

Electronic Supplementary Information (ESI) service

New ambient pressure organic superconductors κ_H - and κ_L -(DMEDO-TSeF)₂[Au(CN)₄](THF)

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Hydrogen bonds in κ_H -(DMEDO-TSeF)₂[Au(CN)₄](THF)

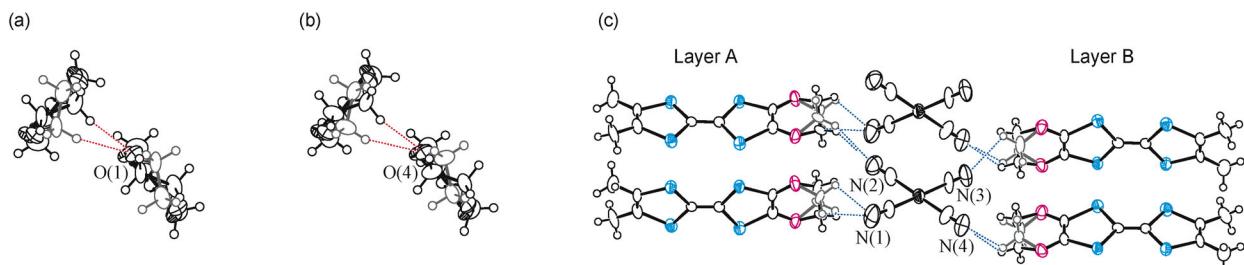


Fig. 1 Hydrogen bonds in κ_H -phase. Red dotted lines indicate $\text{CH}\cdots\text{O}$ hydrogen bonds; (a) in the layer A and (b) in the layer B. (c) Blue dotted lines indicate $\text{CH}\cdots\text{N}$ hydrogen bonds between the donor and the anion. Gray atoms and bonds indicate minor orientation of the disordered ethylene bridge.

Table 1 Hydrogen bond distances^a for κ_H -(DMEDO-TSeF)₂[Au(CN)₄](THF) less than 3.0 Å

Atom	Atom	Distance / Å	ADC ^b	Atom	Atom	Distance / Å	ADC ^b
In layer A							
HCH···O							
For the major orientation of the ethylene group							
O(1)	H(7A1)	2.497	4	O(4)	H(18B)	2.517	55404
For the minor orientation of the ethylene group							
O(1)	H(8B2)	2.675	4	O(4)	H(17D)	2.696	55404
H ₂ CH···O							
O(1)	H(9C)	2.959	2	O(3)	H(20C)	2.883	65703
O(2)	H(9A)	2.989	3				
Molecule A-Anion							
Molecule B-Anion							
HCH···N							
For the major orientation of the ethylene group							
N(1)	H(7A2)	2.810	4	N(3)	H(17A)	2.773	4
N(1)	H(8A1)	2.990	4	N(3)	H(17A)	2.789	1
N(2)	H(8A2)	2.548	1	N(3)	H(18A)	2.698	1
				N(4)	H(18A)	2.671	4
				N(3)	H(18B)	2.808	1
For the minor orientation of the ethylene group							
N(1)	H(7B1)	2.590	4	N(3)	H(17C)	2.671	1
N(1)	H(7B2)	2.592	4	N(3)	H(17C)	2.934	4
N(2)	H(8B1)	2.746	1	N(3)	H(17D)	2.886	1
N(2)	H(8B1)	2.965	4	N(4)	H(18C)	2.905	1
				N(3)	H(18C)	2.535	4
				N(4)	H(18D)	2.844	4
H ₂ CH···N							
N(1)	H(10A)	2.495	55602	N(4)	H(19A)	2.960	66703
				N(4)	H(19C)	2.589	65602

^a The hydrogen bonds shorter than sum of the van der Waals radii are represented by the pink characters; sum of the van der Waals radii; O···H = 2.72 Å; N···H = 2.75 Å; A. Bondi, *J. Phys. Chem.*, 1964, **68**, 411–451. ^b Symmetry operators: (1) +X, +Y, +Z; (2) -X, 1/2+Y, 1/2-Z; (3) -X, -Y, -Z; (4) +X, 1/2-Y, 1/2+Z.

Hydrogen bonds in κ_L -(DMEDO-TSeF)₂[Au(CN)₄](THF)

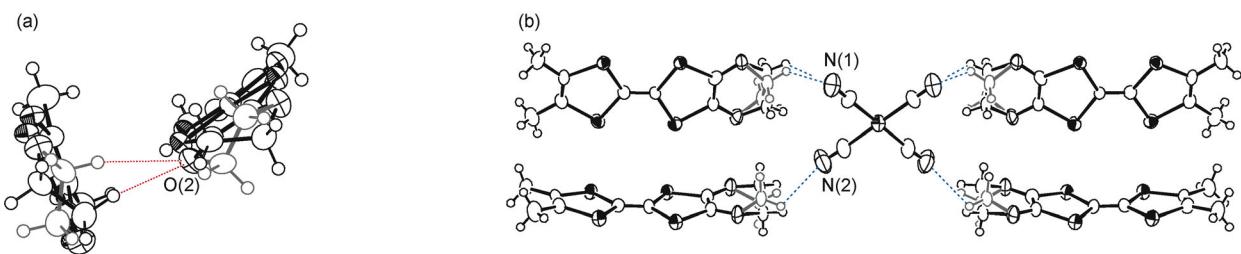


Fig. 2 Hydrogen bonds in κ_L -phase. (a) Red dotted lines indicate $\text{CH}\cdots\text{O}$ hydrogen bonds in the donor layer. (b) Blue dotted lines indicate $\text{CH}\cdots\text{N}$ hydrogen bonds between the donor and the anion. Gray atoms and bonds indicate minor orientation of the disordered ethylene bridge.

