

Supplementary Information

Syntheses, structures and luminescent properties of six divalent metal terephthalate coordination polymers based on three new flexible bis(imidazole) ligands

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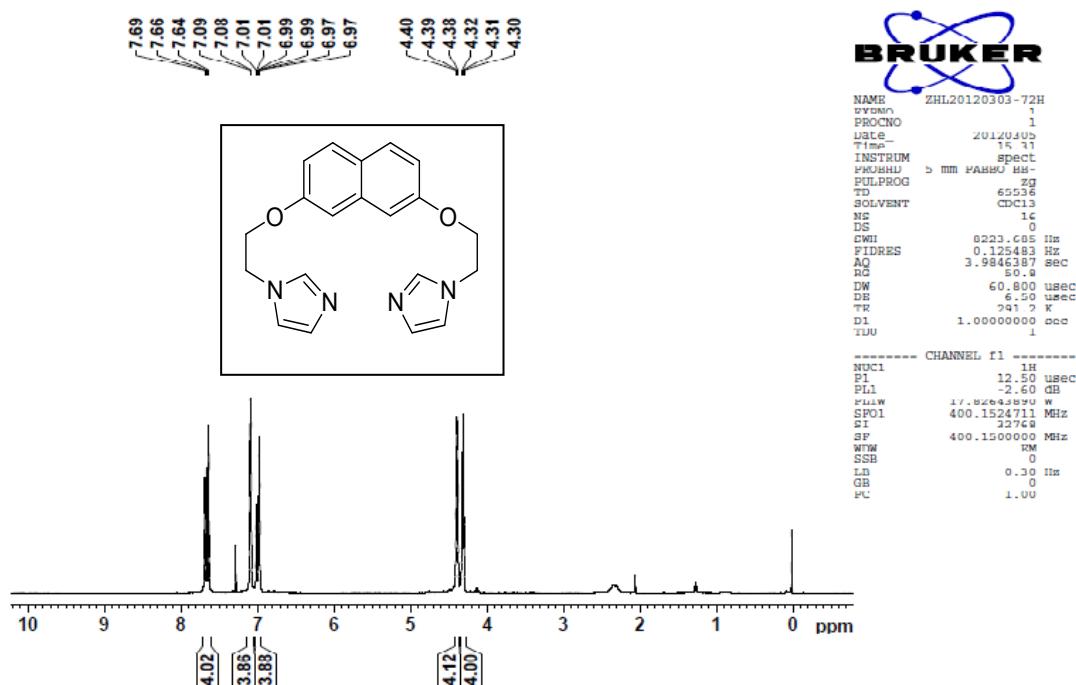


