

## **Electronic Supplementary Information**

**for**

**Molecular botanical garden: assembly of supramolecular silver(I) and mercury(II) complexes of NS<sub>2</sub>-donor macrocycles with flower-, leaf- and tree-shaped structures**

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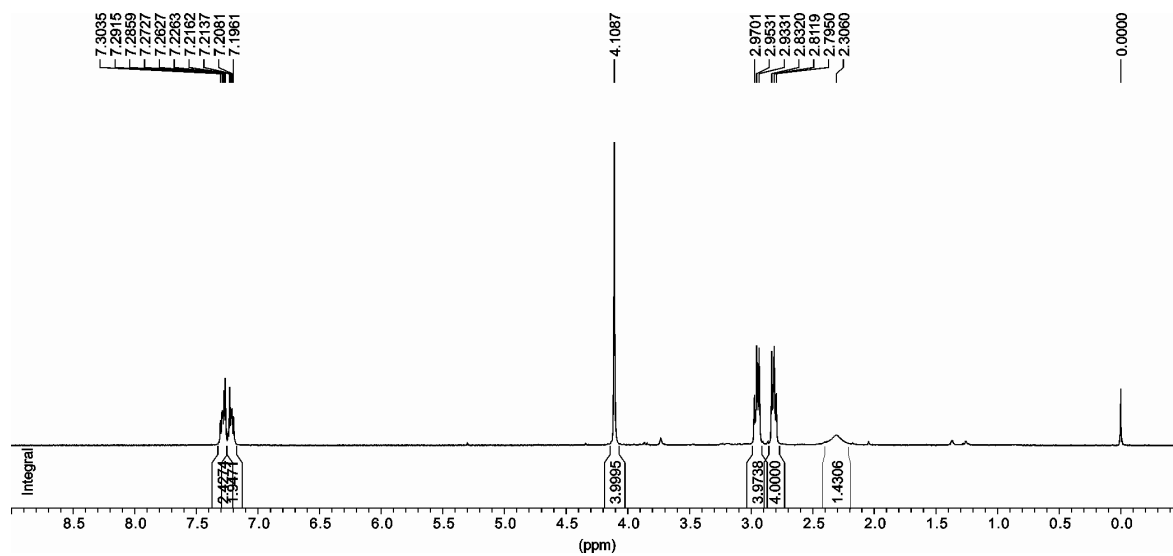


Fig. S1  $^1\text{H}$  NMR spectrum of  $\text{L}^1$  in  $\text{CDCl}_3$ .

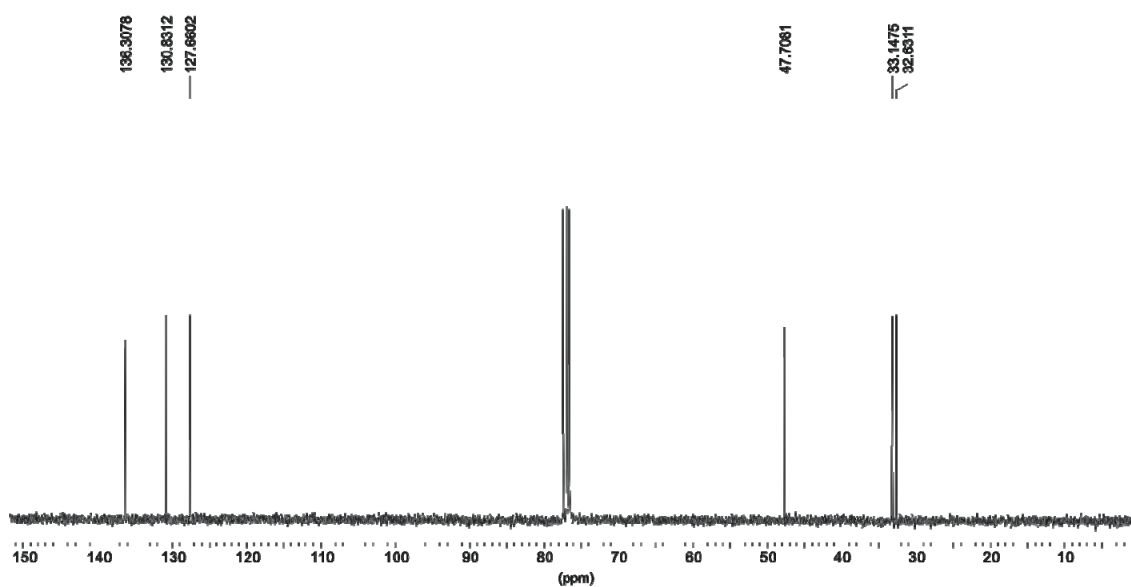


Fig. S2  $^{13}\text{C}$  NMR spectrum of  $\text{L}^1$  in  $\text{CDCl}_3$ .

