Supporting Information

Self-assembly and thermotropic liquid crystal property of a hexavacant germanomolybdate: [Ge₂Mo₁₆O₅₈]¹²⁻

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Fig. S1. The formation of $\{\beta$ -*A*-GeMo₉O₃₄ $\}$.



Fig. S2. The 3D framework of compound 2 along different directions.



Fig. S3. The IR spectra of compound **1**, compound **2**, DODACI and **SEC-Mo**₁₆. The IR spectra of compound **1** shows the characteristic bands at 900-676 cm⁻¹, corresponding to terminal Mo-Ot vibrations, Ge-O vibrations and Mo-O bridges vibrations, respectively. In addition, the vibration bands at 1611-1405 cm⁻¹ can be assigned to the bending vibration bands of -NH₂, -CH₃ and -CH₂. Moreover, the stretching bands of -OH and -NH₂ groups are observed at 3462 and 3131 cm⁻¹, respectively. The IR spectra of compound **2** shows the characteristic bands at 883-654 cm⁻¹, corresponding to terminal Mo-Ot vibrations, Ge-O vibrations and Mo-O bridges vibrations, respectively. In addition, the vibrations and Sec-No₁₆ and Sec-Mo₁₆, we can clearly observe that the difference between **SEC-Mo₁₆** and DODACI is the peaks in the low-wave number region, corresponding to the characteristic peaks of compound **1** and compound **2**, indicating the existence of DODA⁺ and polyoxoanion

 $[Ge_2Mo_{16}O_{58}]^{12-}$ in SEC-Mo₁₆.



Fig. S4. The ¹H NMR spectra of DODACl and SEC-Mo₁₆.

The ¹H NMR spectra of DODACl exhibts five types of hydrogens of different chemical environments, which can also been observed in the ¹H NMR spectra of **SEC-Mo₁₆**, indicating the existence of DODA⁺ in **SEC-Mo₁₆**.



Fig. S5. The TG curve of compound **1**. The TG curve of compound **1** shows two steps of weight loss, the total loss is 11.4 % (calc. 14.7 %).



Fig. S6. The TG curve of SEC-Mo₁₆.



Fig. S7. The PXRD patterns of compound 1.