Fig. S1. The ¹H NMR spectra of L1

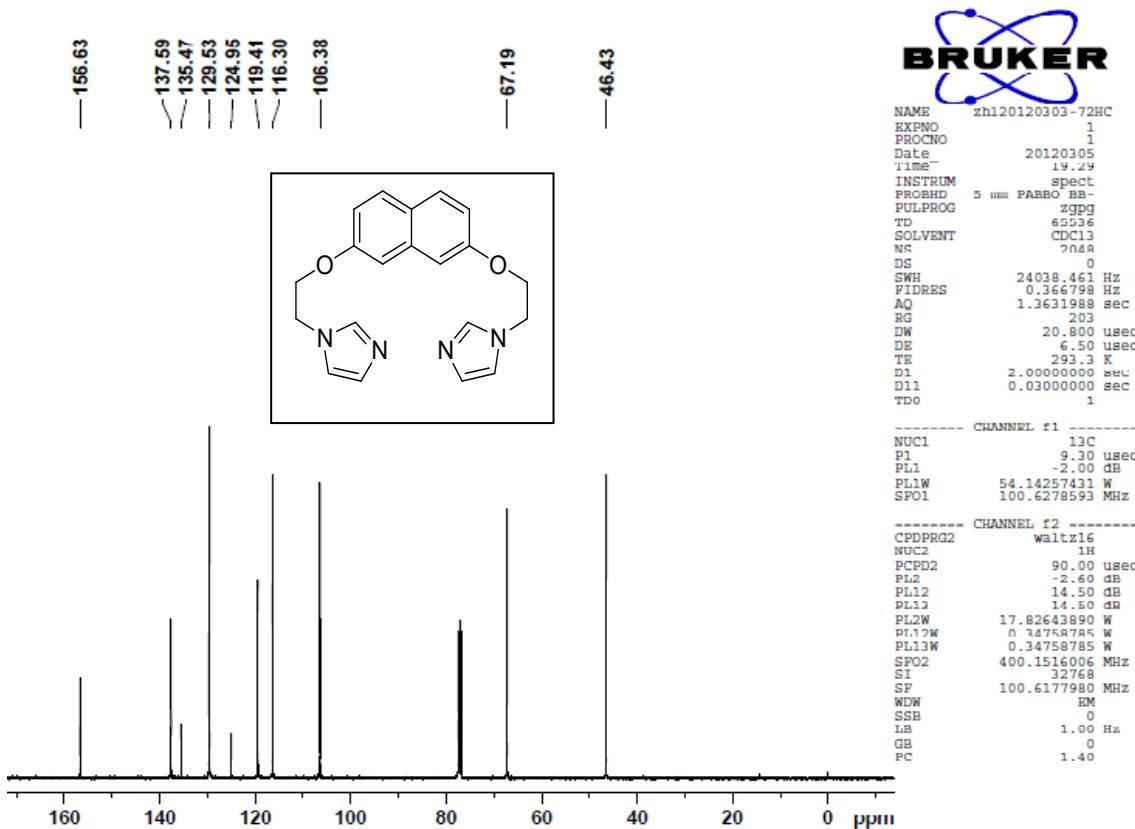


Fig. S2. The ^{13}C NMR spectra of L1

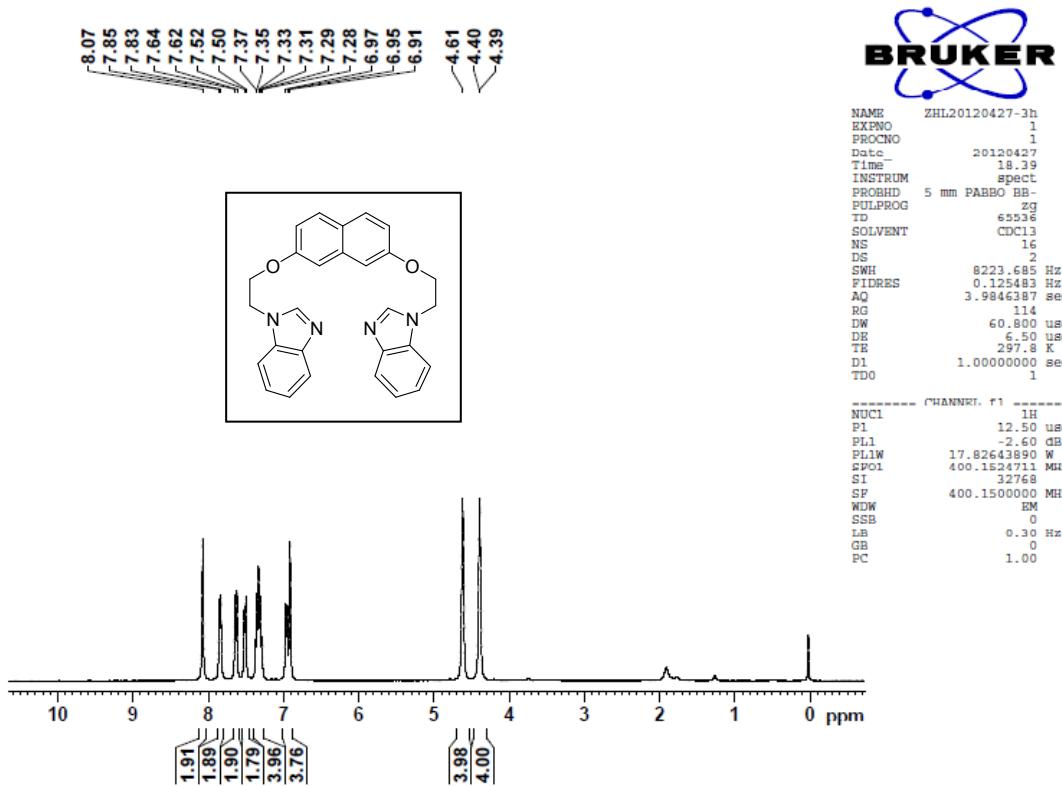


Fig. S3. The ^1H NMR spectra of L2