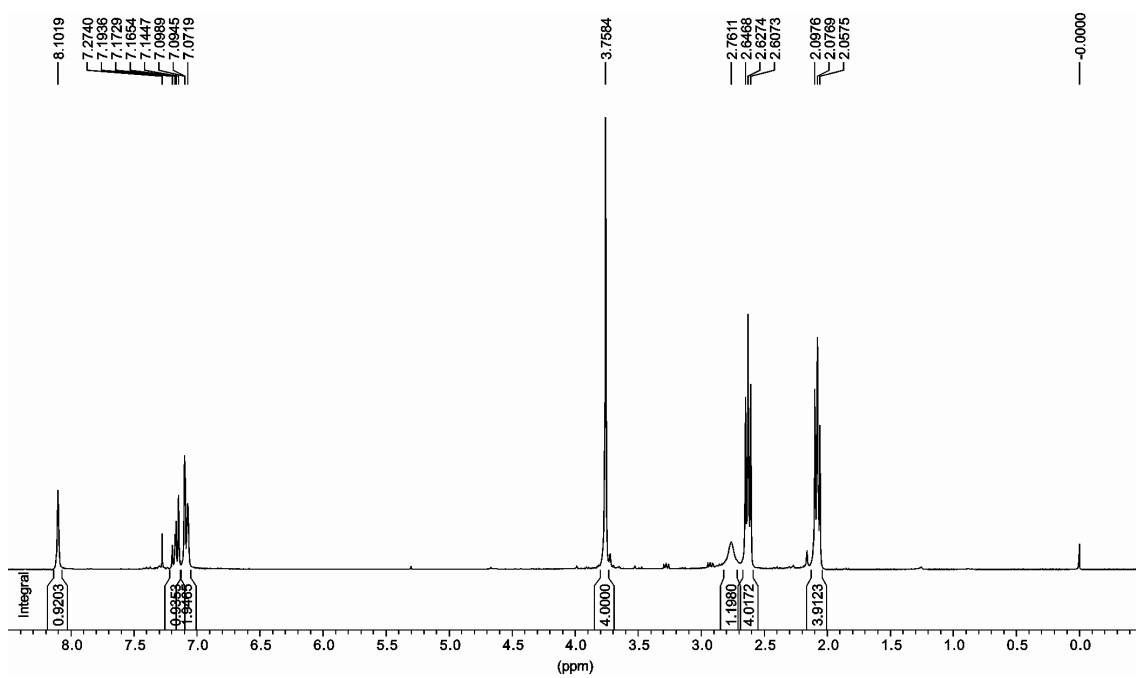


Fig. S3  $^1\text{H}$  NMR spectrum of  $\text{L}^2$  in  $\text{CDCl}_3$ .

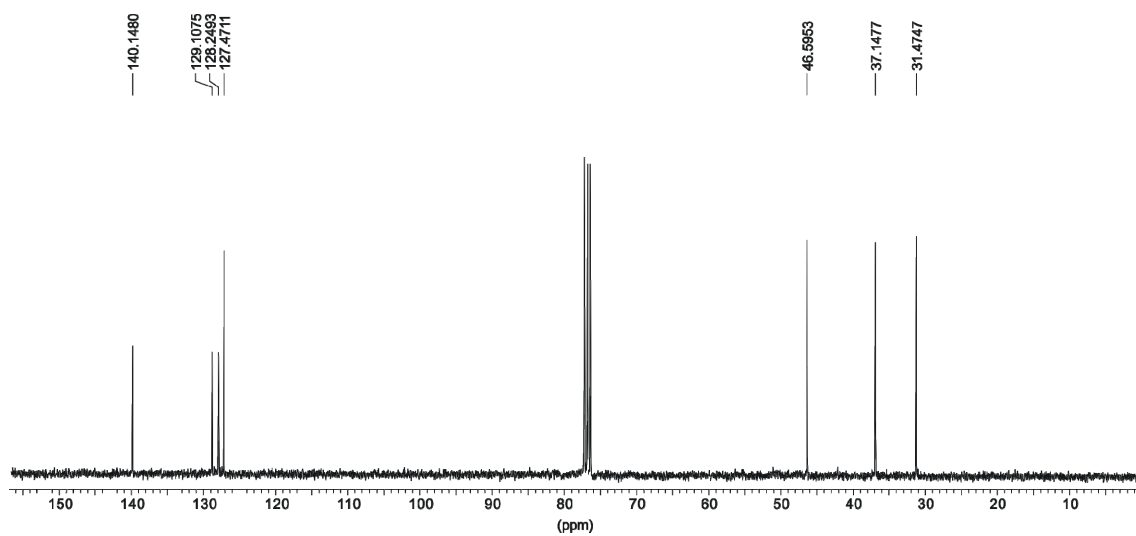


Fig. S4  $^{13}\text{C}$  NMR spectrum of  $\text{L}^2$  in  $\text{CDCl}_3$ .

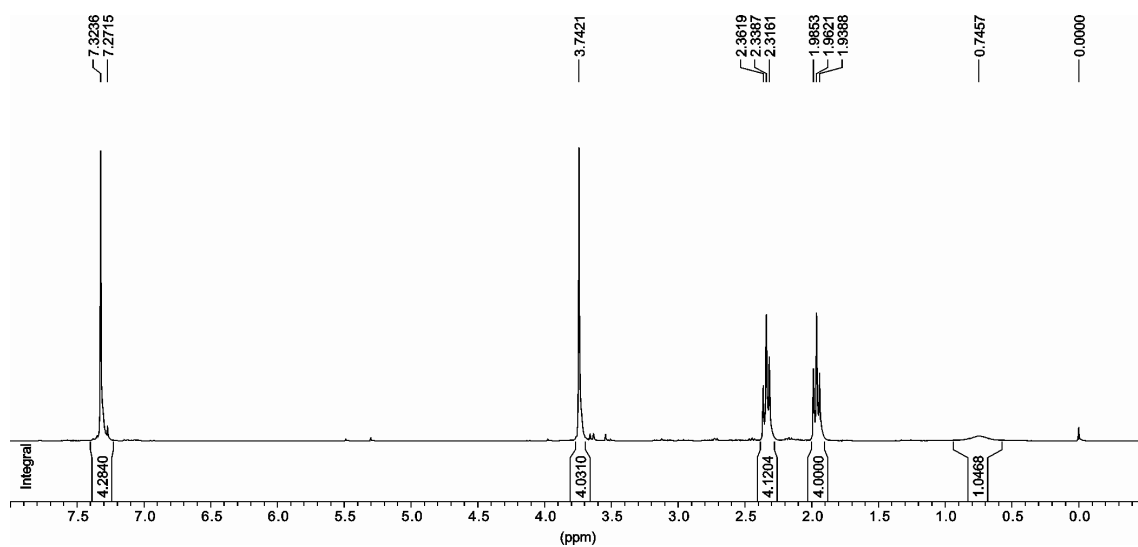


Fig. S5 <sup>1</sup>H NMR spectrum of L<sup>3</sup> in CDCl<sub>3</sub>.

