Electronic supplementary information (ESI)

A Series of 2D to 3D Metal Organic Frameworks Based on Different Polycarboxylate Anions and Flexible 2,2'-bis(1*H*-imidazolyl)ether Ligand

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Ni(1)-N(1)	2.0344(18)	Ni(1)-O(2)	2.0843(18)
Ni(1)-O(1)	2.1644(14)		
$N(1)-Ni(1)-N(1)^{#2}$	88.24(10)	N(1)-Ni(1)-O(2)	95.93(7)
N(1)-Ni(1)-O(1)	157.98(6)	O(2)-Ni(1)-O(1)	62.09(5)
O(1)-Ni(1)-O(1) ^{#2}	93.72(8)	O(2)-Ni(1)-O(2) ^{#3}	161.07(8)

Table S1. Selected bond distances (Å) and angles (°) for 1.

Symmetry codes for 1: ^{#2} -x,y,3/2-z; ^{#3} 1+x,-y,1/2+z

 Table S2. Selected bond distances (Å) and angles (°) for 2.

Cu(1)-N(1)	1.965(3)	Cu(1)-O(1)	1.986(3)
Cu(2)-O(5)	1.982(2)	Cu(2)-N(4)	2.000(3)
N(1)-Cu(1)-N(1) ^{#1}	180	N(1)-Cu(1)-O(1)	89.83(12)
O(1)-Cu(1)-O(1) ^{#1}	180	O(5) ^{#3} -Cu(2)-O(5)	180
O(5)-Cu(2)-N(4)	89.46(11)	$N(4)$ - $Cu(2)$ - $N(4)^{#3}$	180
Symmetry adds for	2 , ^{#1} 2 u 2 u 2 u 2	^{#3} 1 x 1 x 2 -	

Symmetry codes for **2**: ^{#1} 2-x,2-y,2-z; ^{#3} 1-x,1-y,3-z

Table S3. Selected bond distances (Å) and angles (°) for 3.

Cu(1)-O(1)	1.991(2)	Cu(1)-N(4) ^{#1}	1.991(3)
Cu(2)-O(4)	1.978(2)	Cu(2)-N(1) ^{#4}	2.005(3)
Cu(2)-O(3)	2.599(2)		
$O(1) - Cu(1) - O(1)^{\#3}$	180	O(1)-Cu(1)-N(4) ^{#1}	91.44(10)
O(1) ^{#2} -Cu(1)-N(4) ^{#1}	88.56(10)	$N(4)^{\#1}$ -Cu(1)-N(4) $^{\#2}$	180
O(4) ^{#4} -Cu(2)-O(4)	180	O(4) ^{#4} -Cu(2)-N(1)	88.91 (11)
O(4) ^{#4} -Cu(2)-N(1) ^{#4}	91.09(11)	N(1) -Cu(2)-N(1) ^{#4}	180

Symmetry codes for **3**: ^{#1} 1/2-x,1/2+y,1/2-z; ^{#2} -1/2+x,1/2-y,-1/2+z; ^{#3} 1-x,-y,1-z; ^{#4} -x,-y,-z

Table S4. Selected bond distances (Å) and angles (°) for 4.

Co(1)-N(1)	2.1281(18)	Co(1)-N(4) ^{#1}	2.1126(16)
Co(1)-O(1)	2.1030(14)	Co(1)-O(2) ^{#2}	2.1417(16)
Co(1)-O(3)	2.1137(14)	$Co(1)-O(4)^{\#3}$	2.1471(14)

$O(1)-Co(1)-N(4)^{\#1}$	97.54(6)	O(1)-Co(1)-O(3)	169.73(5)
N(4) ^{#1} -Co(1)-O(3)	89.69(6)	O(1)-Co(1)-N(1)	90.75(6)
N(4) ^{#1} -Co(1)-N(1)	94.26(7)	O(3)-Co(1)-N(1)	96.00(6)
O(1)-Co(1)-O(2) ^{#2}	78.53(5)	N(4) ^{#1} -Co(1)-O(2) ^{#2}	91.64(6)
O(3)-Co(1)-O(2) ^{#2}	94.04(6)	N(1)-Co(1)-O(2) ^{#2}	168.37(6)
O(1)-Co(1)-O(4) ^{#3}	94.66(6)	N(4)-Co(1)-O(4) ^{#3}	167.53(6)
O(3)-Co(1)-O(4) ^{#3}	94.56(6)	N(1)-Co(1)-O(4) ^{#3}	88.62(6)
O(2) ^{#2} -Co(1)-O(4) ^{#3}	87.76(6)		

Symmetry codes for **4**: ^{#1} 1-x,-y,1-z; ^{#2} 1-x,-y,-z; ^{#3} -x,-y,-z

Table S5. Selected bond distances (Å) and angles (°) for 5.

