

Figure 1-S. Calculated Powder Patterns of 1 (red = monoclinic, blue = orthorhombic) using 120K data

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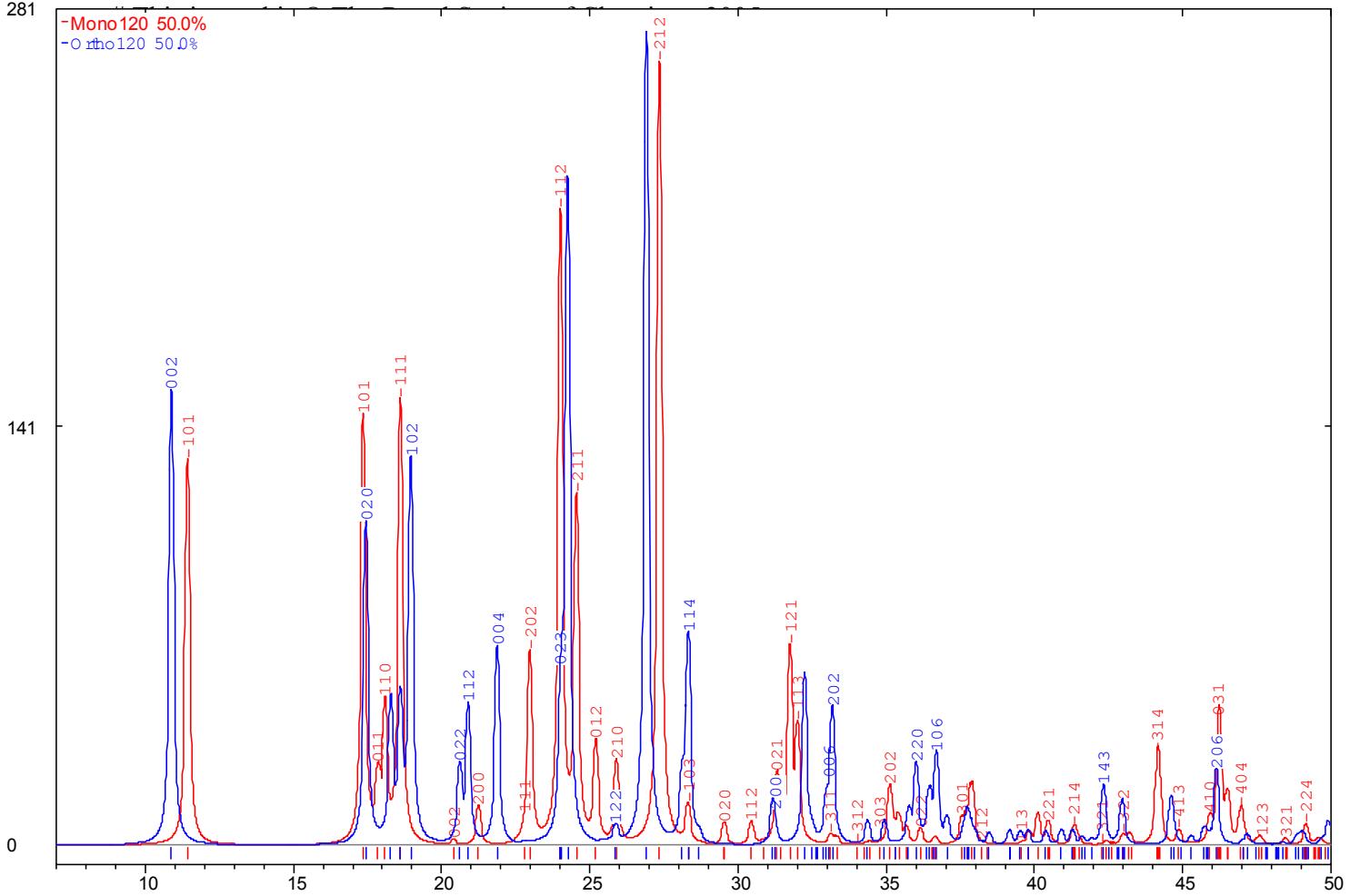


Figure 2-S. Calculated Powder Patterns of 1 (red = monoclinic, blue = orthorhombic) with indices, using 120K data