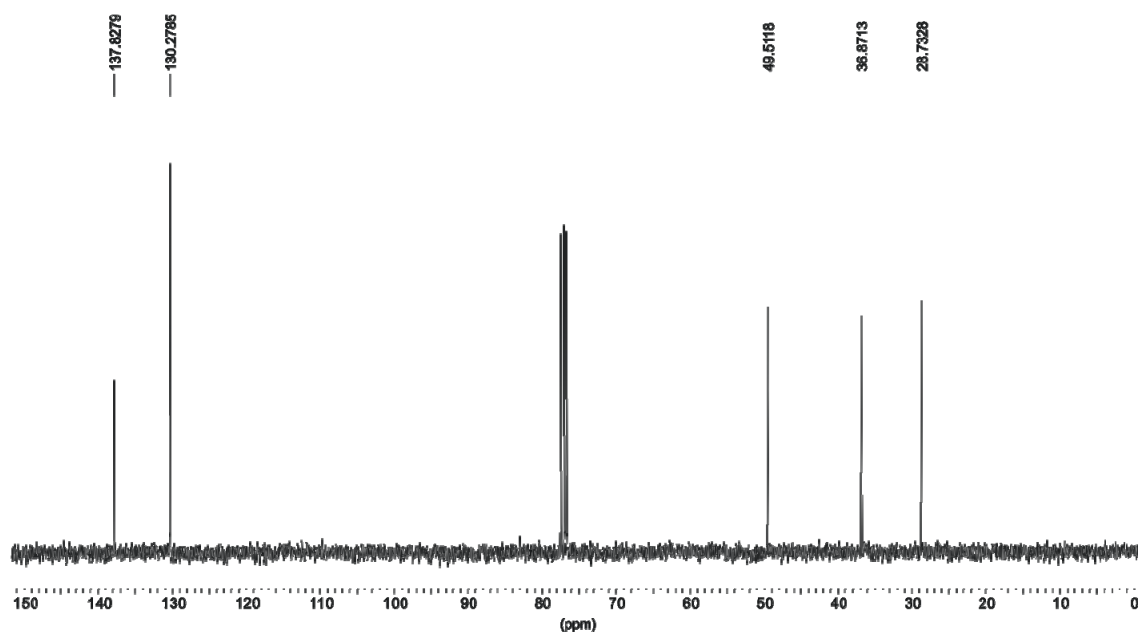


Fig. S6 <sup>13</sup>C NMR spectrum of L<sup>3</sup> in CDCl<sub>3</sub>.

[Ag<sub>6</sub>(L<sup>1</sup>)<sub>6</sub>(PF<sub>6</sub>)](PF<sub>6</sub>)<sub>5</sub> (**1**). Layering a methanol (3 mL) of AgPF<sub>6</sub> (15.9 mg, 0.06 mmol) onto a dichloromethane solution (2 mL) of L<sup>1</sup> (15.2 mg, 0.06 mmol) afforded colourless crystalline **1** suitable for X-ray analysis. M.p. 169.5 °C (decomp.). IR (KBr, cm<sup>-1</sup>): 3336, 2933, 1662, 1456, 1105, 841 (PF<sub>6</sub><sup>-</sup>), 775, 557. Anal. Calc. for C<sub>72</sub>H<sub>102</sub>Ag<sub>6</sub>F<sub>36</sub>N<sub>6</sub>P<sub>6</sub>S<sub>12</sub>: C, 29.28; H, 3.48; N, 2.85. Found: C, 28.93; H, 3.26; N, 3.19%.

[Ag<sub>3</sub>(L<sup>2</sup>)<sub>4</sub>](PF<sub>6</sub>)<sub>3</sub>·C<sub>6</sub>H<sub>5</sub>CH<sub>3</sub> (**2**). Complex **2** was obtained after adding a small amount of toluene to the top layer of dichloromethane solution of L<sup>2</sup> (15.2 mg, 0.06 mmol) then layering this with a methanol (3 mL) of AgPF<sub>6</sub> (15.9 mg, 0.06 mmol), allowing the three phase system to stand. M.p. 198.6 °C (decomp.). IR (KBr, cm<sup>-1</sup>): 3305, 1552, 1425, 1380, 1108, 837 (PF<sub>6</sub><sup>-</sup>), 713. ESI-MS: *m/z* 426.1 [Ag<sub>3</sub>(L<sup>2</sup>)<sub>4</sub>]<sup>3+</sup>.

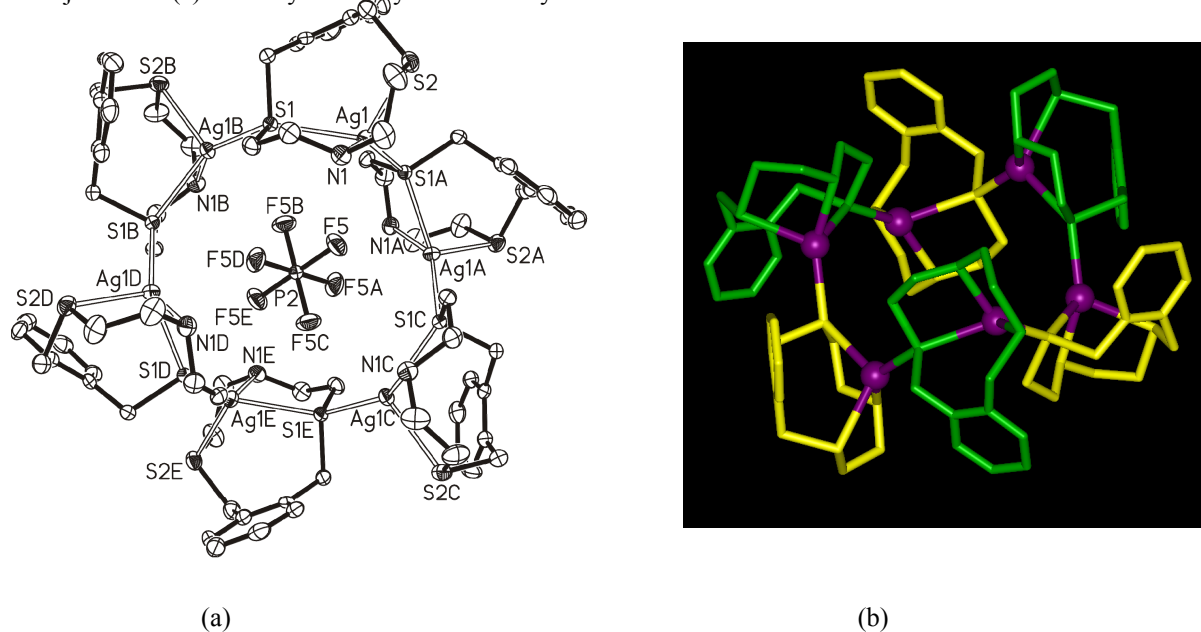
[Hg<sub>2</sub>(L<sup>2</sup>)Br<sub>4</sub>]<sub>*n*</sub> (**3a**). To the stirring solution of L<sup>2</sup> (20.2 mg, 0.08 mmol) in dichloromethane was added to HgBr<sub>2</sub> (29.2 mg, 0.08 mmol) in methanol afforded colourless precipitate. After the filtration, the vapor diffusion of diethyl ether to DMSO solution gave rise to crystalline product, affording colourless single crystals. M.p. 185.1 °C (decomp.), IR (KBr, cm<sup>-1</sup>): 3191, 2358, 1686, 1518, 1410, 1301, 1218, 1047, 952, 712, 455. Anal. Calc. for C<sub>12</sub>H<sub>17</sub>Br<sub>4</sub>Hg<sub>2</sub>NS<sub>2</sub>: C, 15.01; H, 1.78; N, 1.46. Found: C, 15.38; H, 2.08; N, 1.83%.

