

*Electronic Supplementary Information (ESI)*

**Step-by-step synthesis of one Fe<sub>6</sub> wheel and two Fe<sub>10</sub> clusters derived from multidentate triethanoamine ligand**

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## Experimental Section

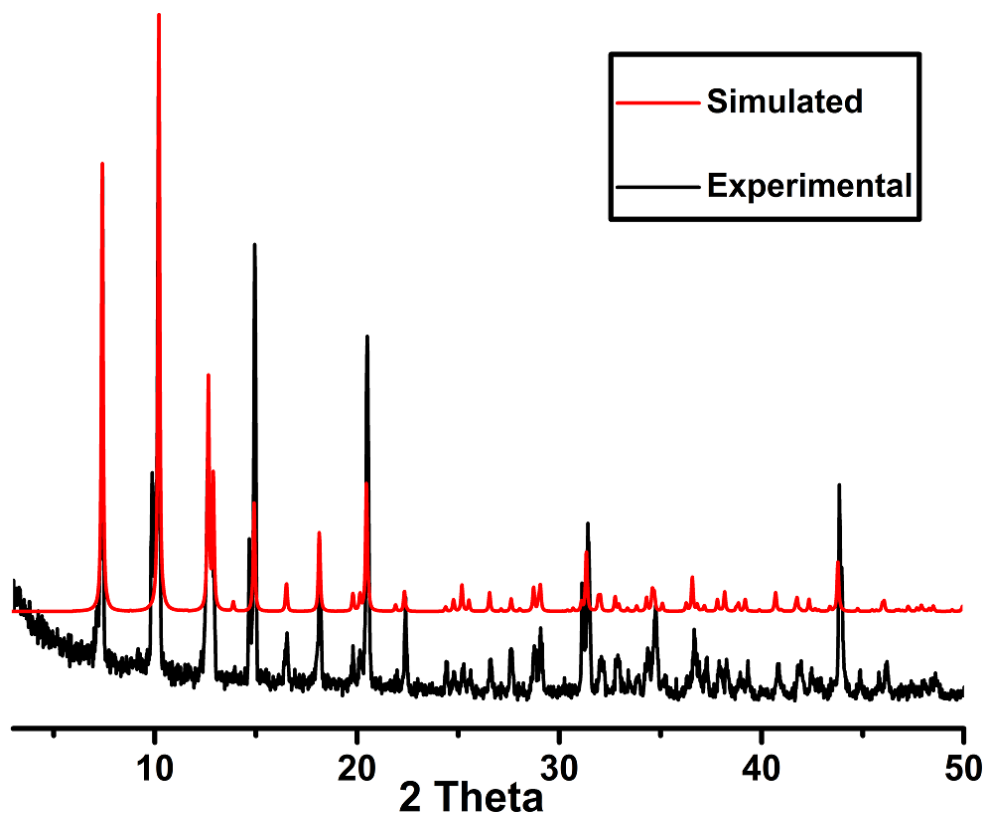
**Materials and methods.** All chemicals were of reagent grade and were used as purchased without further purification. Elemental analysis (C and H) was performed on a Perkin-Elmer 240C analyzer (Perkin-Elmer, USA).  $[\text{Fe}_3\text{O}(\text{HCO}_2)_6(\text{H}_2\text{O})_3]\cdot\text{NO}_3$  and  $[\text{Fe}_3\text{O}(\text{HCOO})_6(\text{H}_2\text{O})_3]\cdot\text{Cl}$  were prepared using the procedures reported previously with small modification.<sup>S1</sup> The XRPD spectrum was recorded on a Rigaku D/Max-2500 diffractometer at 40 kV, 100 mA for a Cu-target tube and a graphite monochromator. Simulation of the XRPD spectrum was carried out by the single-crystal data and diffraction-crystal module of the Mercury (Hg) program available free of charge *via* the Internet at <http://www.iucr.org>. IR spectra were measured in the range of 400-4000  $\text{cm}^{-1}$  on a Tensor 27 OPUS FT-IR spectrometer using KBr pellets (Bruker, German). Magnetic susceptibility was measured by a Quantum Design MPMS superconducting quantum interference device (SQUID). Diamagnetic corrections were estimated by using Pascal constants and background corrections by experimental measurement on sample holders.

