NH vs CH^{...}anion hydrogen bond formation in platinum(tetrakispyridine) complexes

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Titrations tetrakispyridineplatinum(II) tetrafluoroborate 1.



• Tetrabutylammonium iodide.

Graph 1 NMR titration shift graph.



Fit 1 WinNMR fit.



• Tetrabutylammonium bromide.



Graph 2 NMR titration shift graph.



Fit 2 WinNMR fit.



Figure 2 Shift at 3 eq.

• Tetrabutylammonium chloride.



Graph 3 NMR titration shift graph.



Fit 3 WinNMR fit.



Figure 3 Shift at 3 eq.

• Tetrabutylammonium benzoate.



Graph 4 NMR titration shift graph.



Fit 4 WinNMR fit.

Pt²⁺-N H 0.26

• Tetrabutylammonium acetate.



Graph 5 NMR titration shift graph.



Fit 5 WinNMR fit.





Tetrabutylammonium methanesulfonate. •

Graph 6 NMR titration shift graph.



Fit 6 WinNMR fit.





• Tetrabutylammonium hydrogensulfate.

Graph 7 NMR titration shift graph.





Pt²⁺-N H 0.08 Figure 7 Shift at 3 eq.

Titrations tetrakis(m-(1H-pyrrole)pyridine)platinum(II) tetrafluoroborate **2**.



• Tetrabutylammonium iodide.

> CH Pyridine

Graph 8 NMR titration shift graph.



Fit 8 WinNMR fit.

➢ CH pyrrole



Graph 9 NMR titration shift graph.



Fit 9 WinNMR fit.



Figure 8 Shift at 3 eq.





CH Pyridine





Fit 10 WinNMR fit.

➢ CH pyrrole



Graph 11 NMR titration shift graph.



Fit 11 WinNMR fit.



• Tetrabutylammonium chloride.

➢ CH Pyridine



Graph 12 NMR titration shift graph.



Fit 12 WinNMR fit.









Fit 13 WinNMR fit.



Figure 10 Shift at 3 eq.

• Tetrabutylammonium acetate.



Graph 14 NMR titration shift graph.

> CH pyrrole



Graph 15 NMR titration shift graph.





Graph 16 NMR titration shift graph.



Fit 14 WinNMR fit.







Graph 17 NMR titration shift graph.



Fit 15 WinNMR fit.



Graph 18 NMR titration shift graph.



Fit 16 WinNMR fit.



• Tetrabutylammonium hydrogensulfate.



> CH Pyridine





Fit 17 WinNMR fit.



Graph 20 NMR titration shift graph.



Fit 18 WinNMR fit.



Figure 13 Shift at 3 eq.

• Tetrabutylammonium methanesulfonate.



> CH Pyridine





Fit 19 WinNMR fit.





Graph 22 NMR titration shift graph.



Fit 20 WinNMR fit.



Figure 14 Shift at 3 eq.

• Tetrabutylammonium nitrate.



Graph 23 NMR titration shift graph.



Fit 21 WinNMR fit.





Graph 24 NMR titration shift graph.



Fit 22 WinNMR fit.



Tetrabutylammonium triflate. •



> CH Pyridine



Graph 25 NMR titration shift graph.



Figure 16 Shift at 3 eq.

Titrations tetrakis(p-(1H-pyrrole)pyridine)platinum(II) tetrafluoroborate 3.



Tetrabutylammonium iodide. •

 \succ CH α Pyridine





Fit 23 WinNMR fit.



Figure 17 Shift at 3 eq.

• Tetrabutylammonium bromide.



Graph 27 NMR titration shift graph.



Fit 24 WinNMR fit.



Figure 18 Shift at 3 eq.

• Tetrabutylammonium chloride.



 \succ CH α Pyridine

Graph 28 NMR titration shift graph.



Fit 25 WinNMR fit.



Figure 19 Shift at 3 eq.

• Tetrabutylammonium hydrogensulfate.



 \succ CH α Pyridine





Fit 26 WinNMR fit.



Figure 20 Shift at 3 eq.

• Tetrabutylammonium nitrate.



Graph 30 NMR titration shift graph.



Fit 27 WinNMR fit.



Figure 21 Shift at 3 eq.

• Tetrabutylammonium methanesulfonate.









Fit 28 WinNMR fit.



Figure 22 Shift at 3 eq.

• Tetrabutylammonium acetate.



[➤] CHa Pyridine







Graph 33 NMR titration shift graph.

