

SUPPORTING INFORMATION

for

Unusual Reactivities of N-Heterocyclic Carbenes upon Coordination to the Platinum(II)-Dimethyl Moiety

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General Considerations. Unless stated otherwise all reactions were performed in an MBraun drybox or under Schlenk techniques. Solvents were obtained from Sigma Aldrich and were dried and degassed as necessary utilizing conventional methods. (COD)PtMe₂ was obtained from Strem Chemicals and used as received. ^{Me}IMe,^[1] ^{Me}IPr,^[1] ICy,^[2] IMes,^[2] IPr^[3] and ^tBu^[4] were synthesized by literature methods. ¹H and ¹³C NMR were obtained on either a Burker 400 MHz or a Bruker 300 MHz NMR spectrometer. Spectra were referenced to benzene at δ 7.16 (¹³C, δ 128.0), toluene at δ 2.04, or THF-d₈ at δ 3.58 (¹³C, δ 67.6) ppm. Elemental Analyses were performed at the Universität Zürich, Organisch-chemisches Institut.

Synthesis of *cis*-Pt(^{Me}IMe)₂Me₂ (1). Toluene (3 mL) was added to the solids (COD)PtMe₂ (100 mg, 0.3 mmol) and ^{Me}IMe (75.2 mg, 0.6 mmol) in a reaction vial. The resulting solution was stirred for 20 minutes, and then the volatiles were removed in vacuo, yielding a white solid. The compound was washed with pentane (3 x 2 mL), affording 131 mg (92 %) of the desired complex. X-ray quality crystals were obtained by slow evaporation of a saturated CH₂Cl₂/pentane solution at room temperature. Anal. Calcd for C₁₆H₃₀N₄Pt (MW 473.53): C, 40.58; H, 6.39; N, 11.83. Found: C, 40.72; H, 6.16; N, 11.89. ¹H NMR (C₆D₆, 400 MHz, δ): 3.53 (s, 12H, N(1,3)-CH₃), 1.45 (s, 12H, C(4,5)-CH₃), 1.08 (s, 6H, ²J_{Pt} = 65.1 Hz, CH₃). ¹³C NMR (C₆D₆, 400 MHz, δ): 187.3 (s, N-C-N), 122.6 (s, C(4,5)), 34.1 (s, ³J_{Pt} = 47.2 Hz, N(1,3)-CH₃), 9.06 (s, C(4,5)-CH₃), -7.76 (s, ¹J_{Pt} = 553.9 Hz, CH₃).

Synthesis of *cis*-Pt(ICy)₂Me₂ (2). Benzene (25mL) was added to the solids (COD)PtMe₂ (200 mg, 0.6 mmol) and ICy (365 mg, 1.2 mmol) in a reaction vial. The resulting solution was stirred for 1 hour, and then the volatiles were removed in vacuo, yielding a white solid. The compound was washed with pentane (3 x 5 mL), affording 414 mg (85 %) of the desired complex. X-ray quality crystals were obtained by slow evaporation of a saturated CH₂Cl₂/pentane solution at room temperature. Anal. Calcd for C₃₂H₅₄N₄Pt (MW 689.90): C, 55.71; H, 7.89; N, 8.12. Found: C, 56.28 ; H, 7.67 ; N, 8.13. ¹H NMR (C₆D₆, 400 MHz, δ): 6.50 (s, 4H, ⁴J_{Pt} = 6.7 Hz. NCH=CHN), 5.37 (m, 4H, ICy-CH), 2.15

(m, 4H, *p*-ICy-CH₂), 1.82 (m, 4H, *p*-ICy-CH₂), (1.63 – 0.94 (m, 36H, *m/o*-ICy-CH₂), 0.89 (s, 6H, ²J_{Pt} = 65.9 Hz, CH₃). ¹³C NMR (CD₂Cl₂, 400 MHz, δ): 185.55 (s, ¹J_{CPT} = 1700.56Hz, N-C-N), 116.29 (s, ³J_{CPT} = 44.0 Hz, NHC=CHN), 52.58 (s, SIPr-NCH₂CH₂N), 28.67 (s, SIPr-CH), 25.42 (s, SIPr-CH₃) 24.81 (s, SIPr-CH₃), -8.36 (s, ¹J_{CPT} = 1104.4 Hz, CH₃).

Synthesis of *cis*-Pt(^{Me}IPr)₂Me₂ (3). Toluene (10 mL) was added to the solids (COD)PtMe₂ (100 mg, 0.30 mmol) and ^{Me}IPr (108.2 mg, 0.60 mmol) in a reaction vial. The resulting solution was stirred for 1 h, and then the volatiles were removed in vacuo, yielding a light cream colored solid. The compound was washed with pentane (3 x 5 mL), affording 185.4 mg (89 %) of the desired complex. X-ray quality crystals were obtained by slow evaporation of a saturated CH₂Cl₂/pentane solution at room temperature. Anal. Calcd for C₂₄H₄₆N₄Pt (MW 585.34): C, 49.21; H, 7.92; N, 9.57. Found: C, 49.49; H, 7.88; N, 9.67. ¹H NMR (tol-d⁸, 400 MHz, δ): 6.36 (s, 4H, sept, *J* = 7.21 Hz, CH(CH₃)₂), 1.78 (s, 12H, s, C(4,5)-CH₃), 1.38 (d, 12H, *J* = 7.13 Hz, CH(CH₃)₂), 1.02 (d, 12H, *J* = 7.26 Hz, CH(CH₃)₂), 1.00 (s, 6H, ²J_{Pt} = 65.6 Hz, Pt-CH₃). ¹³C NMR (tol-d₈, 400 MHz, δ): 188.2 (s, N-C-N), 122.8 (s, ³J_{Pt} = 17.5 Hz C(4,5)), 51.3 (s, ³J_{Pt} = 52.8 Hz, CH(CH₃)₂), 22.6 (s, CH(CH₃)₂), 20.4 (s, CH(CH₃)₂) 10.2 (s, C(4,5)-CH₃), -7.75 (s, Pt-CH₃).

Synthesis of *cis*-Pt(IMes)₂Me₂ (4).^[5] Benzene (25mL) was added to the solids (COD)PtMe₂ (200 mg, 0.6 mmol) and IMes (365 mg, 1.2 mmol) in a Schenk flask. The resulting colourless solution was stirred for 10 minutes, and then the volatiles were removed in vacuo, yielding a white solid. The compound was washed with pentane (3 x 5 mL), affording 380 mg (76 %) of the desired complex. X-ray quality crystals were obtained by slow evaporation of a saturated benzene/Et₂O solution at room temperature. Anal. Calcd for C₄₄H₅₄N₄Pt (MW 833): C, 63.37; H, 6.53; N, 6.72. Found: C, 63.75 ; H, 6.72 ; N, 6.67 . ¹H NMR (C₆D₆, 400 MHz, δ): 6.68 (s, 4H, IMes-CH-Ar), 6.52 (s, 4H, IMes-CH-Ar), 6.13 (s, 4H, NCH=CHN), 2.38 (s, 12H, IMes-CH₃), 2.22 (s, 12H, IMes-CH₃), 1.87 (s, IMes-CH₃), 0.39 (s, 6H, ²J_{Pt} = 68.1 Hz , Me). ¹³C NMR (C₆D₆, 400 MHz, δ): 193.1 (s,N-C-N), 139.2 (s, SIPr-C), 136.8 (s, SIPr-C), 136.3 (s, SIPr-C), 135.8 (s,

SIPr-C), 130.23 (s, SIPr-C) 129.3 (s, overlapping with C₆D₆, SIPr-C), 123.3 (s, SIPr-NCH₂CH₂N), 21.6 (s, SIPr-CH), 20.4 (s, SIPr-CH₃), -0.03 (s, Pt-CH₃).