### X-ray Crystallography.

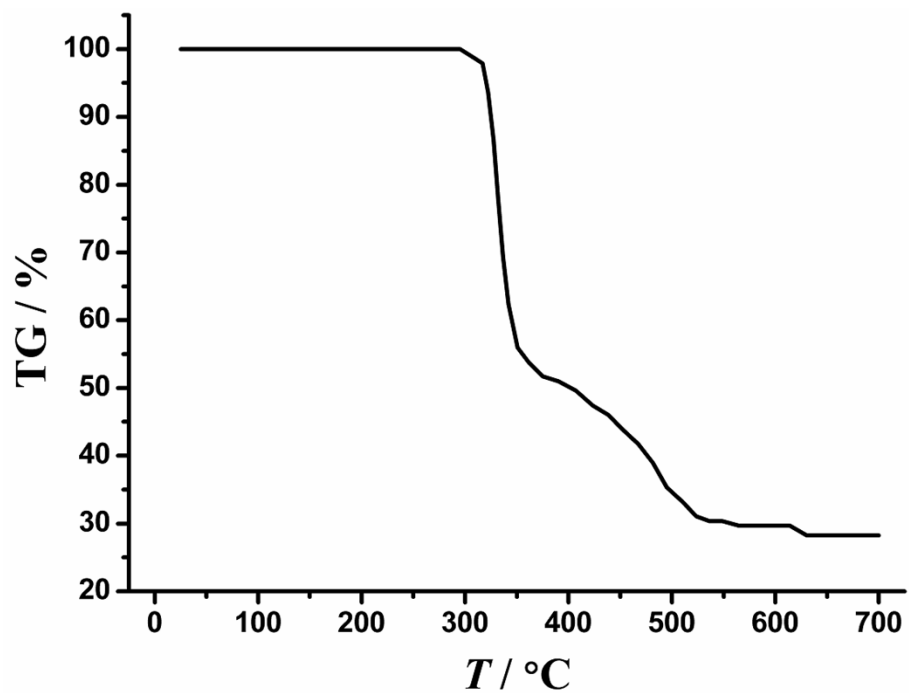
X-ray single-crystal diffraction data were collected on a SCX-Mini diffractometer at 293(2) K (for **1**) and a Rigaku RAXIS-RAPID diffractometer at 113(2) K (for **2** and **3**) with Mo-K $\alpha$  radiation ( $\lambda = 0.71073 \text{ \AA}$ ) by  $\omega$  scan mode. The program *Crystalclear*<sup>S2</sup> was used for the integration of the diffraction profiles. The structures were solved by direct method using the SHELXS program of the SHELXTL package and refined by full-matrix least-squares methods with SHELXL (semi-empirical absorption corrections were applied by using the SADABS program).<sup>S3</sup> The non-hydrogen atoms were located in successive difference Fourier syntheses and refined with anisotropic thermal parameters on  $F^2$ . All hydrogen atoms were generated theoretically at the specific atoms and refined isotropically with fixed thermal factors. The selected bond lengths and angles are given in Tables S1-S3 (ESI).

