

## **Supplementary Information**

### **Structural studies and physicochemical properties of L-valine hydrochloride monohydrate**

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**Table S1. Crystal data and structure refinement parameters for (L.T.)and(R.T.) structures.**

Structure	L.T. (Present study)	R.T.
Empirical formula	C <sub>5</sub> H <sub>12</sub> NO <sub>2</sub> <sup>+</sup> Cl <sup>-</sup> .H <sub>2</sub> O	C <sub>5</sub> H <sub>12</sub> NO <sub>2</sub> <sup>+</sup> Cl <sup>-</sup> .H <sub>2</sub> O
Formula Weight	171.62	171.62
Temperature (K)	120(2)	293(2)
Wavelength (Å)	0.71073	0.71073
Crystal system	Orthorhombic	Orthorhombic
space group	P2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>	P2 <sub>1</sub> 2 <sub>1</sub> 2 <sub>1</sub>
a, b, c (Å)	6.115(2), 6.850(1), 20.781(2)	6.155(1), 6.828(1), 21.099(3)
α, β, γ (°)	90.0, 90.0, 90.0	90.0, 90.0, 90.0
Volume (Å <sup>3</sup> )	870.5(3)	886.7(2)
Z / Density (calc.) (Mg/m <sup>3</sup> )	4 / 1.310	4 / 1.286
Absorption coefficient (mm <sup>-1</sup> )	0.396	0.389
F(000)	368	368
Crystal size (mm <sup>3</sup> )	0.28 × 0.19 × 0.09	0.48 × 0.36 × 0.32
θ range for data collection (°)	1.96 to 24.97	1.93 to 27.48
Limiting indices	-7<=h<=6,-8<=k<=8, -24<=l<=24	-7<=h<=7,-8<=k<=8, -27<=l<=27
Reflections collected / unique	8018/1522[R(int)=0.0372]	2442/2020[R(int)=0.0229]
Completeness to θ (%)	100.0	100.0
Absorption correction	Semi-empirical from equivalents	Psi-scan
Max. and min. transmission	0.96 and 0.91	0.88 and 0.84
Refinement method	Full-matrix least-squares on F <sup>2</sup>	Full-matrix least-squares on F <sup>2</sup>
Data/restraints/parameters	1522 / 0 / 95	2020 / 0 / 148
Goodness-of - fit on F <sup>2</sup>	1.171	0.973
Extinction coefficient	0.066(4)	0.065(5)
Final R indices [I > 2σ(I)]	R1=0.0178, wR2=0.0465	R1=0.355, wR2=0.0809
R indices (all data)	R1=0.0178, wR2=0.0465	R1=0.0433, wR2=0.0832
Largest diff. peak and hole (e.Å <sup>-3</sup> )	0.166 and -0.141	0.295 and -0.159

$R_1 = \sum ||F_o| - |F_c|| / \sum |F_o|$ ,  $wR_2 = [\sum \{(F_o^2 - F_c^2)^2\} / \sum \{w(F_o^2)^2\}]^{1/2}$ ,  $w = 1 / \{\sigma^2(F_o^2) + (aP)^2 + bP\}$ , where  $a = 0.0157$  and  $b = 0.1750$  for (L.T.) and  $a = 0.0465$  and  $b = 0.0$  for (R.T.) structure.  $P = (F_o^2 + 2F_c^2) / 3$  for both the structures.

**Table S2. Hydrogen bonding geometry of C<sub>5</sub>H<sub>14</sub>N<sub>1</sub>O<sub>3</sub>Cl(L.T.) (Å, °).**

D–H···A	d(D–H)	d(H···A)	d(D···A)	D–H···A	Symmetry
O2–H2···Cl1	0.82	2.20	3.010(2)	172	x, 1+y, z
O1W–H1W···Cl1	0.81	2.37	3.178(2)	172	1/2+x, 1/2-y, -z
O1W–H2W···Cl1	0.86	2.31	3.156(2)	168	---
N1–H1C···O1	0.89	2.36	2.873(2)	117	1/2+x, 3/2-y, -z
N1–H1A···O1W	0.89	1.98	2.848(2)	165	---
N1–H1C···O1W	0.89	2.02	2.861(2)	157	1/2+x, 3/2-y, -z
N1–H1B···Cl1	0.89	2.42	3.237(2)	153	1+x, y, z
C1–H1F···Cl1	0.96	2.83	3.750(2)	162	1+x, y, z
C3–H3···O2	0.98	2.54	2.914(2)	103	---