

**Supporting information for:**

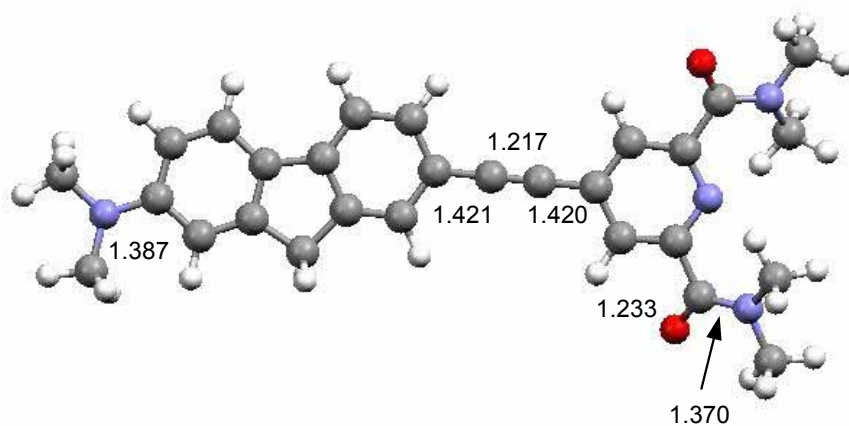
**Synthesis of chromophores combining second harmonic generation and two photon induced fluorescence properties.**

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**Characterisations of 1 and 2.**

**4-(2'-(*N,N*-dihexyl-phenyl)ethynyl-2,6-di(diethylcarbamoyl)-pyridine 1.** Mp = 101.6°C. <sup>1</sup>H NMR (CD<sub>3</sub>CN) : δ = 7.47 (s, 2H); 7.35 (d, 2H, *J* = 9 Hz); 6.64 (d, 2H, *J* = 9 Hz); 3.48 (q, 4H, *J* = 7 Hz) ; 3.3 (m, 4H); 3.3 (m, 4H); 1.55 (m, 4H); 1.3 (m, br, 12H); 1.18 (t, 6H, *J* = 7 Hz) ; 1.08 (t, 6H, *J* = 7 Hz) ; 0.88 (t, 6H, *J* = 6 Hz). <sup>13</sup>C NMR (CD<sub>3</sub>CN) : δ = 166.9; 154.2; 148.6; 133.8; 132.9; 122.2; 110.9; 105.7; 97.4; 84.0 ; 49.9; 42.4; 39.0; 30.9; 26.3 ; 25.8; 21.9; 13.1; 11.6; 12.8. FTIR 2203 cm<sup>-1</sup> (ν<sub>C≡C</sub>); 1625 cm<sup>-1</sup> (ν<sub>C=O</sub>). Elemental analysis calcd (%) for C<sub>35</sub>H<sub>52</sub>N<sub>4</sub>O<sub>2</sub>: C 74.96, H 9.35, N: 9.99; found: C 75.04, H 9.38, N 9.78.

**4-(2'-(*N,N*-dihexyl-7'-amino-9',9'-dihexylfluorene)ethynyl-2,6-di(diethylcarbamoyl)-pyridine 2.** <sup>1</sup>H NMR (CDCl<sub>3</sub>, δ) 7.68 (s, 2H), 7.51-7.38 (m, 4H), 6.59 (d, 1H, *J* = 8.6 Hz), 6.54 (s, 1H), 3.55 (q, 4H, *J* = 7.1 Hz), 3.48-3.32 (m, 8H), 1.94-1.82 (m, 4H), 1.68-1.50 (m, 4H), 1.46-0.96 (m, 36H), 0.91-0.83 (m, 6H), 0.79-0.72 (m, 10H); <sup>13</sup>C NMR (CDCl<sub>3</sub>, δ) 167.73, 153.77, 153.08, 149.91, 148.68, 143.83, 134.31, 131.21, 127.98, 126.00, 125.09, 121.09, 117.87, 116.85, 110.86, 105.99, 98.04, 85.67, 54.79, 51.46, 43.27, 40.57, 40.15, 31.76, 31.51, 29.75, 27.23, 26.86, 23.71, 22.65, 14.27, 14.03, 12.78 ; FTIR 2202 cm<sup>-1</sup> (ν<sub>C≡C</sub>); 1640 cm<sup>-1</sup> (ν<sub>C=O</sub>). Elemental analysis calcd (%) for C<sub>54</sub>H<sub>80</sub>N<sub>4</sub>O<sub>2</sub>·2H<sub>2</sub>O: C 76.01; H 9.92; N 6.57; found C 76.49; H 9.69; N 6.26.