[Hg<sub>2</sub>(L<sup>2</sup>)I<sub>4</sub>]<sub>*n*</sub> (**3b**). To the stirring solution of L<sup>2</sup> (20.2 mg, 0.08 mmol) in dichloromethane was added to HgI<sub>2</sub> (36.3 mg, 0.08 mmol) in methanol afforded colourless precipitate. After the filtration, the vapor diffusion of diethyl ether to DMSO solution gave rise to crystalline product, affording colourless single crystals. M.p. 203.2 °C (decomp.). IR (KBr, cm<sup>-1</sup>): 3177, 2361, 1684, 1520, 1408, 1215, 1136, 1043, 947, 708, 455. Anal. Calc. for C<sub>12</sub>H<sub>17</sub>I<sub>4</sub>Hg<sub>2</sub>NS<sub>2</sub>: C, 12.55; H, 1.49; N, 1.22; S, 5.59. Found: C, 12.93; H, 1.52; N, 1.49; S, 5.76%.

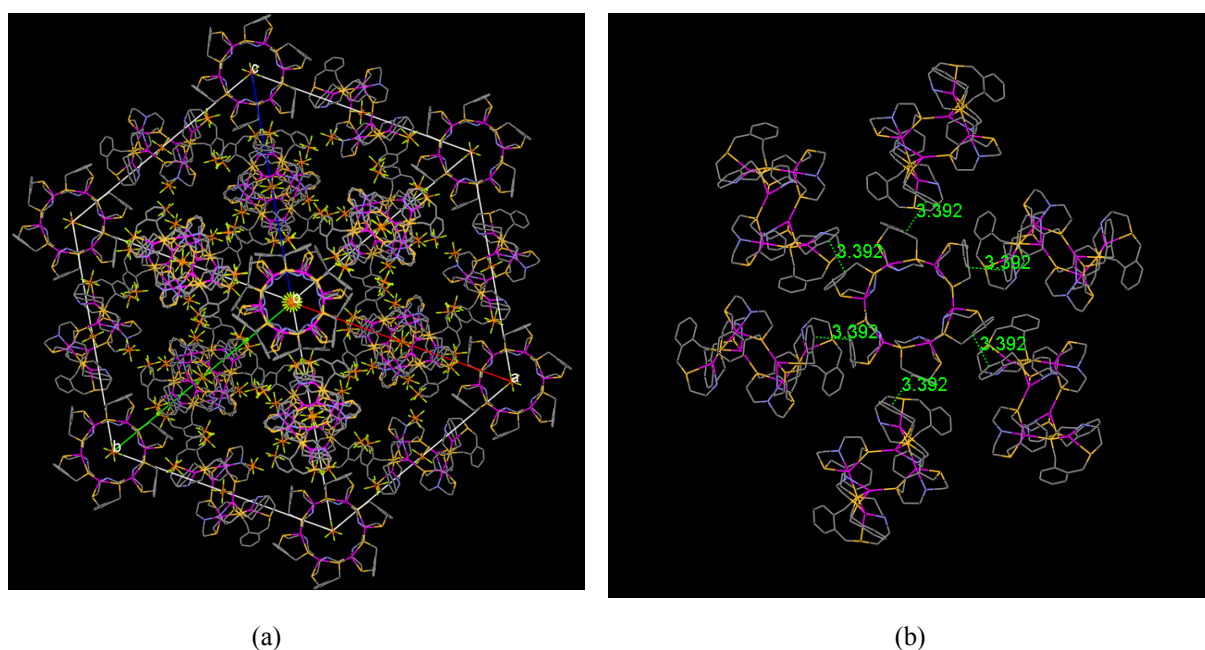
[Hg<sub>2</sub>(L<sup>3</sup>)Br<sub>4</sub>]<sub>*n*</sub> (**4**). Layering a methanol (3 mL) of HgI<sub>2</sub> (28.6 mg, 0.06 mmol) onto a dichloromethane solution (2 mL) of L<sup>3</sup> (15.2 mg, 0.06 mmol) afforded pale yellow-coloured crystalline **4** suitable for X-ray analysis. M.p. 152.6 °C (decomp.). IR (KBr, cm<sup>-1</sup>): 3259, 2912, 1503, 1437, 1408, 1093, 825, 738. Anal. Calc. for C<sub>12</sub>H<sub>17</sub>Br<sub>4</sub>Hg<sub>2</sub>NS<sub>2</sub>: C, 15.01; H, 1.78; N, 1.46. Found: C, 14.94; H, 2.03; N, 1.66%.

