Two novel halogeno(cyano)argentates built by silver halide clusters: molecular structures and luminescent properties

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Theoretical Approach Methodology. The electronic ground states of the two complexes were calculated by using B3LYP density functional theory. 3-21G quality basis sets were employed for all atoms. The ground state geometry was adapted from the truncated X-ray data, $[Ag_{12}(CN)_{11}Br_{10}]^{9-}$ for **1** and $[Ag_{12}(CN)_{11}I_{10}]^{9-}$ for **2**. Based on these geometries, time-dependent DFT (TDDFT) calculation using the B3LYP functional was carried out. The ground-state B3LYP and excited-state TDDFT calculations were carried out using Gaussian2003.ⁱ More accurate data could be obtained with more sophisticated theoretical approaches to gain detailed insights into the energy correlation among various electronic states for nontruncated structures.



Figure S1. Crystal structure of 1 viewing along the *b* axis showing the $[Et_4N]^+$ cations being located between the layers. The red dash lines represent the C-H...Br H-bonds.



Figure S2. 3-D network of **2** viewing along the *c* axis with the $(Pp_4N)^+$ cations locating the channels.



Figure S3. (a) Fluorite topology of 2 (green for octanuclear clusters, red for tetranuclear clusters); (b) CaF_2 structure (pale red for Ca ions, pale blue for F ions).



Figure S4. Electron-density distribution of the lowest unoccupied and highest occupied frontier orbitals calculated for **1** (a) and **2** (b).

D–H···A	D−H···A (Å)	∠(DHA) (°)
C14–H14A…Br1	3.807	17 <mark>4.9</mark>
C15–H15A····Br4	3.744	147.0
C24–H24A····Br2	3.94 <mark>3</mark>	1 <mark>69</mark> .7
C25–H25A····Br4	3.831	139.7
C32–H32A…Br4	3.772	147.8

Table S1. The C–H···Br hydrogen bonds in **1**.

Table S2. The calculated excitation (E), energy, oscillator strength (f), and dominant orbital

Complexes	Ex (nm)	Energy (eV)	f	Dominant configurations
1	315.8	3.93	0.0446	HOMO → LUMO
2	356.0	3.48	0.0023	HOMO \rightarrow LUMO











Figure S6. Thermogravimetric analysis curves for 1 (a) and 2 (b).





Figure S7. IR spectra for 1 (a) and 2 (b).

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