Synthesis of Pt(IMes)(IMes')Me (5).^[5] Toluene (10 mL) was added to the solids (COD)PtMe₂ (50.0 mg, 0.15 mmol) and IMes (91.4 mg, 0.30 mmol) in a Schlenk flask. The sealed flask was removed from the glove box and attached to a Schlenk line under argon. The clear colorless solution was heated with stirring for 16 hr at 60 °C. The final color of the solution was light yellow. The solvent was removed in vacuo and washed with cold pentane (3 x 2 mL) followed by one wash with cold toluene (1 mL) to yield 46.2 mg (46 %) of an off white colored powder. X-ray quality crystals were obtained by slow evaporation of a saturated benzene/Et₂O solution at room temperature. Anal. Calcd for C₄₃H₅₀N₄Pt (MW 817.96) C, 63.14; H, 6.16; N, 6.85. Found: C, 63.57; H, 6.34; N, 6.34. ¹H NMR (C₆D₆, 400 MHz, δ): 7.00 (s, Mes-CH); 6.76 (s, 1H, Mes-CH); 6.70 (s, 1H, Mes-CH); 6.64 (s, 1H, Mes-CH); 6.60 (s, 1H, Mes-CH); 6.57 (s, 2H, CH=CH) 6.55 (s, 1H, Mes-CH); 6.47 (s, 1H, Mes-CH); 6.15 (s, 1H, Mes-CH); 6.09 (d, 1H); 6.00 (d, 1H, J = 2.2 Hz); 2.59 (s, 3H, Mes-CH₃); 2.40 (s, 3H, Mes-CH₃); 2.25 (s, 3H, Mes-CH₃); 2.23 (s, 6H, Mes-CH₃); 2.22 (s, 2H, Pt-CH₂); 2.15 (s, 3H, Mes-CH₃); 1.97 (s, 3H, Mes-CH₃); 1.92 (s, 3H, Mes-CH₃); 1.83 (s, 3H, Mes-CH₃); 1.39 (s, 3H, Mes-CH₃); 0.66 (s, 3H, ²J_{Pt} = 70.0 Hz, Pt-Me). ¹³C NMR (C₆D₆, 400 MHz, δ): 193.7 (s, ¹J_{Pt} = 887.7 Hz, N-C-N), 188.1 (s, ¹J_{Pt} = 945.08 Hz, N-C-N) 138.89 (s, Mes-C), 138.72 (s, Mes-C), 138.33 (s, Mes-C), 138.13 (s, Mes-C), 137.72 (s, Mes-C), 136.95 (s, Mes-C), 136.58 (s, Mes-C), 136.42 (s, Mes-C), 136.26 (s, Mes-C), 135.60 (s, Mes-C), 134.05 (s, Mes-C), 133.98 (s, Mes-C), 130.85 (s, Mes-C), 130.05 (s, Mes-C), 129.09 (s, Mes-C), 128.90 (s, Mes-C), 126.95 (s, Mes-C), 123.35 (s, NCH=CHN), 122.30 (s, NCH=CHN), 120.99 (s, NCH=CHN) 120.43 (s, NCH=CHN), 23.04 (s, Pt-CH₂-Mes), 21.86 (s, Mes-CH₃), 21.80 (s, Mes-CH₃), 21.51 (s, Mes-CH₃), 21.26 (s, Mes-CH₃), 20.09 (s, Mes-CH₃), 19.83 (s, Mes-CH₃), 19.56 (s, Mes-CH₃), 19.27 (s, Mes-CH₃), 19.08 (s, Mes-CH₃), 18.82 (s, Mes-CH₃), -4.85 (s, Pt-Me).

Synthesis of Pt(IPr)₂ (6). Toluene (25mL) was added to the solids (COD)PtMe₂ (200 mg, 0.6 mmol) and IPr (466 mg, 1.2 mmol) in a Schlenk flask. The resulting yellow

solution was heated at 80 °C for 12 h, and then the volatiles were removed in vacuo, yielding a bright yellow solid. The compound was washed with pentane (3 x 5 mL), affording 390 mg (67 %) of the desired complex. X-ray quality crystals were obtained by slow evaporation of a saturated Et₂O solution at room temperature. Anal. Calcd for C₅₄H₇₂N₄Pt (MW 972.28): C, 66.71; H, 7.96; N, 5.76. Found: C, 66.65; H, 8.09; N, 5.92. ¹H NMR (C₆D₆, 400 MHz, δ): 7.31 (m, 4H, *p*-Ar-H), 7.09 (d, 8H, *J* = 7.6 Hz, *o*-Ar-H) 6.23 (s, 4H, C(4,5)-H), 2.94 (sept, 4H, *J* = 6.9 Hz, CH(CH₃)₂), 1.24 (d, 24H, *J* = 6.9 Hz, CH(CH₃)₂), 1.12 (d, 24H, *J* = 7.0 Hz, CH(CH₃)₂). ¹³C NMR (C₆D₆, 400 MHz, δ): 198.3 (s, ¹J_{C_{Pt}} = 1880 Hz, N-C-N), 146.27 (s, IPr-C), 139.43 (s, IPr-C), 129.03 (s, overlapping with C₆D₆, IPr-C), 123.79 (s, IPr-C), 121.59 (s, NCH=CHN), 29.10 (s, IPr-CH), 25.47 (s, IPr-CH₃), 24.41 (s, IPr-CH₃).

Synthesis of Pt(*n*^tBu)(*a*^tBu)₂Me₂ (7). Et₂O (10 mL) was added to the solids (COD)PtMe₂ (200 mg, 0.6 mmol) and ^tBu (21.6 mg, 1.2 mmol) in a reaction vial. The resulting solution was stirred for 2 weeks or until an appearance of a white solid. The solution was decanted and the white precipitate was washed with Et₂O (3 x 5 mL), affording 157 mg (51 % yield) of Pt(*n*^tBu)(*a*^tBu)₂Me₂. X-ray quality crystals were obtained by slow evaporation of a saturated CH₂Cl₂/Et₂O solution at room temperature. Anal. Calcd for C₂₄H₄₆N₄Pt (MW 585.74): C, 49.21; H, 7.92; N, 9.57. Found: C, 49.32 ; H, 7.66 ; N, 9.60. ¹H NMR (C₆D₆, 400 MHz, δ): 7.28 (d, 1H, ⁴J_H = 2.2; Hz, ⁴J_{Pt} = 7.2 Hz, C(2)-H), 6.82 (d, 1H, ⁴J_H = 2.2; ³J_{Pt} = 17.1 Hz, C(4)=CH) 6.77 (s, 2H, ⁴J_{Pt} = 6.1 Hz C(4,5)=CH), 1.97 (s, 18H, C(CH₃)₃), 1.70 (s, 9H, C(CH₃)₃), 1.10 (s, 3H, ²J_{Pt} = 67.0 Hz, CH₃), 0.89 (s, 3H, ²J_{Pt} = 67.5 Hz, CH₃), 0.87 (s, 9H, C(CH₃)₃). ¹³C NMR (THF-d₈, 400 MHz, δ): 191.8 (s, N-C-N), 165.2 (s, Pt-C=C), 125.9 (s, ³J_{Pt} = 19.6 Hz, N-C(H)-N), 124.7 (s, ³J_{Pt} = 59.4 Hz, C(H)=C-Pt), 117.4 (s, ³J_{Pt} = 27.6 Hz, C(4,5)-*n*ItBu), 59.1 (s, N-C(CH₃)₃), 58.9 (s, N-C(CH₃)₃), 56.5 (s, N-C(CH₃)₃), 32.3 (s, C(CH₃)₃), 31.1 (s, C(CH₃)₃), 30.3 (s, C(CH₃)₃), -5.89 (s, ¹J_{Pt} = 604.2 Hz Pt-CH₃), -9.05 (s, ¹J_{Pt} = 616.2 Hz, Pt-CH₃).