### References

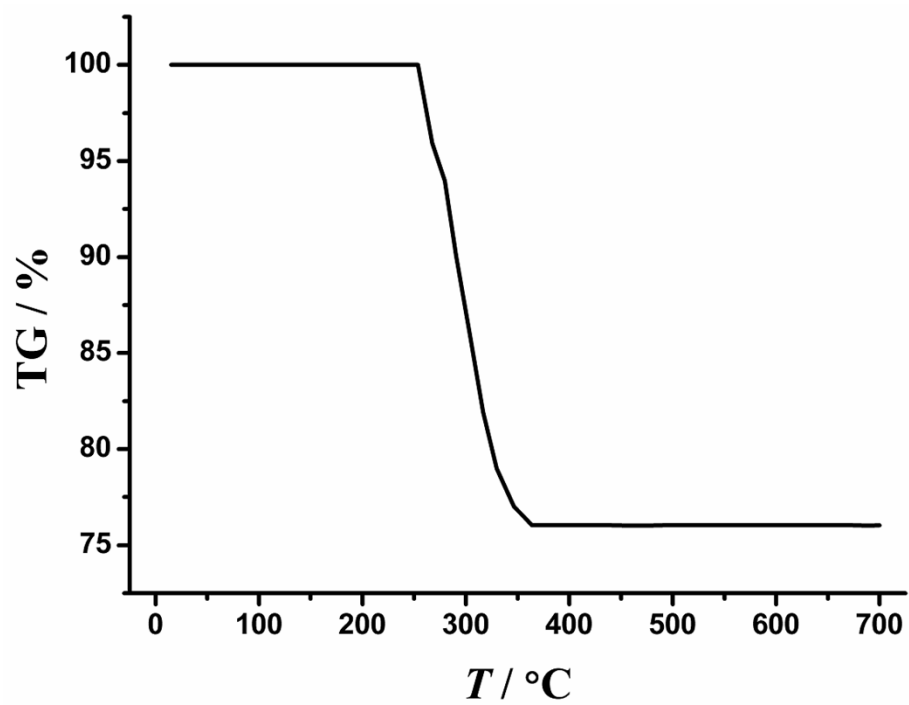
- S1. (a) M. K. Johnson, D. B. Powell and R. D. Cannon, *Spectrochim. Acta.*, 1981, **37**, 995; (b) A. M. Bond, R. J. H. Clark, D. G. Humphrey, P. Panayiotopoulos, B. W. Skelton and A. H. White, *J. Chem. Soc., Dalton Trans.*, 1998, 1845.
- S2. Rigaku, Process-Auto; Rigaku Americas Corporation: The Woodlands, Texas, 1998.
- S3. G. M. Sheldrick, *SHELXL97, Program for Crystal Structure Refinement*; University of Göttingen: Göttingen, Germany, 1997.