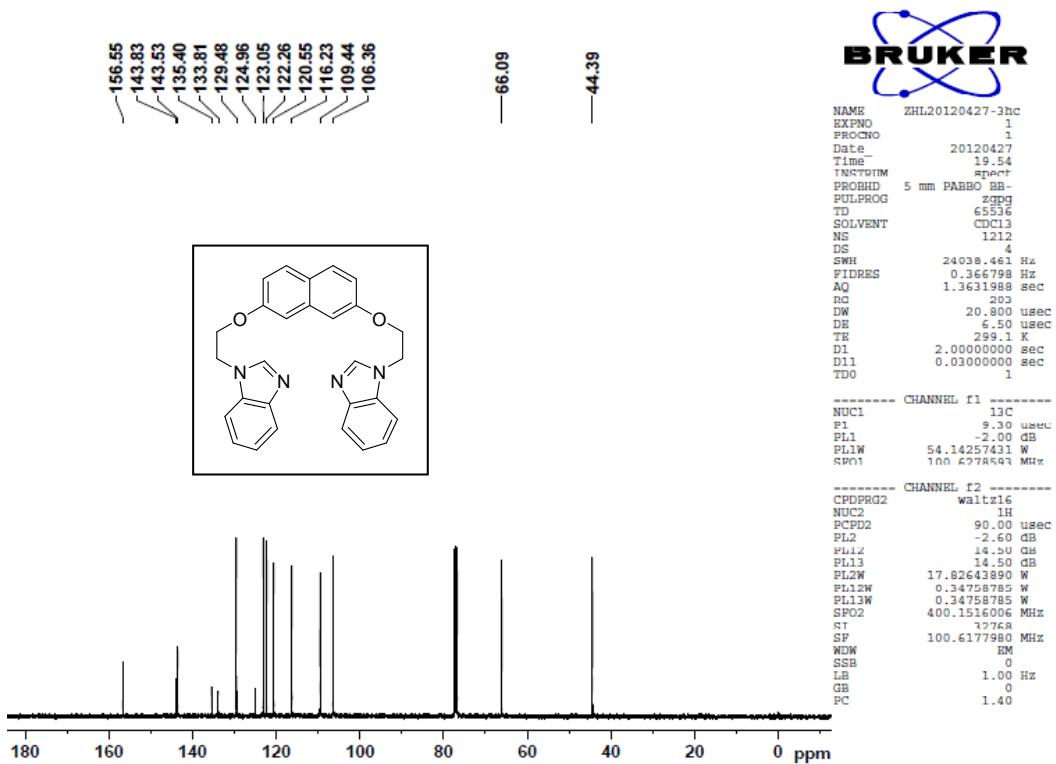


Fig. S4. The ^{13}C NMR spectra of **L2**

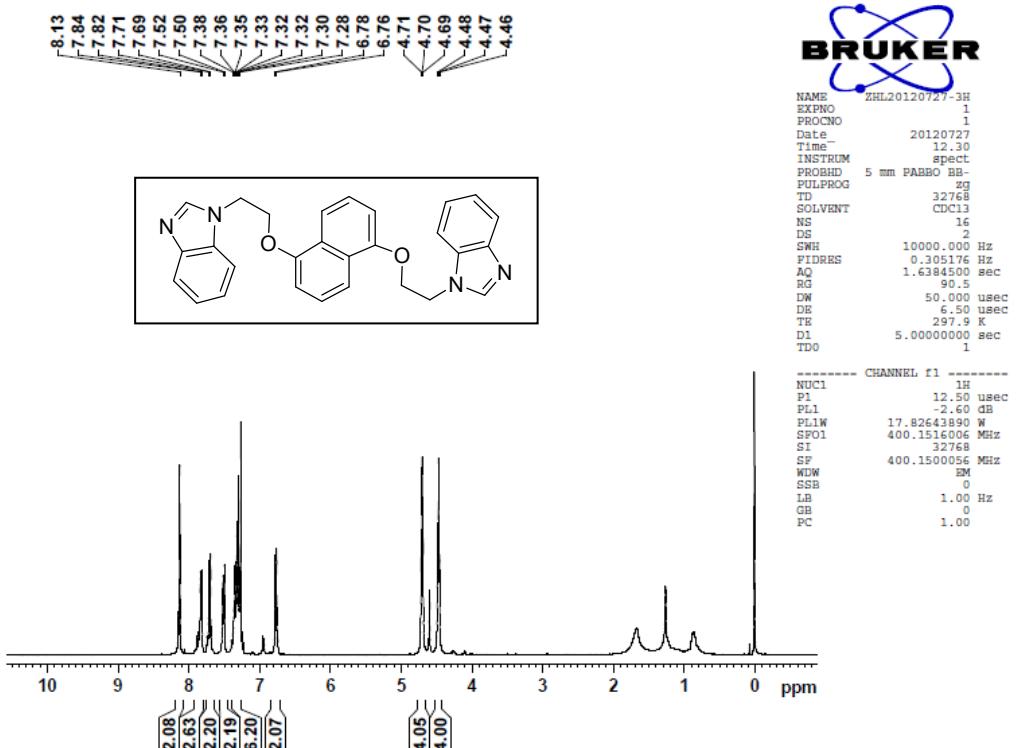


Fig. S5. The ^1H NMR spectra of **L3**

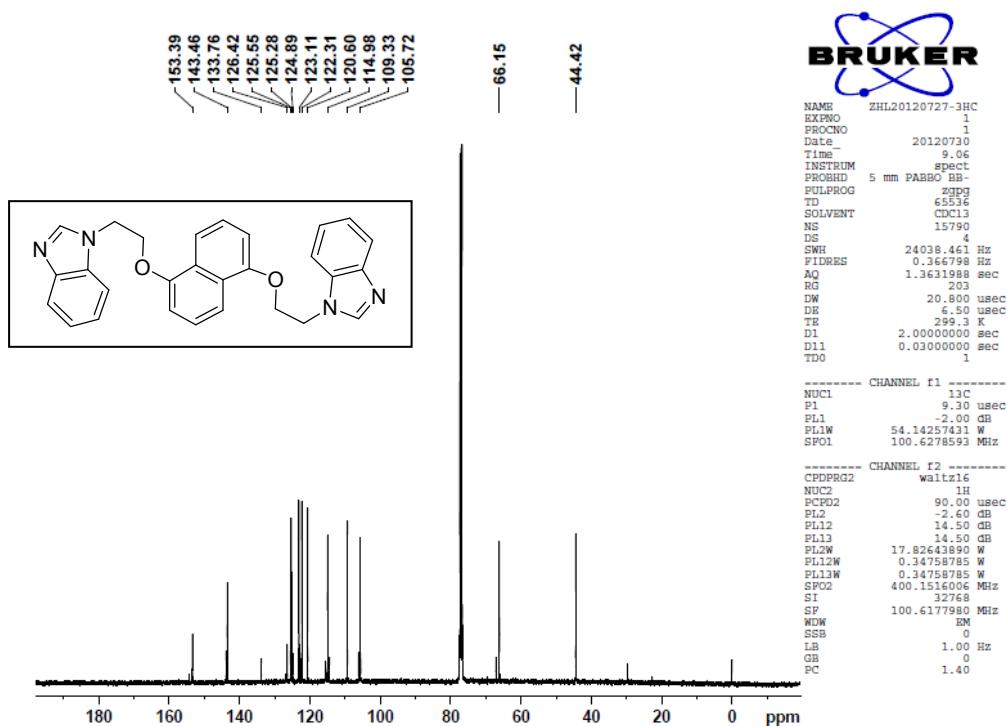


Fig. S6. The ^{13}C NMR spectra of **L3**

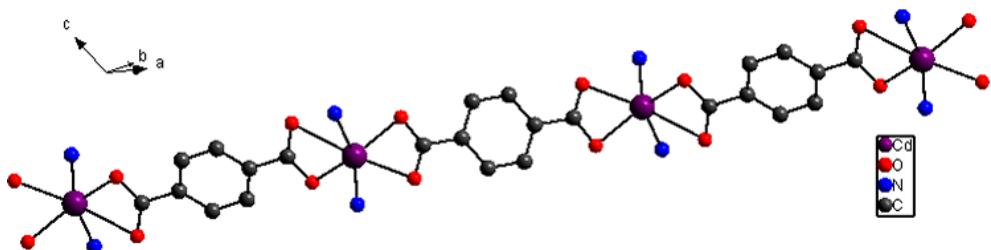


Fig. S7. View of the linear chain constructed by $p\text{-bdc}^{2-}$ -ligands and Cd ions in compound **1**

