

# Supporting information

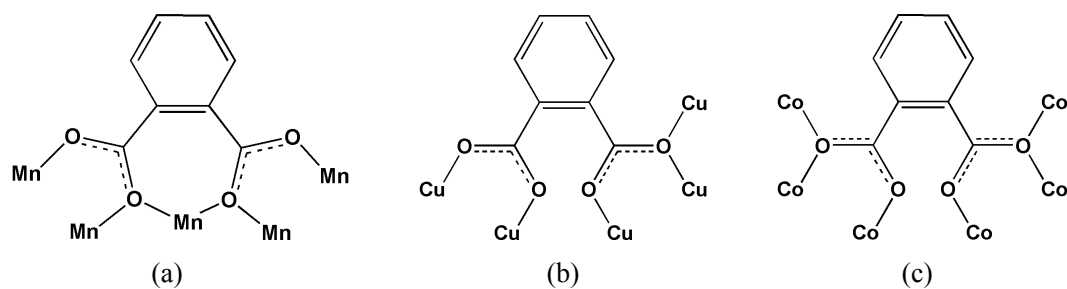
## **Mn<sup>II</sup>, Cu<sup>II</sup> and Co<sup>II</sup> coordination polymers showing antiferromagnetism, coexistence of spin frustration and long range magnetic ordering**

*Ya-Min Li<sup>a\*</sup>, Chang-Yu Xiao<sup>a</sup>, Xu-Dong Zhang<sup>b</sup>, Yan-Qing Xu<sup>c</sup>, Hui-Jie Lun<sup>a</sup>, Jing-Yang Niu<sup>a\*</sup>*

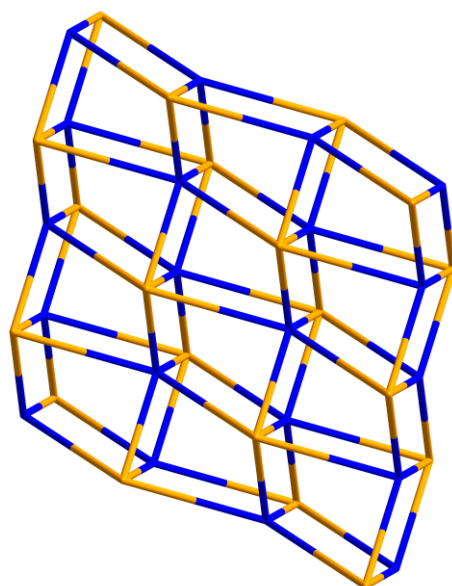
<sup>a</sup> Henan Key Laboratory of Polyoxometalate; Institute of Molecular and Crystal Engineering, College of Chemistry and Chemical Engineering, Henan University, Kaifeng, Henan 475004, PR China; E-mail: liyamin@henu.edu.cn

<sup>b</sup> State Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fuzhou, Fujian 350002, PR China

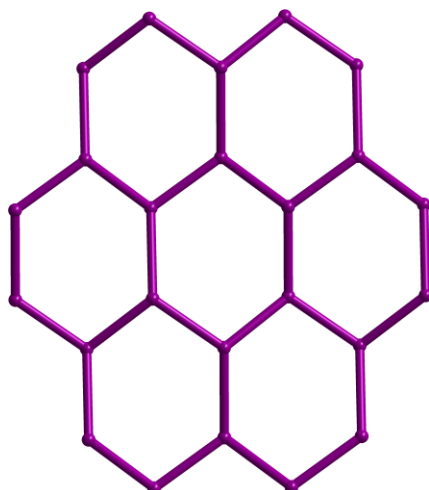
<sup>c</sup> School of Chemistry, Key Laboratory of Cluster Science, Ministry of Education of China, Beijing Institute of Technology, Beijing, 100081, PR China