**Fig S1.** Optimized geometry of **2'** at the B3LYP/6-31G\* level of Density Functional Theory. The optimized AM1 geometry has been used as guess. Selected bond lengths are given in Angströms.

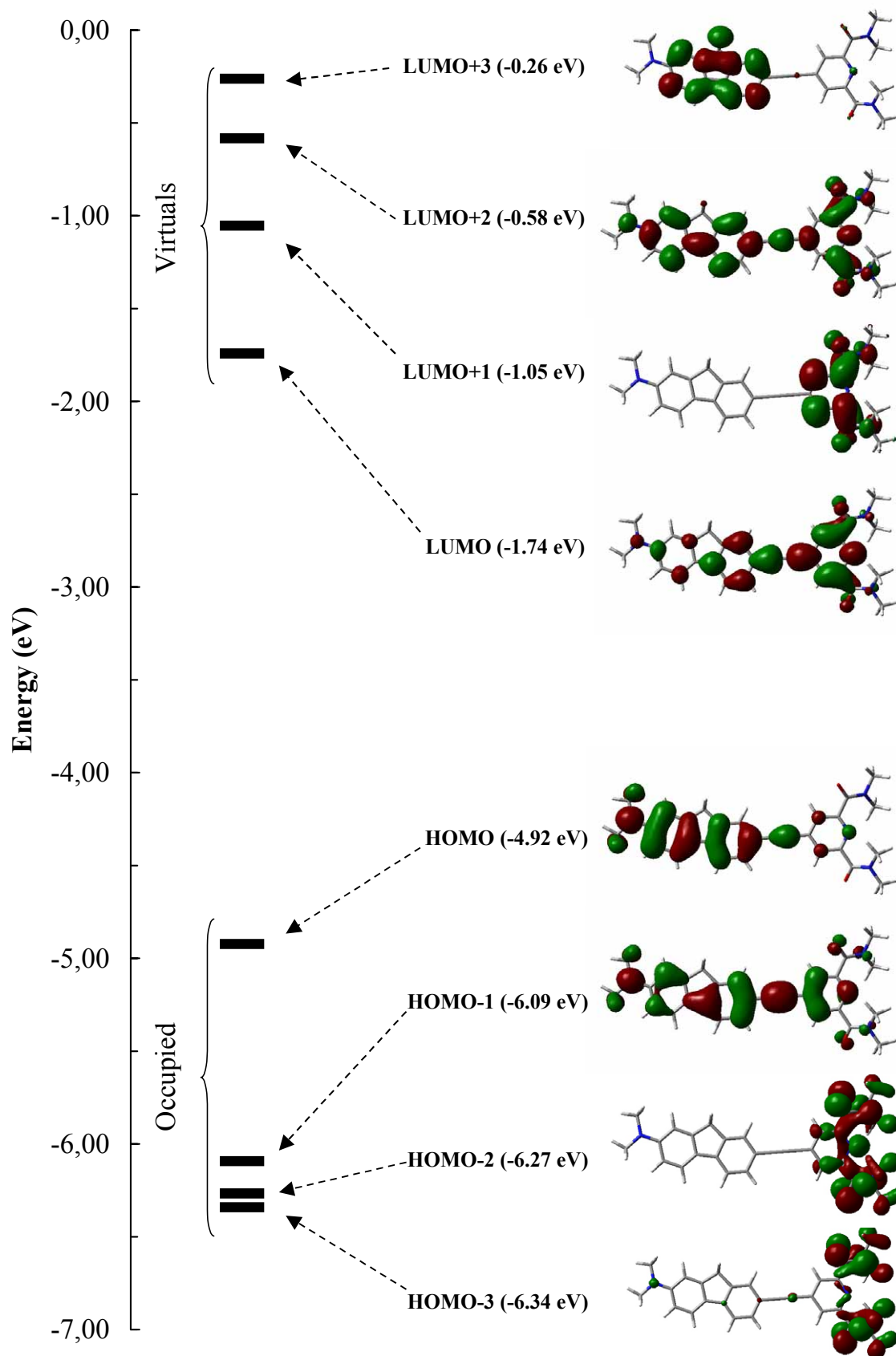
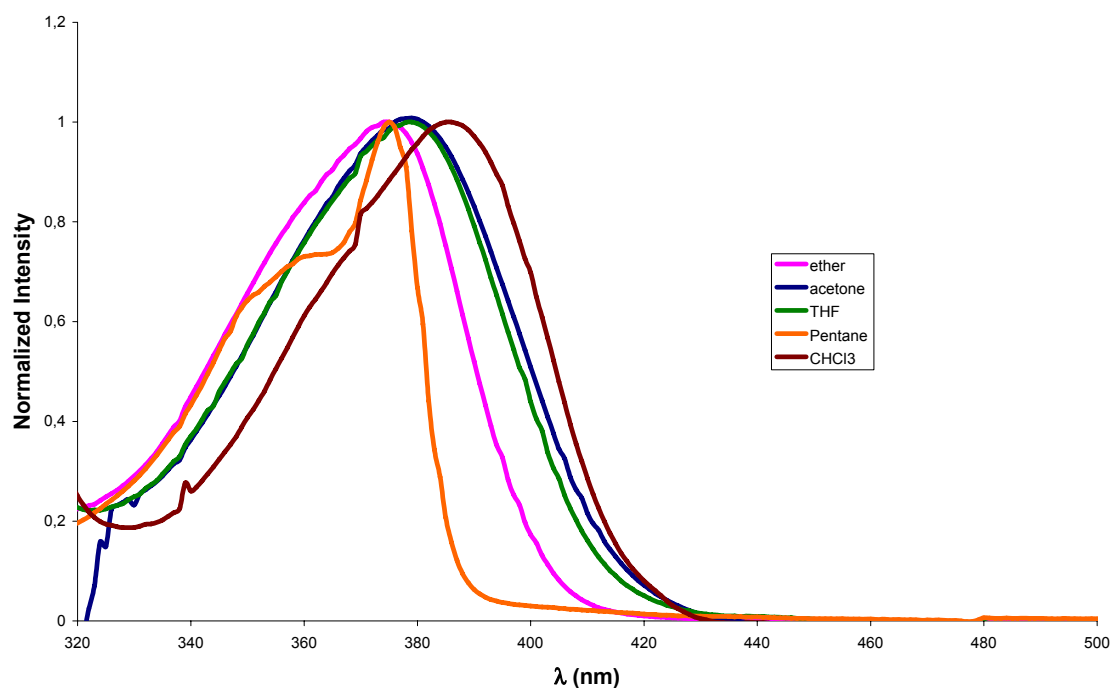