**Fig. S1** X-ray powder diffraction patterns of **1**.

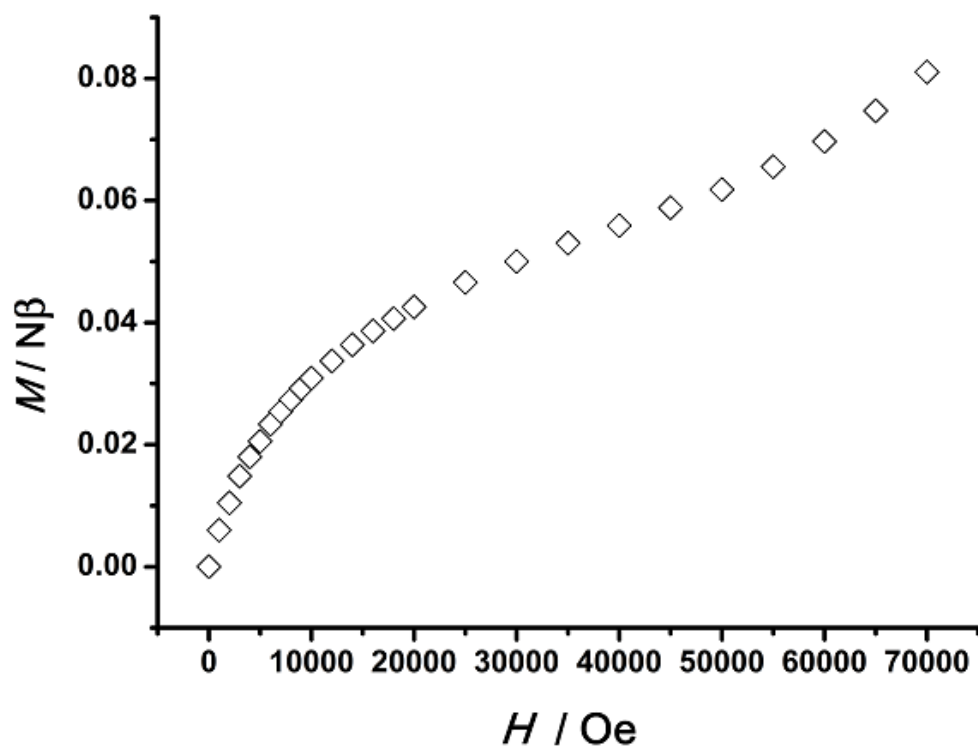


(a)



(b)

Fig. S2 TGA curves for 2 and 3.



**Fig. S3** The  $M$  vs.  $H$  curve of **1**.

**Table S1.** Selected bond lengths (Å) and angles (°) for complex **1**<sup>a</sup>

Fe1—O1	1.923(4)	Fe1—O3 <sup>#1</sup>	2.033(3)
Fe1—O3	1.990(3)	Fe1—O2 <sup>#2</sup>	2.068(3)
Fe1—O2	1.993(3)	Fe1—N1	2.211(4)
O1—Fe1—O3	93.27(15)	O2—Fe1—O2 <sup>#2</sup>	100.9(2)
O1—Fe1—O2	106.56(15)	O3 <sup>#1</sup> —Fe1—O2 <sup>#2</sup>	92.88(14)
O3—Fe1—O2	96.43(14)	O1—Fe1—N1	80.17(16)
O1—Fe1—O3 <sup>#1</sup>	102.91(16)	O3—Fe1—N1	109.25(15)
O3—Fe1—O3 <sup>#1</sup>	163.26(19)	O2—Fe1—N1	153.20(14)
O2—Fe1—O3 <sup>#1</sup>	75.05(13)	O3 <sup>#1</sup> —Fe1—N1	78.16(14)
O1—Fe1—O2 <sup>#2</sup>	151.05(15)	O2 <sup>#2</sup> —Fe1—N1	79.54(15)
O3—Fe1—O2 <sup>#2</sup>	74.34(13)		

<sup>a</sup>Symmetry codes: #1: x-y, x, -z+1; #2: y, -x+y, -z+1.

**Table S2.** Selected bond lengths (Å) and angles (°) for complex **2**

Fe1—O4	1.884(11)	Fe8—O18	2.041(11)
Fe1—O6	1.920(10)	Fe8—O20	2.040(10)
Fe1—O7	1.929(10)	Fe8—O25	2.057(11)
Fe1—O5	1.966(11)	Fe9—O3	1.922(10)
Fe1—N1	2.269(12)	Fe9—O16	2.005(11)
Fe2—O1	1.984(11)	Fe9—O17	2.020(10)
Fe2—O8	2.014(11)	Fe9—O18	2.029(10)
Fe2—O9	2.029(10)	Fe9—O2	2.146(10)
Fe2—O7	2.037(10)	Fe9—N5	2.264(14)
Fe2—O2	2.180(11)	Fe10—O21	1.908(11)
Fe2—N2	2.256(13)	Fe10—O17	1.931(11)
Fe3—O1	1.905(10)	Fe10—O20	1.981(12)
Fe3—O26	1.957(11)	Fe10—O19	1.989(10)
Fe3—O13	2.000(10)	Fe10—N6	2.254(13)
Fe3—O6	2.049(10)	Fe7—O3	1.938(10)
Fe3—O8	2.091(11)	Fe7—O30	1.996(11)
Fe3—O22	2.120(10)	Fe7—O11	2.012(10)
Fe4—O1	1.895(11)	Fe7—O15	2.020(10)
Fe4—O28	1.951(12)	Fe7—O24	2.073(10)
Fe4—O14	2.026(9)	Fe7—O19	2.099(10)
Fe4—O10	2.075(10)	Fe8—O3	1.945(10)
Fe4—O23	2.098(10)	Fe8—O32	1.996(12)
Fe4—O5	2.131(10)	Fe8—O12	2.024(10)
Fe5—O11	1.984(10)	Fe5—N3	2.177(11)
Fe5—O2	1.989(10)	Fe6—O13	1.983(11)
Fe5—O12	2.000(10)	Fe6—O14	1.990(10)
Fe5—O9	2.022(10)	Fe6—O2	2.023(11)
Fe5—O10	2.055(10)	Fe6—O16	2.060(10)