**Table S1** Crystal and experimental data

	<b>1</b>	<b>2</b>	<b>3a</b>	<b>3b</b>	<b>4</b>
Formula	C <sub>72</sub> H <sub>102</sub> Ag <sub>6</sub> F <sub>36</sub> N <sub>6</sub> P <sub>6</sub> S <sub>12</sub>	C <sub>55</sub> H <sub>76</sub> Ag <sub>3</sub> F <sub>18</sub> N <sub>4</sub> P <sub>3</sub> S <sub>8</sub>	C <sub>12</sub> H <sub>17</sub> Br <sub>4</sub> Hg <sub>2</sub> NS <sub>2</sub>	C <sub>12</sub> H <sub>17</sub> Hg <sub>2</sub> I <sub>4</sub> NS <sub>2</sub>	C <sub>12</sub> H <sub>17</sub> Br <sub>4</sub> Hg <sub>2</sub> NS <sub>2</sub>
Formula weight	2953.36	1808.20	960.21	1148.17	960.21
Temperature (K)	173(2)	173(2)	173(2)	173(2)	173(2)
Crystal system	Cubic	Triclinic	Orthorhombic	Orthorhombic	Monoclinic
Space group	<i>Ia-3d</i>	<i>P1</i>	<i>Pbca</i>	<i>Pbca</i>	<i>P2<sub>1</sub>/c</i>
Z	16	1	8	8	4
<i>a</i> (Å)	36.322(3)	11.4605(5)	18.2813(8)	18.9518(19)	13.5645(11)
<i>b</i> (Å)	36.322(3)	11.6121(6)	8.1124(4)	8.5212(9)	11.9094(10)
<i>c</i> (Å)	36.322(3)	13.1785(6)	26.4524(12)	27.755(3)	12.4053(10)
$\alpha$ (°)	90	90.294(1)	90	90	90
$\beta$ (°)	90	99.549(1)	90	90	104.114(2)
$\gamma$ (°)	90	94.075(1)	90	90	90
<i>V</i> (Å <sup>3</sup> )	47921(8)	1724.87(14)	3923.0(3)	4482.2(8)	1943.5(3)
<i>D<sub>x</sub></i> (g/cm <sup>3</sup> )	1.637	1.741	3.251	3.403	3.282
$2\theta_{\max}$ (°)	56.92	54	56.56	54	53
<i>R</i>	0.0433	0.0535	0.0463	0.0427	0.0778
<i>wR</i>	0.1194	0.1340	0.1018	0.1084	0.1129
GOF	1.112	1.049	1.016	1.118	1.002
No. of reflection used [>2σ( <i>I</i> )]	5010 [ <i>R</i> <sub>int</sub> = 0.1461]	8741 [ <i>R</i> <sub>int</sub> = 0.0118]	4740 [ <i>R</i> <sub>int</sub> = 0.1028]	4888 [ <i>R</i> <sub>int</sub> = 0.0589]	3943 [ <i>R</i> <sub>int</sub> = 0.0857]
Diffractometer	Bruker SMART CC D	Bruker SMART CC D	Bruker SMART CC D	Bruker SMART CC D	Bruker SMART CC D
Structure determination	SHELXTL	SHELXTL	SHELXTL	SHELXTL	SHELXTL
Refinement	full-matrix	full-matrix	full-matrix	full-matrix	full-matrix



**Fig. S7** Molecular structure of **1**,  $[\text{Ag}_6(\text{L}^1)_6(\text{PF}_6)](\text{PF}_6)_5$ : (a) top view (Ortep) and (b) general view (ball-and-stick, the anion at the centre was removed). Thermal ellipsoids are drawn at the 30% probability level. Symmetry operations: (A)  $1/2 - x, 1/2 + y, z$ , (B)  $-1/2 + x, y, 1/2 - z$ , (C)  $1/2 - x, -y, 1/2 + z$ , (D)  $-1/2 + x, 1/2 - y, -z$ , (E)  $1 - x, -y, -z$



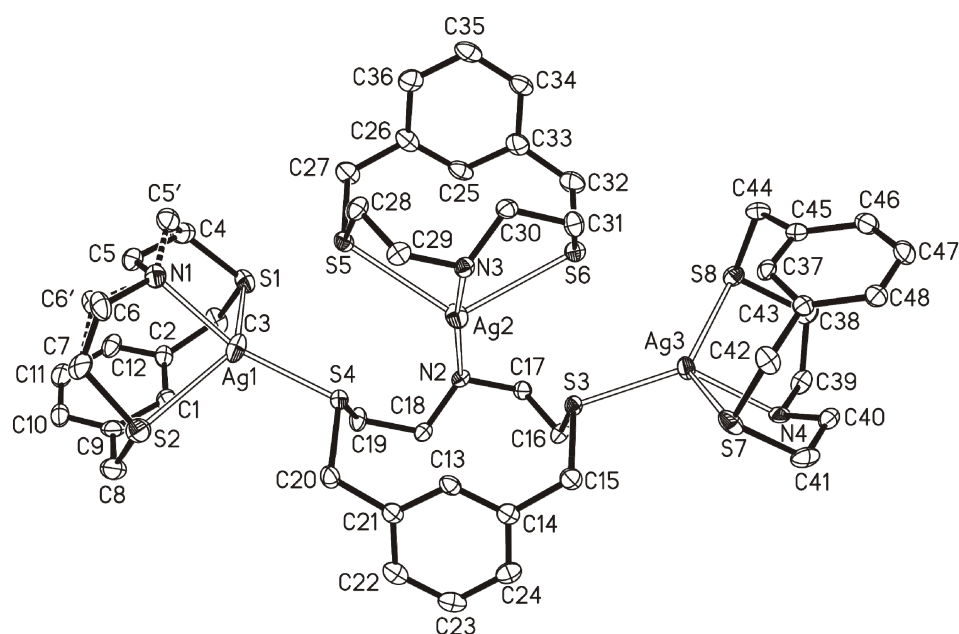
**Fig. S8** (a) Packing diagram and (b) the  $\pi$ - $\pi$  stacking interactions (dashed lines) of **1**,  $[\text{Ag}_6(\text{L}^1)_6(\text{PF}_6)](\text{PF}_6)_5$ .