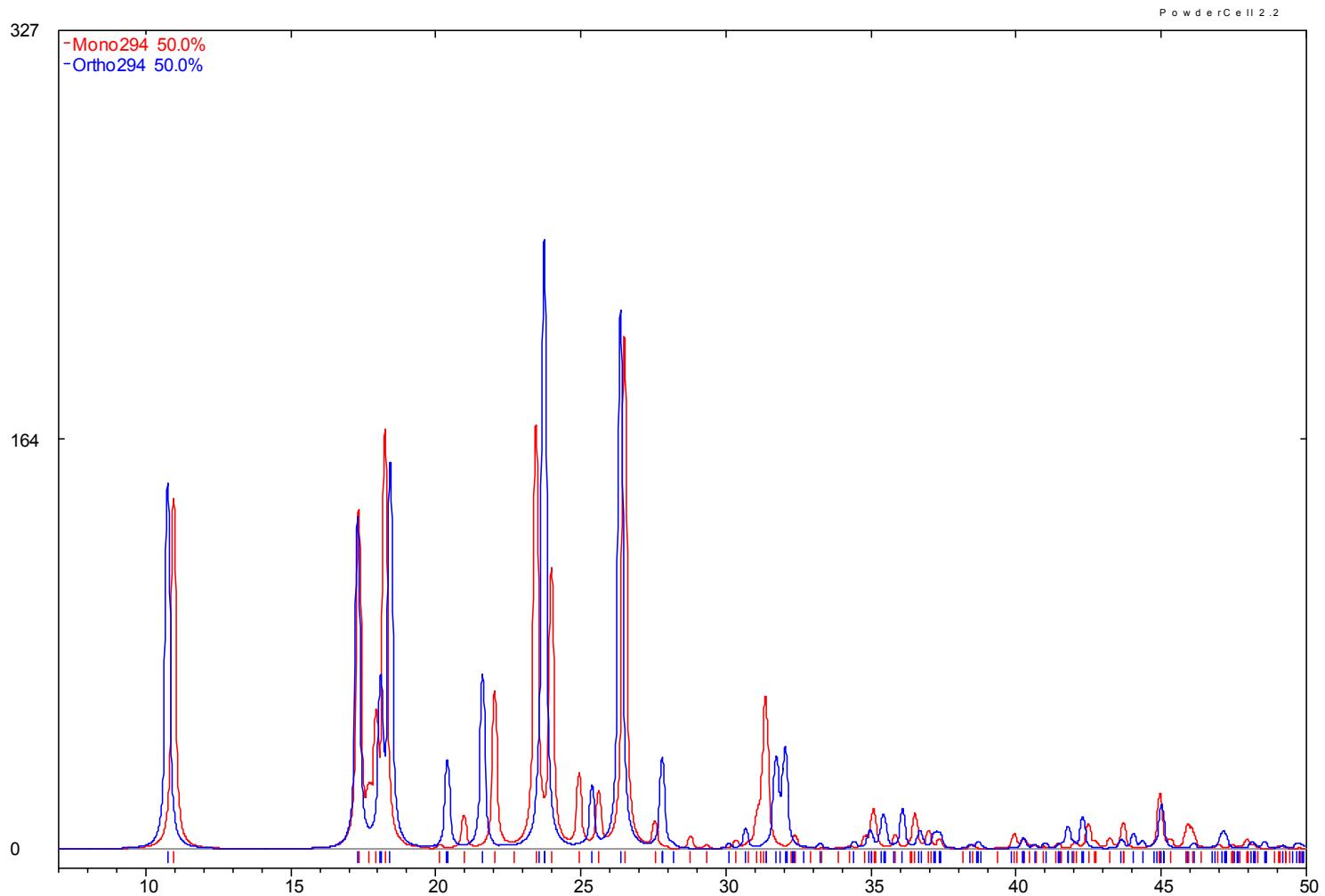


Figure 3-S. Calculated Powder Patterns of 1 (red = monoclinic, blue = orthorhombic), using 294K data

