

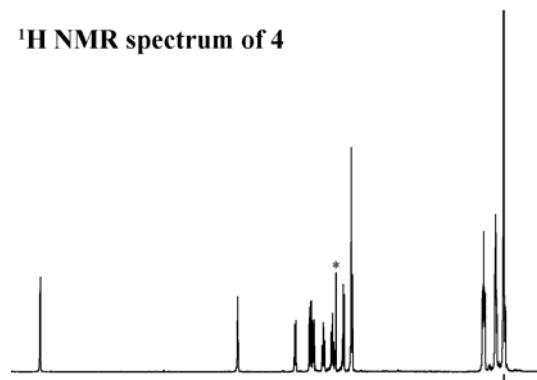
## Electronic Supplementary Information

### Crown ether-containing *N*-salicylidene aniline derivatives: synthesis, characterization and optical properties

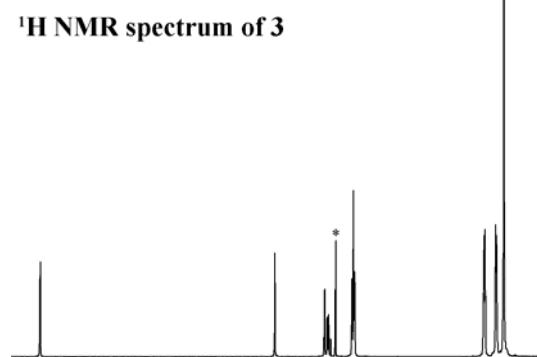
Damir A. Safin,<sup>a</sup> Koen Robeyns<sup>a</sup> and Yann Garcia\*<sup>a</sup>

<sup>a</sup> Institute of Condensed Matter and Nanosciences, MOST – Inorganic Chemistry, Université Catholique de Louvain, Place L. Pasteur 1, 1348 Louvain-la-Neuve, Belgium. Fax: +32(0) 1047 2330; Tel: +32(0) 1047 2831; E-mail: damir.safin@ksu.ru, yann.garcia@uclouvain.be

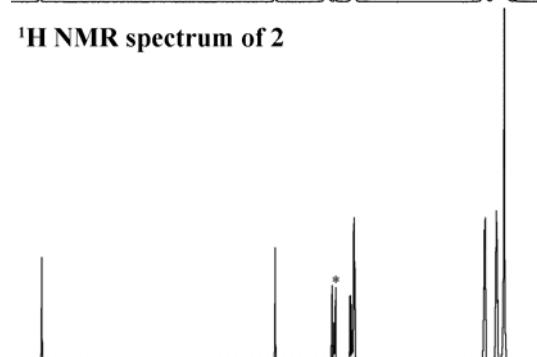
<sup>1</sup>H NMR spectrum of 4



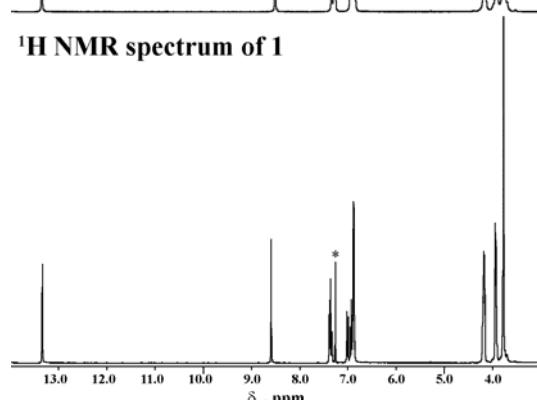
<sup>1</sup>H NMR spectrum of 3



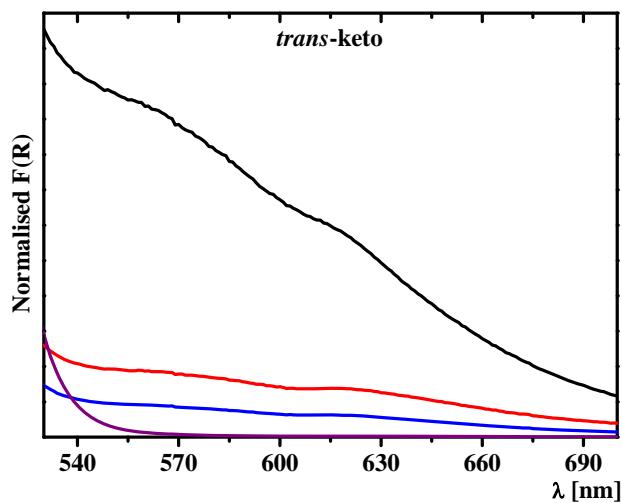
<sup>1</sup>H NMR spectrum of 2



<sup>1</sup>H NMR spectrum of 1



**Fig. S1** <sup>1</sup>H NMR spectra of **1–4**. The solvent ( $\text{CHCl}_3$ ) is marked by an asterisk.



**Fig. S2** A zoomed range of the *trans*-keto form of the Kubelka-Munk spectra of **1** (black), **2** (red), **3** (blue) and **4** (purple) at 298 K.