**Table S2** Selected bond lengths (Å) and bond angles (°) for **1**

Ag1-N1	2.338(3)	S1-Ag1B	2.452(1)
Ag1-S1A	2.452(1)	Ag1-S2	2.613(1)
Ag1-S1	2.614(1)		
N1-Ag1-S1A	144.2(1)	N1-Ag1-S1	79.8(1)
S1A-Ag1-S1	117.6(1)	N1-Ag1-S2	80.0(1)
S1A-Ag1-S2	116.8(1)	S1-Ag1-S2	110.9(1)
Ag1B-S1-Ag1	137.0(1)		

Symmetry Code: (A)  $1/2 - x, 1/2 + y, z$ , (B)  $-1/2 + x, y, 1/2 - z$ .

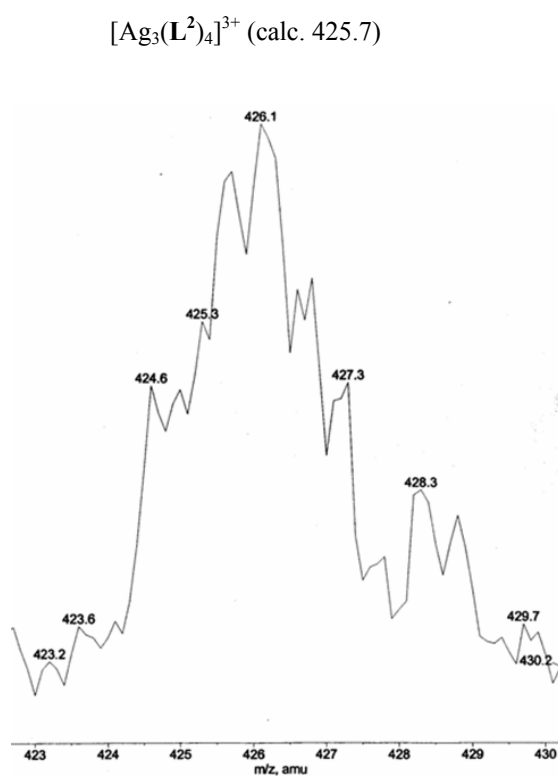




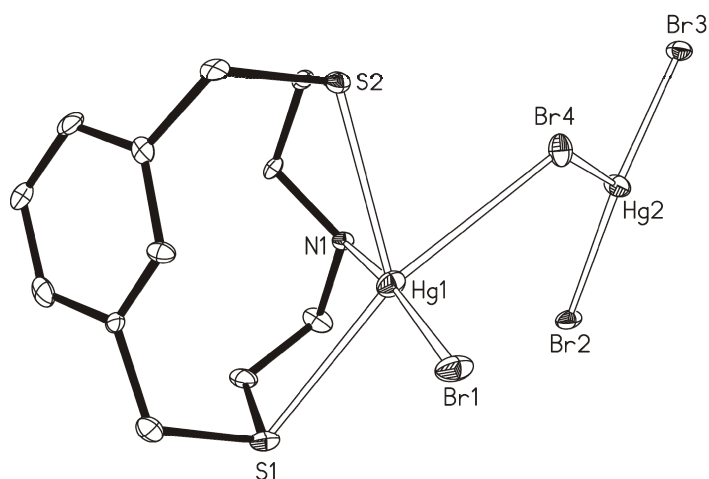
**Fig. S9** Molecular structure of **2**,  $[\text{Ag}_3(\text{L}^2)_4](\text{PF}_6)_3 \cdot \text{C}_6\text{H}_5\text{CH}_3$ . Hydrogen atoms, noncoordinating anions and solvent are omitted. Thermal ellipsoids are drawn at the 30% probability level.

**Table S3** Selected bond lengths (Å) and bond angles (°) for **2**

Ag1-N1	2.315(8)	Ag1-S4	2.486(2)
Ag1-S1	2.761(2)	Ag1-S2	2.773(2)
Ag2-N2	2.269(6)	Ag2-N3	2.280(6)
Ag2-S5	2.900(2)	Ag2-S6	2.928(2)
Ag3-N4	2.352(7)	Ag3-S3	2.516(2)
Ag3-S8	2.669(2)	Ag3-S7	2.708(2)
N1-Ag1-S4	156.2(2)	S4-Ag1-S1	108.2(1)
N1-Ag1-S1	80.2(2)	N1-Ag1-S2	79.8(2)
S4-Ag1-S2	109.4(1)	S1-Ag1-S2	127.6(1)
N2-Ag2-N3	166.3(2)	N2-Ag2-S5	114.3(2)
N3-Ag2-S5	77.2(2)	N2-Ag2-S6	101.0(2)
N3-Ag2-S6	78.5(2)	S5-Ag2-S6	118.1(1)
N4-Ag3-S3	132.2(2)	N4-Ag3-S8	79.8(2)
S3-Ag3-S8	127.8(1)	N4-Ag3-S7	80.0(2)
S3-Ag3-S7	100.1(1)	S8-Ag3-S7	129.0(1)



