

**Nanoporous organic layered crystals of double-headed bis(trifluorolactate)s.
Hydrogen-bonded systematic crystal structures controlled by the
symmetries of molecular components.**

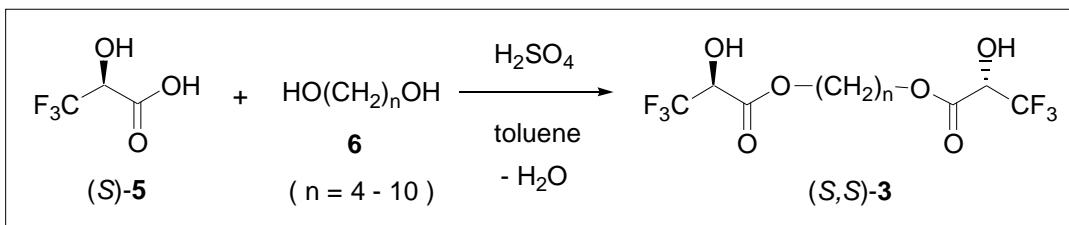
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Electronic Supplementary Information (ESI)

- (i) Compound data for (*S,S*)-**3**
- (ii) ORTEP plots for **3**
- (iii) Additional figures of **3**
- (iv) ^1H NMR spectra of dissolved crystals **3f**, **3g**, and **4f** in CDCl_3
- (v) TG curves of inclusion compounds **3f**, **3g**, and **4f**

Synthesis of double-headed (*S,S*)-bis(trifluorolactate)s 3



Typical synthetic procedure for **3** and the compound data for **3a** are described in the experimental section.

Pentamethylene-(*S,S*)-bis(3,3,3-trifluorolactate) [(*S,S*)-3b]: 78% yield. Colourless crystal. Mp 100-101 °C. IR (KBr): 3432, 2968, 1754, 1742, 1462, 1134 cm⁻¹. ¹H NMR (CDCl₃): δ 4.48 (quintet, *J* = 7.2 Hz, 2H), 4.39 (dt, *J*₁ = 10.8 Hz, *J*₂ = 6.6 Hz, 2H), 4.29 (dt, *J*₁ = 10.8 Hz, *J*₂ = 6.6 Hz, 2H), 3.47 (d, *J* = 7.8 Hz, 2OH), 1.71-1.82 (m, 4H), 1.41-1.53 (m, 2H). ¹⁹F NMR (CDCl₃): δ 85.0 (d, *J* = 7.2 Hz, 6F). ¹³C NMR (CD₃OD): δ 168.2, 124.4 (q, *J* = 281 Hz), 71.2 (q, *J* = 32 Hz), 67.1, 28.9, 23.0. MS: m/z 145 (1), 99 (4), 69 (100), 68 (17), 51(5), 41(49); Anal. Calcd for C₁₁H₁₄F₆O₆: C 37.09, H, 3.96, Found: C 36.83, H, 4.19. [α]²⁵_D -6.61 (c 1.0, acetone).

Hexamethylene-(*S,S*)-bis(3,3,3-trifluorolactate) [(*S,S*)-3c]: 80% yield. Colourless crystal. Mp 113-114 °C. IR (KBr): 3444, 2980, 1750, 1738, 1228, 1130 cm⁻¹. ¹H NMR (CDCl₃): δ 4.48 (q, *J* = 7.2 Hz, 2H), 4.36 (dt, *J*₁ = 10.8 Hz, *J*₂ = 6.6 Hz, 2H), 4.30 (dt, *J*₁ = 10.8 Hz, *J*₂ = 6.6 Hz, 2H), 1.65-1.80 (m, 4H), 1.35-1.50 (m, 4H), OHs were not observed. ¹⁹F NMR (CDCl₃): δ 85.0 (d, *J* = 6.9 Hz, 6F). ¹³C NMR (CDCl₃): δ 167.5, 122.2 (q, *J* = 282 Hz), 69.8 (q, *J* = 32.5 Hz), 67.3, 28.1, 25.0. MS: m/z 145 (2), 127 (5), 117 (1), 99 (9), 83 (80), 67 (34), 55 (100); Anal. Calcd for C₁₂H₁₆F₆O₆: C 38.93, H 4.36, Found: C 38.8, H 4.5. [α]²⁵_D -5.80 (c 1.1, acetone).