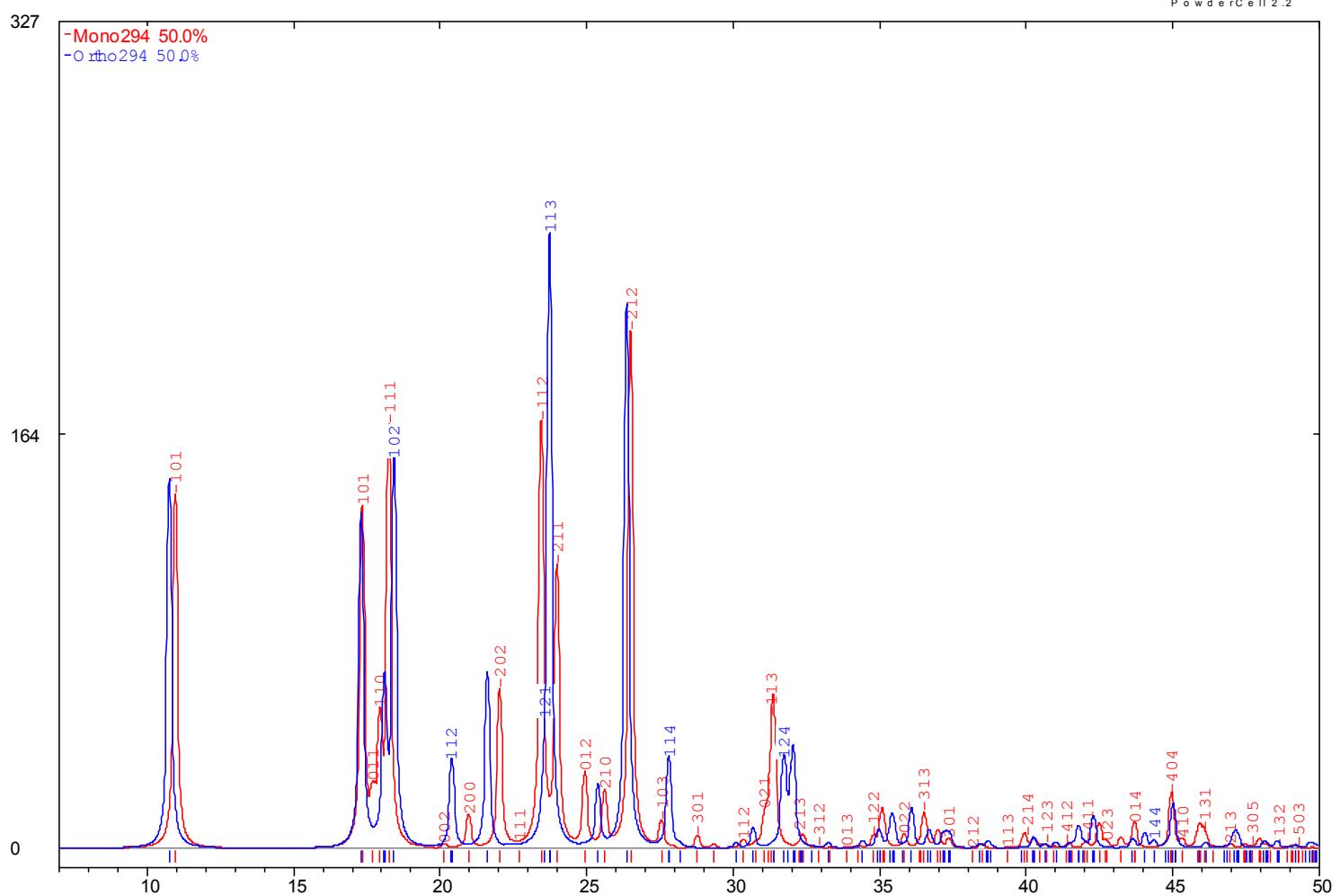
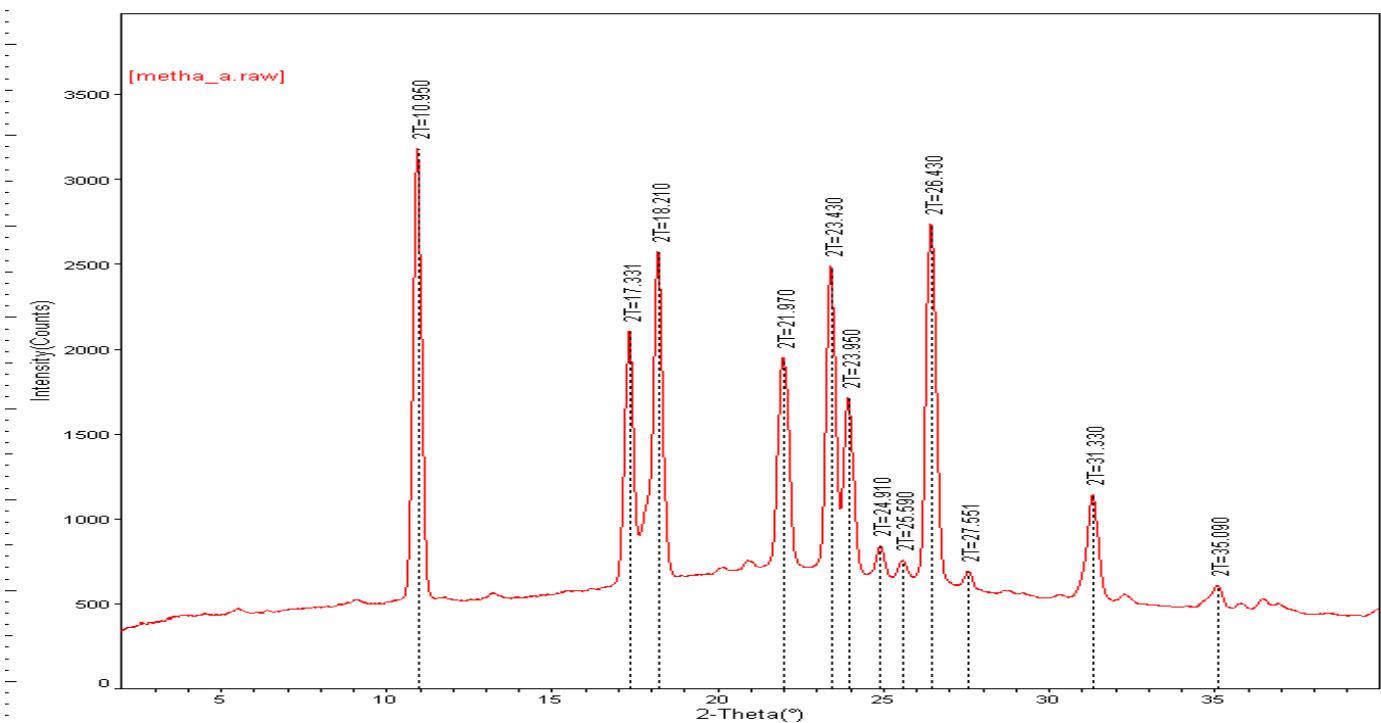
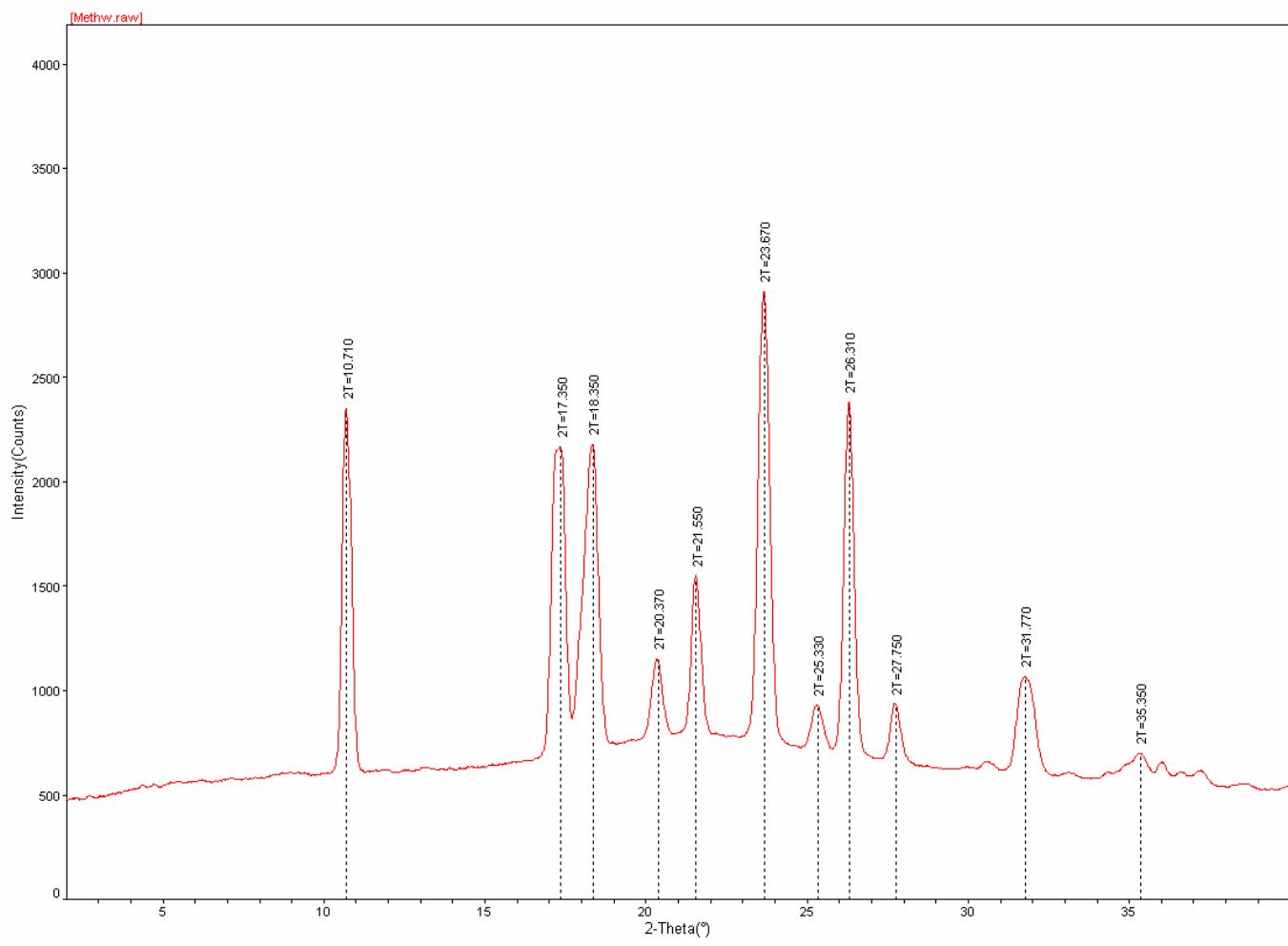


