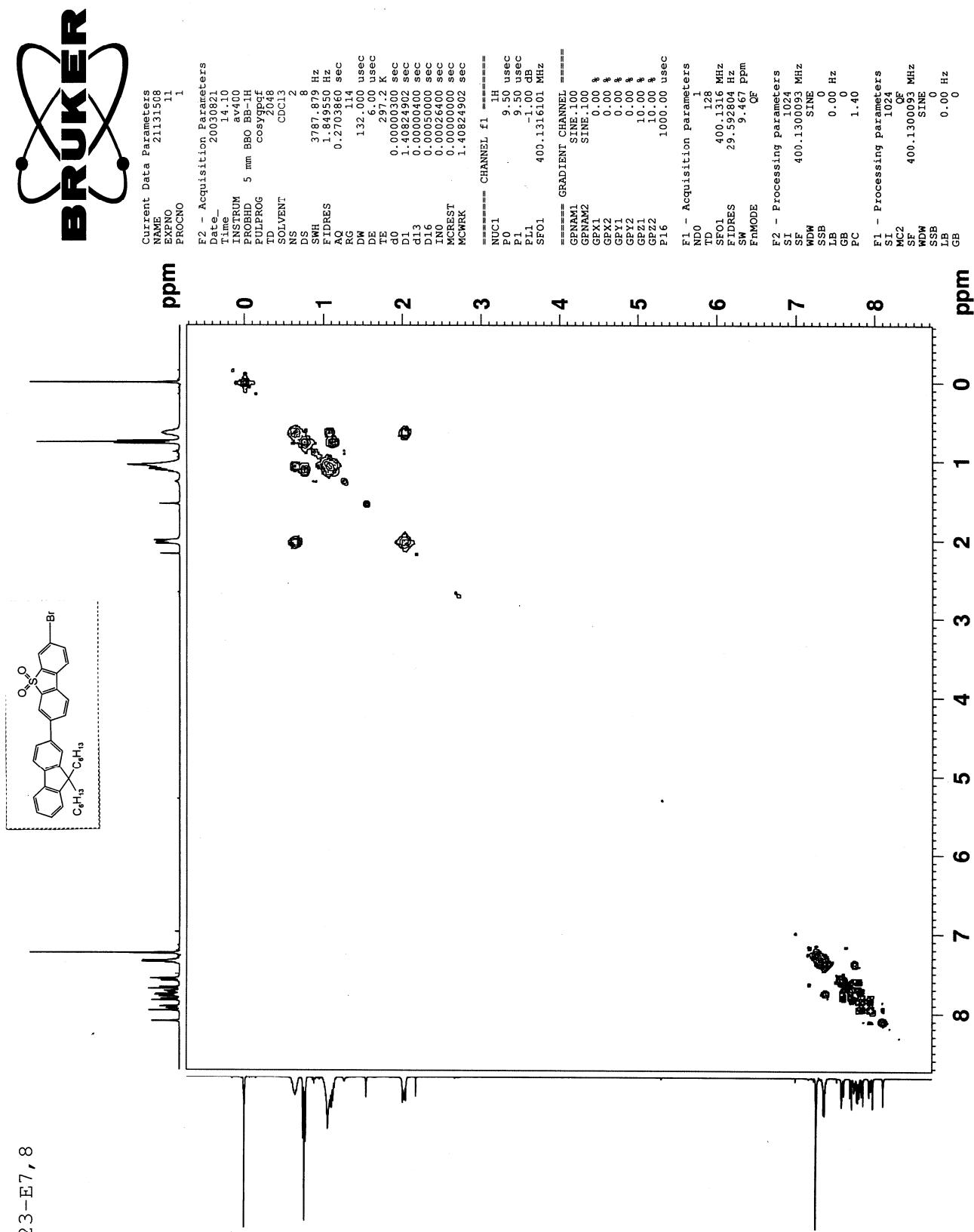
1) Hold for 1.0 min at 25.00°C
 2) Heat from 25.00°C to 800.00°C at 10.00°C/min



IR-12E; 3,6-dibromobenzothiophene-S,S-dioxide

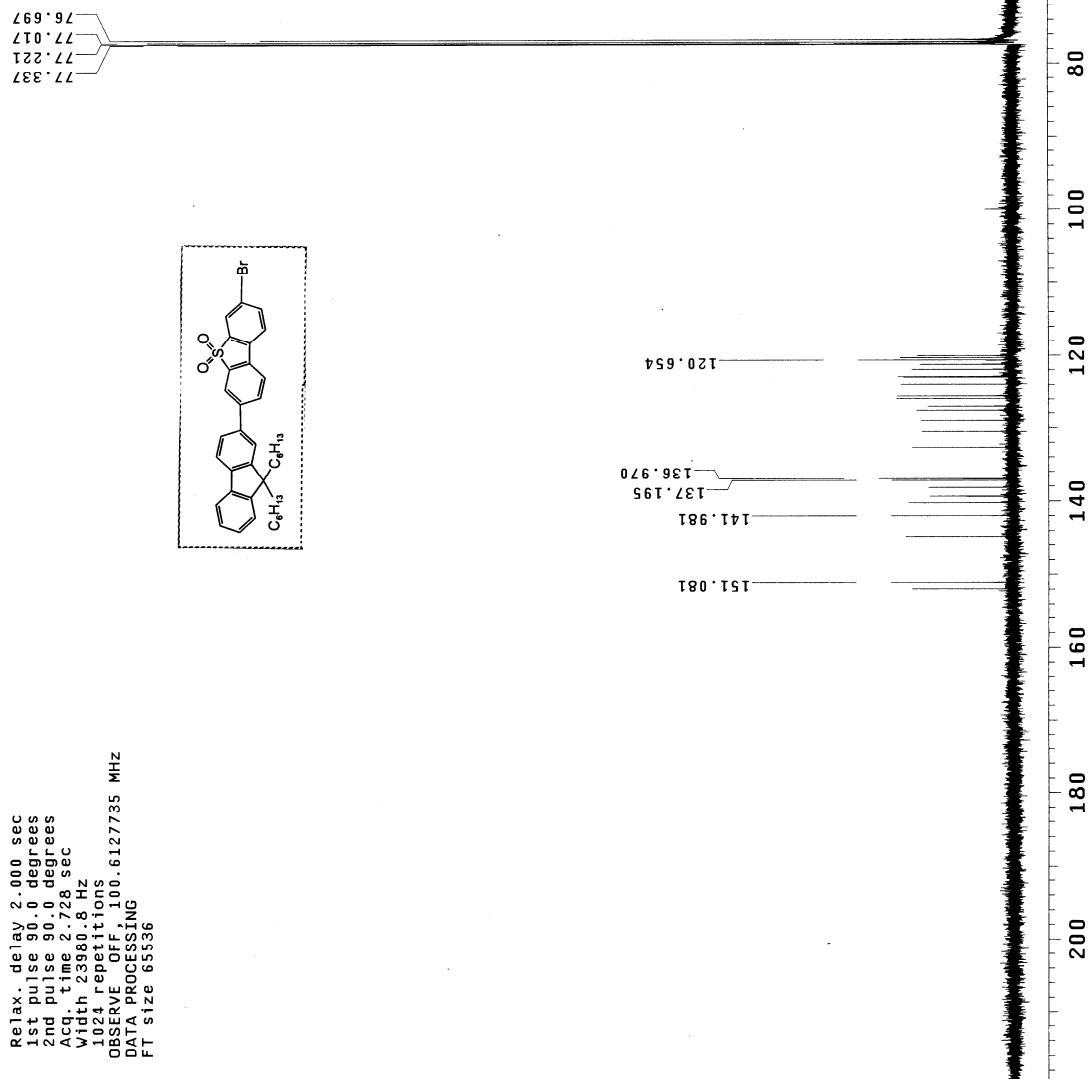
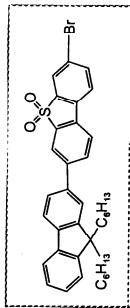
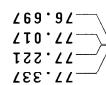