Heptamethylene-(*S,S*)-bis(3,3,3-trifluorolactate) [(*S,S*)-3d]: 92% yield. Colourless crystal. Mp 93-94 °C. IR (KBr): 3448, 2952, 2872, 1750, 1740, 1270, 1190, 1132 cm⁻¹. ¹H NMR (CDCl₃): δ 4.48 (quintet, *J* = 6.6 Hz, 2H), 4.35 (dt, *J*₁ = 10.8 Hz, *J*₂ = 6.6 Hz, 2H), 4.29 (dt, *J*₁ = 10.5 Hz, *J*₂ = 6.6 Hz, 2H), 3.51 (d, *J* = 6.9 Hz, 2OH), 1.6-1.8 (m, 4H), 1.25-1.45 (m, 6H). ¹⁹F NMR (CDCl₃): δ 85.0 (d, *J* = 6.9 Hz, 6F). ¹³C NMR (CDCl₃): δ 167.5, 122.2 (q, *J* = 284 Hz), 69.8 (q, *J* = 32.1 Hz), 67.5, 28.4, 28.1, 25.3. MS: m/z 141 (2), 99 (4), 98 (4), 97 (48), 81 (11), 69 (19), 68 (29), 67 (19), 55

(100), 41 (25); Anal. Calcd for C₁₃H₁₈F₆O₆: C 40.63, H, 4.72, Found: C 40.79, H, 5.0. [α]²⁵_D -4.33 (c 1.1, acetone).

Octamethylene-(S,S)-bis(3,3,3-trifluorolactate) [(S,S)-3e]: 90% yield. Colourless crystal. Mp 71-72 °C. IR (KBr): 3452, 2948, 1750, 1734, 1224, 1198, 1134 cm⁻¹. ¹H NMR (CDCl₃): δ 4.47 (quintet, J = 7.2 Hz, 2H), 4.36 (dt, J₁ = 10.5 Hz, J₂ = 6.6 Hz, 2H), 4.30 (dt, J₁ = 10.8 Hz, J₂ = 6.6, 2H), 3.41 (d, J = 7.8 Hz, 2OH), 1.71 (quintet, J = 6.6 Hz, 4H), 1.35 (brs, 8H). ¹⁹F NMR (CDCl₃): δ 85.0 (d, J = 6.9 Hz, 6F). ¹³C NMR (CDCl₃): δ 167.5, 122.2 (q, J = 284 Hz), 69.7 (q, J = 33.3 Hz), 67.6, 28.7, 28.1, 25.3. MS: m/z 145 (1), 129 (1), 111 (21), 110 (12), 98 (8), 95 (7), 83 (10), 82 (41), 81 (22), 69 (100), 68 (31), 67 (42), 56 (10), 55 (84), 54 (37), 43 (16), 42 (16), 41 (62); Anal. Calcd for C₁₄H₂₀F₆O₆: C 42.22, H 5.06, Found: C 42.0, H 5.29. [α]²⁵_D -4.38 (c 1.1, acetone). The ee was determined to be >99% ee by HPLC analysis of the dibenzoate ester (Daicel Chiralcel® OD-H, 100:1 hexane:*i*-PrOH, 1.0 mL / min, 254 nm, *t*_{ss}(major) = 13.3 min, *t*_{RS}(minor) = 15.6 min, *t*_{RR}(minor) = 19.1 min).

Nonamethylene-(S,S)-bis(3,3,3-trifluorolactate) [(S,S)-3f]: 71% yield. Colourless crystal. Mp 48-49 °C. IR (KBr): 3444, 2940, 1752, 1740, 1344, 1280, 1228, 1132 cm⁻¹. ¹H NMR (CDCl₃): δ 4.48 (quintet, J = 7.2 Hz, 2H), 4.35 (dt, J₁ = 10.5 Hz, J₂ = 6.6 Hz, 2H), 4.29 (dt, J₁ = 10.5 Hz, J₂ = 6.6 Hz, 2H), 3.44 (d, J = 7.8 Hz, 2OH), 1.71 (quintet, J = 6.9 Hz, 4H), 1.25-1.40 (brs, 10H). ¹⁹F NMR (CDCl₃): δ 85.0 (d, J = 6.9 Hz, 6F). ¹³C NMR (CDCl₃): δ 167.5, 122.2 (q, J = 281 Hz), 69.8 (q, J = 33.2 Hz), 67.7, 29.1, 28.8, 28.2, 25.4; MS: m/z 145 (1), 125 (5), 124 (6), 96 (23), 95 (15), 83 (41), 82 (34), 81 (23), 69 (100), 68 (37), 67 (34), 57 (19), 56 (13), 55 (84), 54 (26), 43 (15), 42 (11), 41 (45); Anal. Calcd for C₁₅H₂₂F₆O₆: C 43.69, H 5.38, Found: C 43.72, H 5.34. [α]²⁵_D -3.93 (c 1.1, acetone).