Figure 4-S. Calculated Powder Patterns of 1 (red = monoclinic, blue = orthorhombic) with indices, using 294K data



**Figure S-5. Experimental Powder Diffraction Pattern of Form I**



Materials Data, Inc.

Tuesday, Jul 20, 2004 04:56p (MDI/JADE5)

**Figure S-6. Experimental Powder Diffraction Pattern of Form II**

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For the structure of 2-carbamoyl-1-phenyl thiolato-ethenyl acetate, (R. L. Beddoes, D. MacLeod, D. Moorcraft, P. Quayle, Y. Zhao and G. M. Davies, *Tetrahedron Lett.*, 1992, **33**, 417-420, CSD refcode KOVROC), the data in CSD strongly suggest an incorrect structure, with a calculated C–NH<sub>2</sub> distance of 1.197 Å and C=O distance of 1.355 Å. Switching the identity of the N and O atoms, and moving the misplaced H atoms to the “new” N atom produces a chemically appropriate structure, with an *s-cis* conformation, and a reasonable hydrogen bonding pattern compared to that derived from the published coordinates.

PLATON reports the following (not very sensible) hydrogen bond patterns for KOVROC:

---

Donor	---	H....Acceptor	[	ARU	]	D - H	H...A	D...A	D - H...A
N(1)	--H(2)	.. ?				0.94			
N(1)	--H(3)	..S(1)	[		]	0.95	2.03	2.8340	142
N(1)	--H(3)	..O(2)	[	8554.01	]	0.95	2.36	2.8238	110
C(11)	--H(11)	..O(1)	[		]	0.95	2.56	2.9105	102
Translation of ARU-code to Equivalent Position Code									
[ 8554. ] = x,1/2-y,-1/2+z									