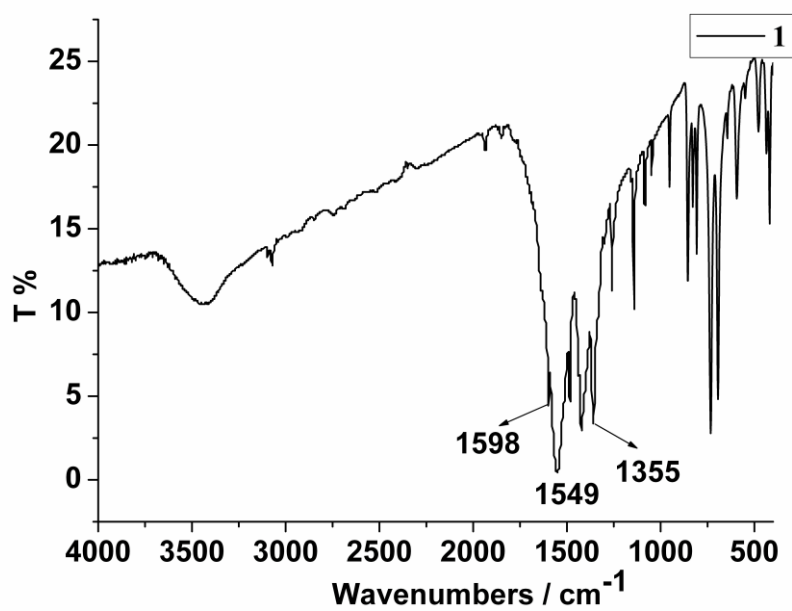
Scheme S1 Different coordination modes of the ligand 1,2- $\text{bdc}^{2-}$  (a) in complex **1**, (b) in complex **2** and (c) in complex **3**.



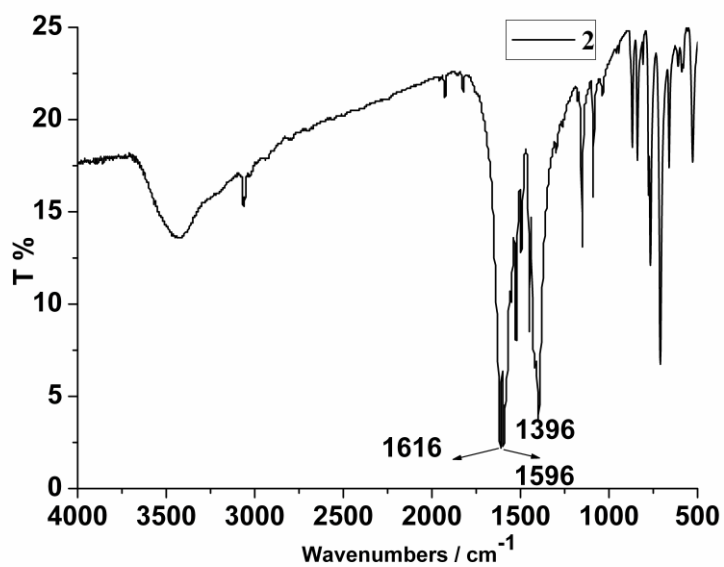
**Fig. S1** A five-connected 2D network with point symbol of  $\{4^8.6^2\}$  in **1** with  $\text{Mn}^{\text{II}}$  being represented as blue balls and the ligand  $\text{bdc}^{2-}$  as orange balls..



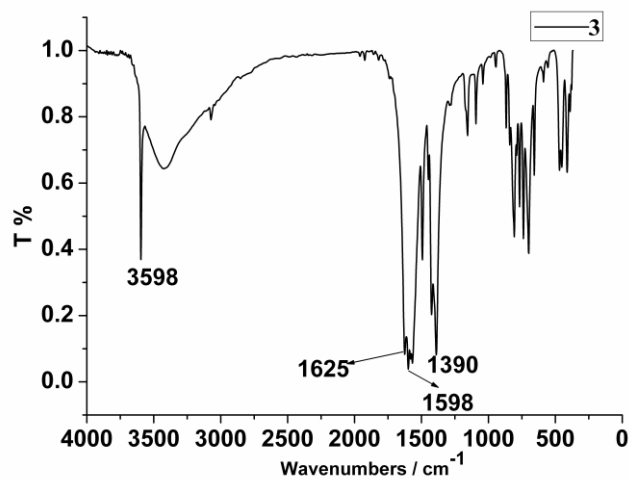
**Fig. S2** A 3-connected **hcb** topological net in **3** with  $\text{Co}_3(\text{OH})$  being represented by purple balls.