Decamethylene-(S,S)-bis(3,3,3-trifluorolactate) [(S,S)-3g]: 90% yield. Colourless crystal. Mp 56 °C. IR (KBr): 3440, 1752, 1748, 1467, 1344, 1224, 1190, 1130 cm⁻¹. ¹H NMR (CDCl₃): δ 4.47 (q, J = 6.9 Hz, 2H), 4.35 (dt, J₁ = 10.8 Hz, J₂ = 6.6 Hz, 2H), 4.29 (dt, J₁ = 10.5 Hz, J₂ = 6.6 Hz, 2H), 3.41 (d, J = 7.5 Hz, 2OH), 1.71 (quintet, J = 7.2 Hz, 4H), 1.25-1.43 (brs, 12H). ¹⁹F NMR

(CDCl₃): δ 85.0 (d, *J* = 6.9 Hz, 6F). ¹³C NMR (CDCl₃): δ 167.6, 122.3 (q, *J* = 282 Hz), 69.8 (q, *J* = 33.3 Hz), 67.7, 29.2, 28.9, 28.2, 25.4. MS: m/z 145 (2), 138 (4), 110 (14), 97 (22), 96 (22), 95 (17), 83 (60), 82 (37), 81 (25), 69 (57), 68 (36), 67 (32), 57 (22), 56 (15), 55 (100), 54 (28), 43 (21), 42 (13), 41 (53); Anal. Calcd for C₁₆H₂₄F₆O₆: C 45.07, H 5.67, Found: C 45.2, H 5.57. [α]²⁵_D -3.78 (*c* 1.1, acetone).

Fig. 1S ORTEP plots for **3** at 50% probability level. Disordered moieties are omitted for clarity.