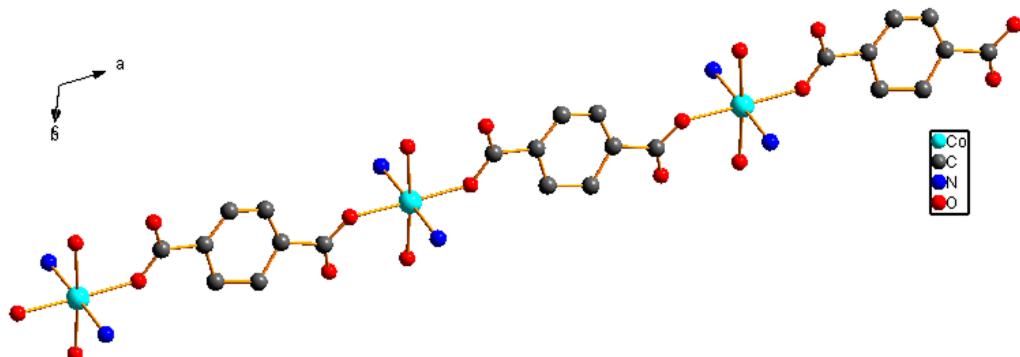


Fig. S8. View of the linear chain constructed by $p\text{-bdc}^{2-}$ -ligands and Co ions in compound **2**

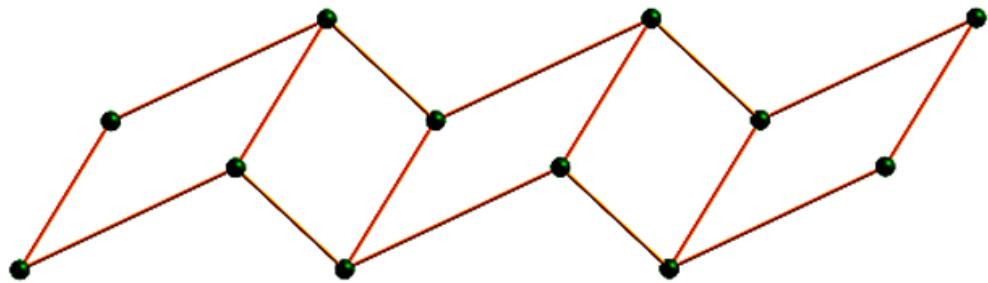


Fig. S9. Topological representation of 1D network of compound 3

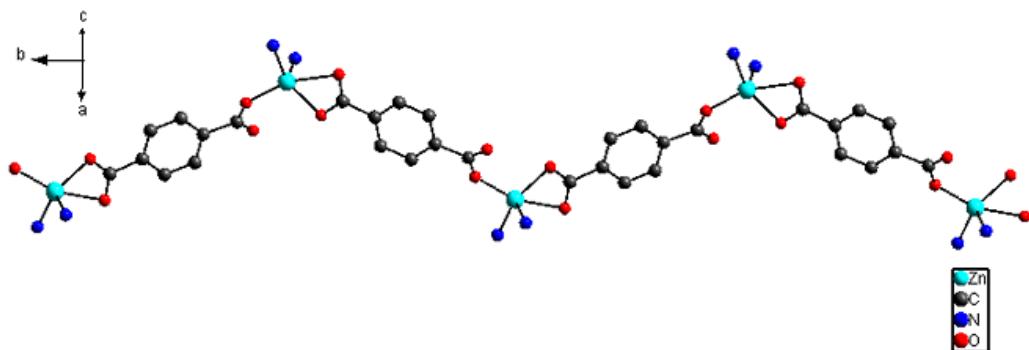


Fig. S10. Schematic view of a wavelike chain constructed by *p*-bdc²⁻ ligands and Zn cations in compound 4