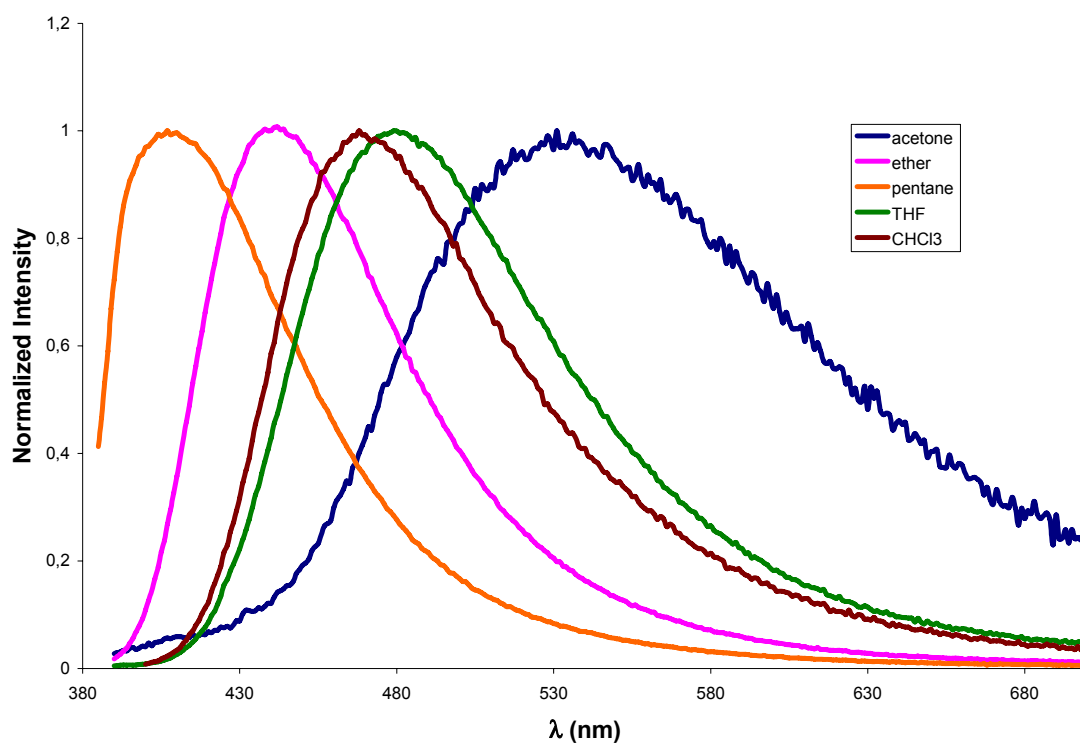
Fig S2. Kohn-Sham orbitals diagram of 2' in its optimized geometry (fig S1).

