Supporting Information



Fig. S1. Ball-and-stick representation of the propeller-shaped structures of the $[MnH(bix)_4]^{3+}$ complex fragments in compound 1. (all of the hydrogen atoms are omitted for clarity).



Fig. S2. The IR spectra of compounds 1-4.

	Temperature	first weight loss	Temperature	second weight
				loss
Compound 1	45°C to 125°C	1.20%	150 to 358 °C	33.65%
		(calcd 1.28%)		(calcd 33.78%)
Compound 2	180°C to 240°C	1.19%	260 to 580 °C	33.26%
		(calcd 1.25%)		(calcd 33.15%)
Compound 3	160°C to 200°C	2.35%	210°C to 480°C	32.87%
		(calcd 2.48%)		(calcd 32.76%)
Compound 4	100°C to 150°C	2.28%	200°C to 350°C	31.25%
		(calcd 2.37%)		(calcd 31.37%)

The TG analysis



Fig. S3. The cyclic voltammograms of compounds 1-4 in 1M H₂SO₄ at different scan rates.



Fig. S4. Changes in C_t/C_0 plot of RhB solutions (2 × 10⁻⁵ mol L⁻¹) versus reaction time in the presence of 5 mg compounds 1-4.

Mo(1)-O(1)	1.686(6)	Mo(2)-O(3)	1.679(7)
Mo(1)-O(9)	1.806(6)	Mo(2)-O(6)	1.802(6)
Mo(1)-O(5)	1.817(6)	Mo(2)-O(7)	1.811(6)
Mo(1)-O(8)	2.057(6)	Mo(2)-O(8)	2.053(6)
Mo(1)-O(11)	2.065(6)	Mo(2)-O(11)#1	2.072(6)
Mo(1)-O(4)	2.395(6)	Mo(2)-O(4)	2.457(6)
V(2)-O(2)	1.616(6)	Mn(1)-N(1)	2.167(9)
V(2)-O(7)#2	1.902(6)	Mn(1)-N(1)#1	2.167(9)
V(2)-O(9)#3	1.902(6)	Mn(1)-N(1)#4	2.167(9)
V(2)-O(6)	1.939(6)	Mn(1)-N(1)#5	2.167(9)
V(2)-O(5)	1.940(6)	Mn(1)-O(10)#5	2.193(9)
V(1)-O(10)	1.634(9)	Mn(1)-O(10)	2.193(9)
V(1)-O(11)	1.896(6)	P(1)-O(4)#1	1.533(6)
V(1)-O(11)#1	1.896(6)	P(1)-O(4)#3	1.533(6)
V(1)-O(8)	1.970(6)	P(1)-O(4)	1.533(6)
V(1)-O(8)#1	1.970(6)	P(1)-O(4)#2	1.533(6)

d lengths (Å) for compound 1.
d lengths (Å) for compound 1.

Symmetry transformations used to generate equivalent atoms:

4,x-1/4,-z+3/4 #3 y+1/4,-x-1/4,-z+3/4

#4 -y-1/4,x-1/4,-z+7/4 #5 y+1/4,-x-1/4,-z+7/4

P(1)-O(9)#1	1.532(3)	P(1)-O(9)#2	1.532(3)
P(1)-O(9)	1.532(3)	P(1)-O(9)#3	1.532(3)
Zn(1)-N(1)	2.124(4)	Zn(1)-N(1)#5	2.124(4)
Zn(1)-O(1)	2.209(4)	Zn(1)-N(1)#4	2.124(4)
Zn(1)-O(1)#5	2.209(4)	Zn(1)-N(1)#2	2.124(4)
V(1)-O(1)	1.624(4)	V(1)-O(3)	1.981(3)
V(1)-O(10)#2	1.897(3)	V(1)-O(3)#2	1.981(3)
V(1)-O(10)	1.897(3)	Mo(1)-O(8)	1.678(3)
Mo(3)-O(2)	1.619(3)	Mo(1)-O(7)	1.813(3)
Mo(3)-O(7)#3	1.907(3)	Mo(1)-O(6)	1.816(3)
Mo(3)-O(4)#3	1.913(3)	Mo(1)-O(3)	2.043(3)
Mo(3)-O(6)	1.937(3)	Mo(1)-O(10)	2.054(3)
Mo(3)-O(11)#2	1.950(3)	Mo(1)-O(9)#2	2.457(3)
Mo(3)-O(9)#2	2.477(3)	Mo(2)-O(10)	2.048(3)
Mo(2)-O(5)	1.686(3)	Mo(2)-O(3)#2	2.056(3)
Mo(2)-O(4)	1.802(3)	Mo(2)-O(9)	2.394(3)
Mo(2)-O(11)	1.812(3)		

Table S2. Selected bond lengths (Å) for compound 2.