Fe6—N4	2.231(12)	Fe6—O15	2.068(10)
O4—Fe1—O6	108.9(5)	O32—Fe8—O12	96.4(5)
O4—Fe1—O7	92.6(5)	O3—Fe8—O18	76.8(4)
O6—Fe1—O7	104.9(5)	O32—Fe8—O18	100.5(5)
O4—Fe1—O5	124.1(5)	O12—Fe8—O18	92.2(4)
O6—Fe1—O5	118.7(5)	O3—Fe8—O20	83.9(4)
O7—Fe1—O5	101.3(5)	O32—Fe8—O20	90.6(5)
O4—Fe1—N1	81.9(5)	O12—Fe8—O20	170.2(5)
O6—Fe1—N1	81.3(5)	O18—Fe8—O20	93.3(5)
O7—Fe1—N1	172.9(4)	O3—Fe8—O25	94.6(4)
O5—Fe1—N1	78.3(5)	O32—Fe8—O25	88.3(5)
O1—Fe2—O8	77.9(4)	O12—Fe8—O25	85.8(4)
O1—Fe2—O9	133.3(4)	O18—Fe8—O25	171.2(4)
O8—Fe2—O9	148.0(4)	O20—Fe8—O25	87.6(5)
O1—Fe2—O7	80.4(4)	O3—Fe9—O16	134.0(5)
O8—Fe2—O7	102.6(4)	O3—Fe9—O17	79.9(4)
O9—Fe2—O7	91.9(4)	O16—Fe9—O17	90.6(4)
O1—Fe2—O2	90.2(4)	O3—Fe9—O18	77.6(4)
O8—Fe2—O2	97.6(4)	O16—Fe9—O18	147.8(4)
O9—Fe2—O2	77.9(4)	O17—Fe9—O18	103.7(4)
O7—Fe2—O2	155.3(4)	O3—Fe9—O2	91.6(4)
O1—Fe2—N2	144.9(5)	O16—Fe9—O2	78.7(4)
O8—Fe2—N2	79.4(5)	O17—Fe9—O2	155.3(4)
O9—Fe2—N2	75.6(5)	O18—Fe9—O2	96.9(4)
O7—Fe2—N2	78.8(4)	O3—Fe9—N5	141.6(5)
O2—Fe2—N2	119.2(4)	O16—Fe9—N5	77.1(5)
O1—Fe3—O26	175.1(5)	O17—Fe9—N5	77.8(4)
O1—Fe3—O13	88.9(4)	O18—Fe9—N5	77.9(5)
O26—Fe3—O13	95.4(4)	O2—Fe9—N5	120.3(4)
O1—Fe3—O6	84.8(4)	O21—Fe10—O17	90.9(5)
O26—Fe3—O6	91.3(4)	O21—Fe10—O20	107.9(6)
O13—Fe3—O6	170.7(5)	O17—Fe10—O20	105.2(5)
O1—Fe3—O8	77.8(4)	O21—Fe10—O19	121.7(5)
O26—Fe3—O8	99.5(5)	O17—Fe10—O19	102.9(5)
O13—Fe3—O8	92.3(4)	O20—Fe10—O19	121.7(5)
O6—Fe3—O8	93.1(4)	O21—Fe10—N6	80.2(4)
O1—Fe3—O22	94.0(4)	O17—Fe10—N6	170.8(4)
O26—Fe3—O22	88.7(5)	O20—Fe10—N6	79.8(5)
O13—Fe3—O22	87.5(4)	O19—Fe10—N6	80.3(5)
O6—Fe3—O22	86.1(5)	O13—Fe6—O14	96.6(4)
O8—Fe3—O22	171.8(4)	O13—Fe6—O2	115.4(4)
O1—Fe4—O28	168.6(5)	O14—Fe6—O2	97.2(4)
O1—Fe4—O14	88.3(4)	O13—Fe6—O16	86.7(4)
O28—Fe4—O14	98.2(5)	O14—Fe6—O16	176.5(4)