Cu(1)-N(1)	1.976(5)	Cu(1)-O(4)	2.006(4)
Cu(1)-O(5)	2.614(1)	Cu(2)-O(3)	2.596(1)
Cu(2)-N(4)	1.986(5)	Cu(2)-O(2)	2.030(4)
N(1)-Cu(1)-N(1) ^{#1}	180	N(1)-Cu(1)-O(4)	89.2(2)
N(1) ^{#1} -Cu(1)-O(4)	90.8(2)	O(4) ^{#1} -Cu(1)-O(4)	180
N(4) ^{#3} -Cu(2)-N(4)	180	N(4) ^{#3} -Cu(2)-O(2)	91.00(19)
N(4)-Cu(2)-O(2)	89.0(2)	O(2)-Cu(2)-O(2) ^{#3}	180

Symmetry codes for **5**: ^{#1} 1-x,2-y,1-z; ^{#3} -x,2-y,-z

Table S6. Selected bond distances (Å) and angles (°) for 6.

Co(1)-O(2)	1.983(2)	Co(1)-O(5) ^{#3}	1.9980(19)
Co(1)-N(1) ^{#4}	2.033(2)	Co(1)-N(4)	2.014(3)
Co(2)-O(9) ^{#1}	1.980(2)	Co(2)-N(5)	2.021(3)
Co(2)-O(6)	2.018(2)	Co(2)-N(8) ^{#2}	2.022(3)
O(2)-Co(1)-O(5) ^{#3}	112.09(9)	O(2)-Co(1)-N(4)	121.33(10)
O(5) ^{#3} -Co(1)-N(4)	110.26(9)	O(2)-Co(1)-N(1) ^{#4}	104.15(9)
O(5) ^{#3} -Co(1)-N(1) ^{#4}	98.28(9)	N(4)-Co(1)-N(1) ^{#4}	107.86(10)
O(9) ^{#1} -Co(2)-O(6)	97.76(9)	O(9) ^{#1} -Co(2)-N(5)	105.45(10)

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O(6)-Co(2)-N(5)	107.52(10)	$O(9)^{\#1}$ -Co(2)-N(8) $^{\#2}$	112.21(10)
O(6)-Co(2)-N(8) ^{#2}	112.18(10)	N(5)-Co(2)-N(8) ^{#2}	119.36(11)

Symmetry codes for **6**: ^{#1} x,3/2-y,-1/2+z; ^{#2} 2-x,1-y,1-z; ^{#3} x,1/2-y,-1/2+z;

^{#4} 1-x,-y,1-z

Table S7. Selected bond distances (Å) and angles (°) for 7.

Cu(1)-O(4)	1.936(2)	Cu(1)-O(5)	1.967(2)
Cu(1)-O(6)	1.973(2)	Cu(1)-N(4)	1.992(3)
Cu(1)-O1W ^{#1}	2.541(6)	Cu(2)-O3 ^{#2}	2.456(4)
Cu(2)-O6 ^{#3}	2.533(8)	Cu(2)-O(1)	1.962(2)
Cu(2)-O(5) ^{#3}	1.9766(19)	Cu(2)-O(8) ^{#4}	1.952(2)
Cu(2)-N(1)	1.985(3)	Cu(3)-O(2)	1.985(2)
Cu(3)-O(3) ^{#7}	2.257(2)	Cu(3)-O(5) ^{#6}	1.978(2)
O1W ^{#1} O7	2.643(5)		
O(4)-Cu(1)-O(5)	96.26(9)	O(4)-Cu(1)-O(6)	174.05(13)
O(5)-Cu(1)-O(6)	86.68(8)	O(4)-Cu(1)-N(4)	87.46(11)
O(5)-Cu(1)-N(4)	167.84(11)	O(6)-Cu(1)-N(4)	88.63(10)
O(8) ^{#4} -Cu(2)-O(1)	169.68(10)	O(8) ^{#4} -Cu(2)-O(5) ^{#7}	93.04(9)
O(1)-Cu(2)-O(5) ^{#7}	89.49(9)	O(8) ^{#4} -Cu(2)-N(1)	88.16(10)
O(1)-Cu(2)-N(1)	90.42(10)	O(5) ^{#7} -Cu(2)-N(1)	173.73(10)
O(5) ^{#7} -Cu(3)-O(5) ^{#6}	180	O(5) ^{#7} -Cu(3)-O(2) ^{#5}	88.75(9)
O(5) ^{#7} -Cu(3)-O(2)	91.25(9)	O(5) ^{#6} -Cu(3)-O(2)	88.75(9)
O(2) ^{#5} -Cu(3)-O(2)	180	$O(5)^{\#7}$ -Cu(3)-O(3) $^{\#7}$	94.36(8)
O(5) ^{#6} -Cu(3)-O(3) ^{#7}	85.64(8)	O(2) ^{#5} -Cu(3)-O(3) ^{#7}	94.72(10)
O(2)-Cu(3)-O(3) ^{#7}	85.28(10)	O(5) ^{#7} -Cu(3)-O(3) ^{#6}	85.64(8)
O(5) ^{#6} -Cu(3)-O(3) ^{#6}	94.36(8)	O(2) ^{#5} -Cu(3)-O(3) ^{#6}	85.28(10)
$O(2)-Cu(3)-O(3)^{\#6}$	94.72(10)	O(3) ^{#7} -Cu(3)-O(3) ^{#6}	180