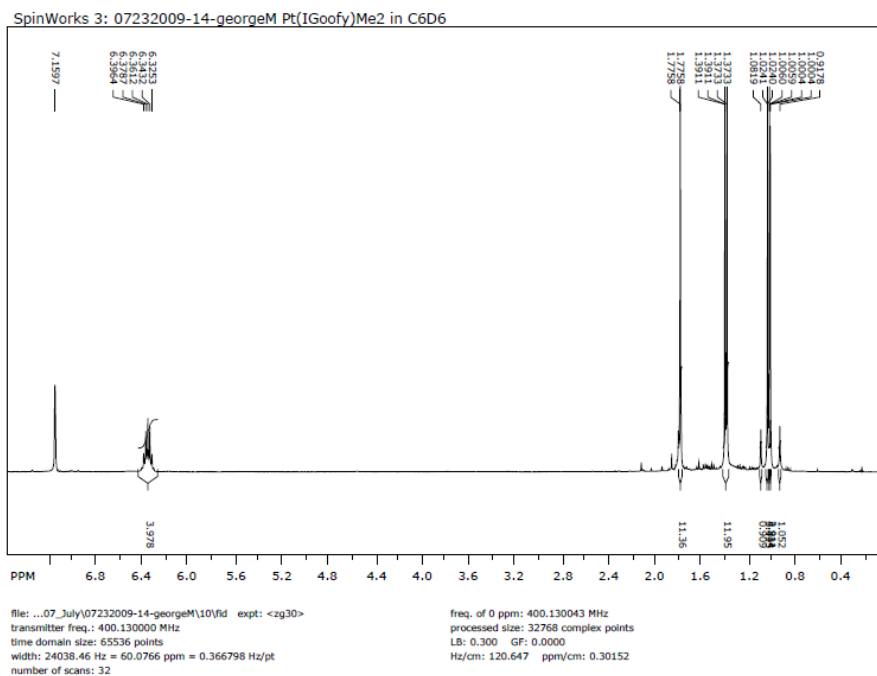
Crystallographic Structure Determinations. Single-crystal x-ray diffraction data for *cis*-Pt(^{Me}Ime)₂Me₂ (**1**) were measured using an Enraf-Nonius Kappa CCD 4-circle diffractometer. Data for the remaining complexes (**2-7**) were measured using a Bruker SMART 1K CCD 3-circle diffractometer. The crystals were cooled and maintained at the data collection temperature using a stream of cold N₂ gas. Integrated intensities were calculated from the raw detector data using the program *SAINTE*^[6]. Empirical corrections for crystal absorption were calculated using the program *SADABS*^[6], and the structures were solved and refined using *SHELX97*^[7]. *cis*-Pt(ICy)₂Me₂, **2**, which crystallizes in the noncentrosymmetric space group, Cc, has a Flack^[8] parameter of 0.021(5). For each complex, all hydrogen atoms were located in difference Fourier maps. However, independent refinement yielded unrealistic geometries of some of the hydrogen atoms due to the influence of the highly absorbing Pt atom. Therefore, hydrogen positions were constrained to idealized geometry using a model in which the hydrogen atoms ride on the atoms to which they are attached. However, for the (IMes)(IMes*)Pt(Me) complex, **5**, the hydrogen parameters on C42 were refined, since the constrained geometry may be inappropriate because of the C42-Pt1 interaction. To improve the initial convergence, ‘rigid bond’ restraints^[9] on the anisotropic thermal displacement parameters were included in the refinements of complexes **2-7** with weighting factors of 0.007 Å². Refinement of the thermal displacement parameters of the N3 atom of complex **5** yielded values corresponding to a non-positive definite tensor. To correct this result, isotropic restraints on the thermal parameters on N3 were imposed with a weighting factor of 0.005 Å².

Table S1. Selected bond lengths and angles for complexes **1-7**

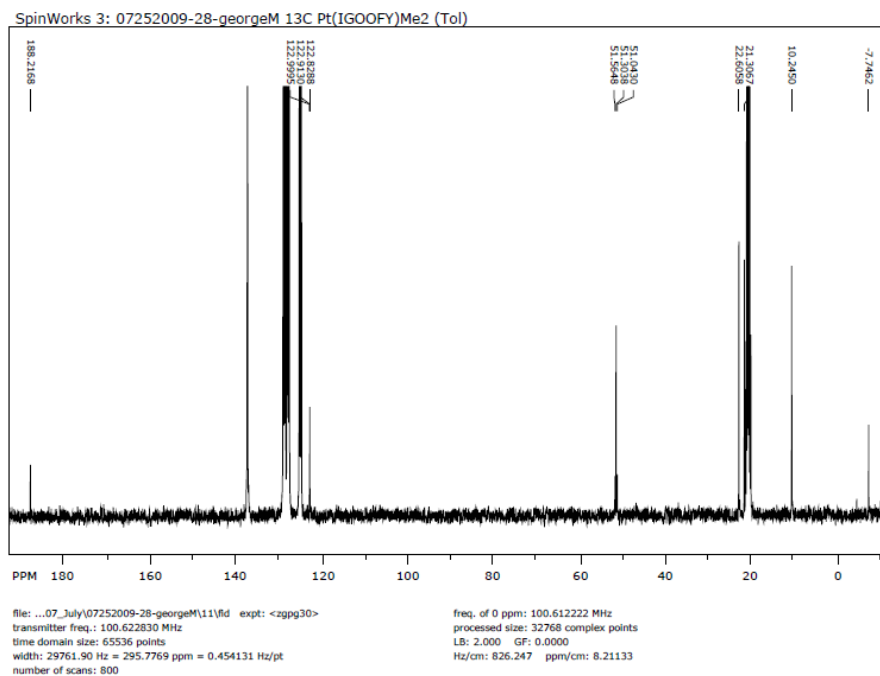
<i>Complex</i>	Pt–NHC	Pt–NHC	Pt–C	Pt–C
1	2.013(4)	2.027(5)	2.090(6)	2.117(5)
2	2.032(3)	2.065(3)	2.099(3)	2.107(3)
3	2.040(2)	2.053(2)	2.093(2)	2.095(2)
4	2.043(3)	2.044(3)	2.091(3)	2.093(3)
5	2.027(4)	2.053(4) ^[a]	2.105(4) ^[a]	2.096(4)
6	1.968(6)	1.985(6)	---	---
7	2.058(4)	2.074(4) ^[b]	2.079(4)	2.101(4)
<i>Complex</i>	NHC–Pt–NHC	NHC–Pt–C (<i>cis</i>)	C–Pt–C	
1	92.8(2)	90.5(2), 91.2(2)	85.6(2)	
2	94.0(1)	88.6(1), 93.3(1)	84.4(1)	
3	96.7(1)	88.5(1), 90.6(1)	84.5(1)	
4	113.0(1)	82.4(1), 83.6(1)	81.1(1)	
5	108.4(2)	87.3(2), 79.8(2) ^[a]	84.3(2) ^[a]	
6	178.3(2)	---	---	
7	104.3(2)	84.9(2), 83.9(2) ^[b]	87.0(2)	

^[a] Bond lengths and angles of cyclometallated IMes. ^[b] Bond lengths and angles of abnormally bound I^tBu.