O1—Fe4—O10	82.0(4)	O2—Fe6—O16	80.3(4)
O28—Fe4—O10	88.5(5)	O13—Fe6—O15	148.3(4)
O14—Fe4—O10	90.8(4)	O14—Fe6—O15	100.7(4)
O1—Fe4—O23	98.9(4)	O2—Fe6—O15	88.7(4)
O28—Fe4—O23	90.8(5)	O16—Fe6—O15	76.9(4)
O14—Fe4—O23	86.7(4)	O13—Fe6—N4	78.6(4)
O10—Fe4—O23	177.4(4)	O14—Fe6—N4	81.2(4)
O1—Fe4—O5	84.1(4)	O2—Fe6—N4	166.0(5)
O28—Fe4—O5	90.4(5)	O16—Fe6—N4	100.5(5)
O14—Fe4—O5	169.8(4)	O15—Fe6—N4	78.0(4)
O10—Fe4—O5	94.9(4)	O3—Fe7—O30	173.1(5)
O23—Fe4—O5	87.6(4)	O3—Fe7—O11	90.5(4)
O11—Fe5—O2	94.4(4)	O30—Fe7—O11	93.0(4)
O11—Fe5—O12	97.0(5)	O3—Fe7—O15	81.9(4)
O2—Fe5—O12	113.7(4)	O30—Fe7—O15	92.1(5)
O11—Fe5—O9	176.6(5)	O11—Fe7—O15	91.2(4)
O2—Fe5—O9	82.6(4)	O3—Fe7—O24	95.5(4)
O12—Fe5—O9	85.7(4)	O30—Fe7—O24	90.6(5)
O11—Fe5—O10	98.7(4)	O11—Fe7—O24	87.5(4)
O2—Fe5—O10	91.8(4)	O15—Fe7—O24	177.1(4)
O12—Fe5—O10	148.7(4)	O3—Fe7—O19	82.8(4)
O9—Fe5—O10	79.7(4)	O30—Fe7—O19	94.6(4)
O11—Fe5—N3	82.6(4)	O11—Fe7—O19	169.4(4)
O2—Fe5—N3	167.9(5)	O15—Fe7—O19	95.9(4)
O12—Fe5—N3	78.4(4)	O24—Fe7—O19	85.1(4)
O9—Fe5—N3	99.9(4)	O3—Fe8—O32	173.7(5)
O10—Fe5—N3	77.1(4)	O3—Fe8—O12	89.4(4)