(a)

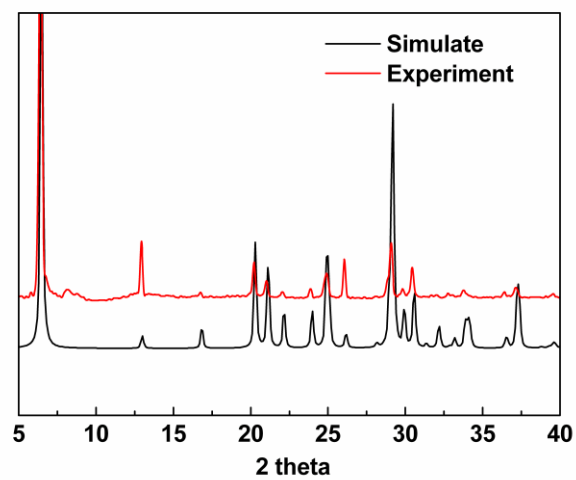


(b)

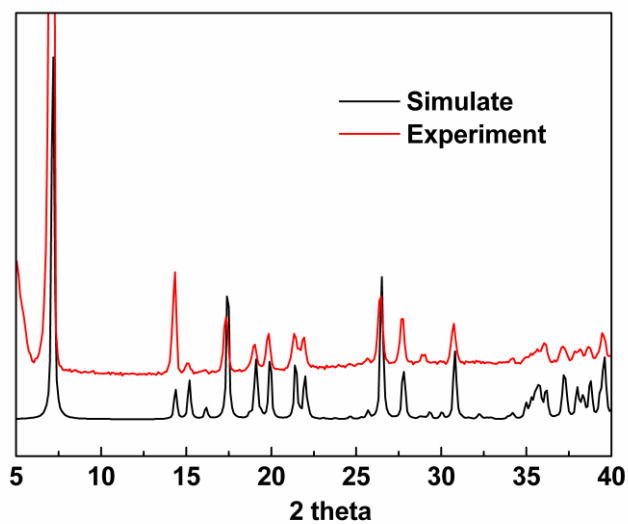


(c)

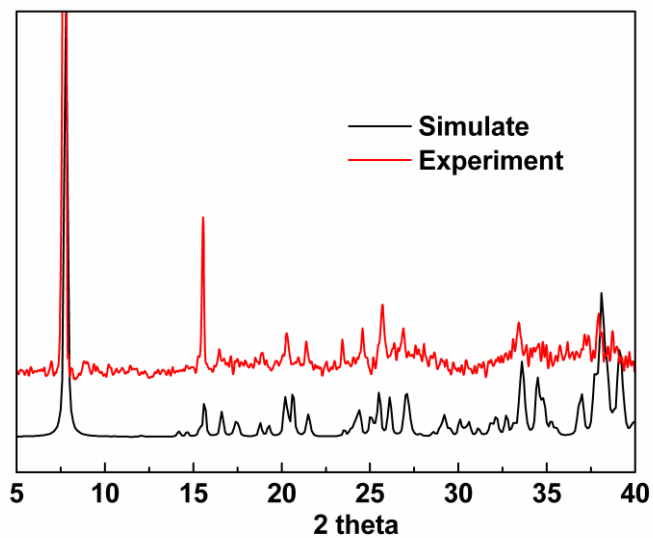
Fig. S3 IR spectra for three compounds.



(a)



(b)



(c)

Fig. S4 PXR D curves for (a) 1, (b) 2, (c) 3.

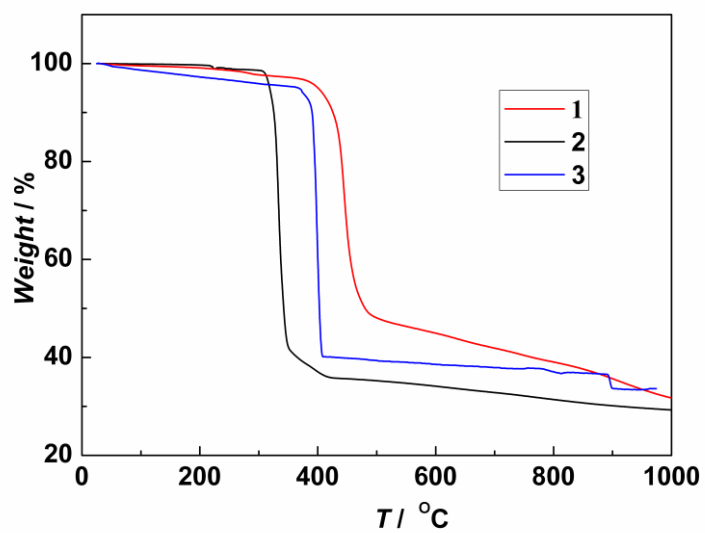


Fig. S5 TGA curves for three compounds.