---

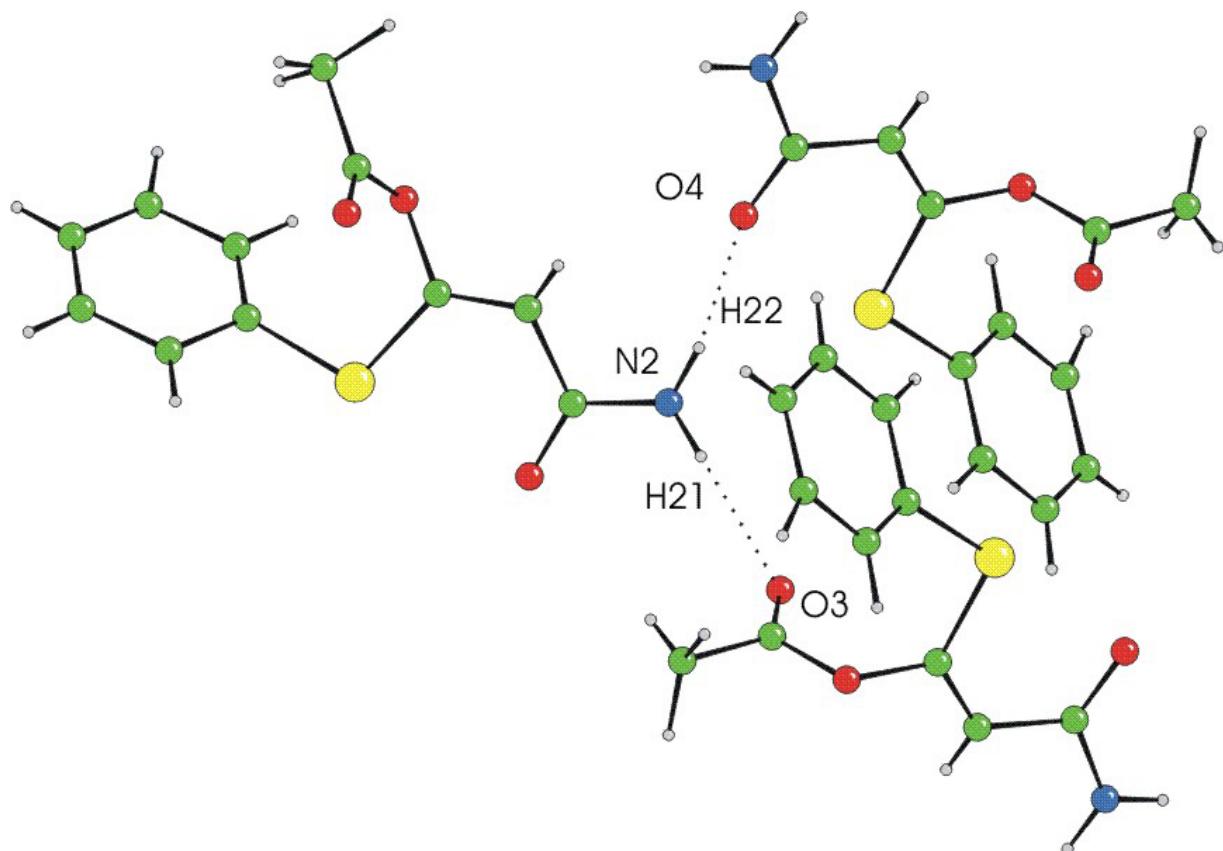
If the amide N and O atoms are simply switched, the H atoms on the “old” N atom removed, and H atoms placed in calculated positions on the “new” N atom, PLATON reports a quite reasonable H-bonding pattern:

---

Donor	---	H....Acceptor	[	ARU	]	D - H	H...A	D...A	D - H...A
N(2)	--H(21)	..O(3)	[	4655.01	]	0.88	2.18	3.0607	176
N(2)	--H(22)	..O(4)	[	8555.01	]	0.88	1.96	2.8238	168
C(11)	--H(11)	..O(1)	[		]	0.95	2.56	2.9105	102
Translation of ARU-code to Equivalent Position Code									
[ 4655. ] = 1-x,1/2+y,1/2-z									
[ 8555. ] = x,1/2-y,1/2+z									

---

The hydrogen bonds for the revised assignment are shown in the Figure on the following page:



**Figure S-7. Hydrogen bonds for the revised (*s-trans*) structure of KOVROC**

A CIF file for the revised structure follows:

```
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_audit_creation_date          04-12-15
_audit_creation_method CRYSTALS_ver_12.51

_oxford_structure_analysis_title  'CSD_CIF_KOVROC_revised'
_chemical_name_systematic        ?
_chemical_melting_point         ?

_cell_length_a                  15.8100(10)
_cell_length_b                  16.2630(10)
_cell_length_c                  9.4360(10)
_cell_angle_alpha                90
_cell_angle_beta                 90
_cell_angle_gamma                90
_cell_volume                     2426.2(3)

_symmetry_cell_setting          'Orthorhombic'
_symmetry_space_group_name_H-M   'P b c a '
_symmetry_space_group_name_Hall   '?'
loop_
_symmetry_equiv_pos_as_xyz
'x, y, z'
```

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'-x,-y,-z'  
'-x+1/2,y+1/2,z'  
'x+1/2,-y+1/2,-z'  
'x,-y+1/2,z+1/2'  
'-x,y+1/2,-z+1/2'  
'-x+1/2,-y,z+1/2'  
'x+1/2,y,-z+1/2'  
  
loop_  
_atom_type_symbol  
_atom_type_scat_dispersion_real  
_atom_type_scat_dispersion_imag  
_atom_type_scat_Cromer_Mann_a1  
_atom_type_scat_Cromer_Mann_b1  
_atom_type_scat_Cromer_Mann_a2  
_atom_type_scat_Cromer_Mann_b2  
_atom_type_scat_Cromer_Mann_a3  
_atom_type_scat_Cromer_Mann_b3  
_atom_type_scat_Cromer_Mann_a4  
_atom_type_scat_Cromer_Mann_b4  
_atom_type_scat_Cromer_Mann_c  
_atom_type_scat_source  
C      0.0033  0.0016  2.3100  20.8439  1.0200  10.2075  1.5886  0.5687  
     0.8650  51.6512  0.2156 'International Tables Vol C 4.2.6.8 and 6.1.1.4'  
H      0.0000  0.0000  0.4930  10.5109  0.3229  26.1257  0.1402  3.1424  
     0.0408  57.7998  0.0030 'International Tables Vol C 4.2.6.8 and 6.1.1.4'  
N      0.0061  0.0033  12.2126  0.0057  3.1322  9.8933  2.0125  28.9975  
     1.1663  0.5826 -11.5290 'International Tables Vol C 4.2.6.8 and 6.1.1.4'  
O      0.0106  0.0060  3.0485  13.2771  2.2868  5.7011  1.5463  0.3239  
     0.8670  32.9089  0.2508 'International Tables Vol C 4.2.6.8 and 6.1.1.4'  
S      0.1246  0.1234  6.9053  1.4679  5.2034  22.2151  1.4379  0.2536  
     1.5863  56.1720  0.8669 'International Tables Vol C 4.2.6.8 and 6.1.1.4'  
  
_cell_formula_units_Z          8  
  
# Given Formula = C11 H11 N1 O3 S1  
# Dc =      1.30 Fooo =    992.00 Mu =      2.58 M =    237.28  
# Found Formula = C11 H11 N1 O3 S1  
# Dc =      1.30 FOOO =    992.00 Mu =      2.58 M =    237.28  
  
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_chemical_formula_moiety       'C11 H11 N1 O3 S1'  
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_cell_measurement_temperature  0  
  
_exptl_crystal_description    'c'  
_exptl_crystal_colour         'colorless'  
_exptl_crystal_size_min       ?  
_exptl_crystal_size_mid       ?  
_exptl_crystal_size_max       ?  
  