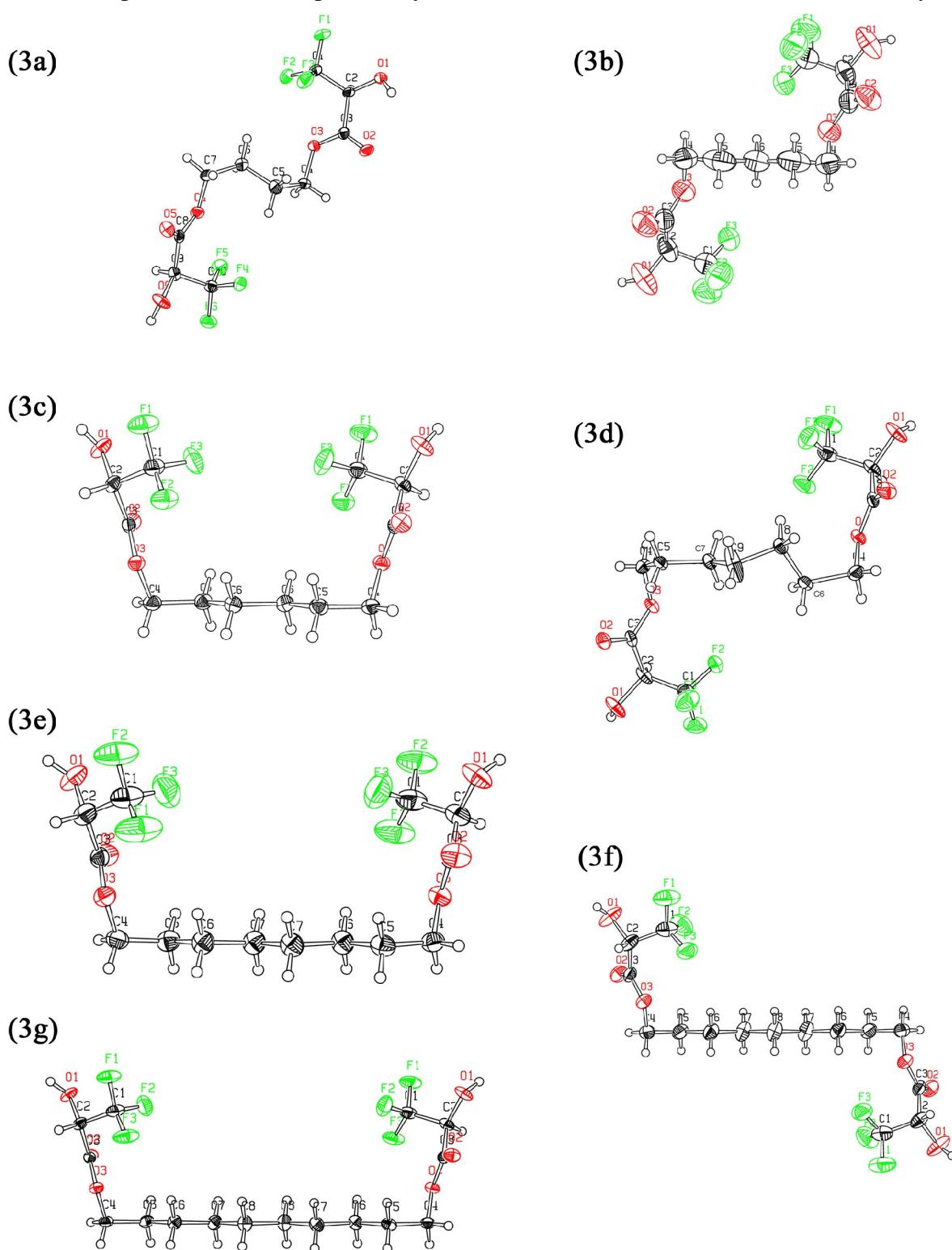


Fig. 2S Crystal structure of **3**. H-atoms of methylene groups and disordered moieties are omitted for clarity. Hash bonds are hydrogen bonds.

