Chemical Communications Electronic Supporting Information

Impact of a Conformationally Restricted Receptor on the BF₂ Chelated Azadiyrromethene Fluorosensing Platform

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General Experimental

THF was distilled under N_2 over sodium wire and benzophenone. Cyclohexane, 1,4dioxane and chloroform were distilled over K_2CO_3 . DMF was distilled under reduced pressure over K_2CO_3 .

Solutions for the solvent studies were prepared from a stock solution of **1** (0.005 mmol in 10 mL THF). 1 mL was diluted into 25 mL of either cyclohexane, DMF, 1,4-dioxane or THF to provide a second stock solution from which further dilutions were made to give the desired concentration.

Fluorescence measurements were recorded with the following setting; both slit widths 5 nm, excitation wavelength 630 nm.

Cremophor EL formulation procedure.

1a or **1b** (1 mg) was dissolved in THF (0.5 mL) to which Cremophor EL (0.2 mL) was added and the mixture was sonicated for 30 minutes. The THF was removed under reduced pressure and the resulting blue oil was made up to 25 mL with saline solution to give a stock solution. Further dilutions were made as required to give the necessary concentration.

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4-(4-Dimethylamino-2,6-dimethyl-phenyl)-2-phenyl-1*H*-pyrrole.

3-(4-Dimethylamino-2,6-dimethyl-phenyl)-5-phenyl-1*H*-pyrrole-2-carboxylic acid ethyl ester¹ (1.0 g, 2.76 mmol) was dissolved in ethanol (20 mL), 10% NaOH solution (10 mL) was added and the solution was heated under reflux for 1 hour. The reaction was allowed to cool to room temperature, concentrated to half the original volume, neutralized with 1M HCl and the resulting precipitate was isolated by filtration. The precipitate was air-dried. This crude material was dissolved in ethanolamine (3 mL) and the solution was heated under reflux for 1 hour. The reaction was allowed to cool, poured into cold water (50 mL) and the resulting precipitate was isolated by filtration (0.74 g, 92%), m.p. 109-111 °C. ¹H NMR (300 MHz, DMSO-d₆) δ : 11.25 (bs, 1H), 7.67 (d, *J* = 7.3 Hz, 2H), 7.35-7.40 (m, 2H), 7.17 (t, *J* = 7.3 Hz, 1H), 6.68 (m, 1H), 6.50 (s, 2H), 6.42 (d, *J* = 1.7 Hz, 1H), 2.90 (s, 6H), 2.14 (s, 6H). ¹³C NMR (75 MHz, DMSO-d₆) δ : 149.3, 137.3, 133.4, 131.0, 129.0, 125.6, 125.2, 123.5, 123.0, 118.7, 112.2, 107.8, 40.8, 22.1. IR (KBr disc) cm⁻¹: 3432. ES-MS: *m/z* [M + H]⁺ 291.1. HRMS Calcd for C₂₀H₂₃N₂ [M + H]⁺: 291.1861, found 291.1873.

1. J. Killoran, J.F. Gallagher, P.V. Murphy, D.F. O'Shea, *New J. Chem.* 2005, **29**, 1258.

2-Nitroso-3-(4-dimethylamino-2,6-dimethyl-phenyl)-5-phenyl-1*H*-**pyrrole** (3b). [3,5-Dimethyl-4-(5-phenyl-1*H*-pyrrol-3-yl)-phenyl]-dimethyl-amine (0.21 g, 0.7 mmol) was dissolved in acetic acid (5 mL) and the solution was cooled on an ice bath. Concentrated HCl (0.1 mL) was added and the reaction mixture was cooled on an ice bath. A solution of sodium nitrite (0.06 g, 0.79 mmol) in water (5 mL) was added drop-wise over 10 minutes to give a brown orange solution that was left to stir for 20 minutes. The solution was poured over a saturated sodium acetate solution (30 mL) and the resulting brown precipitate was isolated by filtration. Recrystallization from ether/cyclohexane gave **3b** as a dark brown solid (0.12 g, 52%), m.p. 189-191 °C. ¹H NMR (300 MHz, CDCl₃) δ : 7.82-7.85 (m, 2H), 7.46-7.49 (m, 3H), 6.82 (s, 1H), 6.55 (s, 2H), 2.98 (s, 6H), 2.25 (s, 6H). ¹³C NMR (75 MHz, CDCl₃) δ : 163.6, 150.6, 144.6, 138.1, 131.2, 129.6, 129.5, 129.4, 127.0, 119.9, 117.1, 111.8, 40.5, 21.8. IR (KBr disc) cm⁻¹: 3281, 1500. ES-MS: *m/z* [M + H]⁺ 320.2. HRMS Calcd for C₂₀H₂₂N₃O [M + H]⁺: 320.1763, found 320.1770.