**Fig. S10** Partial ESI-mass spectrum of **2** showing the existence of  $[\text{Ag}_3(\text{L}^2)_4]^{3+}$  species.

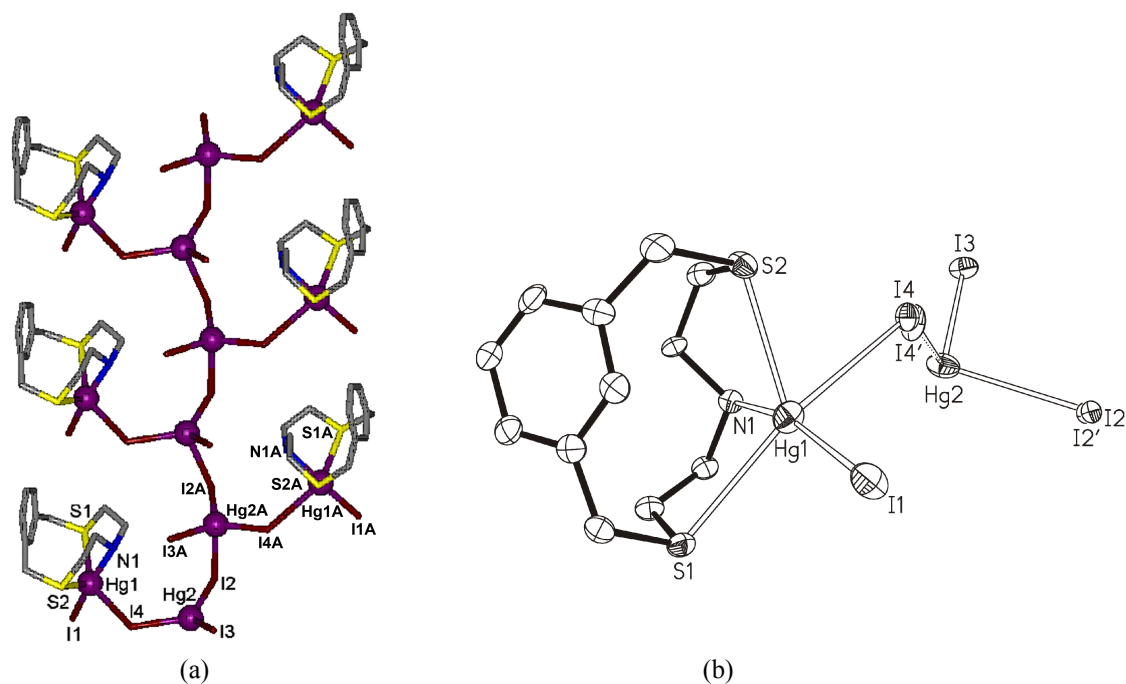


**Fig. S11** Asymmetric unit of **3a**,  $[\text{Hg}_2(\text{L}^2)\text{Br}_4]_n$ . Hydrogen atoms are omitted. Thermal ellipsoids are drawn at the 30% probability level.

**Table S4** Selected bond lengths (Å) and bond angles (°) for **3a**

Hg1-N1	2.198(8)	Hg1-S2	2.864(3)
Hg1-Br1	2.439(1)	Hg1-S1	2.928(3)
Hg2-Br2	2.527(1)	Hg2-Br3	2.541(1)
Hg2-Br4	2.600(1)	Br2-Hg2B	2.978(1)
Hg2-Br2A	2.978(1)		
N1-Hg1-Br1	170.8(2)	Br1-Hg1-S2	109.3(1)
N1-Hg1-S2	79.8(2)	N1-Hg1-S1	78.6(2)
Br1-Hg1-S1	94.2(1)	S2-Hg1-S1	127.6(1)
Br2-Hg2-Br3	131.8(1)	Br2-Hg2-Br4	118.3(1)
Br3-Hg2-Br4	107.9(1)	Br2-Hg2-Br2A	94.6(1)
Br3-Hg2-Br2A	89.6(1)	Br4-Hg2-Br2B	100.6(1)
Hg2-Br2-Hg2B	99.6(1)		

Symmetry Code: (A) -  $x - 1/2, y + 1/2, z$ , (B) -  $x - 1/2, y - 1/2, z$ .

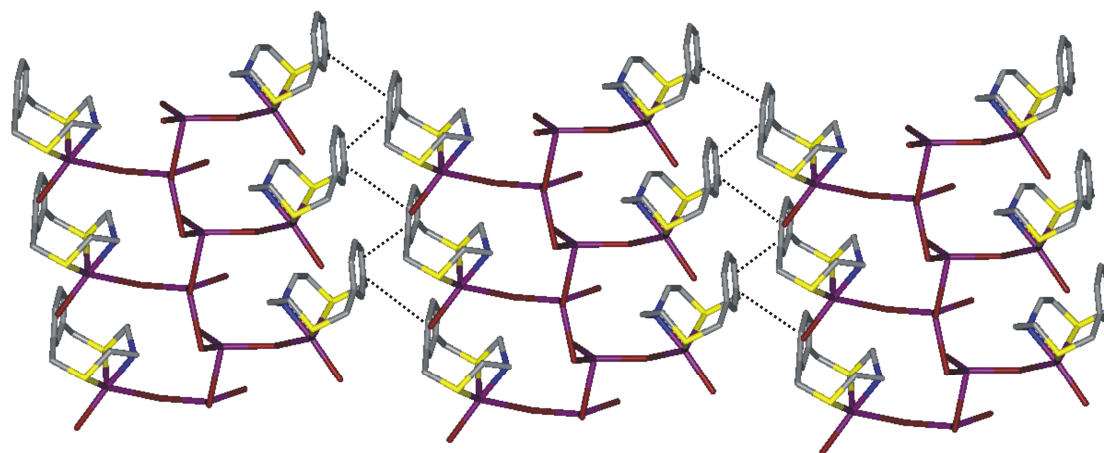


**Fig. S12** (a) Leaf-shaped infinite 1D structure and (b) asymmetric unit of **3b**,  $[\text{Hg}_2(\text{L}^2)\text{I}_4]_n$ . Thermal ellipsoids are drawn at the 30% probability level. Symmetry operations: (A)  $-x - 1/2, y + 1/2, z$

