

Electronic Supplementary Information

Thiophosphorylated bis-thioureas for competitive bulk liquid membrane transport of some metal ions

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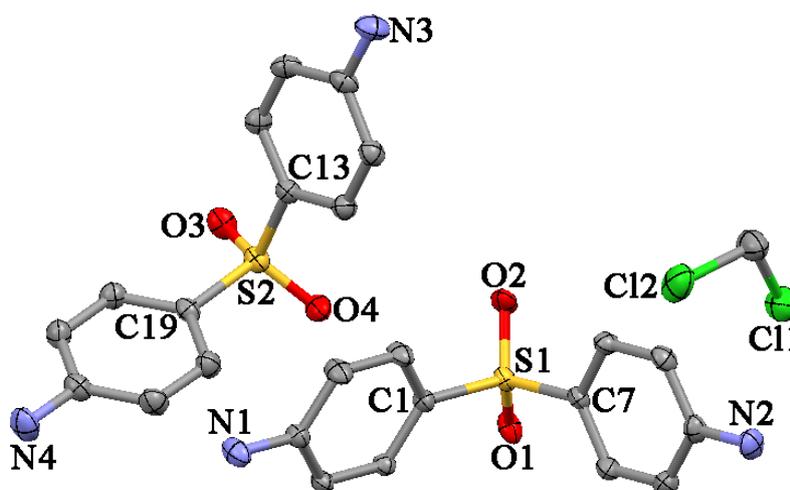


Fig. S1 Molecular structure of 2(4-aminophenylsulfone)·CH₂Cl₂. Thermal ellipsoids are drawn at the 50% probability level. Hydrogen atoms were omitted for clarity.

Table S1 Selected bond lengths (Å) and bond angles (°) for 2(4-aminophenylsulfone)·CH₂Cl₂

<i>Bond lengths</i>					
S(1)–O(1)	1.458(2)	S(1)–C(1)	1.748(2)	N(1)–C(4)	1.353(3)
S(1)–O(2)	1.442(2)	S(1)–C(7)	1.748(2)	N(2)–C(10)	1.372(3)
S(2)–O(3)	1.458(2)	S(2)–C(13)	1.748(2)	N(3)–C(16)	1.375(3)
S(2)–O(4)	1.444(2)	S(2)–C(19)	1.752(2)	N(4)–C(22)	1.360(3)
<i>Bond angles</i>					
O(1)–S(1)–O(2)	117.42(9)	O(2)–S(1)–C(7)	108.13(9)	O(3)–S(2)–C(19)	107.39(9)
O(1)–S(1)–C(1)	108.14(9)	C(1)–S(1)–C(7)	106.56(9)	O(4)–S(2)–C(13)	107.71(9)
O(1)–S(1)–C(7)	108.21(9)	O(3)–S(2)–O(4)	117.98(9)	O(4)–S(2)–C(19)	107.46(9)
O(2)–S(1)–C(1)	107.89(9)	O(3)–S(2)–C(13)	107.16(9)	C(13)–S(2)–C(19)	108.89(9)

Table S2 Selected hydrogen bond lengths (Å) and angles (°) for **2(4-aminophenylsulfone)·CH₂Cl₂**

D–H···A	<i>d</i> (D–H)	<i>d</i> (H···A)	<i>d</i> (D···A)	∠(DHA)
N(1)–H(1B)···O(3)#1	0.86	2.13	2.962(3)	164
N(1)–H(1C)···O(2)#2	0.86	2.21	3.005(3)	155
N(2)–H(2B)···O(1)#3	0.86	2.28	3.086(3)	156
N(3)–H(3B)···O(3)#3	0.86	2.24	3.053(3)	157
N(3)–H(3C)···O(1)#4	0.86	2.54	3.216(3)	136
N(4)–H(4B)···O(1)#1	0.86	2.60	3.333(3)	144
N(4)–H(4C)···O(4)#1	0.86	2.16	3.014(3)	171

Symmetry transformations used to generate equivalent atoms: #1 $-x, 1/2 + y, 3/2 - z$; #2 $1 - x, 1/2 + y, 3/2 - z$; #3 $1 + x, y, z$; #4 $x, 3/2 - y, -1/2 + z$