> NH Pyrrole



Graph 34 NMR titration shift graph.



Fit 29 WinNMR fit.



• Tetrabutylammonium benzoate.



Graph 35 NMR titration shift graph.



Fit 30 WinNMR fit.

> CH β Pyridine



Graph 36 NMR titration shift graph.



Fit 31 WinNMR fit.



Graph 37 NMR titration shift graph.



Fit 32 WinNMR fit.



Titrations tetrakis(m-(1H-pyrrole)pyridine)platinum(II) tetrafluoroborate **2** in other solvents.

$MeNO_2-d_3$.

• Tetrabutylammonium methanesulfonate.



Graph 38 NMR titration shift graph.

➢ CH Pyridine



Graph 39 NMR titration shift graph.

MeCN-d₃.



• Tetrabutylammonium methanesulfonate.

Graph 40 NMR titration shift graph.



Graph 41 NMR titration shift graph.

Ligands synthesis



Scheme 1 Compounds 4, 5 and 6 prepared from the same bibliographic source^{a1}. Final compound 7 prepared using the Paal Knoor reaction^{b2,3}.



Scheme 2 Compounds 8, 9 and 10 prepared from the same bibliographic source¹. Final compound 11 prepared using the Paal Knoor reaction^{2,3}.

^b 2. Kruse, C. G.; Bouw, J. P.; Vanhes, R.; Vandekuilen, A.; Denhartog, J. A. J., New Methods for the Synthesis of 2-Arylpyrroles. *Heterocycles* 1987, *26*, (12), 3141-3151.
3. Noland, W. E.; Cole, K. P.; Britton, D., Five (1H-pyrrol-2-yl)pyridines. *Acta*

^a 1. Baxendale, I. R.; Brusotti, G.; Matsuoka, M.; Ley, S. V., Synthesis of nornicotine, nicotine and other functionalised derivatives using solid-supported reagents and scavengers. *Journal of the Chemical Society-Perkin Transactions 1* **2002**, (2), 143-154.

Crystallographica Section C-Crystal Structure Communications 2003, 59, O263-O267.

N-methoxy-N-methylnicotinamide 4. This compound was prepared following the literature method¹. Yield, 8.5 g, or 54%. ¹H NMR 300 MHz in CDCl₃ (ppm): 8.83 (dd, *J* 2.3, *J* 0.8, 1H), 8.56 (dd, *J* 4.9, *J* 1.9, 1H), 7.90 (dd, *J* 7.9, *J* 2.3, 1H), 7.24 (ddd, *J* 7.9, *J* 4.9, *J* 0.8, 1H), 1.30 (s, 3H), 1.30 (s, 3H). ¹³C NMR 75.4 MHz in CDCl₃ (ppm): 166.6 (CO), 150.6 (CH), 148.5 (CH), 135.3 (CH), 129.1 (C), 122.6 (CH), 60.5 (CH₃), 32.4 (CH₃).

3-(1,3-dioxolan-2-yl)-1-(pyridin-3-yl)propan-1-one 5. This compound was prepared following the literature method¹. Yield 2.7 g of yellow oil, or 74%. ¹H NMR 300 MHz in CDCl₃ (ppm): 9.12 (dd, J 1.5, J 0.8, 1H), 8.71 (dd, J 4.9, J 1.9, 1H), 8.18 (ddd, J 7.9, J 1.9, J 1.5, 1H), 7.36 (ddd, J 7.9, J 4.9, J 0.8, 1H), 4.95 (t, J 4.1, 1H), 3.90 (m, 2H), 3.80 (m, 2H), 3.07 (t, J 7.1, 2H), 2.15 (td, J 7.1 J 4.1, 2H). ¹³C NMR 75.4 MHz in CDCl₃ (ppm): 198.2 (CO), 153.4 (CH), 149.6 (CH), 135.4 (CH), 132.2 (C), 123.6 (CH), 103.1 (CH), 65.0 (2 × CH₂), 32.7 (CH₂), 27.7 (CH₂).

4-oxo-4-(pyridin-3-yl)butanal 6. This compound was prepared following the literature method¹. Yield 0.6 g of yellow oil, or 73%. ¹H NMR 300 MHz in CDCl₃ (ppm): 9.89 (s, 1H), 9.19 (d, *J* 0.8, 1H), 8.78 (dd, *J* 0.8, *J* 3.6, 1H), 8.24 (dd, *J* 5.5, *J* 1.3, 1H), 7.42 (dd, *J* 3.6 *J* 5.5, 1H), 3.31 (t, *J* 4.5, 2H), 2.96 (t, *J* 4.7, 2H). ¹³C NMR 75.4 MHz in CDCl₃ (ppm): 200.1 (CO), 196.8 (CO), 153.8 (CH), 149.7 (CH), 135.5 (CH), 131.9 (C), 123.8 (CH), 37.5 (CH₂), 31.3 (CH₂).