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(3,5-Diphenyl-1*H*-pyrrol-2-yl)-[3-(4-dimethylamino-2,6-dimethyl-phenyl)-5-

phenyl-pyrrol-2-ylidene]-amine (4b). [3,5-Dimethyl-4-(2-nitroso-5-phenyl-1Hpyrrol-3-yl)-phenyl]-dimethyl-amine **3b** (0.09 g, 0.28 mmol) and 2,4-diphenyl-1Hpyrrole 2b (0.075 g, 0. 34mmol) were dissolved in acetic acid (5 mL) and acetic anhydride (1 mL) and heated at 100 °C for 1 hours. During the course of the reaction an intense blue color formed. The reaction mixture was allowed to cool to room temperature, washed with a saturated sodium bicarbonate solution (100 mL) and water (50 mL) and the resulting precipitate was isolated by filtration, leaving a colorless filtrate. Purification by silica gel column chromatography eluting with CH₂Cl₂/hexane (1:1) gave **4b** as a dark colored solid (0 016g, 11%), m.p. 209-211°C. ¹H NMR (300 MHz, CDCl₃) δ : 7.96-8.01 (m, 4H), 7.88 (d, J = 7.32 Hz, 2H), 7.40-7.56 (m, 6H), 7.19-7.23 (m, 3H), 7.13 (s, 1H), 6.95 (s, 1H), 6.56 (s, 2H), 3.00 (s, 6H), 2.25 (s, 6H), (NH not observed). ¹³C NMR (75 MHz, CDCl₃) δ: 159.9, 153.2, 150.4, 150.1, 147.0, 146.0, 139.0, 137.7, 133.6, 132.7, 131.9, 130.3, 129.4, 129.1, 129.0, 128.2, 128.1, 127.4, 126.9, 125.9, 122.6, 120.2, 112.3, 111.8, 40.8, 21.8. IR (KBr disc) cm⁻¹: 1604, 1542. λ_{max} (CHCl₃) nm: 587. HRMS Calcd for C₃₆H₃₃N₄ [M + H]⁺: 521.2705, found 521.2690.

BF₂ Chelated-(3,5-Diphenyl-1H-pyrrol-2-yl)-[3-(4-dimethylamino-phenyl)-5-(3,5-Diphenyl-1*H*-pyrrol-2-yl)-[3-(4phenyl-pyrrol-2-ylidene]-amine (**1**a). dimethylamino-phenyl)-5-phenyl-pyrrol-2-ylidene]-amine² 4a (0.076 g, 0.15 mmol) was dissolved in dry CH₂Cl₂ (25 mL), treated with diisopropylethylamine (0.27 mL, 1.54 mmol) and boron trifluoride diethyletherate (0.27 mL, 2.16 mmol), and stirred at room temperature under nitrogen for 6 hours. The mixture was diluted with CH₂Cl₂ (30 mL), washed with water (2 x 50 mL), and organic layer was dried over sodium sulfate and evaporated to dryness. Purification by column chromatography on silica eluting with CH₂Cl₂/hexane (3:1) gave 1a as a metallic brown solid (0.067 g, 93%), m.p. 244-245 °C. ¹H NMR (300 MHz, CDCl₃) δ : 8.12 (d, J = 8.9 Hz, 2H), 7.99-8.07 (m, 6H), 7.38-7.49 (m, 9H), 6.96 (s, 1H), 6.84 (s, 1H), 6.74 (d, J = 8.9 Hz, 2H), 3.09 (s, 6H). ¹³C NMR (75 MHz, CDCl₃) δ: 155.6, 151.5, 146.2, 144.2, 141.4, 133.1, 132.3, 131.9, 131.2, 130.7, 129.9, 129.5, 129.3, 128.7, 128.5, 128.4, 128.4, 120.5, # Supplementary Material (ESI) for Chemical Communications # This journal is © The Royal Society of Chemistry 2006

117.8, 115.5, 111.9, 40.1. IR (KBr disc) cm⁻¹: 1600, 1488. λ_{max} (CHCl₃) nm: 610, 726. HRMS Calcd for C₃₄H₂₈N₄BF₂ [M + H]⁺: 541.2375, found 541.2380.