***cis*-Pt(^{Me}Pr)₂Me₂ (3)**
¹H NMR

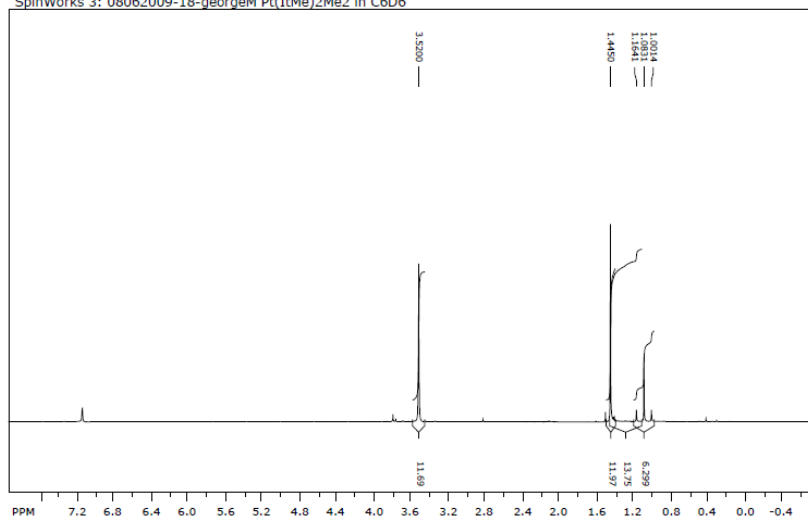


¹³C NMR



***cis*-Pt(^{Me}IME)₂Me₂(1)**
¹H NMR

SpinWorks 3: 08062009-18-georgeM Pt(IME)₂Me₂ in C₆D₆

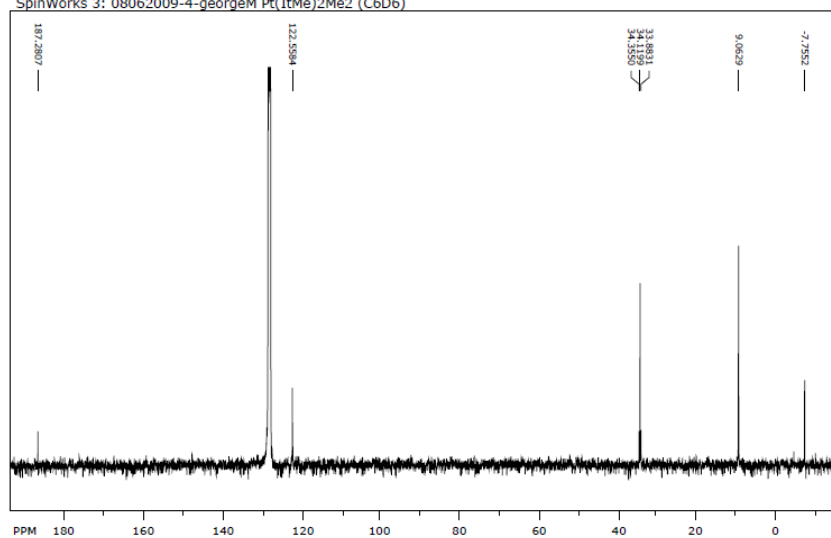


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number of scans: 32

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processed size: 32768 complex points
LB: 0.300 GF: 0.0000
Hz/cm: 139.246 ppm/cm: 0.34800

¹³C NMR

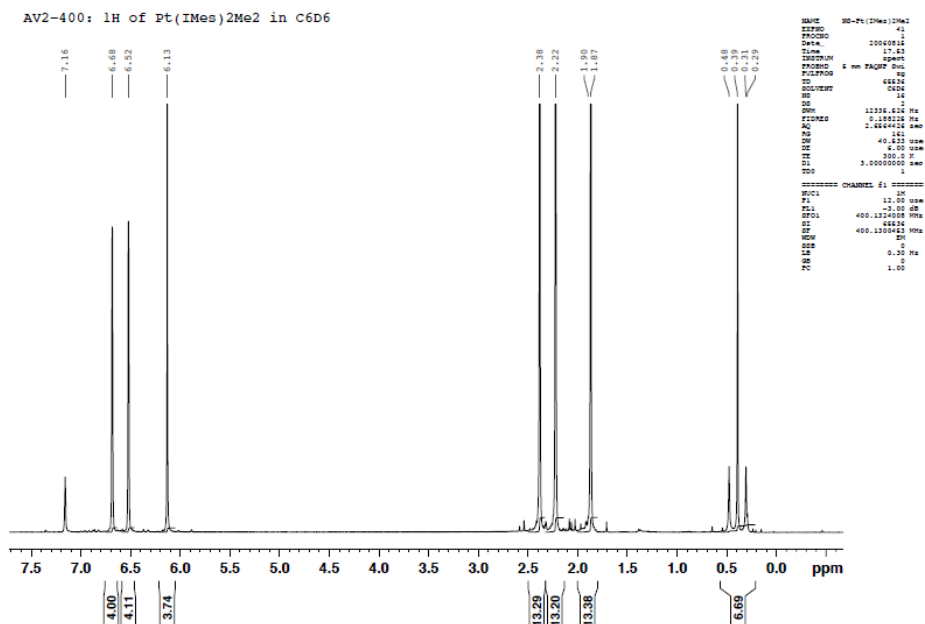
SpinWorks 3: 08062009-4-georgeM Pt(IME)₂Me₂ (C₆D₆)



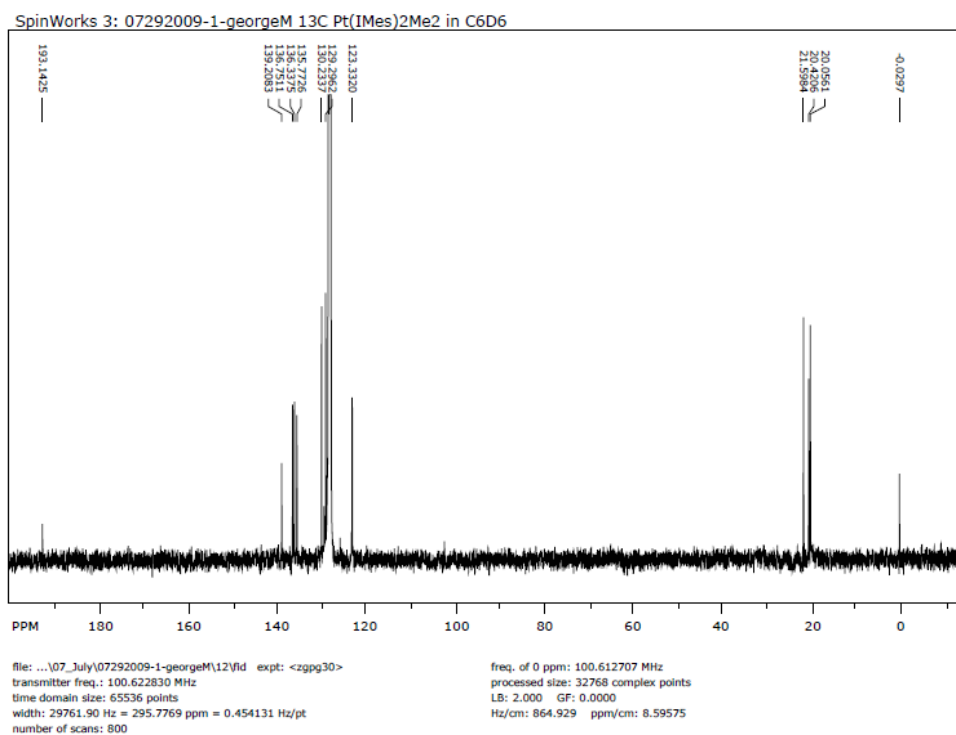
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number of scans: 800

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Hz/cm: 847.717 ppm/cm: 8.42470