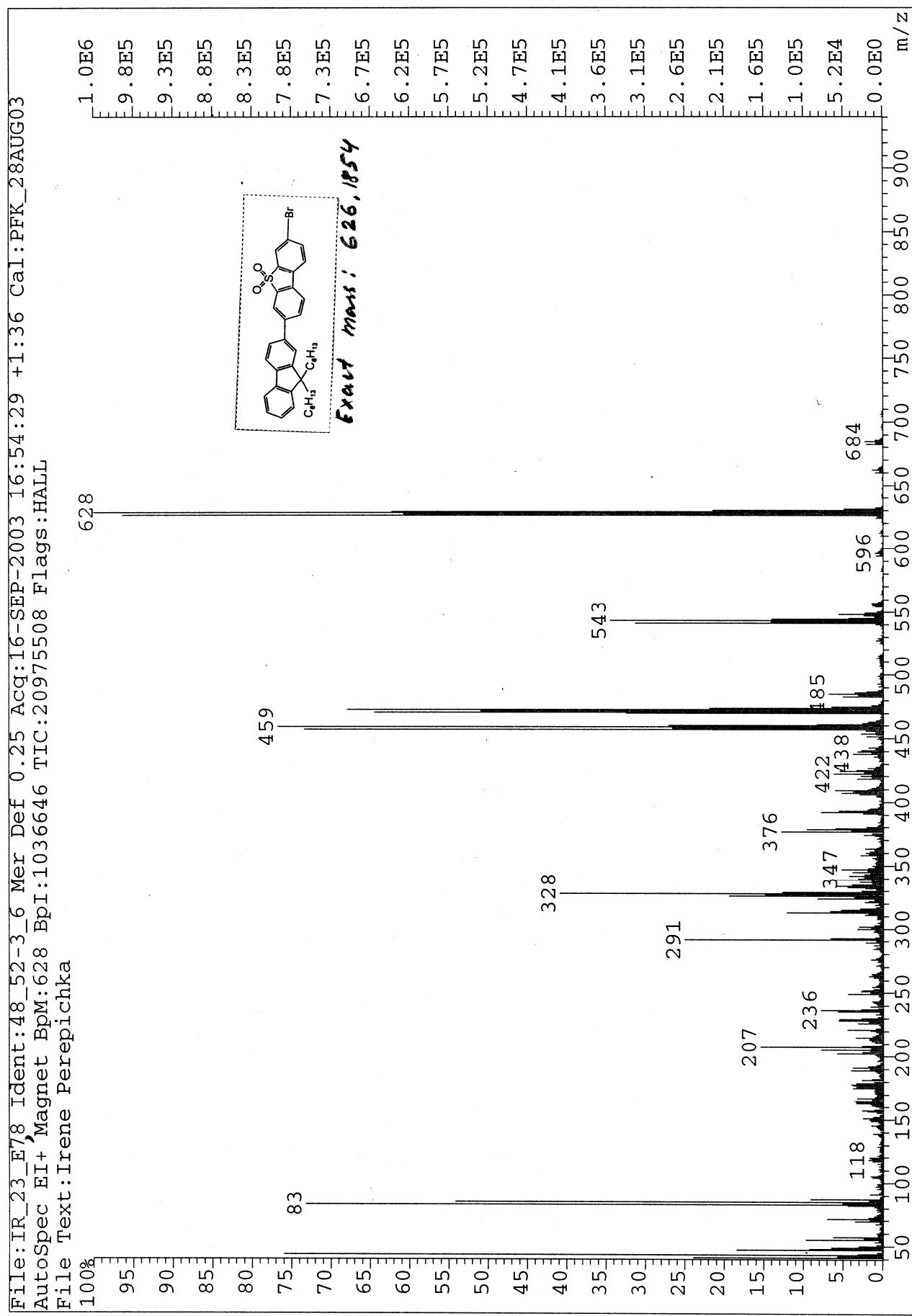


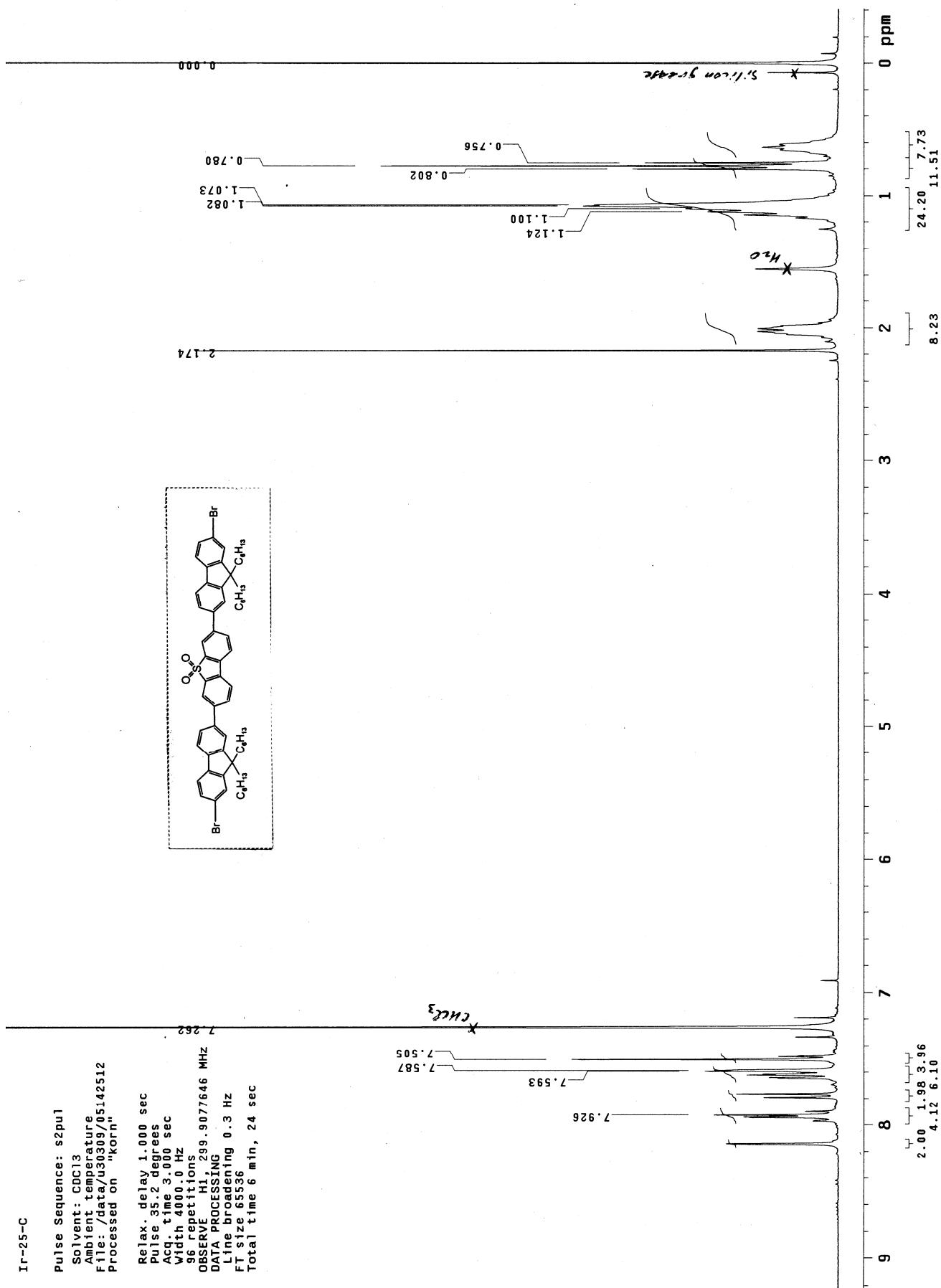


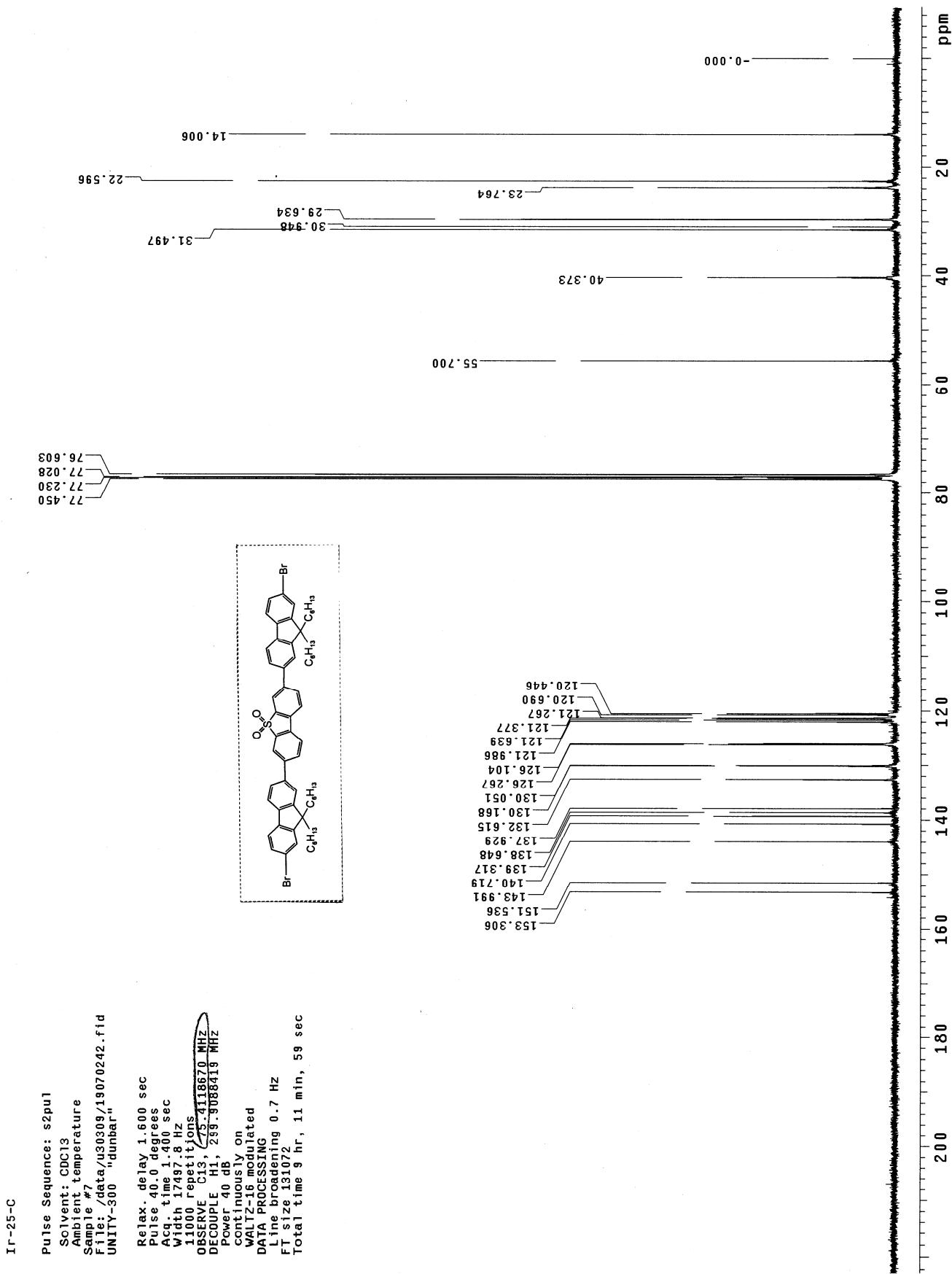
Bruker data converted from file
 /data/data/walkup/nmr/21202922/10
 IR-23-E7,8