_exptl_crystal_density_diffrn 1.299  
_exptl_crystal_density_meas   ?  
# Non-dispersive F(000):  
_exptl_crystal_F_000          992  
_exptl_absorpt_coefficient_mu 0.258
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_exptl_absorpt_correction_T_max    1.0000
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_diffrrn_radiation_monochromator   'graphite'
_diffrrn_radiation_type           'Mo K\alpha'
_diffrrn_radiation_wavelength     0.71073
_diffrrn_measurement_method       \w/2\q

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_computing_data_reduction           'USER DEFINED DATA REDUCTION'
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_computing_structure_solution       'USER DEFINED STRUCTURE SOLUTION'
_computing_structure_refinement    'CRYSTALS (Betteridge et al., 2003)'
_computing_publication_material     'CRYSTALS (Betteridge et al., 2003)'
_computing_molecular_graphics       'CAMERON (Watkin et al., 1996)'

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_diffrrn_standards_interval_count  ?
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_diffrrn_standards_decay_%         ?

_diffrrn_ambient_temperature       0
_diffrrn_reflns_number             0
_reflns_number_total               0
_diffrrn_reflns_av_R_equivalents  0.000
# Number of reflections with Friedels Law is 0
# Number of reflections without Friedels Law is 0

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_diffrrn_reflns_theta_max          0.000
_diffrrn_measured_fraction_theta_max NaN

_diffrrn_reflns_theta_full         0.000
_diffrrn_measured_fraction_theta_full 1.000

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_diffrrn_reflns_limit_h_max***** 
_diffrrn_reflns_limit_k_min***** 
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_diffrrn_reflns_limit_l_max***** 

_oxford_diffrrn_Wilson_B_factor    0.00
_oxford_diffrrn_Wilson_scale       0.00

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# _atom_sites_solution_secondary   difmap
_atom_sites_solution_hydrogens     geom

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_refine_diff_density_max           ?

_refine_ls_number_reflns           0
_refine_ls_number_restraints       0
_refine_ls_number_parameters       0

#_refine_ls_R_factor_ref 0.0006
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_refine_ls_goodness_of_fit_ref 0.0000

#_reflns_number_all 0
_refine_ls_R_factor_all      0.0000
_refine_ls_wR_factor_all     0.0000

# The I/u(I) cutoff below was used for refinement as
# well as the _gt R-factors:
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_reflns_number_gt              0
_refine_ls_R_factor_gt        0.0000
_refine_ls_wR_factor_gt       0.0000

_refine_ls_shift/su_max       0.000000

# choose from: rm (reference molecule of known chirality),
# ad (anomalous dispersion - Flack), rmad (rm and ad),
# syn (from synthesis), unk (unknown) or . (not applicable).
_chemical_absolute_configuration  '.'

_refine_ls_structure_factor_coef F
_refine_ls_matrix_type         full
_refine_ls_hydrogen_treatment  noref          # none, undef, noref, refall,
                                         # refxyz, refU, constr or mixed

# WARNING. The IUCr will not accept Unit Weights
_refine_ls_weighting_scheme    calc
_refine_ls_weighting_details
;
Method= Quasi-Unit weights
W = 1.0 or 1./2F
;

