Supporting information

Novel zinc oxide twins with perfect mirror symmetry by solvothermal synthesis method

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Experimental Methods

1. Materials

Chemicals were used as received: zinc nitrate hexahydrate $(Zn(NO_3)_2 \cdot 6H_2O, Merck)$, julolidine (98%, Acros); N, N-diethylformamide (DEF, 99%, ABCR) as standard solvent. In some experiments N, N-Dimethylformamide (DMF, water<50 ppm, Acros), methanol (99.9%, Roth) were used as solvent.

2. Synthesis of ZnO crystals

The molar ratio of the solution composition for synthesis of ZnO crystals was 1 $Zn(NO_3)_2 \cdot 6H_2O$: 5 julolidine : (2~10) DEF. a typical synthesis is presented as followed, 0.35 g Julolidine was dissolved in 5.0 g DEF, and then 0.12 g zinc nitrate hexahydrate was added with stirring. After thoroughly stirring for 5 minutes, the mixture was introduced in a Teflon-lined stainless autoclave and heated at 175 °C for 2~6 days in an oven. After crystallization, the solids were filtered, washed several times with DMF, and then dried at 373 K.

3. Characterization of ZnO crystals

Scanning electron microscopy (SEM) micrographs were taken on a JEOL JSM-6700F with a cold field emission gun operating at 2 kV and 10 μ A. Phase purity and crystallinity of the as-synthesized crystals were confirmed by powder The X-ray diffraction (XRD) patterns were recorded at room temperature under ambient conditions with a PANalytical instrument (X'Pert-MPD, Cu K_a with k = 1.5418 Å) at 40 kV, 50 mA.



Fig. S1 Typical SEM image (a) and XRD pattern of the ZnO crystals prepared absence of julolidine in DEF.



Fig. S 2. Typical XRD pattern of the ZnO disks prepared in DMF solvent.



Fig. S3. Typical XRD pattern of the ZnO disks prepared in methanol solvent.



Fig. S4. Typical XRD pattern of the ZnO crystals prepared in the mixed DEF and DMF.



Fig. S5. Typical XRD pattern of the ZnO crystals prepared in the mixed DEF and methanol.



Fig. S6. Typical SEM image of the dumbbell-like shaped ZnO twins prepared in DEF at 473 K for 4 days.