Symmetry transformations used to generate equivalent atoms:

#1 y+1/4,-x+3/4,-z-1/4 #2 -x+1,-y+1/2

#2 -x+1,-y+1/2, z+0 #3 -y+3/4,x-1/4,-z-1/4

#4 -y+3/4,x-1/4,-z+3/4 #5 y+1/4,-x+3/4,-z+3/4

Cu(1)-N(1)#1	2.038(7)	Cu(1)-N(1)#3	2.038(7)
Cu(1)-N(1)#2	2.038(7)	Cu(1)-O(1)#3	2.353(8)
Cu(1)-N(1)	2.038(7)	Cu(1)-O(1)	2.353(8)
Mo(1)-O(8)	1.695(6)	Mo(2)-O(7)	1.680(6)
Mo(1)-O(4)	1.793(6)	Mo(2)-O(2)	1.803(6)
Mo(1)-O(6)	1.813(6)	Mo(2)-O(5)	1.805(6)
Mo(1)-O(10)	2.042(5)	Mo(2)-O(9)	2.051(6)
Mo(1)-O(9)	2.057(6)	Mo(2)-O(10)#1	2.051(5)
Mo(1)-O(3)	2.398(6)	Mo(2)-O(3)	2.454(6)
Mo(3)-O(11)	1.607(6)	Mo(3)-O(5)	1.943(6)
Mo(3)-O(2)#4	1.909(6)	Mo(3)-O(6)	1.957(6)
Mo(3)-O(4)#5	1.915(6)	Mo(3)-O(3)	2.475(5)
V(1)-O(1)	1.599(8)	P(1)-O(3)#4	1.539(6)
V(1)-O(10)	1.902(5)	P(1)-O(3)#1	1.539(6)
V(1)-O(10)#1	1.902(5)	P(1)-O(3)	1.539(6)
V(1)-O(9)	1.979(5)	P(1)-O(3)#5	1.539(6)
V(1)-O(9)#1	1.979(5)	O(2)-V(2)#5	1.909(6)
O(4)-Mo(3)#4	1.915(6)	O(2)-Mo(3)#5	1.909(6)
O(10)-Mo(2)#1	2.051(5)	O(4)-V(2)#4	1.915(6)

Table S3. Selected bond lengths (\AA) for compound 3.

Symmetry transformations used to generate equivalent atoms:

#2 -y+1/4,x+1/4,-z+1/4 #3 y-1/4,-x+1/4,-

$$3 v-1/4 - x+1/4 - z+1/4$$

Mo(1)-O(10)	1.643(8)	Mo(2)-O(12)	1.654(8)
Mo(1)-O(9)	1.847(9)	Mo(2)-O(6)	1.845(8)
Mo(1)-O(15)	1.858(8)	Mo(2)-O(11)	1.846(9)
Mo(1)-O(13)	1.926(8)	Mo(2)-O(17)	1.925(8)
Mo(1)-O(6)	1.931(8)	Mo(2)-O(16)	1.951(9)
Mo(1)-O(21)	2.435(13)	Mo(2)-O(2)	2.403(13)
Mo(1)-O(2)	2.454(13)	Mo(2)-O(1)	2.407(14)
Mo(3)-O(5)	1.650(8)	Mo(4)-O(4)	1.649(8)
Mo(3)-O(19)	1.837(9)	Mo(4)-O(14)	1.791(8)
Mo(3)-O(17)#1	1.847(9)	Mo(4)-O(16)#1	1.862(9)
Mo(3)-O(18)	1.951(10)	Mo(4)-O(9)	1.928(9)
Mo(3)-O(15)#1	1.957(9)	Mo(4)-O(8)	1.990(9)
Mo(3)-O(20)	2.417(12)	Mo(4)-O(21)	2.424(12)
Mo(3)-O(2)#1	2.495(13)	Mo(4)-O(1)#1	2.464(14)
Mo(5)-O(3)	1.651(8)	Mo(6)-O(7)	1.659(7)
Mo(5)-O(22)	1.811(10)	Mo(6)-O(8)#1	1.825(9)
Mo(5)-O(13)	1.863(9)	Mo(6)-O(18)	1.843(11)
Mo(5)-O(19)	1.935(9)	Mo(6)-O(11)	1.962(10)
Mo(5)-O(14)	1.997(9)	Mo(6)-O(22)	1.980(10)
Mo(5)-O(20)	2.431(13)	Mo(6)-O(20)	2.433(12)
Mo(5)-O(21)	2.496(13)	Mo(6)-O(1)	2.500(13)
Cu(1)-N(1)#2	1.873(9)	Cu(2)-N(4)	1.876(10)
Cu(1)-N(1)	1.873(9)	Cu(2)-N(5)	1.879(10)
P(1)-O(1)	1.529(13)	P(1)-O(21)#1	1.538(13)
P(1)-O(1)#1	1.529(13)	P(1)-O(21)	1.538(13)
P(1)-O(2)	1.535(13)	P(1)-O(20)	1.581(12)
P(1)-O(2)#1	1.535(13)	P(1)-O(20)#1	1.581(12)

Table S4. Selected bond lengths (Å) for compound 4.