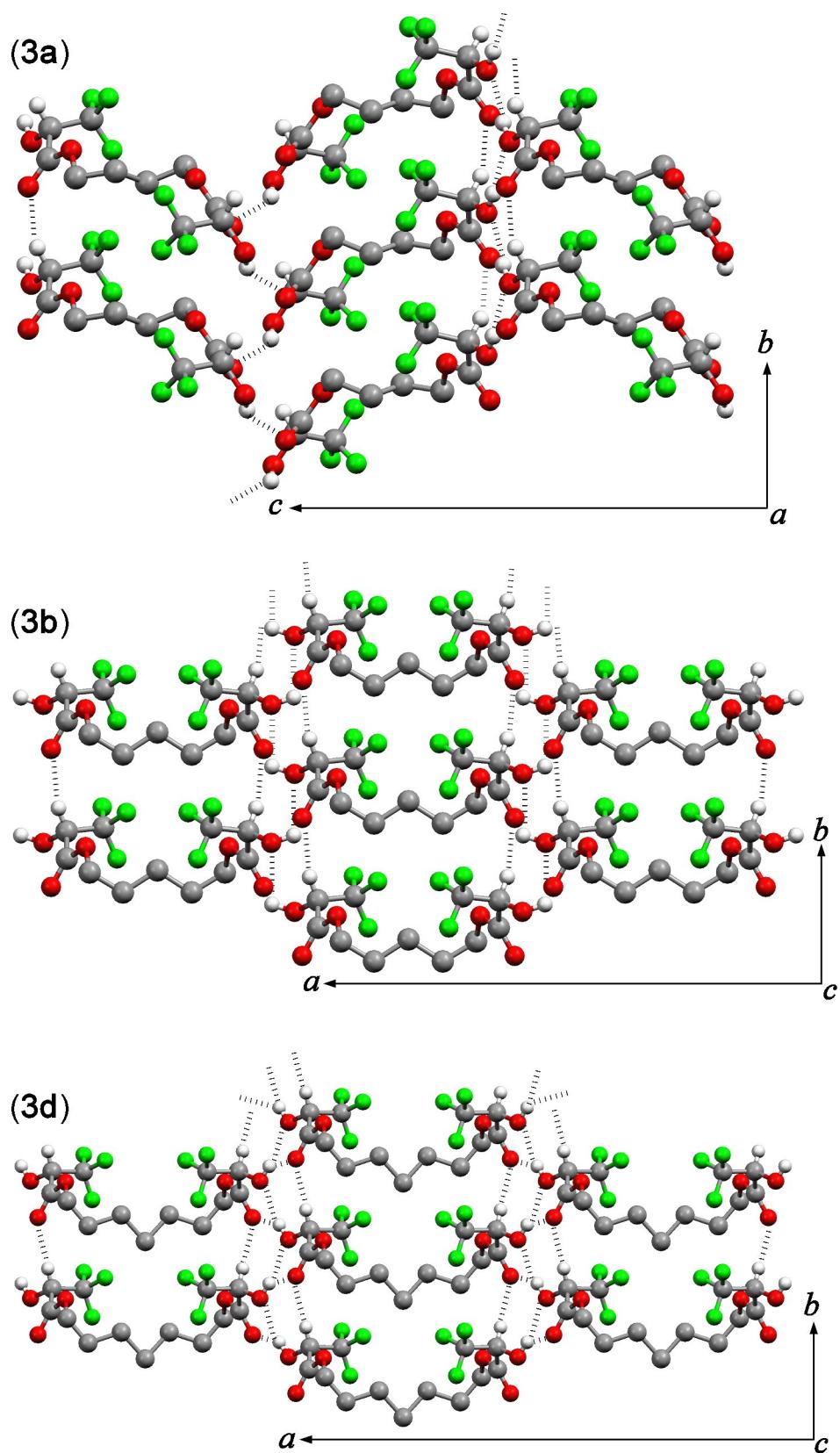


Fig. 2S Crystal structures of **3** (continued).