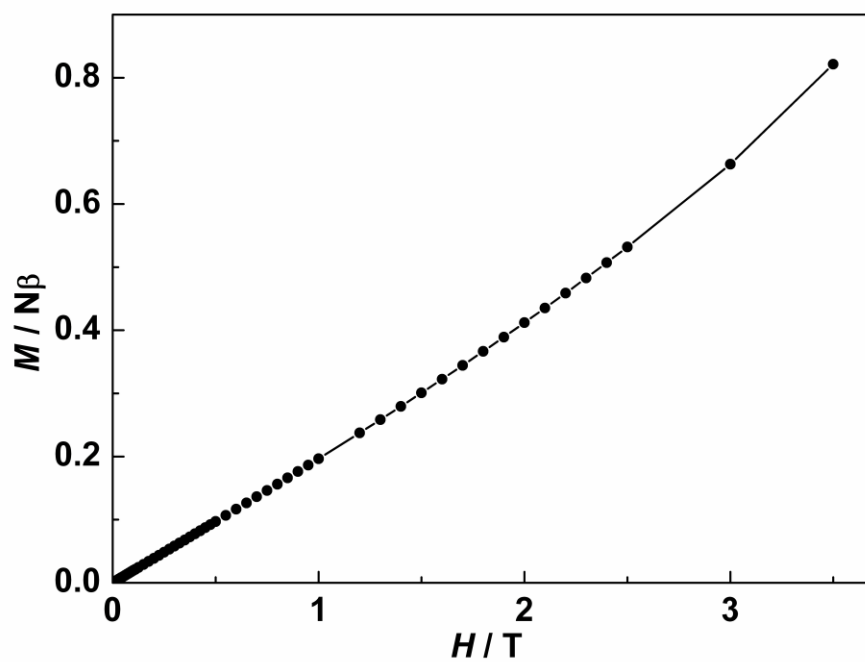


Fig. S6 The curve of magnetization vs. applied fields at 2 K in 1.