Symmetry transformations used to generate equivalent atoms:

#1 -x+1,-y,-z #2 -x+1,-y+1,-z

Table S5. The bond valence sum calculations for V and Mo atoms in compounds 1-4.

Compound 1				
Mo1	6.006	V1	4.199	
Mo2	6.031	V2	4.565	
Compound 2				
Mo1	6.016	V1	4.189	
Mo2	6.058	V2	4.510	
Mo3	6.189			
Compound 3				
Mo1	6.059	Mo3	6.103	
Mo2	6.146	V1	4.246	
Compound 4				
Mo1	6.466	Mo4	6.470	
Mo2	6.430	Mo5	6.343	
Mo3	6.364	Mo6	6.288	

Compound 1			
DonorHAcceptor	D - H	HA	DA
N(9)H(9A)O(3)	0.86	2.10	2.915(14)
N(9)H(9A)O(40)	0.86	2.59	3.120(15)
C(3)H(3B)O(3)	0.97	2.39	3.278(16)
C(4)H(4A)O(8)	0.93	2.60	3.524(18)
C(11)H(11A)O(8)	0.93	2.59	3.504(15)
C(17)H(17A)N(8)	0.93	2.59	2.895(15)
C(17)H(17A)O(10)	0.93	2.50	3.380(15)
C(34)H(34A)O(3)	0.93	2.60	3.316(15)
C(74)H(74A)O(21)	0.97	2.48	3.438(17)
Compound 2			
DonorHAcceptor	D - H	HA	DA
N(5)H(5A)O(5)	0.86	2.05	2.870(12)
C(2)H(2B)O(7)	0.97	2.46	3.425(13)
C(3)H(3A)O(5)	0.93	2.56	3.293(11)
C(18)H(18A)N(4)	0.93	2.59	2.901(11)
C(18)H(18A)O(8)	0.93	2.47	3.348(11)
C(21)H(21A)O(2)	0.93	2.57	3.487(11)
C(23)H(23A)O(5)	0.97	2.38	3.273(10)
Compound 3			
DonorHAcceptor	D - H	HA	DA
C(2)H(2A)O(2)	0.97	2.49	3.444(11)
C(6)H(6A)O(1)	0.93	2.58	3.089(10)
C(12)H(12A) .O(7)	0.93	2.45	3.342(12)
C(14)H(14A) .O(11)	0.93	2.57	3.488(10)
C(26)H(26B)O(8)	0.97	2.36	3.270(11)
Compound 4			
DonorHAcceptor	D - H	НА	DA
C(3)H(3B)O(6)	0.97	2.56	3.39(2)
C(3)H(3B)O(12)	0.97	2.51	3.39(2)
C(4)H(4A)N(2)	0.93	2.55	2.86(2)
C(6)H(6A)O(11)	0.93	2.56	2, 29(2)
	0.75	2.30	3.38(2)
C(6)H(6A)O(22)	0.93	2.53	3.258(18)
C(6)H(6A)O(22) C(10)H(10A)O(5)	0.93 0.93	2.53 2.39	3.258(18) 3.174(17)
C(6)H(6A)O(22) C(10)H(10A)O(5) C(11)H(11A)O(3)	0.93 0.93 0.93 0.93	2.53 2.39 2.50	3.258(18) 3.174(17) 3.289(19)
C(6)H(6A)O(22) C(10)H(10A)O(5) C(11)H(11A)O(3) C(11)H(11A)O(13)	0.93 0.93 0.93 0.93 0.93	2.50 2.53 2.39 2.50 2.53	3.258(18) 3.174(17) 3.289(19) 3.37(2)
C(6)H(6A)O(22) C(10)H(10A)O(5) C(11)H(11A)O(3) C(11)H(11A)O(13) C(13)H(13A)O(6)	0.93 0.93 0.93 0.93 0.93 0.93	2.50 2.53 2.39 2.50 2.53 2.57	3.258(18) 3.174(17) 3.289(19) 3.37(2) 3.496(19)
C(6)H(6A)O(22) C(10)H(10A)O(5) C(11)H(11A)O(3) C(11)H(11A)O(13) C(13)H(13A)O(6) C(14)H(14A)O(9)	0.93 0.93 0.93 0.93 0.93 0.93 0.93 0.93	2.53 2.53 2.50 2.50 2.53 2.57 2.37	3.258(18) 3.174(17) 3.289(19) 3.37(2) 3.496(19) 3.239(18)

Table S6. Hydrogen bonds for compounds 1-4.

Supplementary Material (ESI) for CrystEngComm This journal is (c) The Royal Society of Chemistry 2011

C(25)H(25B)O(16)	0.97	2.58	3.510(18)
C(26)H(26A)O(3)	0.93	2.25	3.017(16)
C(27)H(27A)Ow(1)	0.93	2.54	3.345(18)
C(27)H(27A)O(12)	0.93	2.36	2.939(19)
C(28)H(28A)Ow(1)	0.93	1.87	2.783(19)