3-(1H-pyrrol-2-yl)pyridine 7. This compound was prepared following the literature method^{2,3}. Yield 1.2 g of white solid, or 84%. ¹H NMR 300 MHz in DMSO-*d*₆ (ppm): 11.43 (s, 1H), 8.87 (d, *J* 2.3, 1H), 8.33 (dd, *J* 1.5, *J* 4.9, 1H), 7.96 (ddd, *J* 7.9, *J* 1.5, *J* 2.25, 1H), 7.35 (dd, *J* 7.9 *J* 4.9, 1H), 6.92 (s, 1H), 6.63 (s, 1H), 6.15 (dd, *J* 3.4, *J* 6.0, 1H). ¹³C NMR 75.4 MHz in DMSO-*d*₆ (ppm): 146.5 (CH), 144.9 (CH), 131.6 (CH), 129.3 (C), 128.6 (C), 124.0 (CH), 120.5 (CH), 110.6 (CH), 107.6 (CH).

N-methoxy-N-methylisonicotinamide 8. This compound was prepared following the literature method¹. Yield 8.53 g of yellow oil, or 55%. ¹H NMR 300 MHz in CDCl₃ (ppm): 8.64 (dd, *J* 4.5, *J* 1.5, 2H), 7.45 (dd, *J* 4.5, *J* 1.5, 2H), 3.48 (s, 3H), 3.31 (s, 3H). ¹³C NMR 75.4 MHz in CDCl₃ (ppm): 167.5 (CO), 1149.9 (CH), 141.7 (C), 121.9 (CH), 61.3 $(2 \times CH_3)$.

3-(1,3-dioxolan-2-yl)-1-(pyridin-4-yl)propan-1-one 9. This compound was prepared following the literature method¹. Yield 4.1 g of yellow oil, or 54%. ¹H NMR 300 MHz in CDCl₃ (ppm): 8.76 (dd, *J* 3.1, *J* 1.5, 2H), 7.70 (dd, *J* 3.1, *J* 1.5, 2H), 4.96 (t, *J* 3.4, 1H), 3.91 (m, 2H), 3.82 (m, 2H), 3.07 (t, *J* 5.3, 2H), 2.12 (m, 2H). ¹³C NMR 75.4 MHz in CDCl₃ (ppm): 198.9 (CO), 150.9 (CH), 142.9 (C), 121.1 (CH), 103.1 (CH), 65.1 (CH₂), 32.7 (CH₂), 27.7 (CH₂).

4-oxo-4-(pyridin-4-yl)butanal 10. This compound was prepared following the literature method¹. Yield 1.32 g of yellow oil, or 42%. ¹H NMR 300 MHz in CDCl₃ (ppm): 9.84 (s, 1H), 8.78 (d, *J* 4.5, 2H), 7.72 (dd, *J* 3.4, *J* 1.1, 2H), 3.26 (t, *J* 4.5, 2H), 2.93 (t, *J* 4.9, 2H). ¹³C NMR 75.4 MHz in CDCl₃ (ppm): 206.8 (CO), 199.9 (CO), 150.9 (CH), 142.4 (C), 121.1 (CH), 37.4 (CH₂), 31.3 (CH₂).

4-(1H-pyrrol-2-yl)pyridine 11. This compound was prepared following the literature method^{2,3}. Yield 0.67 g of white solid, or 56%. ¹H NMR 300 MHz in DMSO- d_6 (ppm): 11.59 (s, 1H), 8.45 (d, *J* 4.5, 2H), 7.56 (d, *J* 4.5, 2H), 6.98 (s, 1H), 6.79 (s, 1H), 6.19 (s, 1H). ¹³C NMR 75.4 MHz in DMSO- d_6 (ppm): 149.9 (CH), 139.3 (C), 128.3 (C), 121.6 (CH), 117.3 (CH), 109.7 (CH), 108.7 (CH).

Stack plots





Figure 25 Stack plot tetrakis m-(1H-pyrrol-2-yl)pyridine)platinum(II) tetrafluoroborate **2** with tetrabutylammonium methanesulfonate in DMSO- d_6 .