2. Hall, M.J.; McDonnell, S.O.; Killoran, J.; O'Shea, D.F. J. Org. Chem. 2005, 70, 5571.

BF₂ Chelated-(3,5-diphenyl-1*H*-pyrrol-2-yl)-[3-(4-dimethylamino-2,6-dimethylphenvl)-5-phenvl-pvrrol-2-vlidenel-amine (1b). (3,5-Diphenvl-1*H*-pvrrol-2-vl)-[3-(4-dimethylamino-2,6-dimethyl-phenyl)-5-phenyl-pyrrol-2-ylidene]-amine 4b (0.016 g, 0.03 mmol) was dissolved in dry CH₂Cl₂ (5 mL), treated with diisopropylamine (0.05 mL, 0.3 mmol) and boron trifluoride diethyletherate (0.05 mL, 0.43 mmol), and stirred at room temperature under nitrogen for 6 hours. The mixture was diluted with CH₂Cl₂ (20 mL), washed with water (2 x 25 mL), and organic layer was dried over sodium sulfate and evaporated to dryness. Purification by column chromatography on silica eluting with CH₂Cl₂/hexane (3:1) gave **1b** as a metallic brown solid (0.011 g, 63%), m.p. 231-232 °C. ¹H NMR (CDCl₃) δ: 8.04-8.08 (m, 2H), 7.95-8.02 (m, 4H), 7.46-7.50 (m, 6H), 7.29-7.31 (m, 3H), 6.96 (s, 1H), 6.75 (s, 1H), 6.54 (s, 2H), 3.01 (s, 6H), 2.30 (s, 6H). ¹³C NMR (CDCl₃) δ: 160.0, 158.6, 150.5, 147.5, 144.9, 142.8, 138.1, 132.3, 131.9, 131.5, 130.9, 130.5, 129.7, 129.4, 129.1, 128.8, 128.6, 128.5, 123.6, 118.4, 111.8, 40.5, 22.1. IR (KBr disc) cm⁻¹: 1600, 1513. λ_{max} (CHCl₃) nm: 637. HRMS Calcd for $C_{36}H_{32}BF_2N_4 [M + H]^+$: 569.2688, found 569.2715.

| entry | 1b | DMF | THF | 1,4-dioxane | cyclohexane |
|-------|-----------------------|-----|-----|-------------|-------------|
| 1 | abs. ^a | 635 | 635 | 635 | 636 |
| 2 | abs./H ^{+ b} | 649 | 645 | 645 | 643 |

Table 1. Absorbance Spectral Characteristics of 1b in nm.

^a Concentration of 5 x 10⁻⁶ M. ^b Trifluoroacetic acid added until maximum peak intensity was reached.

Figure 1. Absorbance spectra; organic solvent study of 1b



Absorbance Spectra of **1b** at 5 x 10^{-6} M in DMF (blue), THF (yellow), 1,4-dioxane (black), cyclohexane (green).

Figure 2. Absorbance spectra; organic solvent study of 1b



Absorbance Spectra of **1b** at 5 x 10^{-6} M in DMF + trifluoroacetic acid (blue), THF + trifluoroacetic acid (yellow), 1,4-dioxane + trifluoroacetic acid (black), cyclohexane + trifluoroacetic acid (green).

| entry | 1b | DMF | THF | 1,4-dioxane | cyclohexane |
|-------|-----------------------|-----|-----|-------------|-------------|
| 1 | flu. ^a | 669 | 672 | 672 | 667, 786 |
| 2 | flu./H ^{+ b} | 673 | 669 | 670 | 665 |

 Table 2. Fluorescence Spectral Characteristics of 1b in nm.

^a concentration of 5 x 10^{-7} M.

^b Trifluoroacetic acid added until maximum peak intensity was reached.

Figure 3. Fluorescence Organic solvent study of 1b



Fluorescence Spectra of **1b** at 5 x 10^{-7} M in DMF (blue), THF (yellow), 1,4-dioxane (black), cyclohexane (green).