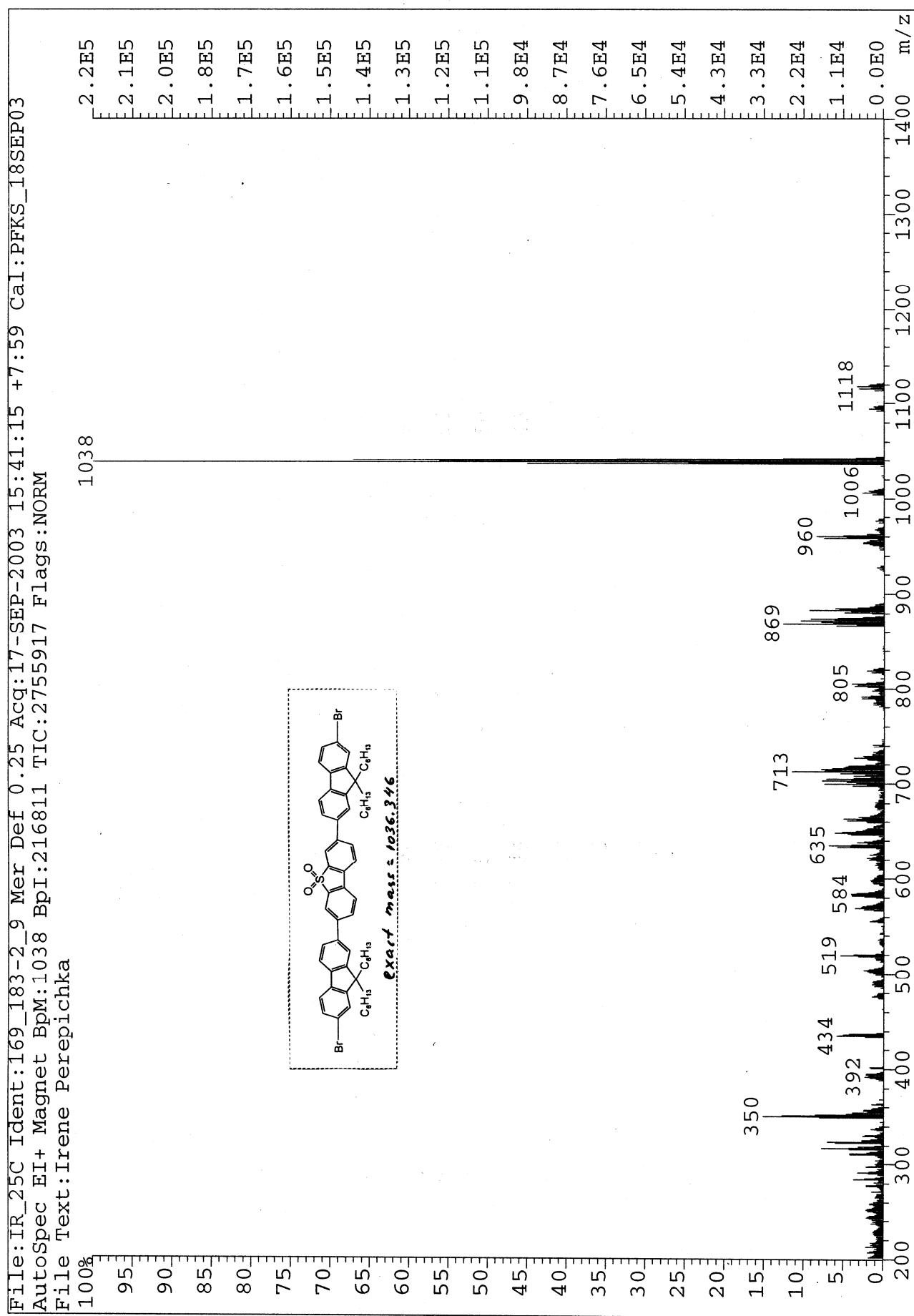
Pulse Sequence: zgpg30
 Solvent: cdd13
 Temp: 24.2 C / 297.3 K
 File: 10_cv
 Processed on "korn"
 Relax. delay 2.000 sec
 1st pulse 90.0 degrees
 2nd pulse 90.0 degrees
 Acq. time 2.728 sec
 width 23980.8 Hz
 1024 repetitions
 OBSERVE OFF 100.6127735 MHz
 DATA PROCESSING
 FT size 65536

