

Table SI1. Catalytic activities of **AsWRuZn-2** with different amounts for the oxidation of n-hexadecane under green conditions.^[a]

Catalyst	Catalyst amount mg ⁻¹	Conv [%]	TOF [h ⁻¹]	Product selectivity [%] and distribution	
				Ketones (7-One:6-One:5-One:4-One:3-One:2-One)	Alcohols (7-Ol:6-Ol:5-Ol:4-Ol:3-Ol:2-Ol)
AsWRuZn-2	2.5	38.38	2796	51.4 (13.5:6.4:8:6.8:7.3:9.4)	22.3 (7.6:3:4.8:2.9:3.2:0.8)
	5	46.98	1711	52.7 (13.6:6.8:8.7:7:7.6:9)	20.7 (6.9:2.2:5.3:2.4:2.9:1)
	10	51.49	938	55.4 (13.8:7.5:9.4:7.3:8.1:9.3)	19.1 (5.6:2.4:4.9:2:3:1.2)
	15	44.87	545	53.9 (13.3:7.5:8.8:7.4:7.9:9)	20.6 (6.5:2:5.1:3:2.9:1.1)

^[a]Reaction conditions: n-hexadecane: 5 mL, airflow rate: 30 mL·min⁻¹, temperature: 150°C, reaction time: 6 h.

Table SI2. Catalytic activities of **AsWRuZn-2** with different reaction time for the oxidation of n-hexadecane under green conditions.^[a]

Catalyst	Reaction time [h]	Conv [%]	TOF [h ⁻¹]	Product selectivity [%] and distribution	
				Ketones (7-One:6-One:5-One:4-One:3-One:2-One)	Alcohols (7-Ol:6-Ol:5-Ol:4-Ol:3-Ol:2-Ol)
AsWRuZn-2	0.5	2.23	487	55.6 (12.7:6.7:6.9:7.4:13.3:8.6)	38.2 (15.5:4.4:6.4:2:6.9:1.2)
	1.5	8.99	655	48.9 (12.6:6.5:7:6.2:8.4:8.2)	32.1 (12.5:3.9:5.7:4.5:4.9:0.6)
	3	24.78	903	51.5 (13:7:8.1:6.7:7.8:8.9)	23.6 (8.2:3.1:4.9:2.9:3.6:0.9)
	4	35.47	969	50.3 (13.7:6.2:8.1:6.5:7.5:8.3)	20.4 (6.7:2.3:4.4:2.6:3.4:1)
	5	47.29	1064	54.1 (13.4:7.4:8.3:7.8:8.7:8.5)	16.6 (5.3:1.8:4.1:1.8:2.5:1.1)
	6	51.49	938	55.4 (13.8:7.5:9.4:7.3:8.1:9.3)	19.1 (5.6:2.4:4.9:2:3:1.2)
	7	55.84	871	50.4 (12.1:7.3:8.9:6.7:7:8.4)	16.8 (4.5:2.2:4.8:1.7:2.5:1.1)
	8	59.35	810	58.6 (14.3:8.5:9.6:7.7:8.3:10.2)	17.3 (5.2:2.1:4.6:1.8:2.5:1.1)
	12	73.93	673	53.7 (13.5:7.1:9.6:7.1:7.8:8.6)	17.1 (4.9:2.4:4.8:1.7:2.2:1.3)

^[a]Reaction conditions: n-hexadecane: 5 mL, airflow rate: 30 mL·min⁻¹, temperature: 150°C, catalyst: 10mg.

Table SI3. Reported catalytic activity of polyoxometalates as catalysts for n-hexadecane oxidation with air.^[a]

Catalyst	Conv [%] ^[c]	TOF [h ⁻¹] ^[d]	Product selectivity [%]	
			Ketones	Alcohols
Blank ^[1,2]	3.9	-	55	24
Cu20 ^[1]	7.3	279	35	37
KNa-1 ^[2]	38.3	636	52	22
CsNa-2 ^[2]	33.7	570	50	29
CsNa-3 ^[2]	34.0	605	51	25
Na-4 ^[2]	31.8	570	50	23
Fe ₄ Se ₂ W ₁₈ ^[3]	3.0	17	51.2	

^[a] Identical reaction conditions: n-hexadecane: 5 mL, airflow rate: 30 mL·min⁻¹, temperature: 150°C, catalyst amount: 10mg, reaction time: 6h

References:

- [1] Chen, L. F.; Hu, J. C.; Mal, S. S.; Kortz, U.; Jaensch, H.; Mathys, G.; Richards, R. M. *Chem. Eur. J.* **2009**, *15*, 7490.
- [2] Bi, L. H.; Al-Kadamany, G.; Chubarova, E. V.; Dickman, M. H. L.; Chen, F.; Gopala, D. S.; Richards, R.; Keita, M. B.; Nadjo, L.; Jaensch, H.; Mathys, G.; Kortz, U. *Inorg. Chem.* **2009**, *48*, 10068.
- [3] Chen, L. F.; Zhu, K. K.; Bi, L. H.; Suchopar, A.; Reicke, M.; Mathys, G.; Jaensch, H.; Kortz, U.; Richards, R. M. *Inorg. Chem.* **2007**, *46*, 8457.