**Table S1.** Excitation spectrum of **2'** in its optimized geometry (fig S1) calculated at the B3LYP/6-31G\* level of Time-Dependent Density Functional Theory. C represents the coefficient of the current configuration in the current excited state.

Excitation	Main contributions			Energy (eV)	Wavelength (nm)	Oscillator strength
	Configuration	C	C <sup>2</sup>			
1	<b>HOMO → LUMO</b>	<b>0.67</b>	<b>0.45</b>	2.93	422	1.02
2	<b>HOMO → LUMO+1</b>	<b>0.69</b>	<b>0.48</b>	3.53	351	0.00
3	<b>HOMO-2 → LUMO</b>	<b>0.62</b>	<b>0.38</b>	3.87	320	0.01
	HOMO-4 → LUMO	0.21	0.04			
	HOMO-8 → LUMO	-0.13	0.02			
4	<b>HOMO-1 → LUMO</b>	<b>0.50</b>	<b>0.25</b>	3.93	315	0.65
	HOMO → LUMO+2	0.38	0.14			
	HOMO-3 → LUMO	0.19	0.04			
	HOMO → LUMO+3	0.12	0.01			
5	<b>HOMO → LUMO+3</b>	<b>0.45</b>	<b>0.20</b>	3.94	314	0.03
	<b>HOMO-3 → LUMO</b>	<b>0.42</b>	<b>0.18</b>			
	HOMO → LUMO+2	0.21	0.04			
	HOMO-7 → LUMO	0.13	0.02			
6	<b>HOMO → LUMO+3</b>	<b>0.45</b>	<b>0.20</b>	4.01	309	0.19
	<b>HOMO-3 → LUMO</b>	<b>0.43</b>	<b>0.18</b>			
	HOMO → LUMO+2	-0.20	0.04			
	HOMO-1 → LUMO	-0.14	0.02			
7	<b>HOMO-4 → LUMO</b>	<b>0.60</b>	<b>0.36</b>	4.04	307	0.01
	HOMO-2 → LUMO	-0.26	0.07			
	HOMO-1 → LUMO+1	-0.10	0.01			
8	<b>HOMO → LUMO+2</b>	<b>0.43</b>	<b>0.18</b>	4.08	304	0.01
	HOMO-1 → LUMO	-0.33	0.11			
	HOMO-3 → LUMO	0.25	0.06			
	HOMO-6 → LUMO	-0.19	0.04			
	HOMO → LUMO+3	-0.12	0.01			
9	<b>HOMO-6 → LUMO</b>	<b>0.54</b>	<b>0.29</b>	4.13	300	0.00
	HOMO-2 → LUMO+1	-0.24	0.06			
	HOMO-1 → LUMO	-0.20	0.04			
	HOMO-4 → LUMO+1	-0.15	0.02			
	HOMO → LUMO+2	0.13	0.02			
10	<b>HOMO-5 → LUMO</b>	<b>0.56</b>	<b>0.31</b>	4.15	299	0.00
	HOMO → LUMO+4	0.33	0.11			
	HOMO-7 → LUMO	-0.13	0.02			
	HOMO-5 → LUMO+4	-0.12	0.01			