**Table S1** Selected bond lengths ( $\text{\AA}$ ) and bond angles ( $^\circ$ ) for **1**

<i>Bond lengths</i>					
O(1)–C(2)	1.348(5)	O(25)–C(24)	1.431(4)	C(10)–C(11)	1.404(4)
O(13)–C(12)	1.362(4)	O(25)–C(26)	1.361(4)	C(10)–C(28)	1.373(5)
O(13)–C(14)	1.420(4)	N(9)–C(8)	1.269(4)	C(11)–C(12)	1.377(5)
O(16)–C(15)	1.413(4)	N(9)–C(10)	1.418(4)	C(12)–C(26)	1.402(4)
O(16)–C(17)	1.415(5)	C(2)–C(3)	1.394(6)	C(14)–C(15)	1.495(5)
O(16)–C(17B)	1.42(4)	C(2)–C(7)	1.395(5)	C(17)–C(18)	1.487(8)
O(19)–C(18)	1.425(6)	C(3)–C(4)	1.367(6)	C(17B)–C(18B)	1.49(4)
O(19)–C(18B)	1.41(2)	C(4)–C(5)	1.374(6)	C(20)–C(21)	1.486(6)
O(19)–C(20)	1.422(5)	C(5)–C(6)	1.383(6)	C(23)–C(24)	1.492(5)
O(22)–C(21)	1.409(5)	C(6)–C(7)	1.382(5)	C(26)–C(27)	1.379(4)
O(22)–C(23)	1.426(4)	C(7)–C(8)	1.450(5)	C(27)–C(28)	1.377(5)
<i>Bond angles</i>					
C(12)–O(13)–C(14)	117.8(3)	C(5)–C(6)–C(7)	121.1(3)	O(16)–C(17)–C(18)	113.0(4)
C(15)–O(16)–C(17)	114.7(4)	C(2)–C(7)–C(6)	118.8(3)	O(19)–C(18)–C(17)	116.4(5)
C(15)–O(16)–C(17B)	102.6(13)	C(2)–C(7)–C(8)	121.9(3)	O(19)–C(20)–C(21)	107.1(3)
C(18)–O(19)–C(20)	113.2(3)	C(6)–C(7)–C(8)	119.3(3)	O(22)–C(21)–C(20)	112.4(3)
C(20)–O(19)–C(18B)	125.1(11)	N(9)–C(8)–C(7)	122.8(3)	O(22)–C(23)–C(24)	111.4(3)
C(21)–O(22)–C(23)	116.6(3)	N(9)–C(10)–C(11)	124.3(3)	O(25)–C(24)–C(23)	107.8(3)
C(24)–O(25)–C(26)	117.4(2)	N(9)–C(10)–C(28)	117.4(3)	O(25)–C(26)–C(12)	115.8(3)
C(8)–N(9)–C(10)	122.2(3)	C(11)–C(10)–C(28)	118.3(3)	O(25)–C(26)–C(27)	125.0(3)
O(1)–C(2)–C(3)	118.7(3)	C(10)–C(11)–C(12)	120.6(3)	C(12)–C(26)–C(27)	119.2(3)
O(1)–C(2)–C(7)	121.5(3)	O(13)–C(12)–C(11)	125.1(3)	C(26)–C(27)–C(28)	120.2(3)
C(3)–C(2)–C(7)	119.8(3)	O(13)–C(12)–C(26)	114.9(3)	C(10)–C(28)–C(27)	121.7(3)
C(2)–C(3)–C(4)	120.0(4)	C(11)–C(12)–C(26)	120.0(3)	O(16)–C(17B)–C(18B)	115(3)
C(3)–C(4)–C(5)	120.8(4)	O(13)–C(14)–C(15)	107.3(3)	O(19)–C(18B)–C(17B)	105(2)
C(4)–C(5)–C(6)	119.5(4)	O(16)–C(15)–C(14)	108.2(2)		
<i>Torsion angles</i>					
C(14)–O(13)–C(12)–C(11)	-13.1(4)	O(1)–C(2)–C(7)–C(8)	0.4(5)	N(9)–C(10)–C(28)–C(27)	178.9(3)