***cis*-Pt(IMes)₂Me₂ (4)**
¹H NMR

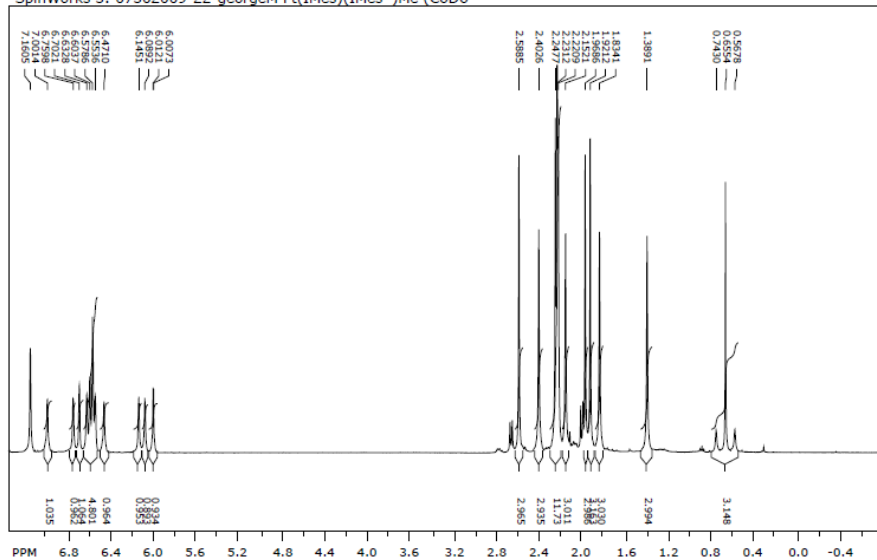


¹³C NMR



Pt(IMes)(IMes')Me (5) ¹H NMR

SpinWorks 3: 07302009-22-georgeM Pt(IMes)(IMes*)Me (C6D6)

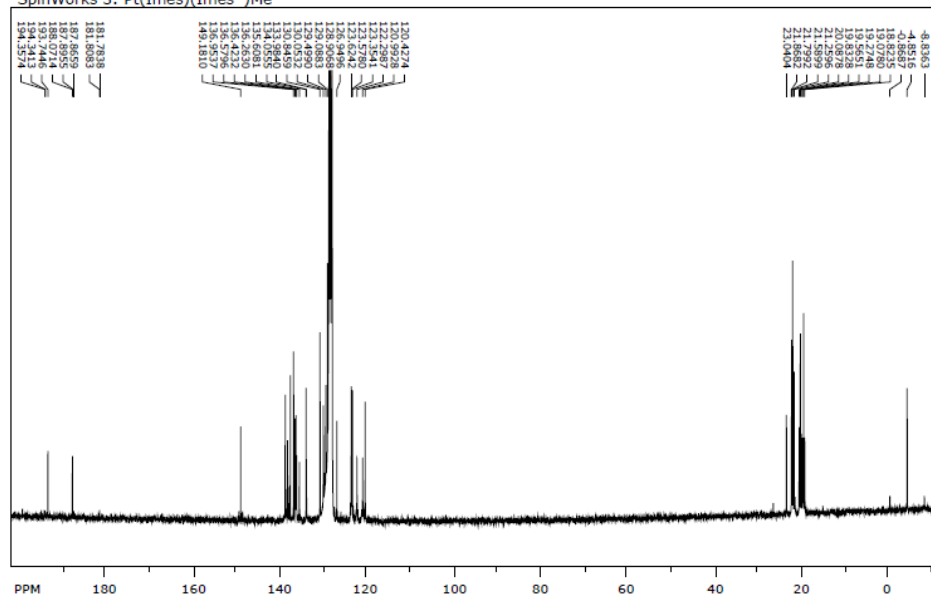


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number of scans: 32

freq. of 0 ppm: 400.130043 MHz
processed size: 32768 complex points
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Hz/cm: 130.795 ppm/cm: 0.32688

¹³C NMR

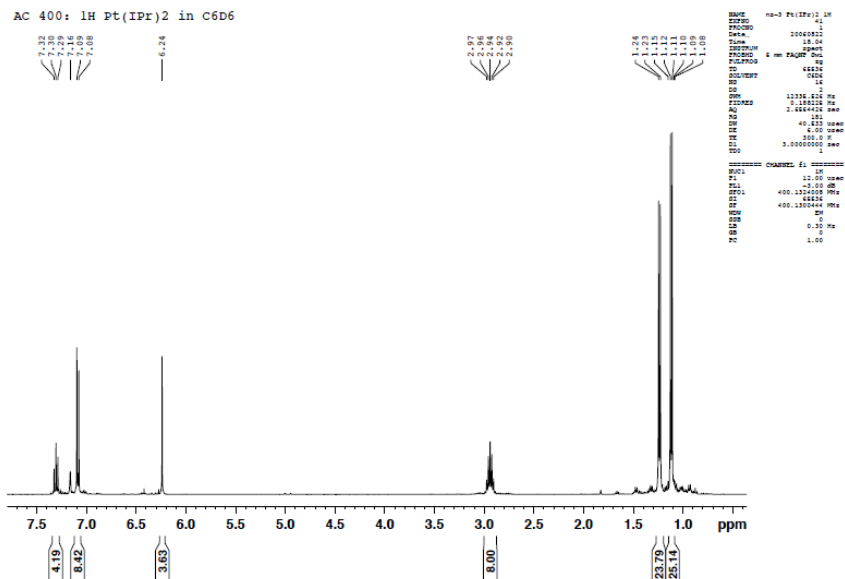
SpinWorks 3: Pt(IMes)(IMes*)Me



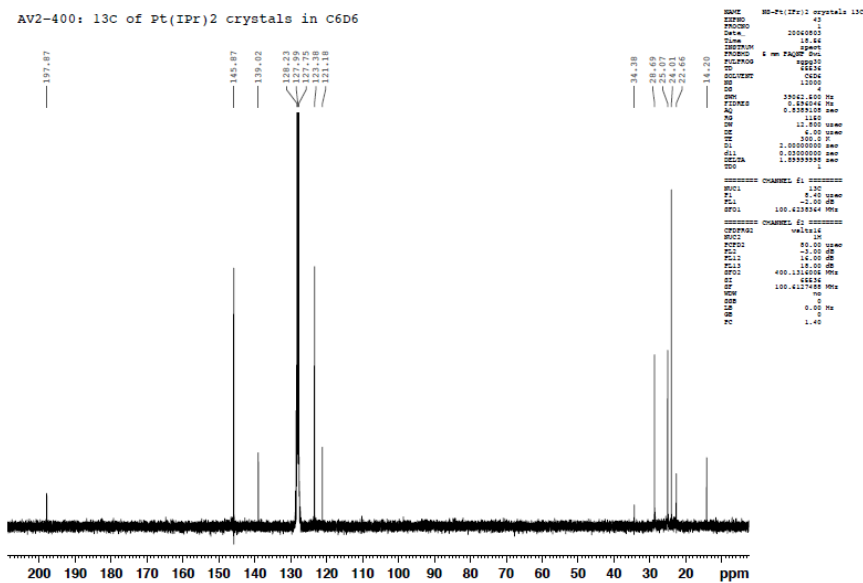
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number of scans: 18432

freq. of 0 ppm: 75.467703 MHz
processed size: 65536 complex points
LB: 1.000 GF: 0.0000
Hz/cm: 647.434 ppm/cm: 8.57814

Pt(IPr)₂ (6) ¹H NMR



¹³C NMR



Buried volume of Me^iPr ligand:

SambVca@Mol.NnaC

SambVca @ Mol.NaC
Results page

S A M B V C A

Buried Volume in Salerno

<http://www.molnac.unisa.it/OM-tools/SambVca>

L. Cavallo et al. email: lcavallo@unisa.it

Molecule from input :

Molecule from input :
./temp/cfca9b5d4c5c4e16f5302ce5366fa247.c3d1

Number of atoms : 33
Atom that is coordinated : 1
Atoms that define the axis : 2
ID of these atoms : 32 33

Radius of sphere (Angs) : 3.500
Distance from sphere (Angs) : 2.050
Mesh step (Angs) : 0.050
H atoms included in the V_{bur} calculation

Cartesian coordinates from input :

Cartesian coordinates from input:

input#		
C	3.48700	7.74700
C	2.65400	9.86200
C	3.50900	9.95900
C	3.99400	9.95900
C	1.57800	

SambVca@Mol.NnaC

H	1.729	
C2	1.99	
C3	1.99	
C	1.99	
N2	1.81	
N3	1.81	
N	1.81	
O	1.78	
F	1.72	
F	2.11	
Cl	2.05	

Coordinates scaled to put the metal at the origin

Coordinates scaled to put the metal at the origin:

C	-0.15721	2.00173
O	2.22081	-0.99021
O	0.48981	-0.99021
C	0.42061	0.34979
O	0.44451	-2.63821
C	-3.54521	2.63973
O	0.67319	-3.26121
C	1.82681	2.26479
O	0.19561	2.91379
C	1.57281	3.05179
O	0.88219	-2.00121
O	0.54381	-2.00121
C	0.43261	1.18279
H	-3.85721	3.54473
H	0.46419	-4.31621
H	0.75819	-3.04821
H	0.51719	-2.63721
H	2.50881	-4.09421
H	1.87381	-3.45321
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H	1.90681	3.86679
H	1.50461	2.47779
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SambVca@Mol.NnaC

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C	5.90900	8.24400
C	1.80300	6.55800
C	0.42600	6.69600
C	2.88100	1.64300
C	1.45500	4.82700
C	1.86600	-0.21300
H	-0.21300	9.29000
H	2.46300	-0.67200
H	2.75700	0.59600
H	3.51600	1.00700
H	-0.51000	-0.45000
H	0.12500	0.19100
H	0.01100	6.45900
H	0.09200	7.51100
H	0.49400	6.12200
H	-0.19400	6.17700
H	3.71200	7.54600
H	3.03100	6.87300
H	2.89500	2.08900
H	1.21200	0.96000
H	0.78600	1.22000
H	2.33400	5.10800
H	2.47900	5.61900
H	1.00600	4.30000
H	1.20200	1.16000
H	1.71100	5.84100
H	2.05000	2.35200
N	1.62600	4.49100
N	1.74600	

Atoms and radius in the parameter file

Atoms and radius in the parameter file

file:///C:/...s/George%20Fortman/Desktop/Nolan/(NCH)2PM62/Xray/Vburr/Equidistance/Pq(dMeIP)2Me2%20with%20H.htm[7/21/2009 1:04:43 PM]
SambVca@Mol.NnaC

H	2.53279	3.23073
H	1.71319	3.90179
H	1.03219	3.28979
H	0.59619	-1.55521
H	0.78681	-2.68421
H	1.21281	-2.42421
H	0.33519	1.46379
H	0.48019	1.97479
H	0.99281	0.65579
H	0.79681	-2.48421
H	0.28781	2.19679
H	0.05119	-1.29221
N	0.37281	0.84679
N	0.25281	0.00000
XX	0.00000	0.00000
XX	0.00000	0.00000

Results : Volumes in Angs^3

Results : Volumes in Angs^3

N of voxels examined :	1436277		
Volume of voxel :	0.1295E-03		
V Free	V Buried	V Total	V Exact
129.419	50.119	179.538	179.534
%V Free	%V_Bur	% Tot/Ex	
72.084	27.916	99.967	

The %V_Bur of your molecule is: 27.9

file:///C:/...s/George%20Fortman/Desktop/Nolan/(NCH)2PM62/Xray/Vburr/Equidistance/Pq(dMeIP)2Me2%20with%20H.htm[7/21/2009 1:04:43 PM] file:///C:/...s/George%20Fortman/Desktop/Nolan/(NCH)2PM62/Xray/Vburr/Equidistance/Pq(dMeIP)2Me2%20with%20H.htm[7/21/2009 1:04:43 PM]

Buried volume of IPr ligand:



Buried volume of nI^tBu ligand:

SambVca@Mol.NnaC

SambVca @ Mol.NaC
Results page

S A M B V C A
 Buried Volume in Salerno
<http://www.molnac.unisa.it/OM-tools/SambVca>
 L. Cavallo et al. email: lcavallo@unisa.it

Molecule from input :
 Molecule from input :
 ./temp/cfca9b5d4c5c4e16f5302ce5366fa247.c3d1

Number of atoms : 33
 Atom that is coordinated : 1
 Atoms that define the axis : 2
 ID of these atoms : 32 33

Radius of sphere (Angs) : 3.500
 Distance from sphere (Angs) : 2.050
 Mesh step (Angs) : 0.050
 H atoms included in the V_{bur} calculation

Cartesian coordinates from input :
 Cartesian coordinates from input :
 input : 8.73100 13.72600
 C 2.77600 7.36900 15.49300
 C -3.29100 8.63400 15.95100
 C -3.32000

SambVca@Mol.NnaC

H	1.29		
C2	1.99		
C3	1.99		
C	1.99		
N2	1.81		
N3	1.81		
N	1.81		
O	1.78		
F	1.72		
P	2.11		
Cl	2.05		

Coordinates scaled to put the metal at the origin
 Coordinates scaled to put the metal at the origin

C	-0.68465	1.83830	-
O	0.48197	-2.04665	3.60530
O	0.39697	-2.04665	3.60530
C	0.02597	-0.78165	4.06330
O	0.50297	1.52635	3.16430
C	1.36535	4.50930	-
C	0.09697	1.78935	3.17030
C	1.00903	2.30735	2.05830
C	1.21997	-3.30665	1.51730
C	0.56697	-3.17365	0.08030
O	0.08597	-3.97565	1.84630
C	1.95397	-4.18165	2.26430
O	0.44503	-2.83065	4.11030
H	1.17297	-0.50465	4.95030
H	1.22297	1.57435	5.24030
H	0.57297	2.94235	4.57530
H	1.07097	1.69935	4.57230
H	2.02497	1.51135	2.31230
H	1.39303	2.74535	3.30730
H	1.17403	1.27735	3.89530
H	1.42603	2.10135	2.08130
H	2.17997		

file:///C:/.../George%20Fortman/Desktop/Nolan/(NCH)2PMe2/xy/Vbur/Equidistance/Pt(tBu)2Me2%20with%20H.htm[7/21/2009 1:04:46 PM] file:///C:/.../George%20Fortman/Desktop/Nolan/(NCH)2PMe2/xy/Vbur/Equidistance/Pt(tBu)2Me2%20with%20H.htm[7/21/2009 1:04:46 PM]