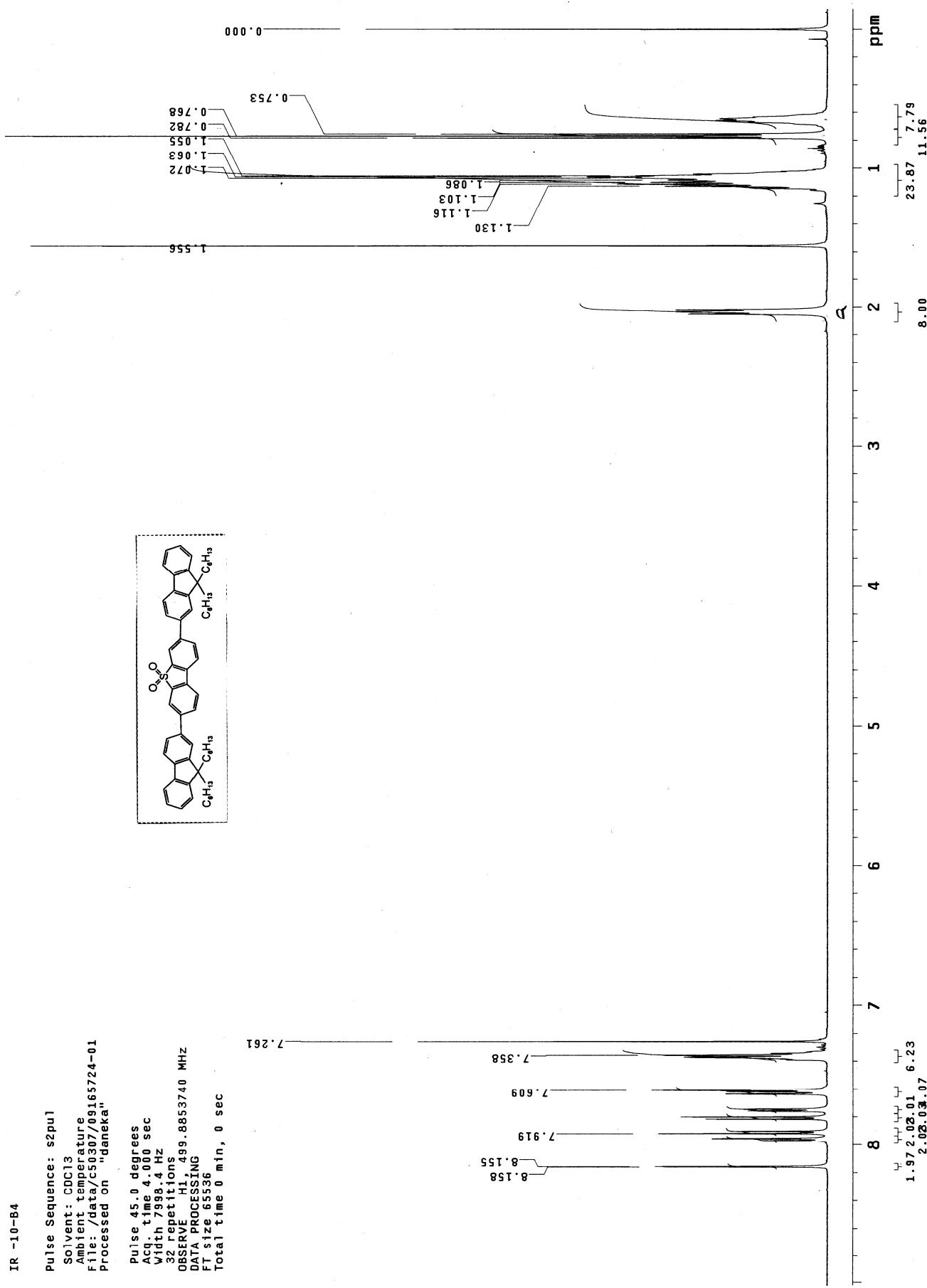








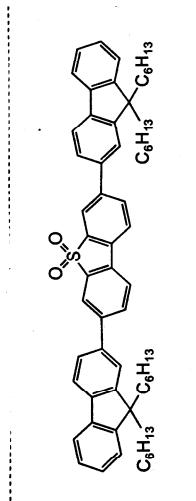
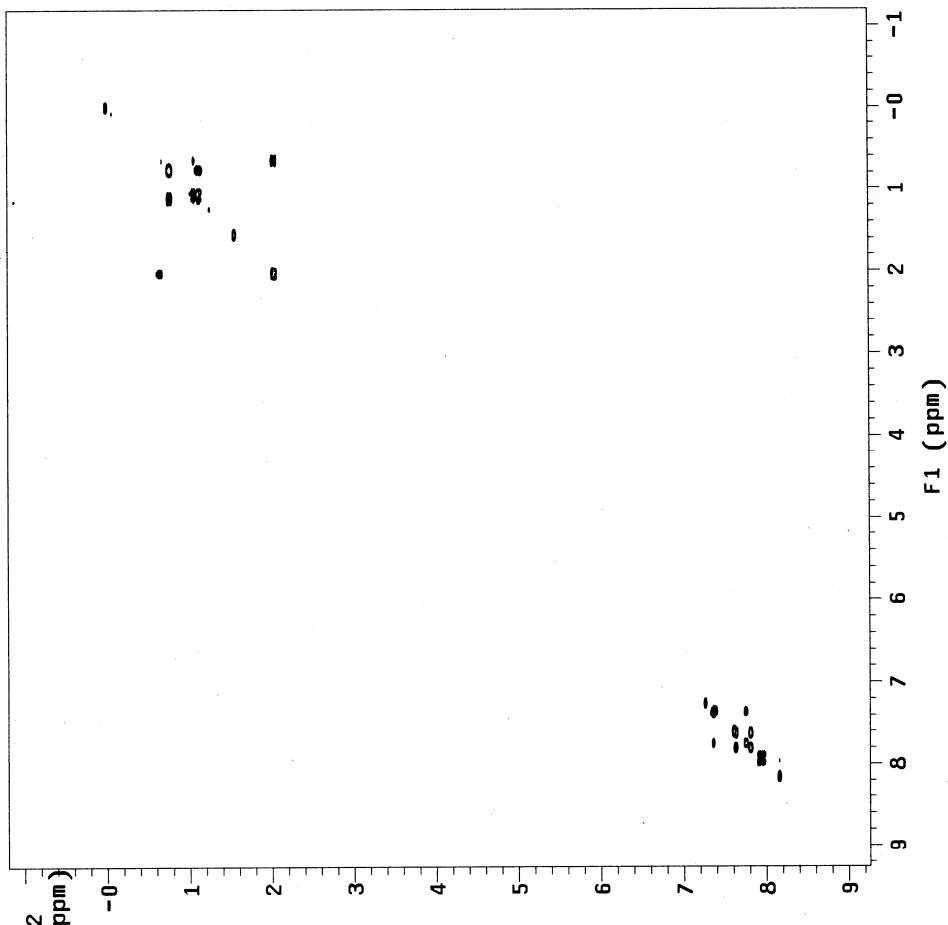


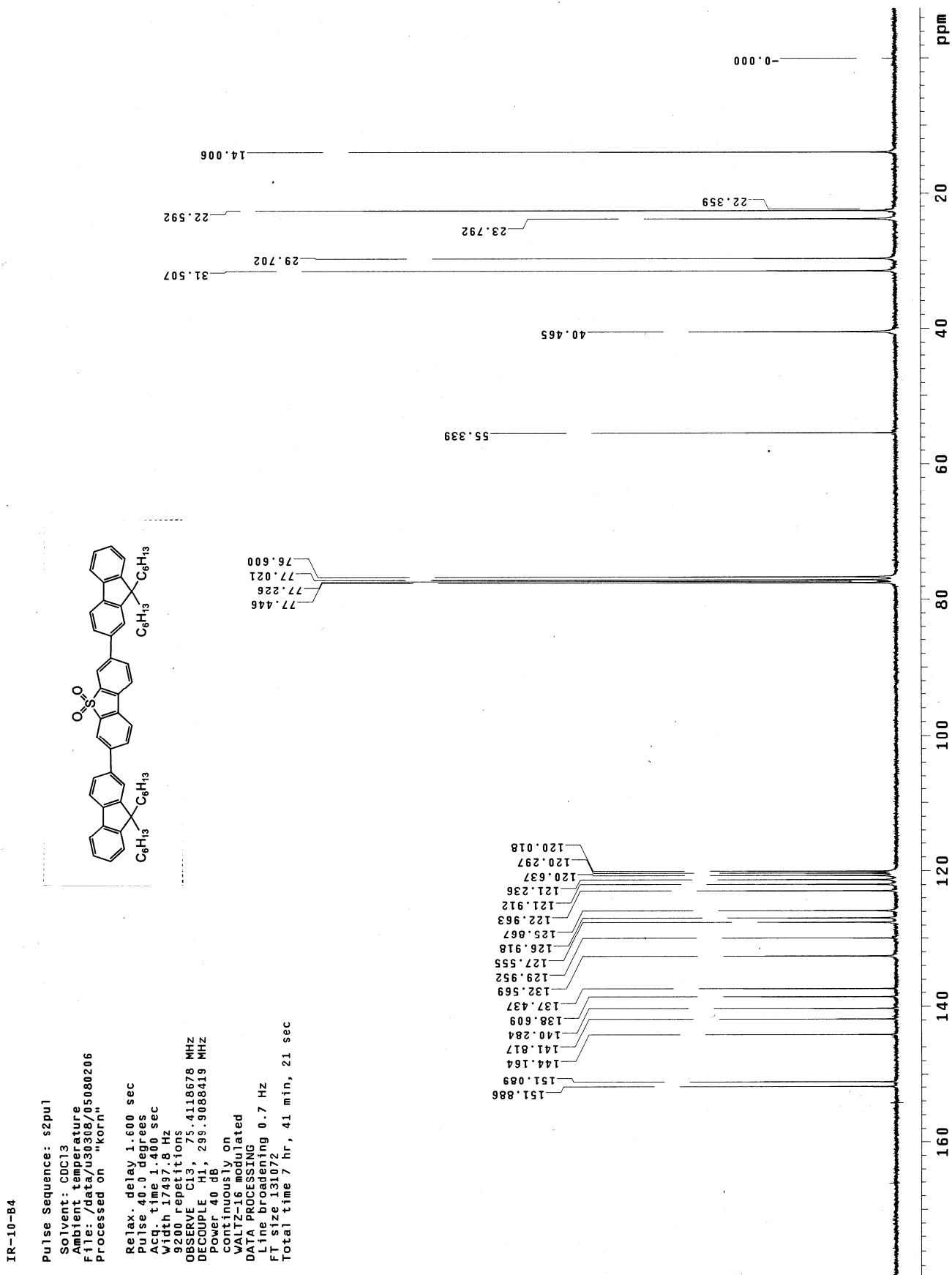


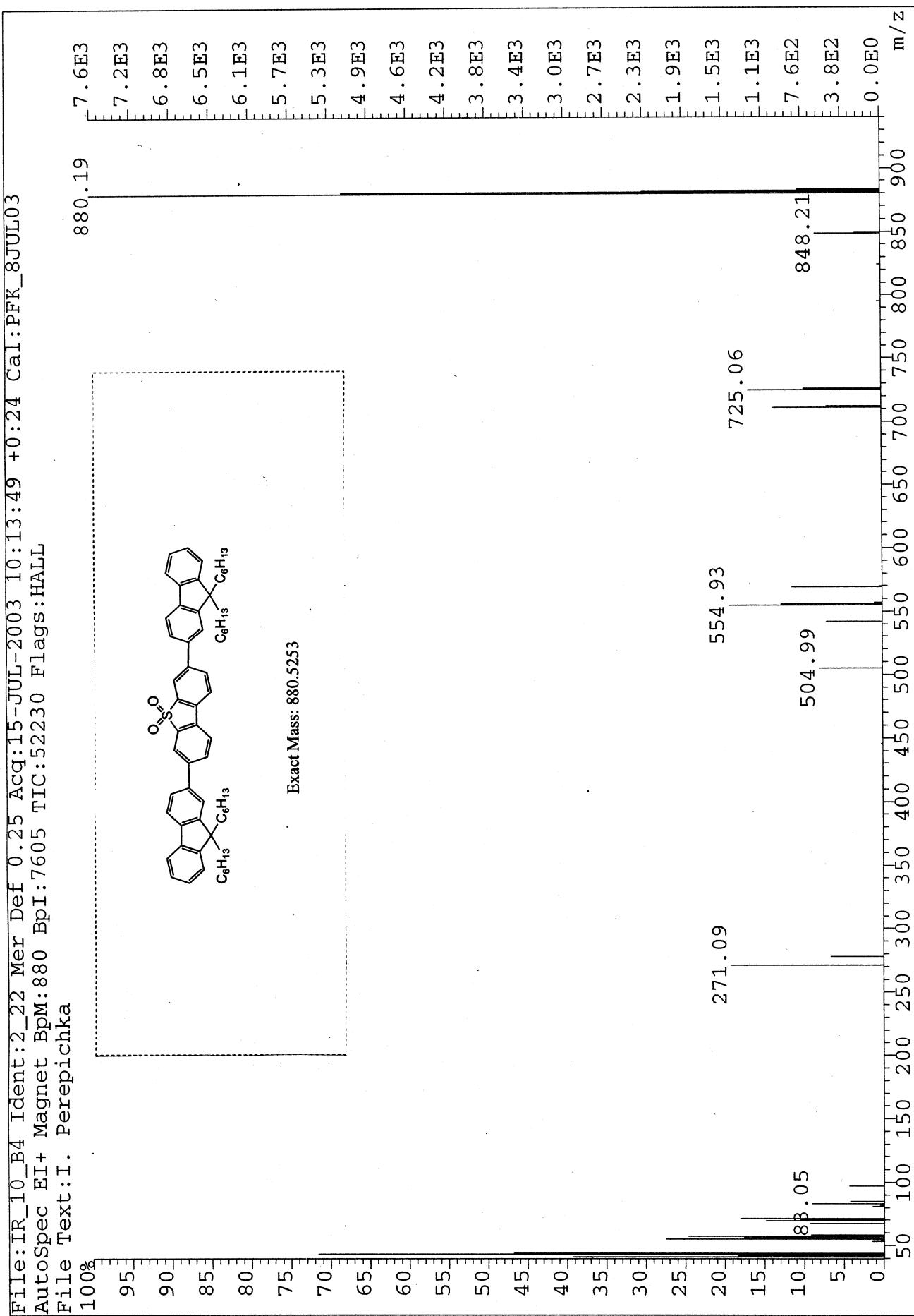
IR -10-B4
 Automation directory: /export/home/vnmr1/automation/wed0903_auto
 File : /data/c50307/09165724-03