Symmetry codes for **7**: ^{#1} -1+x,1/2-y,1-z; ^{#2} -x,-1/2+y,1/2-z; ^{#3} x,1/2-y,-1/2+z;

 $^{\#4} 1/2 + x, -1/2 + y, 1/2 - z; \ ^{\#5} - x, -y, -z; \ ^{\#6} - x, -1/2 + y, 1/2 - z; \ ^{\#7} x, 1/2 - y, -1/2 + z$

Table S8.	Selected	bond	distances ((Å)) and	angles	(°)) for S	8.
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$Cu(1)-N(4)^{#4}$	1.992(5)	Cu(1)-O(1)	1.979(3)
Cu(1)-O(2W)	2.320(6)	Cu(2)-N(1)	1.966(5)
Cu(2)-O(1W)	1.957(4)	$Cu(2)-O(4)^{\#5}$	1.963(3)
Cu(2)-O(5)	1.948(4)	O1W…O6 ^{#2}	2.803(5)
O2W…O3 ^{#6}	2.870(6)		
$O(1)^{\#3}$ -Cu(1)-N(4) ^{#4}	89.62(17)	$O(1)-Cu(1)-N(4)^{\#4}$	91.04(17)
O(1)-Cu(1)-N(4) ^{#1}	89.62(17)	N(4) ^{#4} -Cu(1)-N(4) ^{#1}	166.9(3)
O(1)-Cu(1)-O(2W)	87.10(12)	N(4) ^{#4} -Cu(1)-O(2W)	96.55(14)
O(5)-Cu(2)-O(1W)	91.00(17)	$O(5)-Cu(2)-O(4)^{\#5}$	171.88(16)
O(1W)-Cu(2)-O(4) ^{#5}	87.76(16)	O(5)-Cu(2)-N(1)	91.25(18)
O(1W)-Cu(2)-N(1)	173.1(2)	O(4) ^{#5} -Cu(2)-N(1)	90.92(18)

Symmetry codes for **8**: ^{#1} 1-x,1/2+y,3/2-z; ^{#2} 1-x,y,3/2-z; ^{#3} -x,y,3/2-z;

^{#4} 1/2-x,1/2+y,z; ^{#5} 3/2-x,-1/2+y,z; ^{#6} 1/2-x,-1/2+y,z











III



Scheme S1. Four coordination modes of the BIE ligand with the metals in this paper.



Scheme S2. Coordination modes of L5 anion in compound 7.



Fig. S1. View of the 2D 4^4 -sql network of 1.



Fig. S2. One 6-membered ring is shown for each net shortest closed circuits, and the hcb net of compound 4.



Fig. S3. The 3D complicated framework of compound 6.



Fig. S4. Schematic diagram (OLEX) showing the $(8^3)_2(8^5 \cdot 10)$ network of **6**.



Fig. S5. View of the five coplanar Cu(II) atoms of compound 7.



Fig. S6. View of the 3D complicated framework of 7.



Fig. S7. View of a single layer composed of L6 anions and Cu(II) ions with 6^3 network within compound **8**.

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Fig. S8. TGA curve of 3.



Fig. S9. TGA curve of 4.



Fig. S10. TGA curve of 5.

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Fig. S11. TGA curve of 6.



Fig. S12. TGA curve of 7.



Fig. S13. TGA curve of 8.