**Table S3.** Selected bond lengths (Å) and angles (°) for complex **3**

O18—Fe2	2.061(6)	Fe8—O25	1.940(7)
Fe1—O4	1.989(6)	Fe8—O24	1.956(6)
Fe1—O2	1.991(6)	Fe8—N5	2.257(7)
Fe1—O1	2.000(6)	Fe9—O5	1.982(6)
Fe1—O3	2.007(6)	Fe9—O11	1.986(6)
Fe1—O5	2.143(6)	Fe9—O12	1.993(6)
Fe1—N1	2.277(8)	Fe9—O22	2.048(6)
Fe2—O19	1.914(6)	Fe9—O13	2.066(6)
Fe2—O15	2.003(6)	Fe9—N6	2.201(7)
Fe2—O24	2.047(6)	Fe10—O19	1.913(6)
Fe2—O20	2.088(6)	Fe10—O16	1.993(6)
Fe2—Cl1	2.326(3)	Fe10—O13	2.057(6)
Fe3—O19	1.969(6)	Fe10—O25	2.059(6)
Fe3—O20	1.992(6)	Fe10—O17	2.091(6)
Fe3—O21	1.996(6)	Fe10—Cl4	2.342(3)
Fe3—O22	1.999(6)	Fe5—Cl2	2.324(3)
Fe3—O5	2.164(6)	Fe6—O4	1.887(6)
Fe3—N2	2.293(8)	Fe6—O12	1.999(6)
Fe3—Fe9	2.9270(18)	Fe6—O14	2.056(6)
Fe4—O5	1.992(6)	Fe6—O10	2.072(6)
Fe4—O16	1.992(6)	Fe6—O7	2.073(6)
Fe4—O15	2.012(6)	Fe6—Cl3	2.359(3)
Fe4—O1	2.026(6)	Fe7—O6	1.884(6)
Fe4—O14	2.064(6)	Fe7—O2	1.945(6)
Fe4—N3	2.203(7)	Fe7—O7	1.949(7)
Fe5—O4	1.937(6)	Fe7—O8	1.961(6)
Fe5—O11	2.002(6)	Fe7—N4	2.267(8)
Fe5—O8	2.029(6)	Fe8—O23	1.879(6)
Fe5—O3	2.073(6)	Fe8—O21	1.935(6)
Fe5—O9	2.081(6)		
O18—Fe2—O20	171.8(3)	O16—Fe10—Cl4	95.96(19)
O4—Fe1—O2	80.8(2)	O13—Fe10—Cl4	91.53(18)
O4—Fe1—O1	133.8(3)	O25—Fe10—Cl4	90.49(19)
O2—Fe1—O1	90.7(2)	O17—Fe10—Cl4	91.76(19)
O4—Fe1—O3	78.3(2)	O3—Fe1—N1	77.8(3)
O2—Fe1—O3	104.5(2)	O5—Fe1—N1	120.4(2)
O1—Fe1—O3	146.9(3)	O19—Fe2—O15	86.8(3)
O4—Fe1—O5	90.2(2)	O19—Fe2—O24	84.3(3)
O2—Fe1—O5	153.9(3)	O15—Fe2—O24	168.5(3)
O1—Fe1—O5	77.9(2)	O19—Fe2—O18	94.7(3)
O3—Fe1—O5	97.6(2)	O15—Fe2—O18	85.5(3)
O4—Fe1—N1	143.1(3)	O24—Fe2—O18	87.9(3)
O2—Fe1—N1	78.5(3)	O19—Fe2—O20	77.1(2)

