Electronic Supplementary Information for

Fabrication of hierarchical porous iron oxide films utilizing the Kirkendall effect

Lizhi Zhang,*^{*a*} Jimmy C. Yu,*^{*b*} Zhi Zheng^{*b*} and Cheuk Wan Leung^{*b*}

^a College of Chemistry, Central China Normal University, Wuhan, 430079, People's Republic of China. Fax: 86 27 6786 7535; Tel: 86 27 6786 7535; E-mail: zhanglz@mail.ccnu.edu.cn

^b Department of Chemistry, the Chinese University of Hong Kong, Shatin, N. T., Hong Kong. Fax: 852 2603 5057; Tel: 852 2609 6268; E-mail: jimyu@cuhk.edu.hk

Characterization procedures

Characterization: X-ray powder diffraction patterns were obtained on a Bruker D8 Advance X-ray diffractometer with Cu K α radiation ($\lambda = 1.54178$ Å). Scanning electron microscopy images and energy dispersive X-ray spectrum were performed on a LEO 1450VP scanning electron microscope with an energy-dispersive X-ray instrument. Transmission electron microscopy images were recorded on a Tecnai 20 FEG transmission electron microscope. XPS measurements were performed in a VG Scientific ESCALAB Mark II spectrometer equipped with two ultrahigh-vacuum Supplementary Material for Chemical Communications This journal is © The Royal Society of Chemistry 2005

(UHV) chambers. All binding energies were referenced to the C_{1S} peak at 284.8 eV of the surface adventitious carbon. The nitrogen adsorption and desorption isotherms of the resulting films on substrates at 77 K were measured using a Micromeritics ASAP2010 system after the film samples were vacuum-dried at 110 °C overnight.



Figure S1 EDX spectra of the final iron oxide films.



Figure S2 EDX spectra of the resulting film grown at early stages during hydrothermal reaction.

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Figure S3 SEM image of the film grown in 15 mL water without addition of iodine at 180 ^{o}C for 12 h. Scale bar: 2 $\mu m.$



Figure S4 SEM image of the film grown in CTAB ethanolic solution (0.4 g CTAB in 15 mL water) without addition of iodine at 180 $^{\circ}$ C for 12 h. Scale bar: 2 μ m.

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Figure S5 SEM image of the film grown in CTAB aqueous solution (0.4 g CTAB in 15 mL water) without addition of iodine at 180 °C for 12 h. Scale bar: 2 μ m.



Figure S6 SEM and TEM images of the resulting cobalt oxide film. a): SEM, b) TEM. Scale bar in a: 2

μm.