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_atom_site_fract_x
_atom_site_fract_y
_atom_site_fract_z
_atom_site_U_iso_or_equiv
_atom_site_occularity
_atom_site_adp_type
_atom_site_refinement_flags_posn
_atom_site_refinement_flags_adp
_atom_site_refinement_flags_occularity
_atom_site_disorder_assembly
_atom_site_disorder_group
_atom_site_attached_hydrogens
S1 S 0.3722 0.0756 0.0853 0.0500 1.0000 Uiso . . . . .
O1 O 0.3686 0.0244 0.3559 0.0500 1.0000 Uiso . . . . .
N2 N 0.4666 0.2960 0.3246 0.0500 1.0000 Uiso . . . . .
O3 O 0.4923 -0.0399 0.3185 0.0500 1.0000 Uiso . . . . .
O4 O 0.4447 0.2339 0.1171 0.0500 1.0000 Uiso . . . . .
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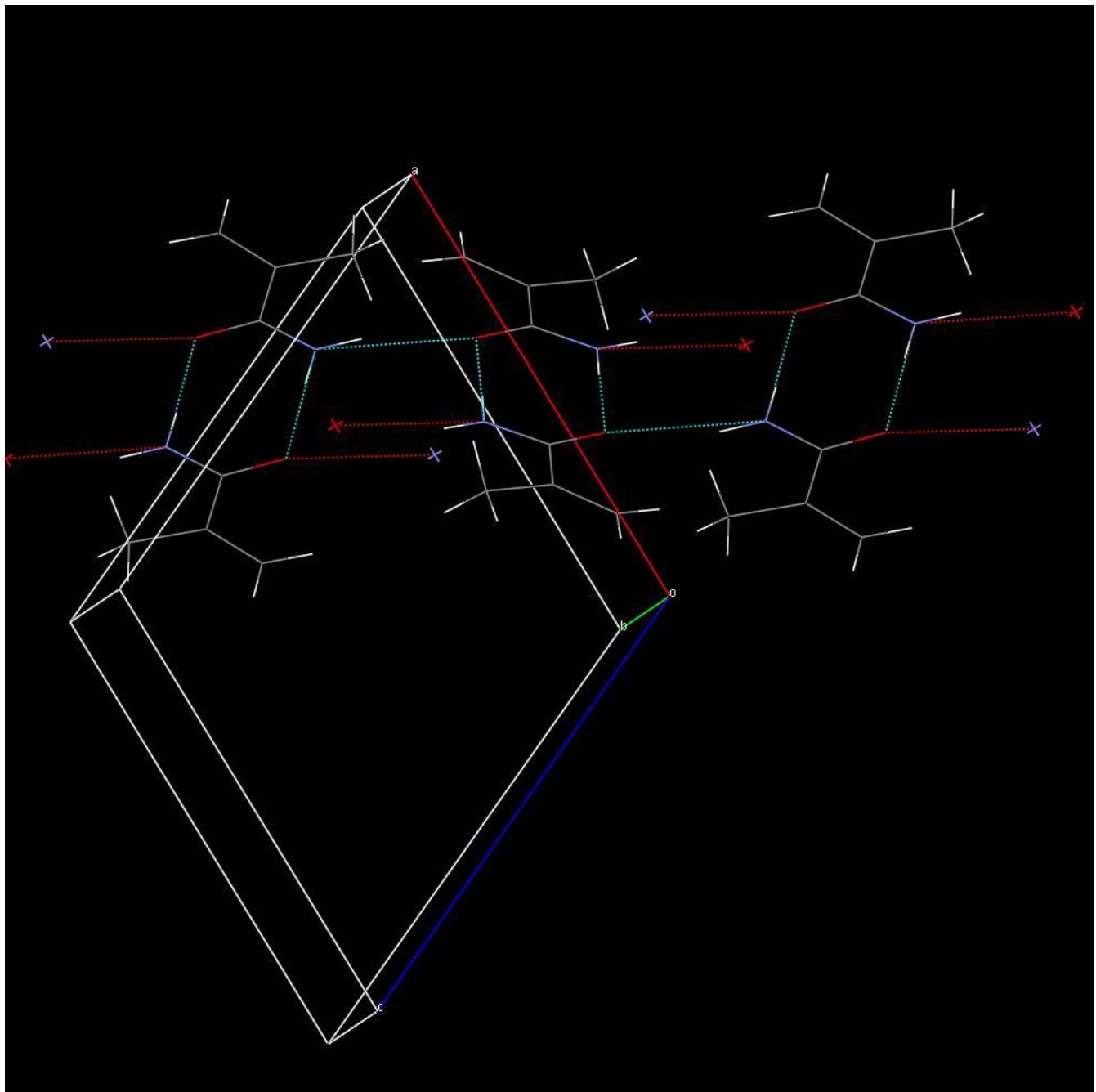
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C11 C 0.2615 -0.0526 0.1430 0.0500 1.0000 Uiso . . . . .  
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H4 H 0.3200 -0.1140 0.4000 0.0500 1.0000 Uiso . . . . .  
H5 H 0.4020 -0.1420 0.4400 0.0500 1.0000 Uiso . . . . .  
H6 H 0.3890 -0.0760 0.5700 0.0500 1.0000 Uiso . . . . .  
H7 H 0.4200 -0.0590 -0.0700 0.0500 1.0000 Uiso . . . . .  
H8 H 0.3550 -0.1860 -0.1100 0.0500 1.0000 Uiso . . . . .  
H9 H 0.2400 -0.2290 0.0170 0.0500 1.0000 Uiso . . . . .  
H10 H 0.1760 -0.1370 0.1840 0.0500 1.0000 Uiso . . . . .  
H11 H 0.2380 -0.0180 0.2140 0.0500 1.0000 Uiso . . . . .  
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H22 H 0.4653 0.2922 0.4176 0.0600 1.0000 Uiso . . . . .