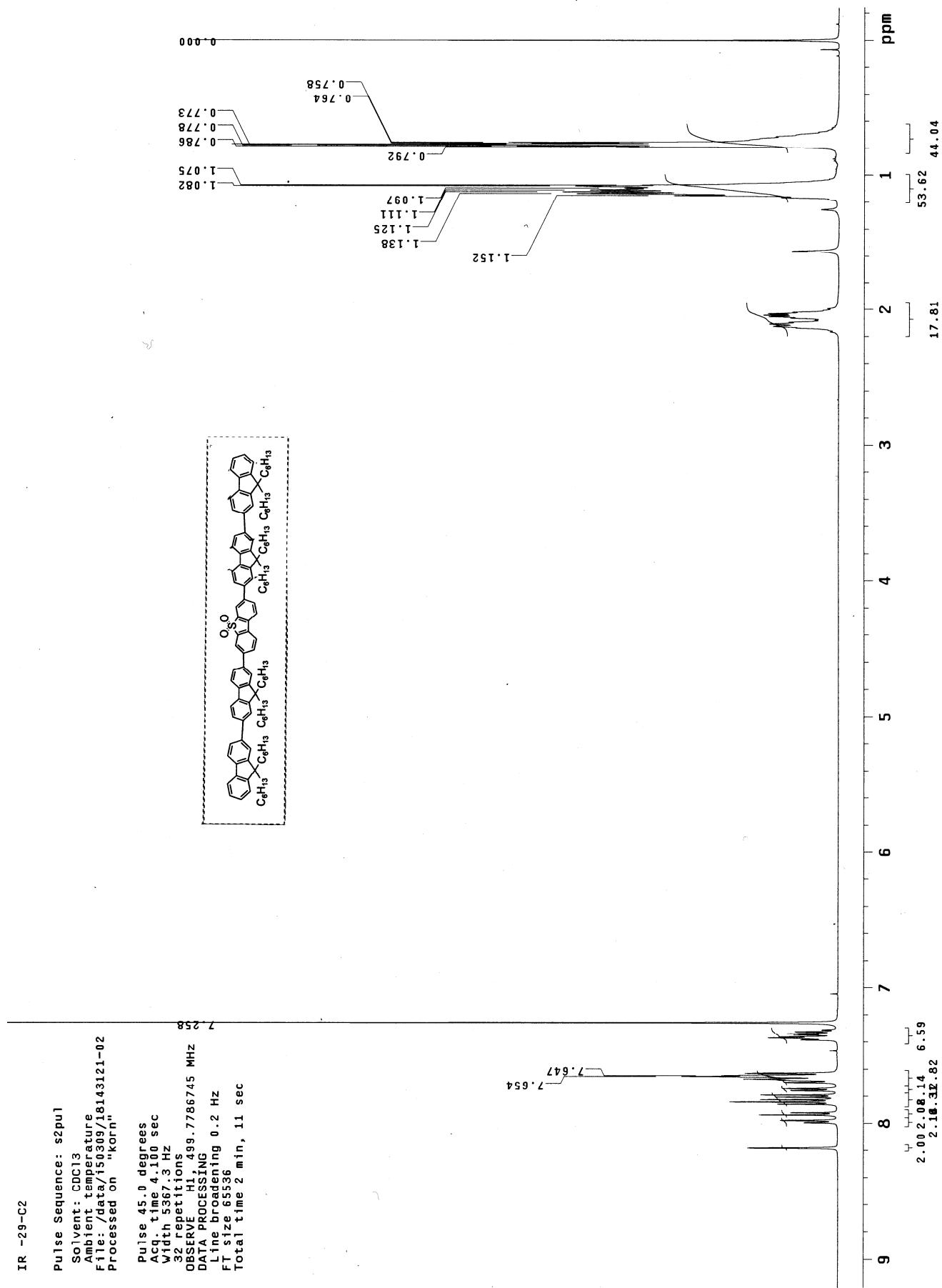
Pulse Sequence: gCOSY
 Solvent: CDCl₃
 Ambient Temperature
 Sample #:1
 File: /data/c50307/09165724-03
 INOVA "Off"

Relax. delay 1.000 sec
 Acq. time 0.196 sec
 Width 5222.6 Hz
 2D width 5222.6 Hz
 2 repetitions
 256 increments
 OBSERVE H1 499.8853740 MHz
 DATA PROCESSING
 Sq. sine bell 0.098 sec
 F1 DATA PROCESSING
 Sq. sine bell 0.025 sec
 FT size 2048 x 2048
 Total time 10 min, 50 sec









IR -29-C2
Automation directory: /export/home/vnmr1/automation/Tue1602_auto

File : /data/150309/18143121-03
Pulse Sequence: gcosy

Solvent: CDCl₃

Ambient temperature

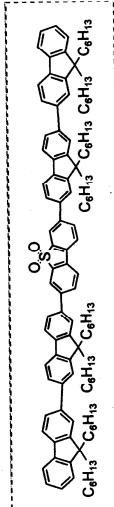
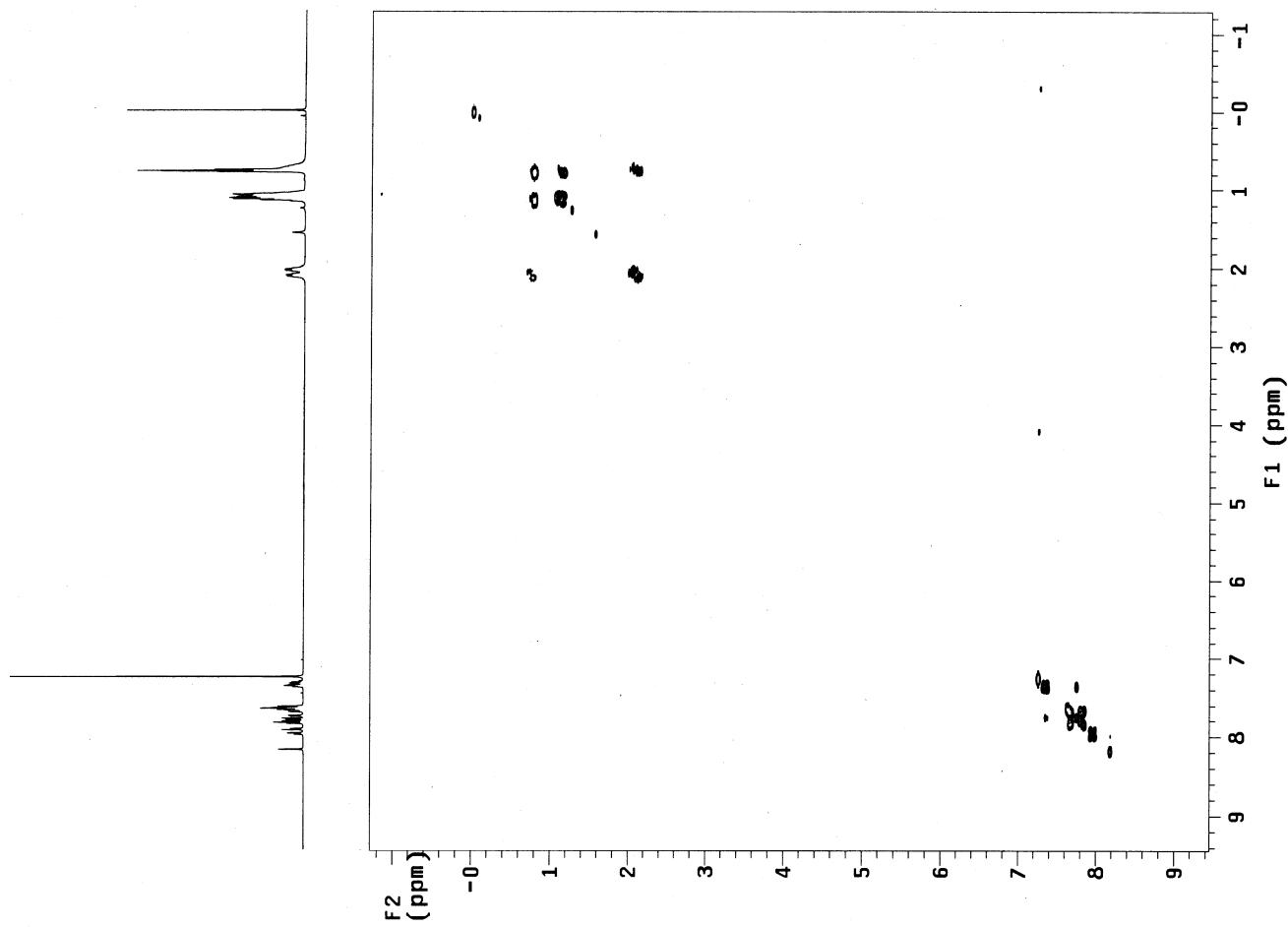
Sample #2