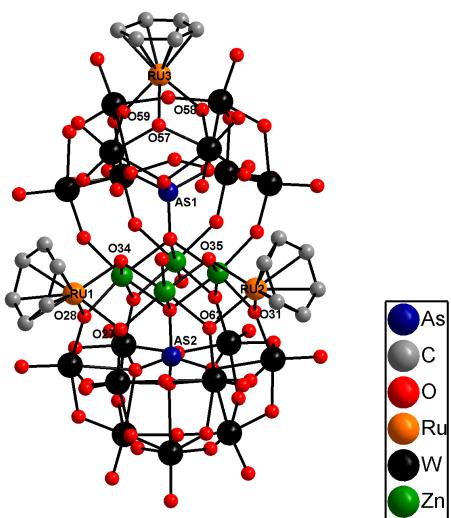


Figure SI1. The atom-labeling scheme of the ball-and-stick representation of polyanion $[\{\text{B}-\alpha\text{-AsW}_9\text{O}_{34}\} \{\text{B}-\beta\text{-AsW}_8\text{O}_{31}\} \{\text{Zn}_4(\text{OH})_2(\text{H}_2\text{O})_2\} \{(\text{RuC}_6\text{H}_6)_3\}]^{6-}$ (AsWRuZn-2).

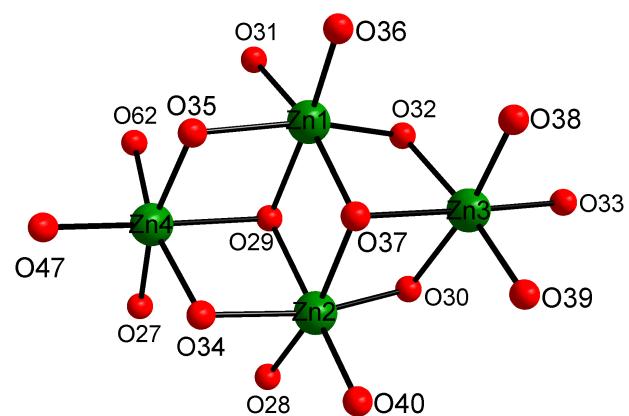


Figure SI2. The atom-labeling scheme of the central tetrameric unit Zn_4O_{16} in **AsWRuZn-2**. The balls represent zinc (olive) and oxygen (red).

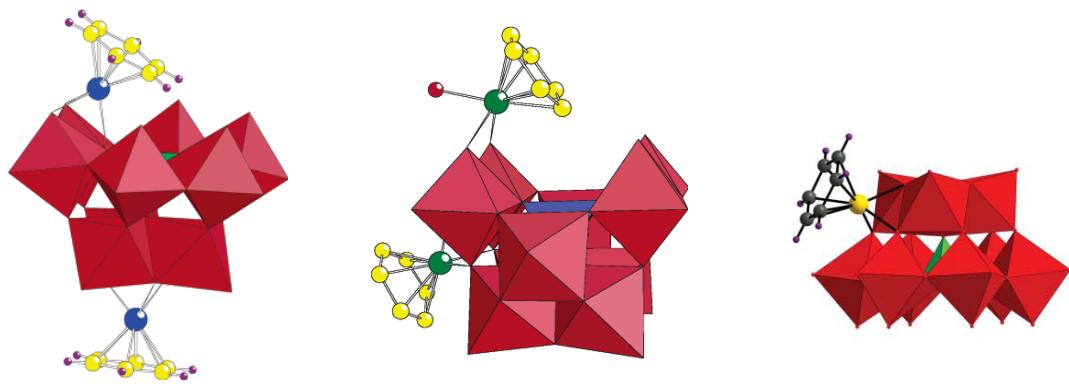


Figure SI3. Polyhedral/ball and stick representations of the polyanions (left) $[(\text{RuC}_6\text{H}_6)_2\text{XW}_9\text{O}_{34}]^{6-}$ (X = Si, Ge) (13), (middle) $[\{\text{Ru}(\text{C}_6\text{H}_6)(\text{H}_2\text{O})\} \{\text{Ru}(\text{C}_6\text{H}_6)\}(\gamma\text{-XW}_{10}\text{O}_{36})]^{4-}$ (X = Si, Ge) (14) and (right). $[(\text{RuC}_6\text{H}_6)\text{XW}_9\text{O}_{34}]^{7-}$ (X = P, As) (15).