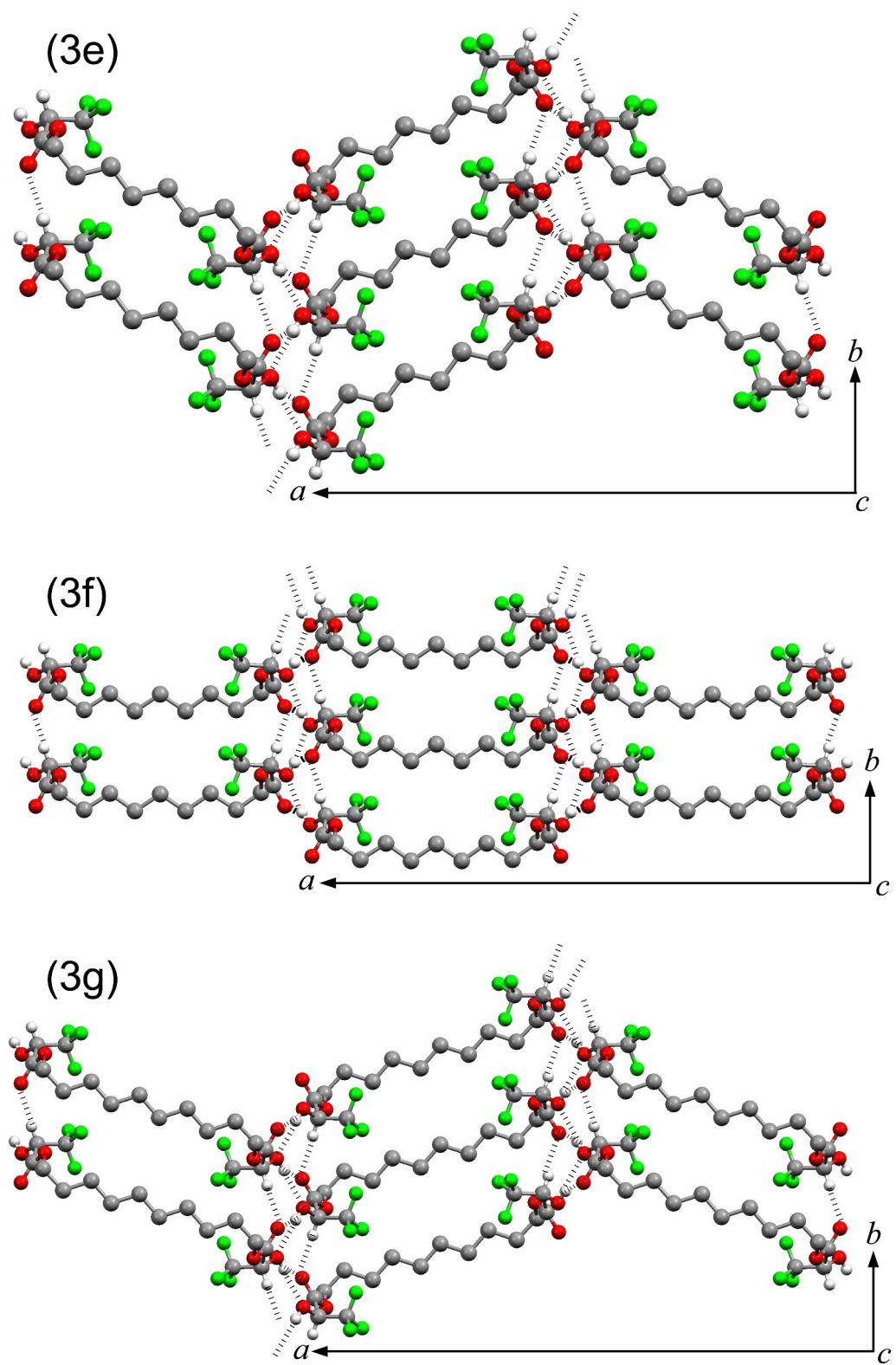


Fig. 3S ^1H NMR chart of **3f** which crystallized from octane/THF (in CDCl_3)

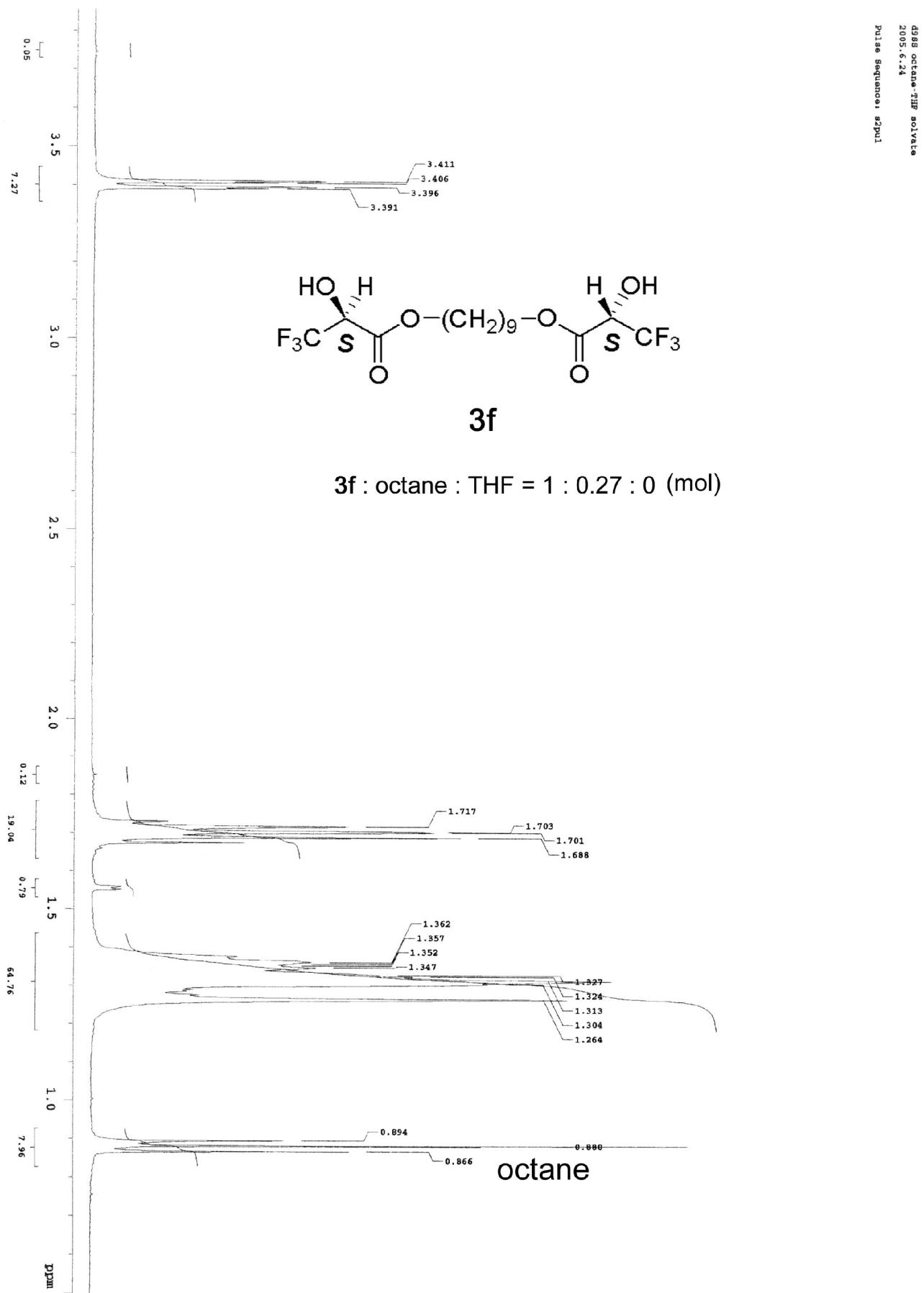


Fig. 4S ^1H NMR spectra of **4f** which crystallized from hexane/ether (in CDCl_3)