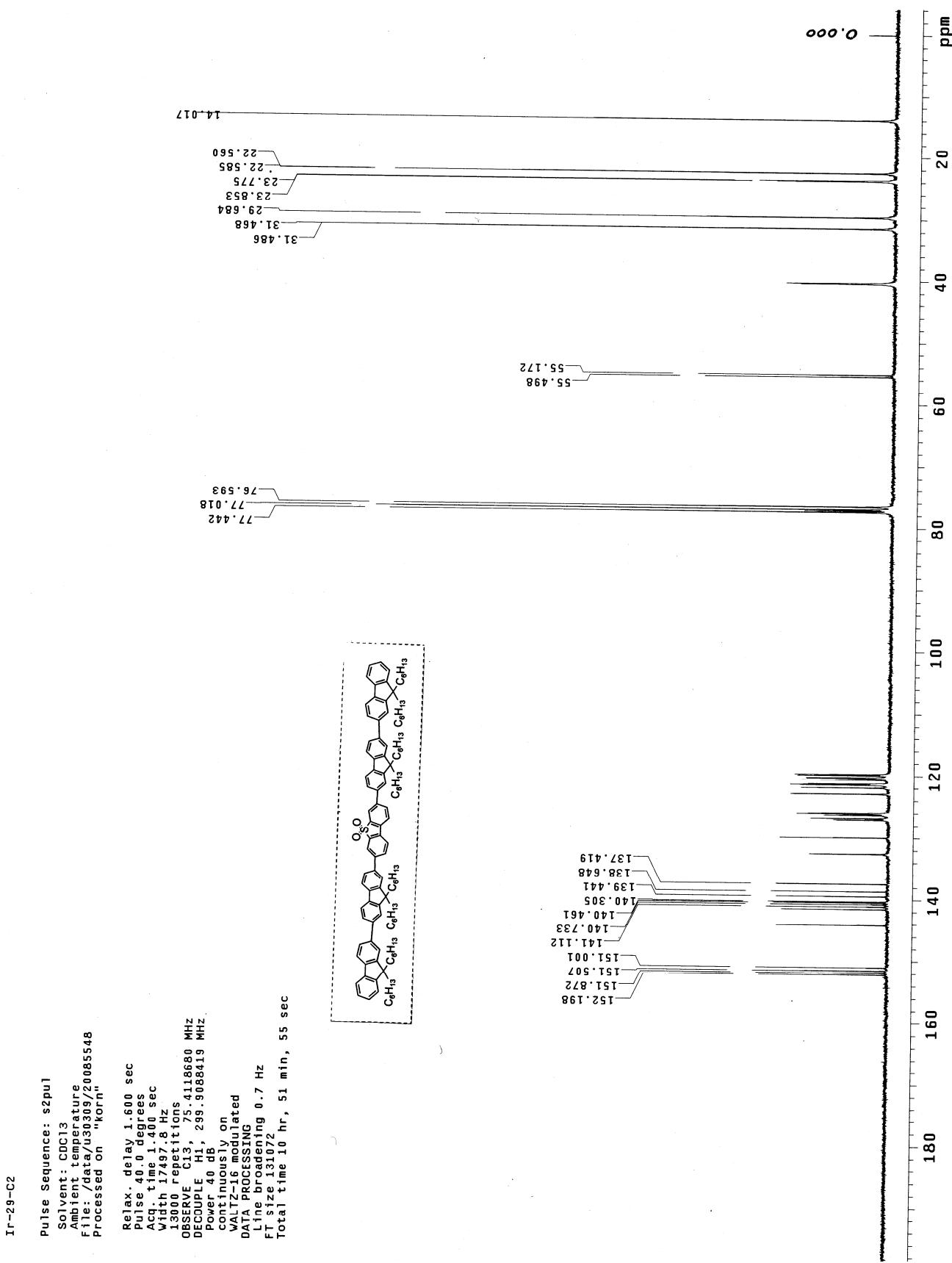
File: /data/150309/18143121-03
INOVA-500 "black"

```

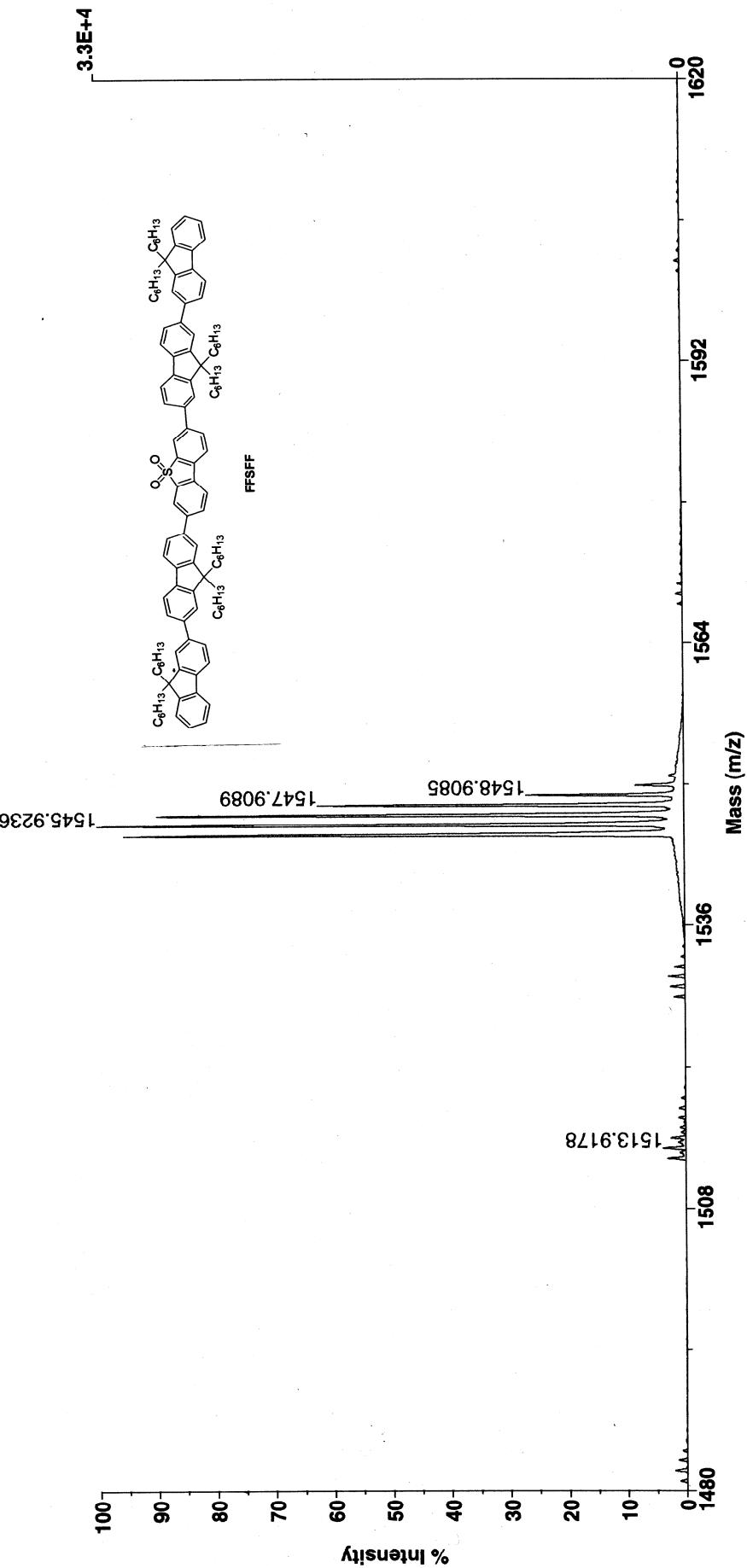
Relaxation delay 1.000 sec
Acq. time 0.191 sec
Width 5367.3 Hz
2D width 5367.3 Hz
2 repetitions
256 increments
OBSERVE H1, 499.7786746 MHz
DATA PROCESSING
S1, sine bell 0.095 sec
F1 DATA PROCESSING
S1, sine bell 0.024 sec
FT size 2048 x 2048
Total time 10 min, 47 sec