**Fig S3.** Solvatochromism in absorption in the case of **1**.



**Fig S4.** Solvatochromism in emission in the case of **1**.

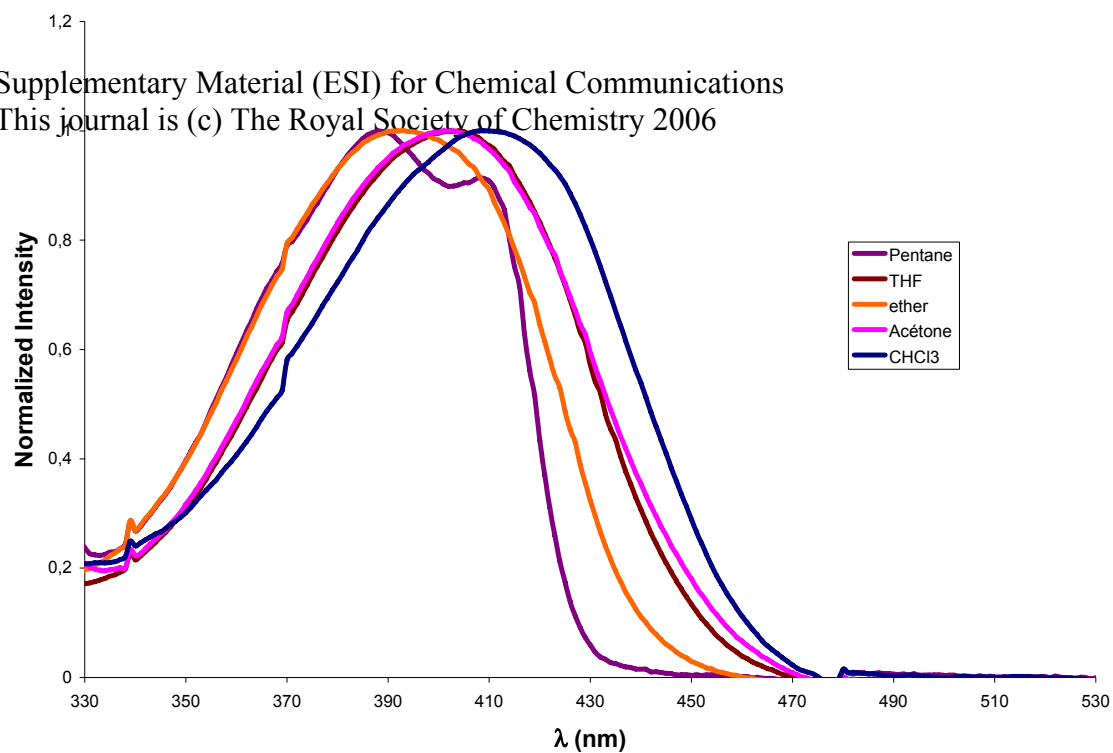


Fig S5. Solvatochromism in absorption in the case of 2.

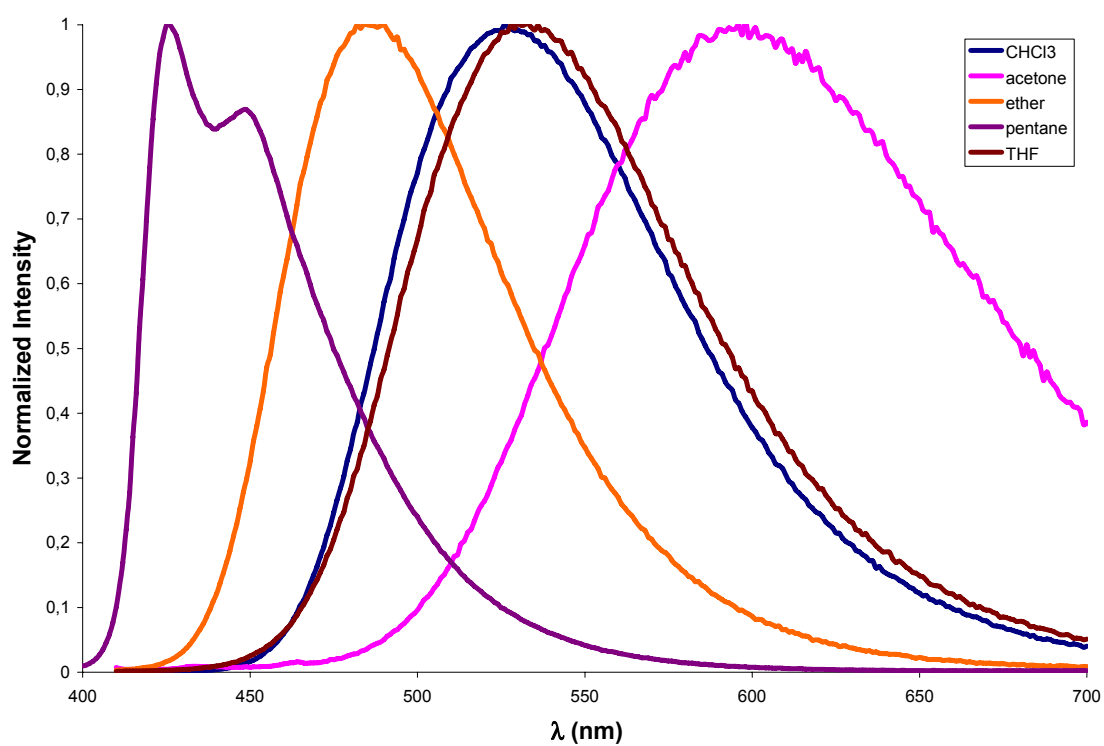


Fig S6. . Solvatochromism in emission in the case of 2.