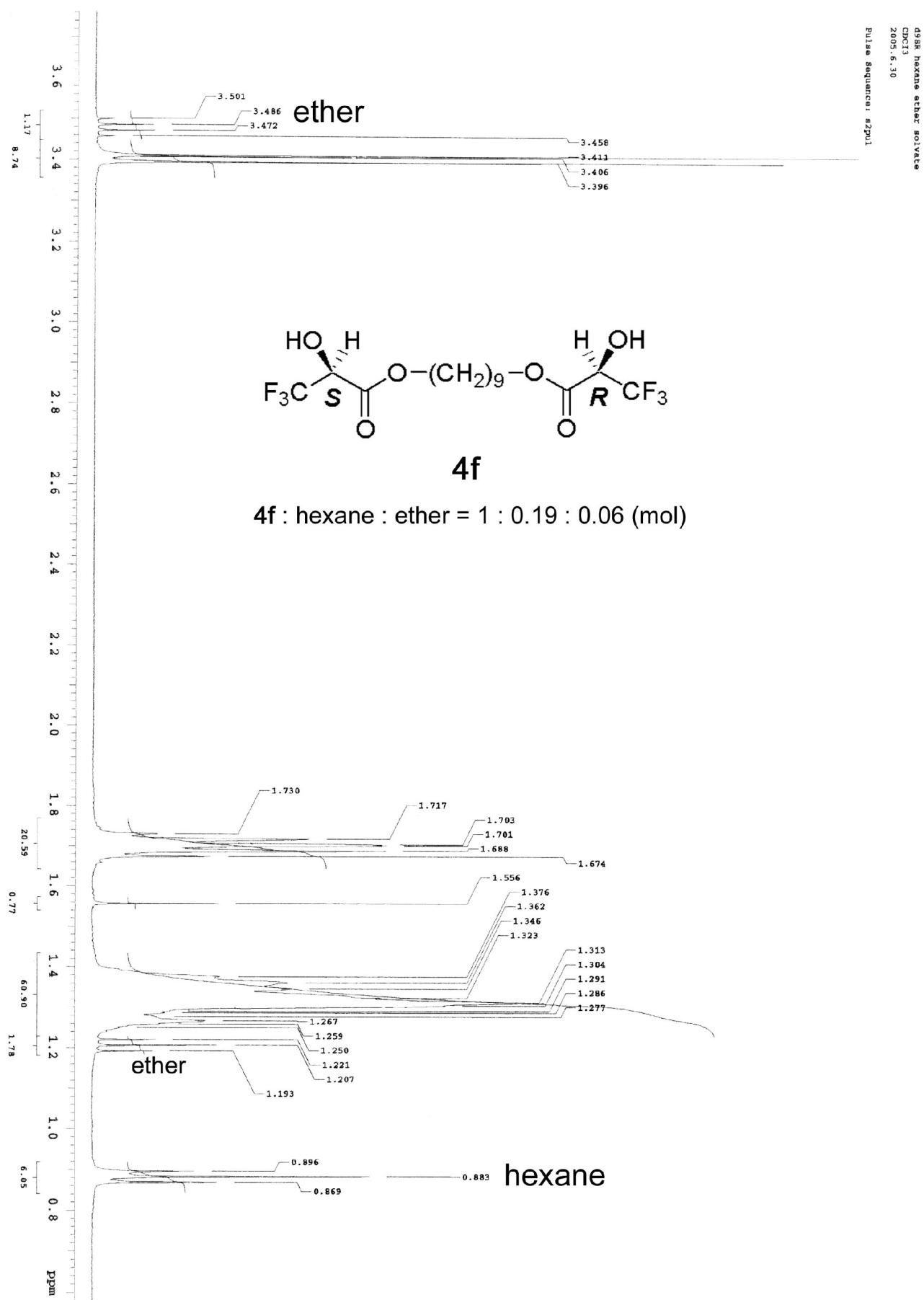


Fig. 5S ^1H NMR spectra of **3g** which crystallized from hexane/ether (in CDCl_3)