Table 2 Hydrogen bond distances^a for κ_L -(DMEDO-TSeF)₂[Au(CN)₄](THF) less than 3.0 Å

Atom	Atom	Distance / Å	ADC ^b	Atom	Atom	Distance / Å	ADC ^b
HCH···O (For the major orientation)							
O(2)	H(7A2)	2.582	55608	HCH···O (For the minor orientation)			
O(2)	H(9C)	2.853	56504	$\text{H}_2\text{CH}\cdots\text{O}$			
HCH···N (For the major orientation)							
N(1)	H(7A1)	2.599	1	N(1)	H(7B1)	2.381	1
N(2)	H(7A1)	2.820	55608	N(1)	H(7B2)	2.823	1
N(2)	H(7A2)	2.792	55608	N(2)	H(7B1)	2.963	55608
N(2)	H(8A1)	2.793	1	N(2)	H(8B1)	2.843	55608
N(2)	H(8A1)	2.826	55608	N(2)	H(8B2)	2.949	1
$\text{H}_2\text{CH}\cdots\text{N}$							
N(1)	H(10A)	2.447	66605	N(1)	H(10C)	2.927	56404

^a The hydrogen bonds shorter than sum of the van der Waals radii are represented by the pink characters; sum of the van der Waals radii; O···H = 2.72 Å; N···H = 2.75 Å. ^b Symmetry operators: (1) +X, +Y, +Z; (2) 1/2+X, 1/2-Y, 1/2-Z; (3) -X, 1/2+Y, -Z; (4) 1/2-X, -Y, 1/2+Z; (5) -X, -Y, -Z; (6) 1/2-X, 1/2+Y, 1/2+Z; (7) +X, 1/2-Y, +Z; (8) 1/2+X, +Y, 1/2-Z.

Molecular packing of κ_L -(DMEDO-TSeF)₂[Au(CN)₄](THF)

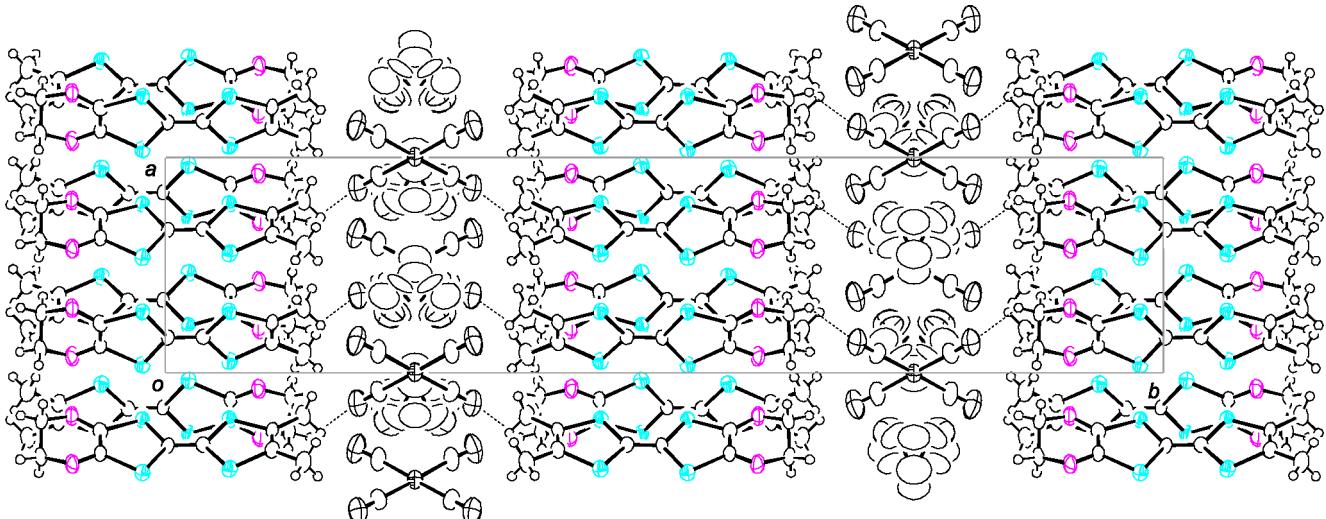


Fig. 3 Molecular packing of the κ_L -phase viewed along the crystallographic *c*-axis. Dotted lines indicate $\text{CH}\cdots\text{N}$ hydrogen bonds less than 2.75 Å. The minor orientations of the disordered ethylene bridge of the donor molecule are omitted for clarity.

Crystal structures for the insulating layer of the κ_H - and κ_L -phase