SambVca@Mol.NnaC

C	10.94200	15.05200	
C	-2.79700	11.38100	16.39700
C	-3.39100	11.20500	15.05800
C	-1.28500	11.72300	13.94600
C	-3.51400	8.10900	13.40500
C	-2.86100	8.24200	11.96800
C	-4.24800	5.44000	13.43400
C	-1.84900	5.23400	14.15200
H	6.58500	5.58500	15.99800
H	3.46700	8.91100	16.83800
H	-3.51700	10.39000	17.12800
H	-2.86700	12.35800	16.46300
H	-3.36500	11.07500	16.46000
H	-4.31900	10.92700	14.20000
H	-0.90100	12.16100	15.19500
H	-1.12000	10.69300	15.78300
H	-0.86800	11.51700	13.96900
H	-4.47200	12.68400	14.08700
H	-3.38300	11.46700	13.07400
H	-3.14600	8.60700	11.96000
H	-1.47000	5.36100	11.53900
H	-2.38300	6.84500	11.47800
H	-2.97700	6.00500	12.95800
H	-4.89200	4.56400	12.99800
H	-4.19800	5.32700	14.36400
H	-4.53700	5.12900	15.08500
H	-2.13200	4.35200	13.72500
H	-1.80000	5.65900	14.12500
H	-0.96600	9.46300	14.87100
N	-3.00700	7.41800	14.14100
N	-2.95400		

Atoms and radius in the parameter file
 Atoms and radius in the parameter file

SambVca@Mol.NnaC

H	3.26635	2.19930	-
H	1.08897	2.05135	1.18630
H	0.85197	-2.80865	0.07230
H	0.82403	-4.05465	-0.34870
H	0.08897	-2.57065	-0.40970
H	0.68297	-3.41065	1.07030
H	2.59797	-4.85165	1.11030
H	1.90397	-4.08865	2.47630
H	2.24297	-4.28665	3.19730
H	0.16203	-5.06365	1.83730
H	0.49403	-3.75665	2.23730
H	1.32803	0.04735	2.98330
N	0.71297	-1.99765	2.25330
N	0.65997	0.00000	0.00000
XX	0.00000	0.00000	0.00000
0.00000			

Results : Volumes in Angs^3
 Results : Volumes in Angs^3
 N of voxels examined : 1436277
 Volume of voxel : 0.125E-03

Y Free	V Buried	V Total	V Exact
113.085	65.449	179.535	179.534
62.988	37.012	99.967	

The %V_Bur of your molecule is: 37.0

file:///C:/.../George%20Fortman/Desktop/Nolan/(NCH)2PMe2/xy/Vbur/Equidistance/Pt(tBu)2Me2%20with%20H.htm[7/21/2009 1:04:46 PM] file:///C:/.../George%20Fortman/Desktop/Nolan/(NCH)2PMe2/xy/Vbur/Equidistance/Pt(tBu)2Me2%20with%20H.htm[7/21/2009 1:04:46 PM]

Buried volume of ^{Me}IME:

SambVca@MoLNac

**SambVca @ MoLNac
Results page**

S A M B V C A
Buried Volume in Salerno
<http://www.molnac.unisa.it/CM-tools/SambVca>
D. Cavallo et al. email: lcavallo@unisa.it

Molecule from input :
Molecule from input :
./temp/cfca9b5d4c5c4e16f5302ce5366fa247.c3d1

Number of atoms : 21
Atom that is coordinated : 1
Atoms that define the axis : 2
ID of these atoms : 20 21

Radius of sphere (Angs) : 3.500
Distance from sphere (Angs) : 2.100
Mesh step (Angs) : 0.050
H atoms included in the V_bur calculation

Cartesian coordinates from input :

input	x	y	z
C	5.27200	1.96000	
4.31300			
C	6.16600	4.07200	
4.39000			
C	4.83200	4.20200	
4.17300			

SambVca@MoLNac

Coordinates scaled to put the metal at the origin

C	0.23819	2.08601	-
0.04297			
C	1.13219	4.19801	-
0.03403			
C	-0.20181	4.32801	-
0.18497			
C	2.67719	2.23501	-
0.28903			
C	2.20619	5.22301	-
0.14303			
C	-1.07981	5.53501	-
0.31297			
C	-2.11881	2.71901	-
0.49697			
H	3.25919	2.49501	-
0.45397			
H	3.06819	2.54101	-
1.13503			
H	2.58519	1.25801	-
0.30703			
H	1.80219	6.11501	-
0.18803			
H	2.73219	5.06101	-
0.95403			
H	2.78019	5.16901	-
0.64197			
H	-0.54281	6.34401	-
0.18497			
H	-1.48681	5.54901	-
1.20497			
H	-1.78481	5.50101	-
0.36603			
H	-2.23181	1.74701	-
0.53697			
H	-2.67381	3.08301	-
0.22503			
H	-2.39281	3.11601	-
1.34997			
H	1.38419	2.83301	-
0.11803			
H	-0.71381	3.03801	-
0.23897			
%V	0.00000	0.00000	
0.00000			

Results : Volumes in Angs^3

Results : Volumes in Angs^3

N of voxels examined	Volume of voxel	V Free	V Buried	V Total	V Exact
1436277	0.1225E+03	133.833	45.702	179.535	179.594

%V_Free %V_Bur % Tot/Ex

file:///C:/...ings/George%20Fortman/Desktop/Nolan/(NCH)2PM2/xy/Vbur/Equidistance/Pt[1Me]2Me%20with%20H.htm[7/21/2009 1:04:46 PM] SambVca@MoLNac

C	7.71100	2.10900	
4.64500			
C	7.24000	5.09700	
4.49900			
C	3.95400	5.40900	
4.04300			
C	2.91500	2.59300	
3.85900			
H	8.29300	2.36900	
3.90200			
H	8.10200	2.41500	
5.49100			
H	7.61900	1.13200	
4.66300			
H	6.83600	5.98900	
4.54400			
H	7.76600	4.93500	
5.31000			
H	7.82400	5.04300	
3.71400			
H	4.49100	6.21800	
4.17100			
H	3.54700	5.42300	
3.15100			
H	3.24900	5.37500	
4.72200			
H	2.80200	1.62100	
3.81900			
H	2.36000	2.85700	
4.58100			
H	2.64100	2.99000	
3.00600			
N	6.41800	2.70700	
4.47400			
N	4.32000	2.91200	
4.11700			

Atoms and radius in the parameter file

Atoms and radius in the parameter file

H	1.29
C2	1.99
C3	1.99
C	1.99
N2	1.81
N3	1.81
N	1.81
O	1.78
P	1.72
Cl	2.11
	2.05

Coordinates scaled to put the metal at the origin

SambVca@MoLNac

74.544	25.456	99.967
--------	--------	--------

The %V_Bur of your molecule is: 25.5

file:///C:/...ings/George%20Fortman/Desktop/Nolan/(NCH)2PM2/xy/Vbur/Equidistance/Pt[1Me]2Me%20with%20H.htm[7/21/2009 1:04:46 PM] SambVca@MoLNac

Buried volume of at^1Bu ligand:

SambVca@Mol.NnaC

SambVca Results for MeIPr in cis-Pt(MeIPr)2Me2 (1)

**SambVca @ Mol.NnaC
Results page**

S A M B V C A
Buried Volume in Salerno
http://www.molnac.unisa.it/CM-tools/SambVca
L. Cavallo et al. email: lcavallo@unisa.it

Molecule from input :
 Molecule from input :
 ./temp/fad3e8cbbf91d363bb5af2ecbe9904a.c3d1

Number of atoms : 33
 Atom that is coordinated : 1
 Atoms that define the axis : 2
 ID of these atoms : 32 33
 Radius of sphere (Angs) : 3.500
 Distance from sphere (Angs) : 2.040
 Mesh step (Angs) : 0.050
 H atoms included in the V_{bur} calculation

Cartesian coordinates from input :
 Cartesian coordinates from input :
 input : 3.48700 7.74700
 C 1.77800 2.65400 9.86200
 C 1.50900 3.99400 9.35900
 C 1.57800

file:///C:/Documents and Settings/George Fortman/Desktop/Nolan/(NCH)2PM2/Xray/Vbur/atlBu-standard.htm(8/7/2009 10:28:05 AM)
 SambVca@Mol.NnaC