Figure 26 Stack plot tetrakis *m*-(1*H*-pyrrol-2-yl)pyridine)platinum(II) tetrafluoroborate **2** with tetrabutylammonium iodide in DMSO- d_6 .

Bromide in DMSO-d₆.



Figure 27 Stack plot tetrakis m-(1H-pyrrol-2-yl)pyridine)platinum(II) tetrafluoroborate **2** with tetrabutylammonium bromide in DMSO- d_6 .



Figure 28 Stack plot tetrakis *m*-(1*H*-pyrrol-2-yl)pyridine)platinum(II) tetrafluoroborate **2** with tetrabutylammonium chloride in DMSO- d_6 .

44

Acetate in DMSO-d₆.



Figure 29 Stack plot tetrakis m-(1H-pyrrol-2-yl)pyridine)platinum(II) tetrafluoroborate **2** with tetrabutylammonium acetate in DMSO- d_6 .

Methanesulfonate in MeNO₂-d₃.



Figure 30 Stack plot tetrakis m-(1H-pyrrol-2-yl)pyridine)platinum(II) tetrafluoroborate **2** with tetrabutylammonium metanesulfonate in MeNO₂- d_3 . Precipitation occurred after 2 equivalents.

Methanesulfonate in MeCN-d₃.



Figure 31 Stack plot tetrakis *m*-(1*H*-pyrrol-2-yl)pyridine)platinum(II) tetrafluoroborate **2** with tetrabutylammonium metanesulfonate in MeCN- d_3 . Precipitation occurred after 1 equivalent.

Chloride in MeNO₂-d₃.



Figure 32 Stack plot tetrakis m-(1H-pyrrol-2-yl)pyridine)platinum(II) tetrafluoroborate **2** with tetrabutylammonium chloride in MeNO₂- d_3 . Precipitation occurred after 1 equivalent.

Bromide in MeNO₂-d₃.



Figure 33 Stack plot tetrakis m-(1H-pyrrol-2-yl)pyridine)platinum(II) tetrafluoroborate **2** with tetrabutylammonium bromide in MeNO₂- d_3 . Precipitation occurred after 1 equivalent.



Iodide in MeNO₂-d₃.

Figure 34 Stack plot tetrakis m-(1*H*-pyrrol-2-yl)pyridine)platinum(II) tetrafluoroborate **2** with tetrabutylammonium iodide in MeNO₂- d_3 . Precipitation occurred after 1 equivalent.

Hydrogensulfate in MeNO₂-d₃.



Figure 35 Stack plot tetrakis m-(1*H*-pyrrol-2-yl)pyridine)platinum(II) tetrafluoroborate **2** with tetrabutylammonium hydrogensulfate in MeNO₂- d_3 . Precipitation occurred after 1 equivalent.



Nitrate in MeNO₂-d₃.

Figure 36 Stack plot tetrakis m-(1H-pyrrol-2-yl)pyridine)platinum(II) tetrafluoroborate with tetrabutylammonium nitrate in MeNO₂- d_3 . Precipitation occurred after 1 equivalent.

Shifts overview

 Table 1 Abstract of ¹H NMR shifts for compounds 1-3.





CH correlation experiments.

HMBQC of compound 2.



Figure 37 Compound 2 CH correlation.

HMBQC of compound 3



Figure 38 Compound 3 CH correlation.

Spectroscopic data.

Compound 1

¹H NMR



Figure 39 ¹H NMR spectra of compound **1**.

¹³C NMR



Figure 40¹³C NMR spectra of compound **1**.

HR Mass spectrum



Figure 41 HRES⁺ of compound **1**.

Compound 2

¹H NMR



ap1805ie1 1 1

400.1324710 MHz

Figure 42 ¹H NMR spectra of compound 1.





tetrakis(m-(1H-pyrrole)pyridine)platinum(II) tetrafluoroborate 2.

Figure 43 ¹³C NMR spectra of compound **2**.



*** Current Data Parameters ***
NAME : apl805ie1
EXPNO : 2
PROCNO : 1
NUC1 : 13C
SF : 100.6129021 MHz



Figure 44 HRES⁺ of compound **2**.

Compound 3

¹H NMR



Figure 45 ¹H NMR spectra of compound 3.





tetrakis(p-(1H-pyrrole)pyridine)platinum(II) tetrafluoroborate 3.

Figure 46¹³C NMR spectra of compound **3**.

HR Mass spectrum



Figure 47 HRES⁺ of compound **3**.