C(14)–O(13)–C(12)–C(26)	166.2(3)	C(3)–C(2)–C(7)–C(6)	1.5(5)	C(11)–C(10)–C(28)–C(27)	-0.1(5)
C(12)–O(13)–C(14)–C(15)	-173.8(3)	C(3)–C(2)–C(7)–C(8)	-178.4(4)	C(10)–C(11)–C(12)–O(13)	178.6(3)
C(17)–O(16)–C(15)–C(14)	-171.8(4)	C(2)–C(3)–C(4)–C(5)	0.8(7)	C(10)–C(11)–C(12)–C(26)	-0.6(5)
C(15)–O(16)–C(17)–C(18)	87.8(6)	C(3)–C(4)–C(5)–C(6)	-0.7(7)	O(13)–C(12)–C(26)–O(25)	-0.1(4)
C(20)–O(19)–C(18)–C(17)	-79.8(5)	C(4)–C(5)–C(6)–C(7)	1.0(6)	O(13)–C(12)–C(26)–C(27)	-179.3(3)
C(18)–O(19)–C(20)–C(21)	-149.6(4)	C(5)–C(6)–C(7)–C(2)	-1.4(6)	C(11)–C(12)–C(26)–O(25)	179.2(3)
C(23)–O(22)–C(21)–C(20)	-84.5(4)	C(5)–C(6)–C(7)–C(8)	178.5(4)	C(11)–C(12)–C(26)–C(27)	0.1(5)
C(21)–O(22)–C(23)–C(24)	124.6(3)	C(2)–C(7)–C(8)–N(9)	0.1(6)	O(13)–C(14)–C(15)–O(16)	68.8(4)
C(26)–O(25)–C(24)–C(23)	178.3(3)	C(6)–C(7)–C(8)–N(9)	-179.8(4)	O(16)–C(17)–C(18)–O(19)	74.0(6)
C(24)–O(25)–C(26)–C(12)	-175.3(3)	C(7)–C(8)–N(9)–C(10)	176.9(3)	O(19)–C(20)–C(21)–O(22)	-169.7(3)
C(24)–O(25)–C(26)–C(27)	3.9(5)	C(11)–C(10)–N(9)–C(8)	-11.0(5)	O(22)–C(23)–C(24)–O(25)	-75.1(3)
O(1)–C(2)–C(3)–C(4)	-180.0(4)	C(28)–C(10)–N(9)–C(8)	170.2(3)	O(25)–C(26)–C(27)–C(28)	-178.6(3)
C(7)–C(2)–C(3)–C(4)	-1.2(6)	N(9)–C(10)–C(11)–C(12)	-178.2(3)	C(12)–C(26)–C(27)–C(28)	0.5(5)
O(1)–C(2)–C(7)–C(6)	-179.8(4)	C(28)–C(10)–C(11)–C(12)	0.6(5)	C(26)–C(27)–C(28)–C(10)	-0.5(6)

**Table S2** Selected hydrogen bond lengths ( $\text{\AA}$ ) and angles ( $^\circ$ ) for **1**

D–H $\cdots$ A	$d(\text{D–H})$	$d(\text{H}\cdots\text{A})$	$d(\text{D}\cdots\text{A})$	$\angle(\text{DHA})$
O(1)–H(1) $\cdots$ N(9)	0.82	1.90	2.627(4)	147

**Table S3** Selected  $\pi\cdots\pi$  interactions for **1**

Cg( <i>I</i> )	Cg( <i>J</i> )	Cg–Cg ( $\text{\AA}$ )	Dihedral angle ( $^\circ$ )	$\beta$ ( $^\circ$ )
Cg(1)	Cg(1)	5.403(2)	85.92	28.23
Cg(1)	Cg(2)	4.665(2)	12.91	37.78
Cg(2)	Cg(1)	4.665(2)	12.91	44.28
Cg(1): C(2)–C(3)–C(4)–C(5)–C(6)–C(7); Cg(2): C(10)–C(11)–C(12)–C(26)–C(27)–C(28)				
$\beta$ : angle Cg( <i>I</i> ) $\rightarrow$ Cg( <i>J</i> ) vector and normal to plane I.				