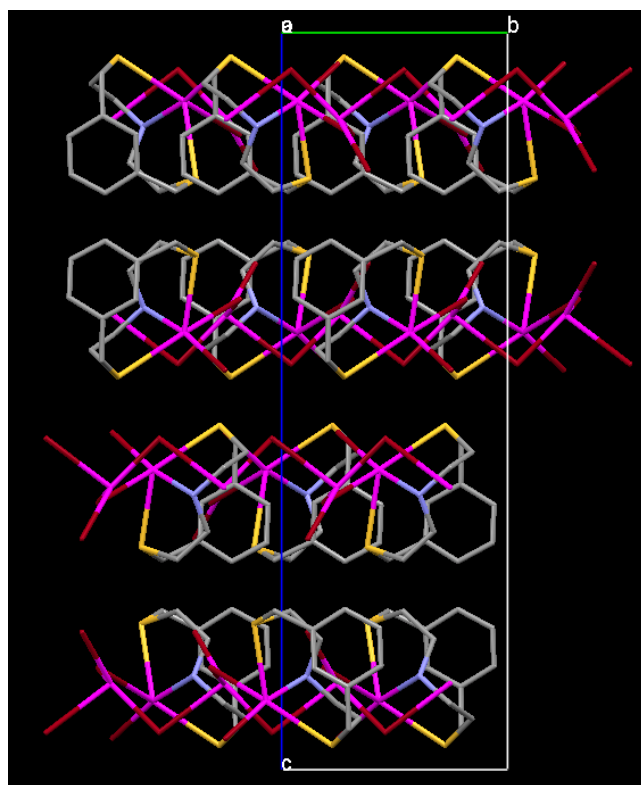
**Table S5** Selected bond lengths (Å) and bond angles (°) for **3b**

Hg1-N1	2.224(9)	Hg1-I1	2.6003(11)
Hg1-S2	2.855(3)	Hg1-S1	3.012(3)
Hg2-I3	2.7029(10)	Hg2-I2A	2.716(15)
Hg2-I4	2.709(16)	Hg2-I2	3.147(15)
I2-Hg2B	2.716(15)		
N1-Hg1-I1	162.7(2)	N1-Hg1-S2	79.6(2)
I1-Hg1-S2	117.73(8)	N1-Hg1-S1	76.3(2)
I1-Hg1-S1	92.22(7)	S2-Hg1-S1	123.09(9)
I3-Hg2-I2A	120.9(3)	I3-Hg2-I4	111.6(3)
I2A-Hg2-I4	122.7(3)	I3-Hg2-I2	98.9(3)
I2A-Hg2-I2	97.6(3)	I4-Hg2-I2	99.9(4)
Hg2B-I2-Hg2	98.5(4)		

Symmetry Code: (A)  $-x - 1/2, y + 1/2, z$ , (B)  $-x - 1/2, y - 1/2, z$ .

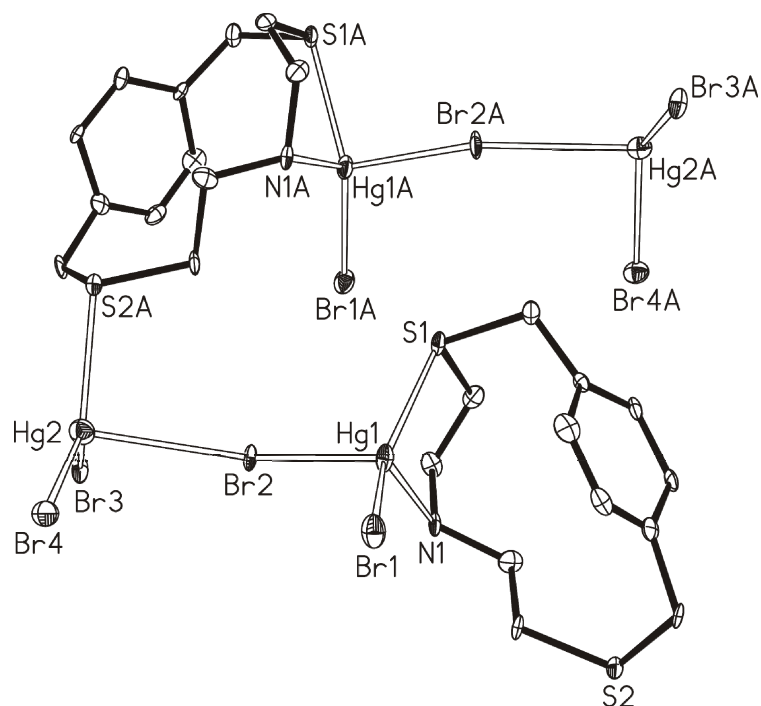


(a)



(b)

**Fig. S13** (a) Packing diagram, showing  $\pi$ - $\pi$  stacking interactions (dashed lines) and (b) projection of the 1D chain for **3**,  $[\text{Hg}_2(\text{L}^2)\text{X}_4]_n$  (**3a**: X = Br and **3b**: X = I) along the a-axis.



**Fig. S14** Coordination environment of **4**,  $[\text{Hg}_2(\text{L}^3)\text{Br}_4]_n$ . Thermal ellipsoids are drawn at the 30% probability level. Thermal ellipsoids are drawn at the 30% probability level. Symmetry operations: (A)  $-x + 2, y + 1/2, -z + 1/2$

**Table S6** Selected bond lengths (Å) and bond angles (°) for **4**

Hg1-N1	2.393(15)	Hg1-Br1	2.494(2)
Hg1-S1	2.524(4)	Hg1-Br2	2.687(2)
Hg2-Br4	2.483(2)	Hg2-Br3	2.557(2)
Hg2-S2A	2.601(4)	Hg2-Br2	2.997(2)
S2-Hg2B	2.601(4)		
N1-Hg1-Br1	119.0(3)	N1-Hg1-S1	83.8(4)
Br1-Hg1-S1	144.4(1)	N1-Hg1-Br2	88.0(3)
Br1-Hg1-Br2	110.3(1)	S1-Hg1-Br2	96.6(1)
Br4-Hg2-Br3	127.3(1)	Br4-Hg2-S2A	123.2(1)
Br3-Hg2-S2A	108.0(1)	Br4-Hg2-Br2	94.0(1)
Br3-Hg2-Br2	93.9(1)	S2A-Hg2-Br2	94.5(1)
Hg1-Br2-Hg2	149.0(1)		

Symmetry Code: (A)  $-x + 2, y + 1/2, -z + 1/2$ , (B)  $-x + 2, y - 1/2, -z + 1/2$ .