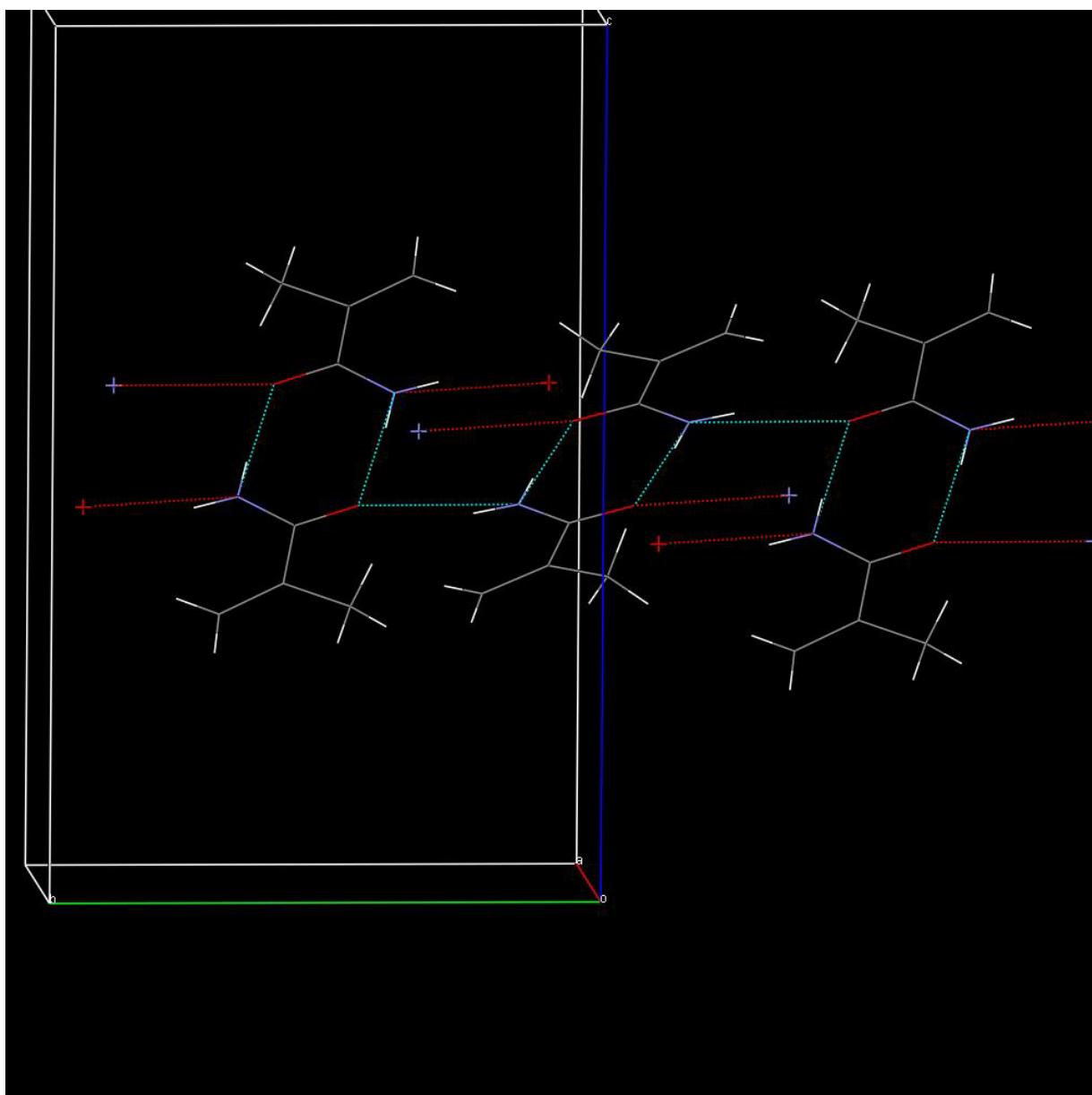
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S1 . C1 . 1.763 yes  
S1 . C6 . 1.780 yes  
O1 . C1 . 1.392 yes  
O1 . C4 . 1.368 yes  
N2 . C3 . 1.355 yes  
N2 . H21 . 0.880 no  
N2 . H22 . 0.880 no  
O3 . C4 . 1.179 yes  
O4 . C3 . 1.197 yes  
C1 . C2 . 1.312 yes  
C2 . C3 . 1.469 yes  
C2 . H1 . 0.769 no  
C4 . C5 . 1.502 yes  
C5 . H4 . 1.192 no  
C5 . H5 . 0.734 no  
C5 . H6 . 1.094 no  
C6 . C7 . 1.379 yes  
C6 . C11 . 1.353 yes  
C7 . C8 . 1.380 yes  
C7 . H7 . 0.935 no  
C8 . C9 . 1.338 yes  
C8 . H8 . 0.852 no  
C9 . C10 . 1.367 yes  
C9 . H9 . 0.932 no  
C10 . C11 . 1.378 yes  
C10 . H10 . 1.049 no  
C11 . H11 . 0.951 no  
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C1 . S1 . C6 . 103.487 yes

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C1 . O1 . C4 . 116.932	yes
C3 . N2 . H21 . 120.000	no
C3 . N2 . H22 . 120.000	no
H21 . N2 . H22 . 120.000	no
S1 . C1 . O1 . 116.263	yes
S1 . C1 . C2 . 124.709	yes
O1 . C1 . C2 . 118.791	yes
C1 . C2 . C3 . 124.053	yes
C1 . C2 . H1 . 117.426	no
C3 . C2 . H1 . 118.326	no
C2 . C3 . N2 . 114.409	yes
C2 . C3 . O4 . 123.859	yes
N2 . C3 . O4 . 121.702	yes
O1 . C4 . O3 . 123.533	yes
O1 . C4 . C5 . 107.748	yes
O3 . C4 . C5 . 128.708	yes
C4 . C5 . H4 . 99.773	no
C4 . C5 . H5 . 104.582	no
H4 . C5 . H5 . 92.297	no
C4 . C5 . H6 . 103.374	no
H4 . C5 . H6 . 125.823	no
H5 . C5 . H6 . 126.883	no
S1 . C6 . C7 . 115.326	yes
S1 . C6 . C11 . 124.202	yes
C7 . C6 . C11 . 120.189	yes
C6 . C7 . C8 . 118.296	yes
C6 . C7 . H7 . 120.742	no
C8 . C7 . H7 . 120.952	no
C7 . C8 . C9 . 121.171	yes
C7 . C8 . H8 . 123.355	no
C9 . C8 . H8 . 114.160	no
C8 . C9 . C10 . 120.845	yes
C8 . C9 . H9 . 130.037	no
C10 . C9 . H9 . 108.452	no
C9 . C10 . C11 . 118.618	yes
C9 . C10 . H10 . 134.076	no
C11 . C10 . H10 . 107.306	no
C10 . C11 . C6 . 120.850	yes
C10 . C11 . H11 . 120.784	no
C6 . C11 . H11 . 118.003	no



**Figure 8-S. Packing Diagram of Form I showing spiralling nature of C(4) chains along 2<sub>1</sub> axis (upper and lower chains are truncated for clarity).**



**Figure 9-S. Packing Diagram for Form II showing spiralling nature of C(4) chains along  $2_1$  axis (upper and lower chains are truncated for clarity).**