```

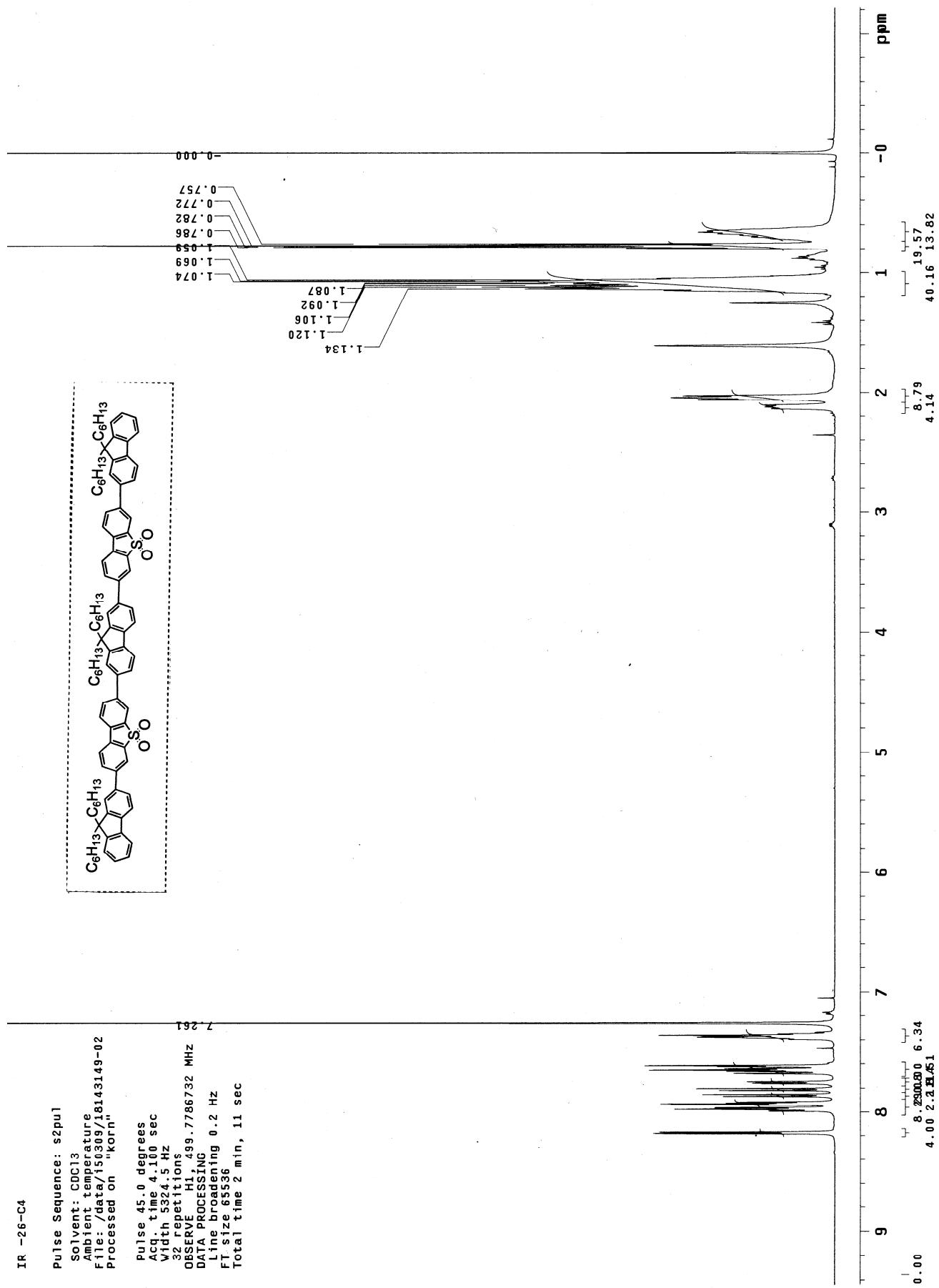




Voyager Spec #1=>NF0.7=>SM5=>AdvBC(32,0.5,0.1)=>MC[BP = 1545.9, 32695]



IR_29_C3.4 (THF), dithranol matrix (THF), layered, reflector mode
C:\...\IR_29_C3.4_0001.dat
Acquired: 15:20:00, October 01, 2003



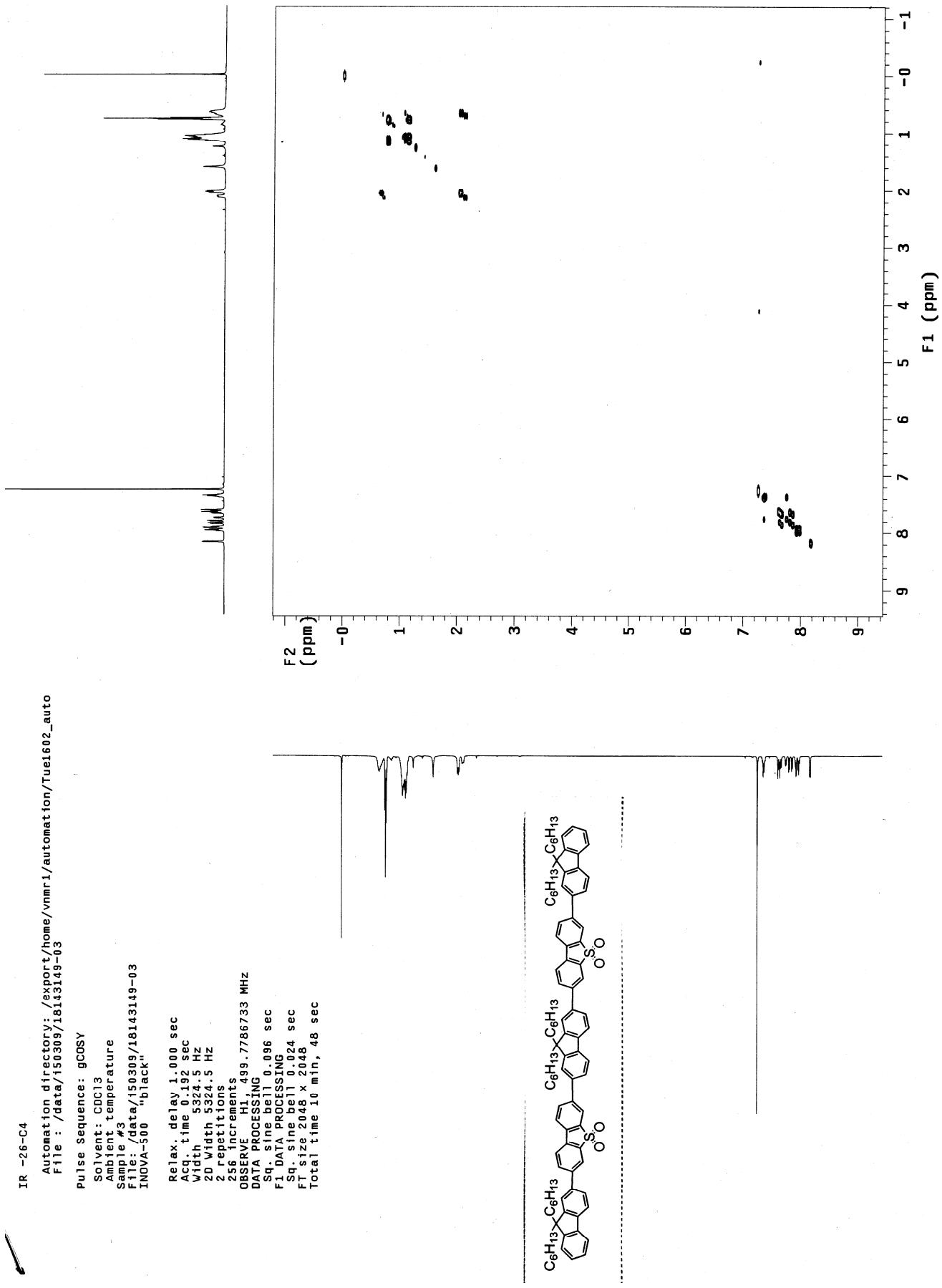
IR -26-C4
Automation directory: /export/home/vmrr1/automation/Tue1602_auto
File : /data/150309/184349-13