C	7.55000	15.35000
H	5.62200	8.28400
C	6.79400	8.17300
C	4.27400	7.51900
C	5.79800	15.28500
C	5.79600	2.36300
C	7.29000	1.61600
C	5.30100	2.37000
C	4.99200	5.06500
H	5.83700	3.85200
H	5.32800	7.84800
H	7.63200	9.21300
H	6.80000	8.26000
H	6.70800	8.02000
H	4.07300	9.13600
H	4.30500	7.76200
H	3.57700	7.02600
H	5.05300	8.43500
H	5.81000	7.07300
H	6.64200	2.55900
H	7.78000	1.43500
H	7.45800	2.96600
H	7.59300	1.80600
H	4.35900	0.69200
H	5.39000	1.73900
H	5.83300	3.01200
H	5.29500	1.45700
H	5.12500	2.51400
H	4.04000	6.16000
N	5.61300	3.98400
N	5.61200	14.81900

Atoms and radius in the parameter file
 Atoms and radius in the parameter file

file:///C:/Documents and Settings/George Fortman/Desktop/Nolan/(NCH)2PM2/Xray/Vbur/atlBu-standard.htm(8/7/2009 4:59:08 PM)

SambVca@Mol.NnaC

H	1.29	
C2	1.99	
C3	1.99	
C	1.99	
N2	1.81	
N3	1.81	
N	1.81	
O	1.78	
F	1.72	
SI	2.45	
P	2.11	
S	2.10	
Cl	2.05	

Coordinates scaled to put the metal at the origin
 Coordinates scaled to put the metal at the origin

C	-1.90611	3.77389
C	-1.16611	1.63689
C	-2.54511	1.70789
C	0.67989	3.54189
C	1.31389	2.91289
C	1.20289	3.20389
C	0.54889	5.06689
C	-4.40611	3.47689
C	-4.60711	3.71189
C	-5.35411	2.39889
C	-4.60011	4.76589
C	-1.90511	4.71589
H	-3.11811	0.95789
H	0.87789	3.17689
H	2.24289	3.21889
H	1.56790	1.28989
H	1.47590	1.04989
H	1.15910	2.25689
H	0.92710	2.16589
H	1.65510	0.79189

file:///C:/Documents and Settings/George Fortman/Desktop/Nolan/(NCH)2PM2/Xray/Vbur/atlBu-standard.htm(8/7/2009 4:59:08 PM)

SambVca@Mol.NnaC

H	0.05589	5.47089
H	0.17910	1.46489
H	0.57790	0.10289
H	1.40990	-4.41111
H	2.54790	-5.53511
H	2.22590	-4.00411
H	2.36900	-5.16411
H	0.87310	-6.27811
H	0.15790	-5.23111
H	0.60090	-3.95811
H	0.06290	-5.51311
H	0.10710	-4.45611
H	1.19210	-0.81011
N	0.38090	-2.98611
N	0.37990	0.00000
XZ	0.00000	0.00000

Results : Volumes in Angs^3
 Results : Volumes in Angs^3
 N of voxels examined : 1436277
 Volume of voxel : 0.125E-03

V Free	V Buried	V Total	V Exact
126.821	52.714	179.535	179.594

%V Free	%V Bur	% Tot/Ex
70.639	29.361	99.967

The V_{bur} of your molecule is: 29.4

file:///C:/Documents and Settings/George Fortman/Desktop/Nolan/(NCH)2PM2/Xray/Vbur/atlBu-standard.htm(8/7/2009 4:59:08 PM)

References:

- [1] N. Kuhn, T. Kratz, *Synthesis* **1993**, 1993, 561.
- [2] A. J. Arduengo, R. Krafczyk, R. Schmutzler, H. A. Craig, J. R. Goerlich, W. J. Marshall, M. Unverzagt, *Tetrahedron* **1999**, 55, 14523.
- [3] L. Jafarpour, E. D. Stevens, S. P. Nolan, *Journal of Organometallic Chemistry* **2000**, 606, 49.
- [4] N. M. Scott, R. Dorta, E. D. Stevens, A. Correa, L. Cavallo, S. P. Nolan, *Journal of the American Chemical Society* **2005**, 127, 3516.
- [5] Reactivity of IMes depends on the source of Platinum-dimethyl precursor used. Reactions run with non-commercial (COD)Pt(Me)₂ gave faster conversion of complex **4** to **5**. More detailed discussion and possible causes are now being investigated.
- [6] *SAINTE* and *SADABS*, **2007**, Bruker AXS Inc., Madison, Wisconsin, USA.
- [7] *SHELX97*, G. M. Sheldrick, *Acta Crystallogr. Sect. A*, **2008**, 64, 112-122.
- [8] H. D. Flack, *Acta Crystallogr. Sect. A*, **1983**, 39, 876-881.
- [9] F. L. Hirshfeld, *Acta Crystallogr. Sect. A*, **1976**, 32, 239-244.

Table S2. Summary of crystallographic data for complexes **1-7**

Complex	1	2	3	4	5	6	7
Chemical Formula	C ₁₆ H ₃₀ N ₄ Pt	C ₃₂ H ₅₄ N ₄ Pt + CH ₂ Cl ₂	C ₂₄ H ₄₆ N ₄ Pt	C ₄₄ H ₅₄ N ₄ Pt	C ₄₃ H ₅₀ N ₄ Pt	C ₅₄ H ₇₂ N ₄ Pt	C ₂₄ H ₄₆ N ₄ Pt
Formula Weight (M)	473.44	774.81	585.74	834.00	817.96	972.25	585.74
Crystal system	monoclinic	monoclinic	monoclinic	monoclinic	monoclinic	monoclinic	monoclinic
Unit-cell dimensions, V _{cell}	a=14.8926(2) Å b=8.1788(1) Å c=16.1119(2) Å β=113.5403(8)° V= 1799.17(4) Å ³	a=15.5185(7) Å b=12.9037(2) Å c=17.4636(8) Å β=105.354(1)° V= 3372.2(3) Å ³	a=15.9482(6) Å b=9.6101(3) Å c=17.6283(6) Å β=105.619(1)° V= 2601.0(2) Å ³	a=16.7156(10) Å b=13.1785(8) Å c=17.6422(10) Å β=99.411(1)° V= 3834.0(4) Å ³	a=10.7787(5) Å b=20.0707(9) Å c=17.2875(8) Å β=91.064(1)° V= 3739.3(3) Å ³	a=12.8035(7) Å b=26.9007(16) Å c=14.1517(8) Å β=92.184(1)° V= 4870.6(5) Å ³	a=12.0260(9) Å b=14.6685(11) Å c=14.7642(10) Å β=91.103(1)° V= 2604.0(3) Å ³
Temperature	160(1) K	150(2) K	150(2) K	150(2) K	150(2) K	150(2) K	173(2)
Space group	P2 ₁ /n	Cc	P2 ₁ /c	P2 ₁ /n	P2 ₁ /n	P2 ₁ /n	P2 ₁ /n
Formula units/unit cell (Z)	4	4	4	4	4	4	4
N _{ref} (measured)	50417	27974	41062	40209	53108	63794	31831
N _{ref} (independent)	5253	10221	7606	11107	9017	10593	6239
R _{int}	0.0753	0.0195	0.0336	0.0373	0.0706	0.0478	0.0706
2θ _{max}	60.0	61.0	60.0	60.0	56.0	54.0	56.0
R _F ²	0.0439	0.0252	0.0223	0.0420	0.0548	0.0584	0.0584
R _F ² (I > 2σ _I)	0.0375	0.0196	0.0187	0.0312	0.0422	0.0475	0.0325
CCDC no.	741809	741810	741808	741811	741812	741813	756172