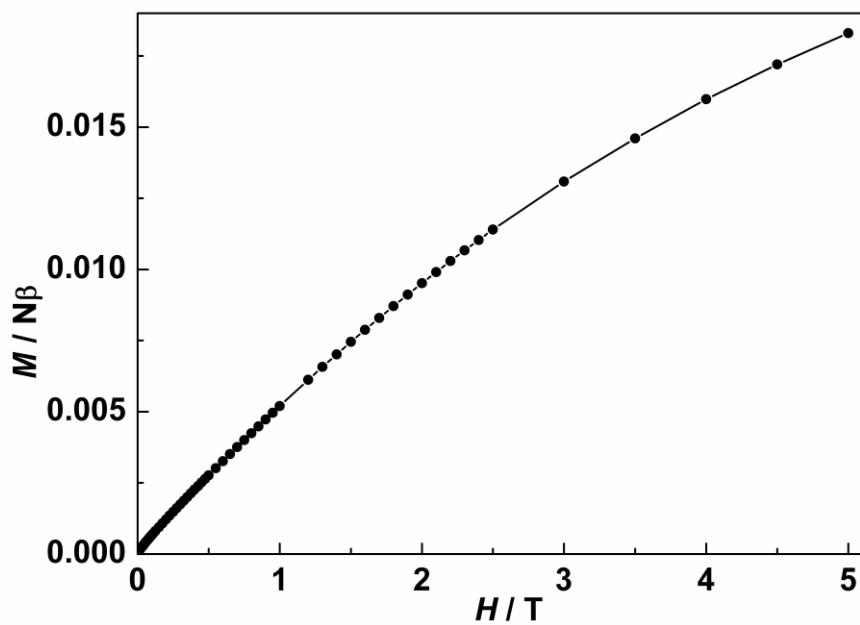
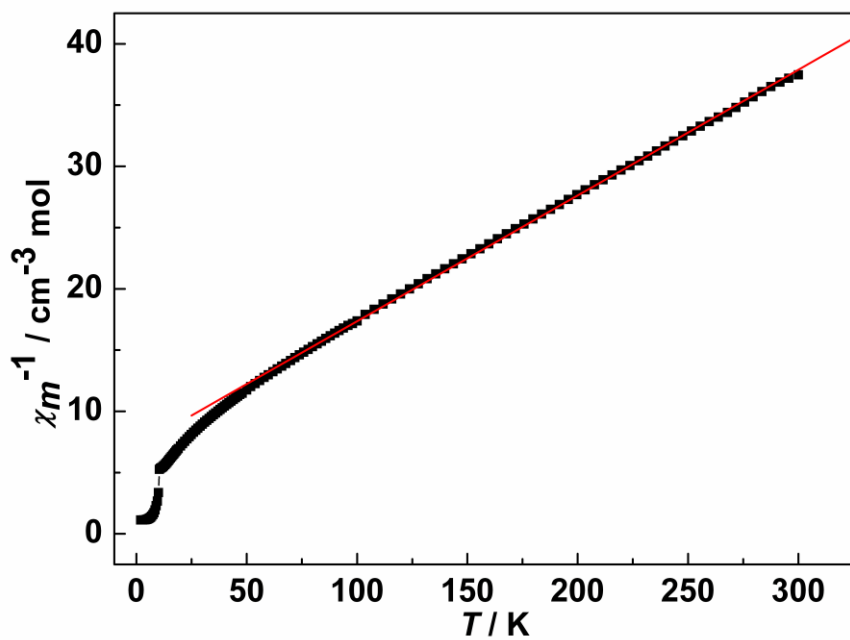
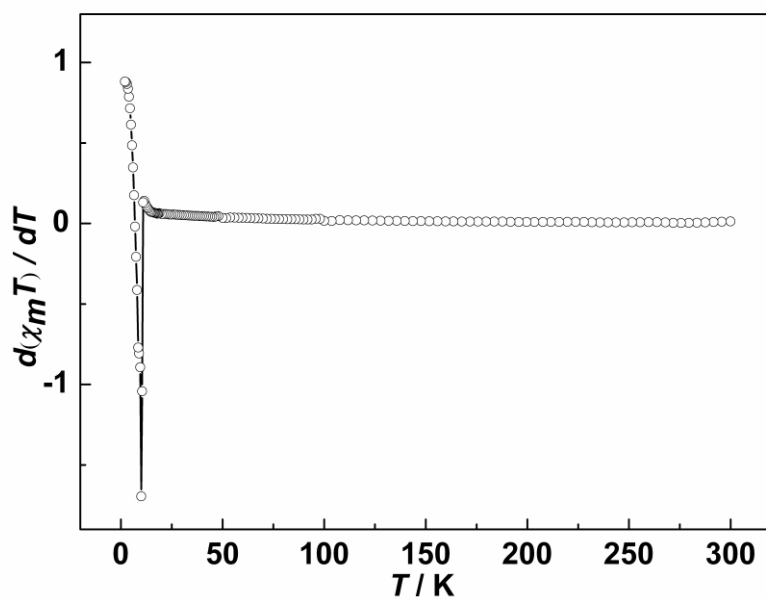


Fig. S7 The curve of magnetization vs. applied fields at 2 K in 2.



**Fig. S8** The  $\chi_m^{-1}$  vs.  $T$  plot of **3** in the range 2-300 K at 1 kOe. The solid line is the best-fit above 50 K according to the Curie-Weiss law.