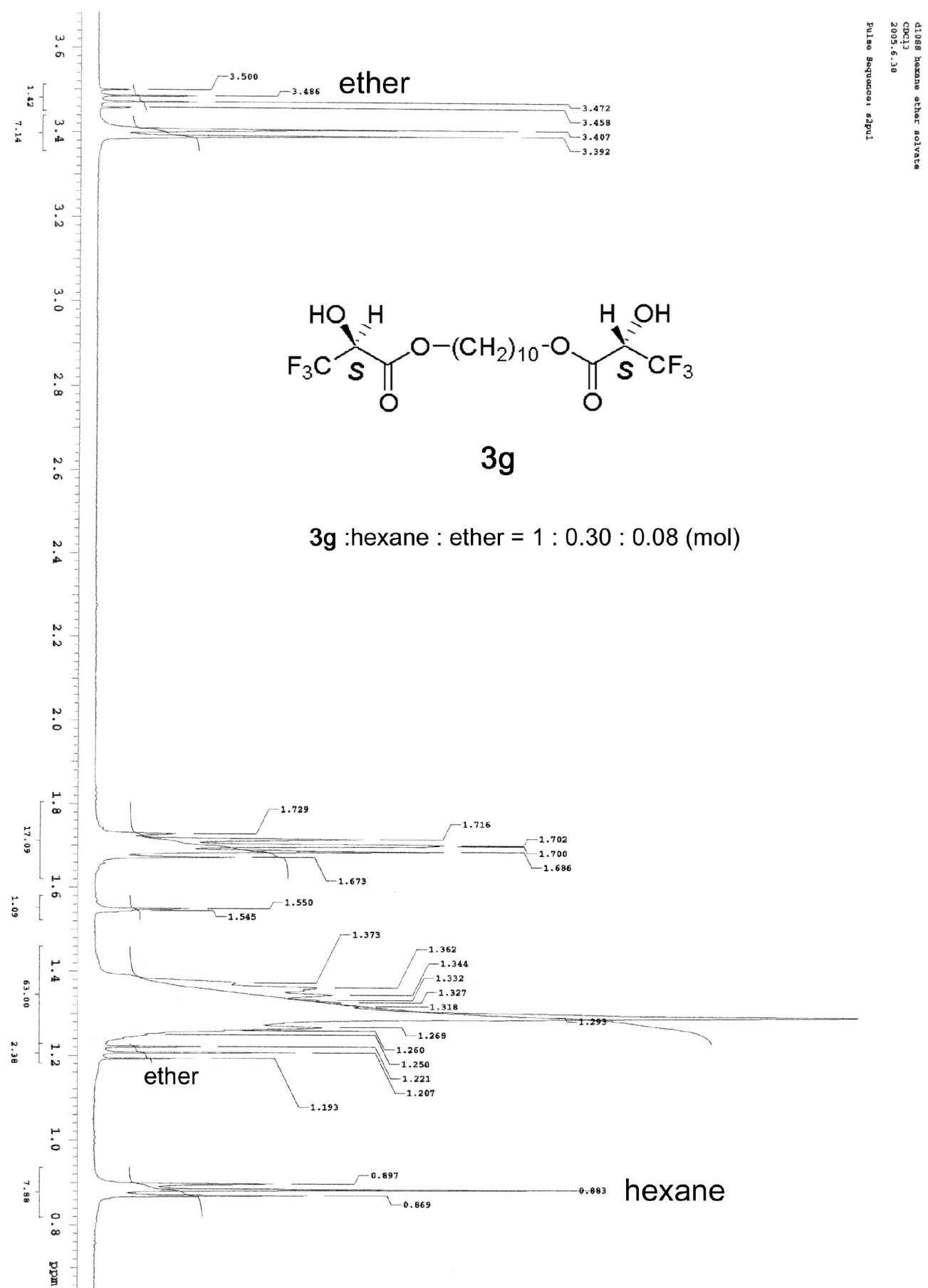


Fig. 6S TG curves for **3f**, **4f** and **3g** crystals which included some crystallization solvents.

Undesirable decomposition of the fluorine compounds with generation of heat was occurred over 140 °C, so TG analysis was stopped at 145 °C.