Figure 4. Fluorescence Organic Solvent Study of 1b.



Fluorescence Spectra of **1b** at 5 x 10^{-7} M in DMF + trifluoroacetic acid (blue), THF + trifluoroacetic acid (yellow), 1,4-dioxane + trifluoroacetic acid (black), cyclohexane + trifluoroacetic acid (green).





1a; Thermal ellipsoids are drawn on the 50% probability level.

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| Identification code | 1a | |
|---|---|--------------------------------|
| Empirical formula | C ₃₄ H ₂₇ B N ₄ F ₂ | |
| Formula weight | 540.41 | |
| Temperature | 113(2) K | |
| Wavelength | 0.71073 Å | |
| Crystal system | Triclinic | |
| Space group | P-1 (#2) | |
| Unit cell dimensions | a = 10.755(3) Å | $\alpha = 91.726(6)^{\circ}$. |
| | b = 10.824(3) Å | $\beta = 113.148(5)^{\circ}$. |
| | c = 12.163(4) Å | $\gamma = 96.565(6)^{\circ}$. |
| Volume | $1288.9(7) Å^{3}$ | • |
| Ζ | 2 | |
| Density (calculated) | 1.392 Mg/m ³ | |
| Absorption coefficient | 0.092 mm ⁻¹ | |
| F(000) | 564 | |
| Crystal size | 0.50 x 0.30 x 0.04 mm ³ | |
| Theta range for data collection | 1.83 to 26.00°. | |
| Index ranges | -13<=h<=13, -13<=k<=1 | 3, -14<=l<=14 |
| Reflections collected | 9788 | |
| Independent reflections | 4977 [R(int) = 0.0530] | |
| Completeness to theta = 26.00° | 97.9 % | |
| Absorption correction | Semi-empirical from equi | valents |
| Max. and min. transmission | 0.9963 and 0.6896 | |
| Refinement method | Full-matrix least-squares | on F ² |
| Data / restraints / parameters | 4977 / 0 / 456 | |
| Goodness-of-fit on F ² | 1.189 | |
| Final R indices [I>2sigma(I)] | R1 = 0.0944, $wR2 = 0.216$ | 58 |
| R indices (all data) | R1 = 0.1443, $wR2 = 0.235$ | 8 |
| Largest diff. peak and hole | 0.380 and -0.367 e.Å ⁻³ | |



ORTEP Diagram of **1b**

1b; Thermal ellipsoids are drawn on the 50% probability level.

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 Table 1. Crystal data and structure refinement for 1b.

| Identification code Empirical formula Formula weight Temperature Wavelength Crystal system Space group Unit cell dimensions | 1b C36 H31 B N4 F2 568.46 293(2) K 0.71073 Å Monoclinic C2 a = 42.555(7) Å b = 5.3195(9) Å | $\alpha = 90^{\circ}.$ $\beta = 103.405(4)^{\circ}$ | |
|--|---|--|--|
| | c = 12.781(2) Å | $\gamma = 90^{\circ}$. | |
| Volume | 2814.3(8) A ³ | | |
| Z | 4 | | |
| Density (calculated) | 1.342 Mg/m ³ | | |
| Absorption coefficient | 0.088 mm ⁻¹ | | |
| F(000) | 1192 | | |
| Crystal size | $0.60 \ge 0.10 \ge 0.01 \text{ mm}^3$ | | |
| Theta range for data collection | 1.64 to 23.98°. | | |
| Index ranges | -48<=h<=48, -5<=k<=5, -14<= | =1<=14 | |
| Reflections collected | 8078 | | |
| Independent reflections | 3777 [R(int) = 0.0602] | | |
| Completeness to theta = 23.98° | 89.7 % | | |
| Absorption correction | Semi-empirical from equivalents | | |
| Max. and min. transmission | 0.9991 and 0.5272 | | |
| Refinement method | Full-matrix least-squares on F ² | | |
| Data / restraints / parameters | 3777 / 1 / 392 | | |
| Goodness-of-fit on F ² | 1.014 | | |
| Final R indices [I>2sigma(I)] | R1 = 0.0509, WR2 = 0.0981 | | |
| R indices (all data) | R1 = 0.0854, WR2 = 0.1109 | | |
| bsolute structure parameter 1.8(13) | | | |
| Largest diff. peak and hole | 0.258 and -0.253 e.Å ⁻³ | | |