**Fig. S9** The  $d(\chi_m T) / dT$  derivative curve of **3**.



**Table S1** Crystallographic data for compound **1~3**.

	<b>1</b>	<b>2</b>	<b>3</b>
Empirical formula	C <sub>8</sub> H <sub>4</sub> MnO <sub>4</sub>	C <sub>8</sub> H <sub>4</sub> CuO <sub>4</sub>	C <sub>16</sub> H <sub>10</sub> Co <sub>3</sub> O <sub>10</sub>
Formula weight	219.05	227.65	539.03
Crystal system	Monoclinic	Monoclinic	Monoclinic
Space group	<i>P2/c</i>	<i>P2<sub>1</sub>/c</i>	<i>P2<sub>1</sub>/c</i>
Unit cell dimensions			
<i>a</i> (Å)	4.664(2)	13.171(5)	11.648(2)
<i>b</i> (Å)	13.625(6)	5.1292(18)	12.091(3)
<i>c</i> (Å)	6.339(2)	11.719(4)	11.846(3)
<i>B</i> (°)	115.82(3)	110.745(6)	103.245(4)
<i>V</i> (Å <sup>3</sup> )	362.6(3)	740.3(5)	1624.0(6)
<i>Z</i>	2	4	4
$\rho$ calcd.(Mg/m <sup>3</sup> )	2.006	2.042	2.205
$\mu$ (mm <sup>-1</sup> )	1.790	2.920	3.090
F(000)	218	452	1068
$\theta$ limits (°)	2.99 to 24.99 deg	1.65 to 25.00	1.80 to 25.00
<i>h, k, l</i> limits	-5 to 5, -15 to 16, -7 to 7	-13 to 15, -3 to 6, -13 to 13	-13 to 12, -14 to 13, -14 to 13
Reflections collected	2341	3523	7928
GOOF	1.218	1.064	0.966
<i>R</i> index [ <i>I</i> > 2 $\sigma$ ( <i>I</i> )]	<i>R</i> <sub>1</sub> = 0.0435, <i>wR</i> <sub>2</sub> = 0.1193	<i>R</i> <sub>1</sub> = 0.0444, <i>wR</i> <sub>2</sub> = 0.1070	<i>R</i> <sub>1</sub> = 0.0409, <i>wR</i> <sub>2</sub> = 0.1045
<i>R</i> (all data)	<i>R</i> <sub>1</sub> = 0.0459, <i>wR</i> <sub>2</sub> = 0.1199	<i>R</i> <sub>1</sub> = 0.0535, <i>wR</i> <sub>2</sub> = 0.1106	<i>R</i> <sub>1</sub> = 0.0601, <i>wR</i> <sub>2</sub> = 0.1136
Largest diff. peak and hole (e·Å <sup>-3</sup> )	1.162 and -0.487	1.357 and -0.781	1.394 and -0.757

$${}^a R = \frac{\sum(|F_o| - |F_c|)}{\sum|F_o|}, {}^b wR = \left\{ \frac{\sum w[(F_o^2 - F_c^2)^2]}{\sum w(F_o^2)^2} \right\}^{1/2}, w = 1/[\sigma^2(F_o^2) + (aP)^2 + bP], P = (F_o^2 + 2F_c^2)/3. \mathbf{1}, a = 0.0694, b = 0.5648; \mathbf{2}, a = 0.0653, b = 0.5533; \mathbf{3}, a = 0.0722, b = 0.0000.$$

**Table S2** Selected bond lengths and bond angles for compound **1-3**.

<b>1</b>			
Mn(1)-O(2)#1	2.114(3)	Mn(1)-O(1)#3	2.170(3)
Mn(1)-O(1)	2.243(3)	Mn(1)-Mn(1)#4	3.4812(11)
O(2)#1-Mn(1)-O(2)#2	101.03(15)	O(2)#1-Mn(1)-O(1)#3	154.82(10)
O(2)#2-Mn(1)-O(1)#3	94.47(10)	O(1)#3-Mn(1)-O(1)#4	78.78(13)
O(2)#1-Mn(1)-O(1)	87.13(10)	O(2)#2-Mn(1)-O(1)	85.16(10)
O(1)#3-Mn(1)-O(1)	114.08(10)	O(1)#4-Mn(1)-O(1)	75.85(10)
O(1)-Mn(1)-O(1)#5	167.86(13)		
<b>2</b>			
Cu(1)-O(2)	1.935(3)	Cu(1)-O(1)	1.953(3)
Cu(1)-O(3)	1.957(3)	Cu(1)-O(4)	2.033(3)
Cu(1)-O(4)#1	2.204(3)	Cu(1)-Cu(1)#2	2.6229(12)
O(2)-Cu(1)-O(1)	168.68(14)	O(2)-Cu(1)-O(3)	86.24(14)
O(1)-Cu(1)-O(3)	91.67(14)	O(2)-Cu(1)-O(4)	88.61(14)
O(1)-Cu(1)-O(4)	91.25(13)	O(3)-Cu(1)-O(4)	167.96(13)
O(2)-Cu(1)-O(4)#1	90.46(13)	O(1)-Cu(1)-O(4)#1	100.65(13)
O(3)-Cu(1)-O(4)#1	111.33(12)		
<b>3</b>			
Co(1)-O(10)	2.005(3)	Co(2)-O(9)#4	2.082(3)
Co(1)-O(9)	2.007(3)	Co(2)-O(10)	2.088(3)
Co(1)-O(6)	2.060(3)	Co(2)-O(4)#1	2.104(3)
Co(1)-O(1)#1	2.064(3)	Co(2)-O(5)	2.106(3)
Co(1)-O(3)#2	2.345(3)	Co(2)-O(7)#4	2.109(3)
Co(1)-O(8)#3	2.367(3)	Co(2)-O(2)	2.109(3)
Co(3)-O(9)	2.088(3)	Co(4)-O(10)	2.082(3)
Co(3)-O(5)#1	2.098(3)	Co(4)-O(2)	2.091(3)
Co(3)-O(3)#2	2.099(3)	Co(4)-O(8)#3	2.118(3)
O(10)-Co(1)-O(9)	173.73(12)	O(9)#4-Co(2)-O(10)	179.07(12)
O(10)-Co(1)-O(6)	95.72(12)	O(9)#4-Co(2)-O(4)#1	86.27(12)
O(9)-Co(1)-O(6)	88.19(12)	O(10)-Co(2)-O(4)#1	92.91(12)
O(10)-Co(1)-O(1)#1	88.58(13)	O(9)#4-Co(2)-O(5)	78.87(11)
O(9)-Co(1)-O(1)#1	95.66(13)	O(10)-Co(2)-O(5)	101.61(11)
O(6)-Co(1)-O(1)#1	99.01(14)	O(4)#1-Co(2)-O(5)	93.22(12)
O(10)-Co(1)-O(3)#2	101.28(11)	O(9)#4-Co(2)-O(7)#4	92.99(12)
O(9)-Co(1)-O(3)#2	74.79(11)	O(10)-Co(2)-O(7)#4	86.66(12)
O(6)-Co(1)-O(3)#2	162.97(12)	O(4)#1-Co(2)-O(7)#4	95.13(12)
O(1)#1-Co(1)-O(3)#2	82.59(13)	O(5)-Co(2)-O(7)#4	167.93(12)
O(10)-Co(1)-O(8)#3	74.81(12)	O(9)#4-Co(2)-O(2)	102.09(12)
O(9)-Co(1)-O(8)#3	100.88(12)	O(10)-Co(2)-O(2)	78.79(12)
O(6)-Co(1)-O(8)#3	83.18(13)	O(4)#1-Co(2)-O(2)	167.42(12)