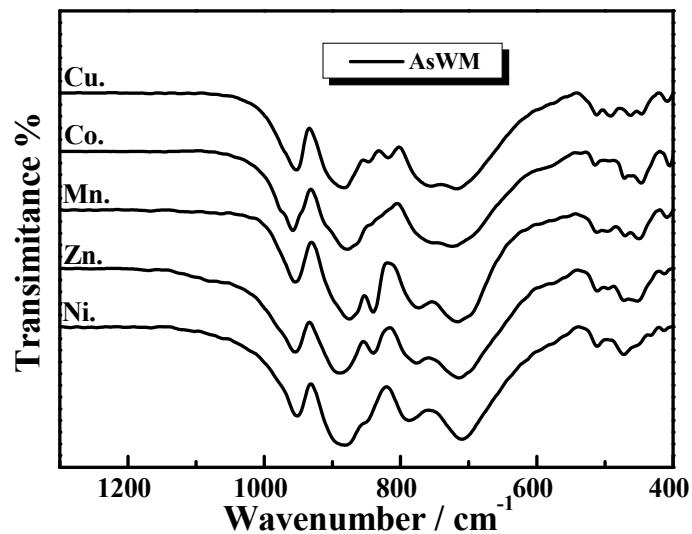


Figure SI4. IR spectra of **AsWM-6-10**.

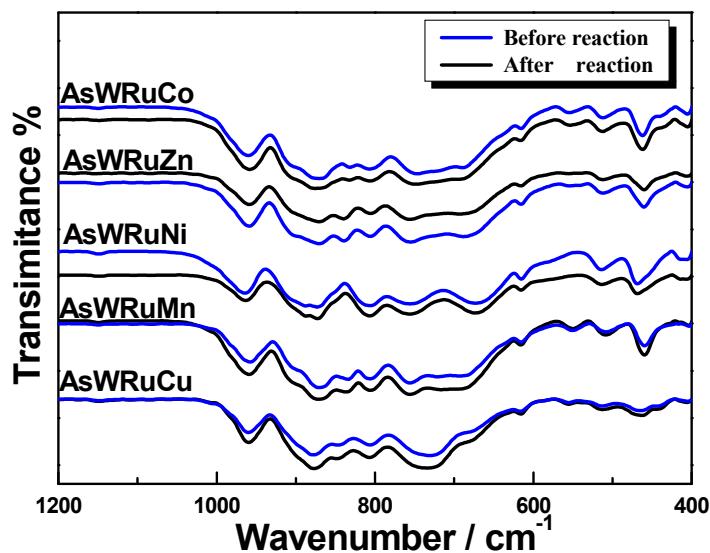


Figure SI5. IR spectra of **AsWRuM-1-5** before (blue) and after (black) catalytic reactions.

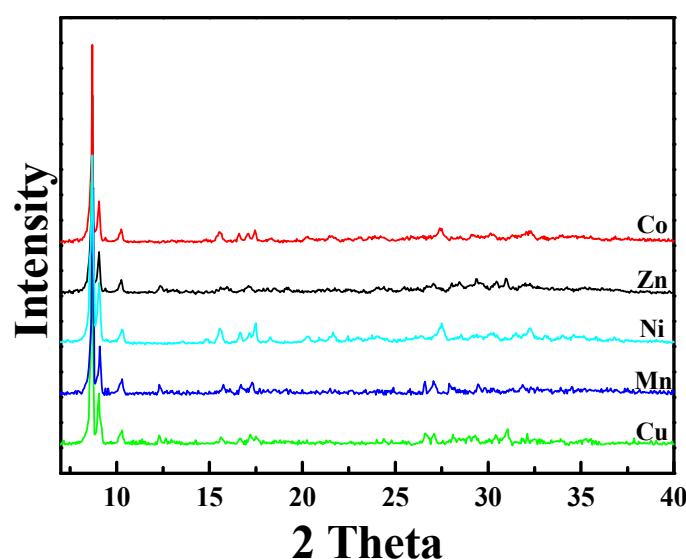


Figure SI6. The powder XRD diffraction patterns of compounds **AsWRuM-1-5** after the following treatment process.

The treatment procedure of the samples for powder XRD measurements is described below:

- 1) The crystals of five compounds were picked out under microscopy to ensure their purities;
- 2) These crystals were dried in vacuum dryer for one week to obtain powder samples;
- 3) The powder samples were further crushed to fine powders for XRD measurements.