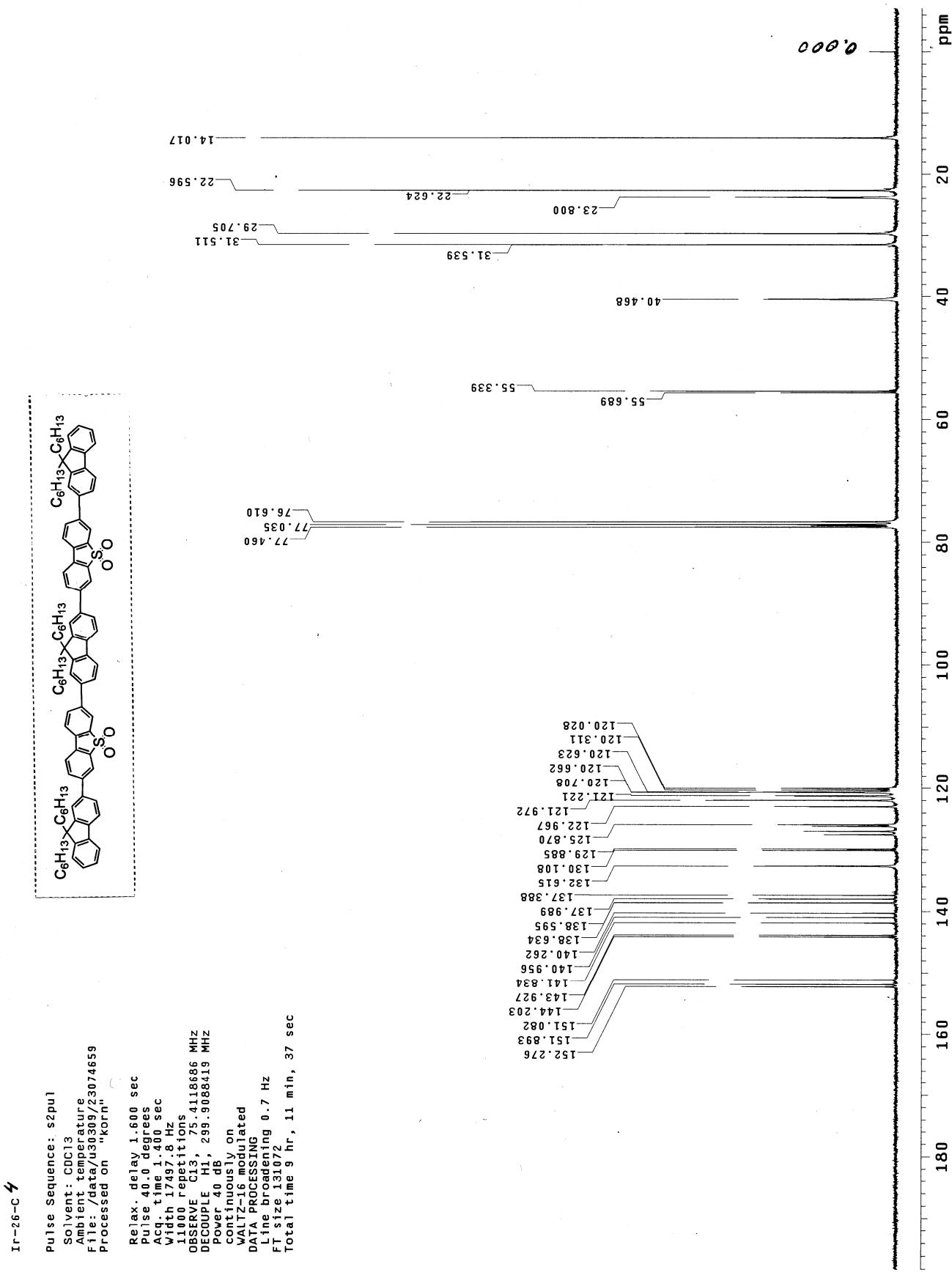
Pulse Sequence: gCOSY
Solvent: CDCl₃
Ambient temperature
Sample #3
File: /data/150309/1843149-03
INOVA-500 "black"

```

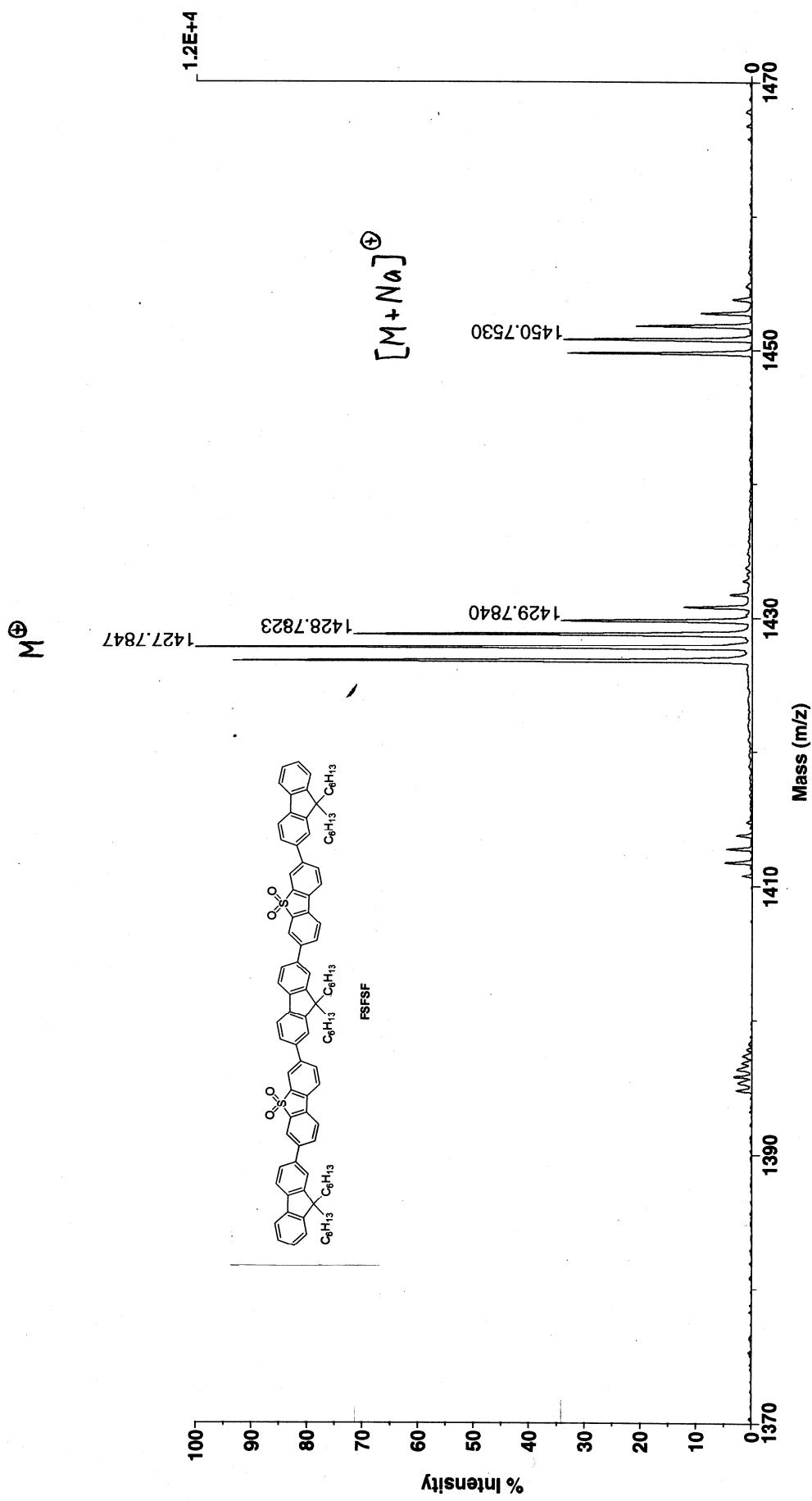
Relax. delay 1.000 sec
Acq. time 1.112 sec
Width 5324.5 Hz
2D Width 5324.5 Hz
    repetitions
    256 increments
OBSERVE H1 499.778573
DATA PROCESSING
SQ., sine bell 0.096 sec
FI, DATA PROCESSING
SQ., sine bell 0.024 sec
FT, size 208 x 2048
Total time 10 min, 48 sec

```

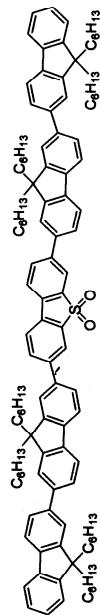
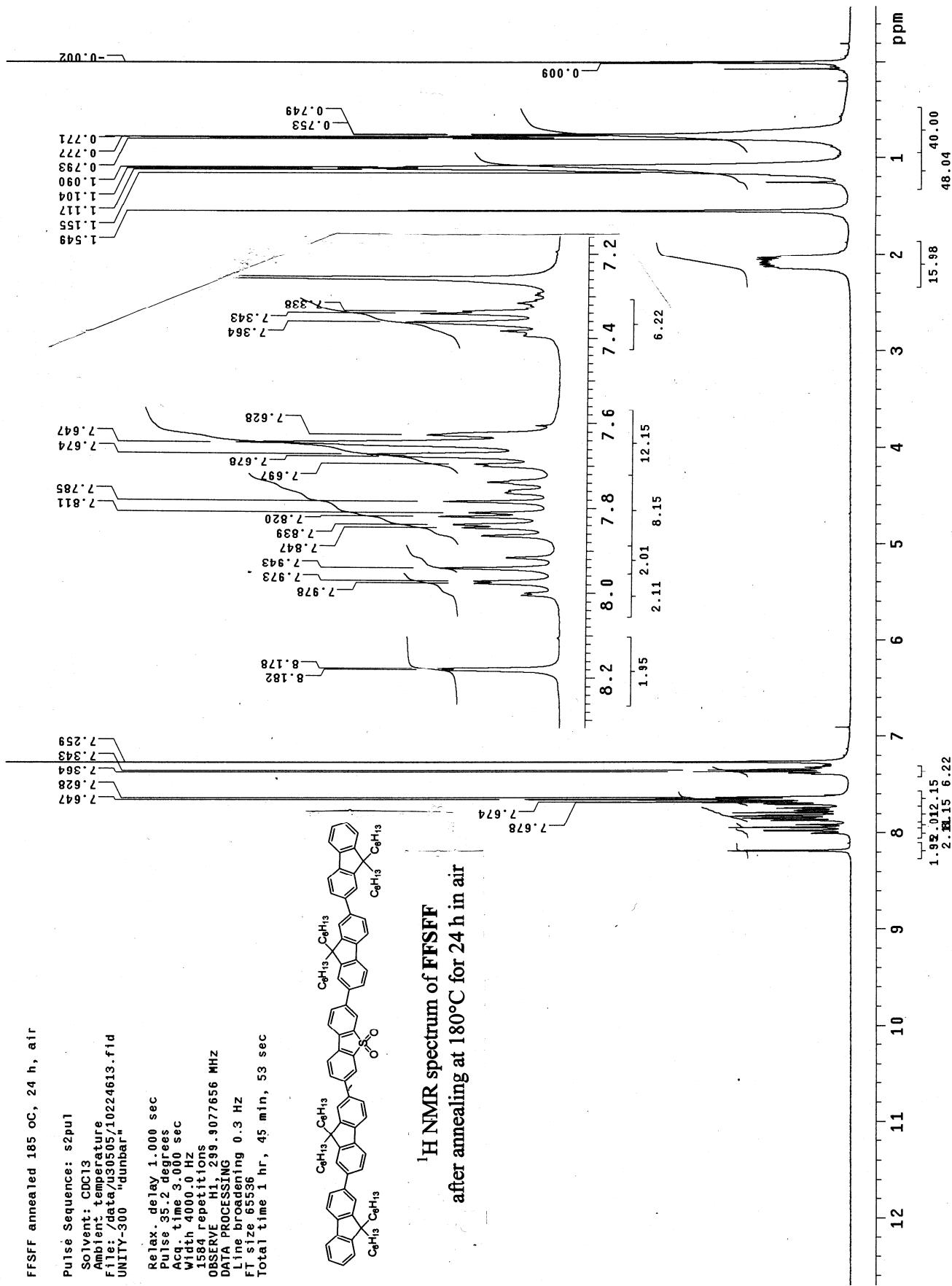




Voyager Spec #1=>NF0.7=>SM5=>AdvBC(32,0.5,0.1)=>MC[BP = 1427.8, 12155]



IR_26_C4 (THF), dithranol matrix (THF), layered, reflector mode
C:\VIR\26_C4_0001.dat
Acquired: 15:15:00, October 01, 2003



¹H NMR spectrum of FFSSFF after annealing at 180°C for 24 h in air