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O(1)#1-Co(1)-O(8)#3	163.38(12)	O(5)-Co(2)-O(2)	79.39(12)
O(3)#2-Co(1)-O(8)#3	100.17(11)	O(7)#4-Co(2)-O(2)	93.81(12)
O(9)#5-Co(3)-O(9)	180.00(12)	O(10)-Co(4)-O(10)#2	180.00(13)
O(9)-Co(3)-O(5)#1	78.92(11)	O(10)-Co(4)-O(2)#2	100.67(11)
O(9)-Co(3)-O(5)#3	101.08(11)	O(10)-Co(4)-O(2)	79.33(11)
O(5)#1-Co(3)-O(5)#3	180.0	O(10)#2-Co(4)-O(2)	100.67(11)
O(9)-Co(3)-O(3)#6	101.24(12)	O(2)#2-Co(4)-O(2)	180.00(8)
O(5)#1-Co(3)-O(3)#6	81.72(12)	O(10)-Co(4)-O(8)#3	78.96(12)
O(9)-Co(3)-O(3)#2	78.76(12)	O(2)-Co(4)-O(8)#3	98.27(12)
O(5)#1-Co(3)-O(3)#2	98.28(12)	O(10)-Co(4)-O(8)#4	101.04(12)
O(5)#3-Co(3)-O(3)#2	81.72(12)	O(2)-Co(4)-O(8)#4	81.73(12)
O(3)#6-Co(3)-O(3)#2	180.0	O(8)#3-Co(4)-O(8)#4	180.00(9)

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Symmetry transformations used to generate equivalent atoms: 1: #1  $x-1, y, z$ ; #2  $-x+1, y, -z+3/2$ ; #3  $x, -y+1, z+1/2$ ; #4  $-x, -y+1, -z+1$ ; #5  $-x, y, -z+3/2$ ; 2, #1  $-x+1, -y+3, -z+1$ ; #2  $-x+1, -y+2, -z+1$ ; 3, #1  $-x, y+1/2, -z+1/2$ ; #2  $-x, -y, -z+1$ ; #3  $x, -y+1/2, z+1/2$ ; #4  $-x, y-1/2, -z+1/2$ ; #5  $-x, -y+1, -z+1$ ; #6  $x, y+1, z$ .