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O1—Fe1—N1	76.7(3)	O15—Fe2—O20	94.5(2)
O24—Fe2—O20	90.7(2)	O4—Fe5—O11	87.8(2)
O21—Fe3—O5	154.0(3)	O19—Fe3—O20	78.2(3)
O22—Fe3—O5	78.1(2)	O19—Fe3—O21	81.6(2)
O19—Fe3—N2	142.9(3)	O20—Fe3—O21	103.8(3)
O20—Fe3—N2	77.2(3)	O19—Fe3—O22	135.3(3)
O21—Fe3—N2	77.8(3)	O20—Fe3—O22	145.6(3)
O22—Fe3—N2	76.1(3)	O21—Fe3—O22	91.3(2)
O5—Fe3—N2	121.4(3)	O19—Fe3—O5	89.3(2)
O5—Fe4—O16	96.8(2)	O20—Fe3—O5	98.0(2)
O5—Fe4—O15	115.0(3)	O11—Fe5—O8	170.2(3)
O16—Fe4—O15	97.0(3)	O4—Fe5—O3	77.9(2)
O5—Fe4—O1	80.9(2)	O11—Fe5—O3	93.5(2)
O16—Fe4—O1	176.7(2)	O8—Fe5—O3	90.8(2)
O15—Fe4—O1	86.1(2)	O4—Fe5—O9	93.9(3)
O5—Fe4—O14	89.4(2)	O11—Fe5—O9	86.1(2)
O16—Fe4—O14	98.2(2)	O8—Fe5—O9	88.3(2)
O15—Fe4—O14	149.3(2)	O3—Fe5—O9	171.8(2)
O1—Fe4—O14	79.5(2)	O4—Fe5—Cl2	172.1(2)
O5—Fe4—N3	166.6(3)	O11—Fe5—Cl2	94.68(19)
O16—Fe4—N3	82.0(3)	O8—Fe5—Cl2	93.73(19)
O15—Fe4—N3	78.3(3)	O3—Fe5—Cl2	94.40(18)
O1—Fe4—N3	99.7(3)	O9—Fe5—Cl2	93.80(19)
O14—Fe4—N3	77.6(3)	O4—Fe5—O8	84.6(2)
O4—Fe6—O12	89.5(3)	O2—Fe7—O7	101.7(3)
O4—Fe6—O14	82.6(3)	O6—Fe7—O8	109.4(3)
O12—Fe6—O14	89.6(2)	O2—Fe7—O8	104.6(3)
O4—Fe6—O10	95.6(3)	O7—Fe7—O8	122.3(3)
O12—Fe6—O10	86.7(2)	O6—Fe7—N4	80.1(3)
O14—Fe6—O10	175.9(2)	O2—Fe7—N4	170.8(3)
O4—Fe6—O7	83.0(3)	O7—Fe7—N4	81.0(3)
O12—Fe6—O7	170.1(2)	O8—Fe7—N4	80.8(3)
O14—Fe6—O7	95.9(2)	O23—Fe8—O21	91.9(3)
O10—Fe6—O7	87.5(2)	O23—Fe8—O25	118.5(3)
O4—Fe6—Cl3	172.29(19)	O21—Fe8—O25	103.6(3)
O12—Fe6—Cl3	96.36(19)	O7—Fe6—Cl3	91.63(19)
O14—Fe6—Cl3	92.43(19)	O6—Fe7—O2	91.0(3)
O10—Fe6—Cl3	89.71(19)	O6—Fe7—O7	120.5(3)
O23—Fe8—O24	113.1(3)	O19—Fe10—O16	90.6(3)
O21—Fe8—O24	102.1(3)	O19—Fe10—O13	82.1(2)
O25—Fe8—O24	120.6(3)	O16—Fe10—O13	90.7(2)
O23—Fe8—N5	80.4(3)	O19—Fe10—O25	83.8(3)
O21—Fe8—N5	172.2(3)	O16—Fe10—O25	170.9(3)
O25—Fe8—N5	80.9(3)	O13—Fe10—O25	95.5(2)
O24—Fe8—N5	80.5(3)	O19—Fe10—O17	94.9(3)

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O5—Fe9—O11	115.6(3)	O16—Fe10—O17	86.5(2)
O5—Fe9—O12	96.7(2)	O13—Fe10—O17	175.9(2)
O11—Fe9—O12	96.1(2)	O25—Fe10—O17	86.9(2)
O5—Fe9—O22	81.3(2)	O19—Fe10—Cl4	170.9(2)
O11—Fe9—O22	85.6(2)	O22—Fe9—O13	79.4(2)
O12—Fe9—O22	177.8(2)	O5—Fe9—N6	166.6(3)
O5—Fe9—O13	89.6(2)	O11—Fe9—N6	77.8(3)
O11—Fe9—O13	148.3(2)	O12—Fe9—N6	81.4(3)
O12—Fe9—O13	99.7(2)	O22—Fe9—N6	100.4(3)

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