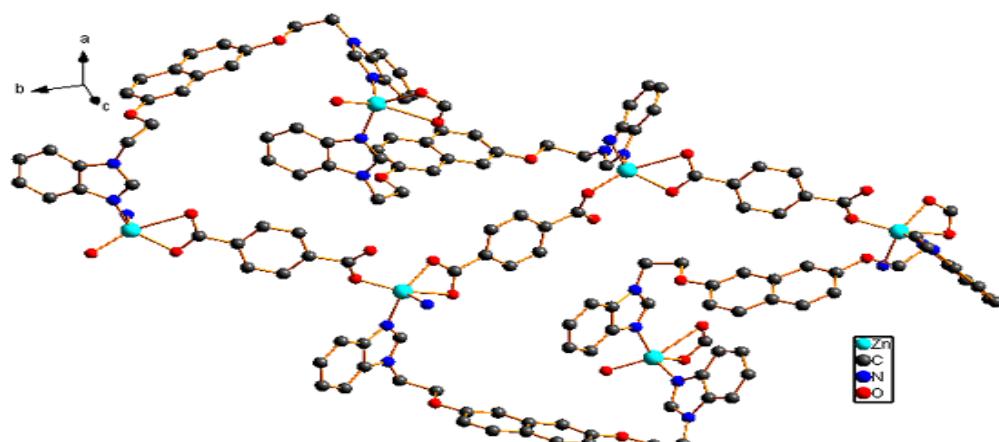


Fig. S11. Schematic representation of the basic building unit of 2D framework in compound 4

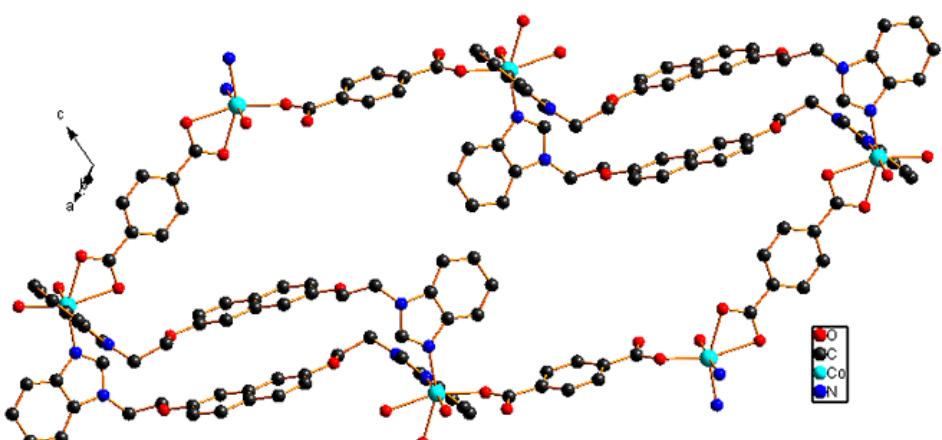


Fig. S12. Schematic representation of the basic building unit of 2D framework in compound 5

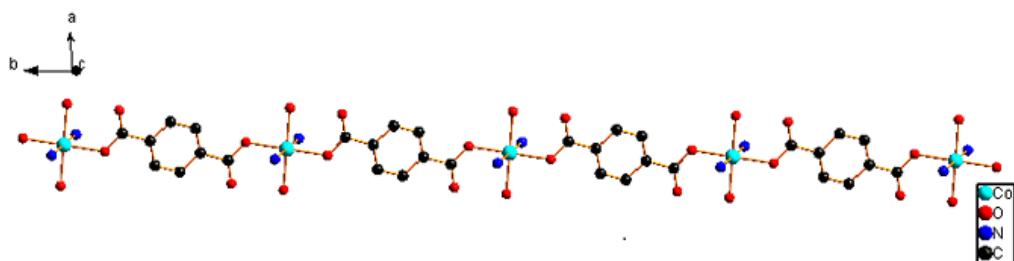


Fig. S13. Schematic view of a linear chain constructed by *p*-bdc²⁻ ligands and Co cations in compound 6

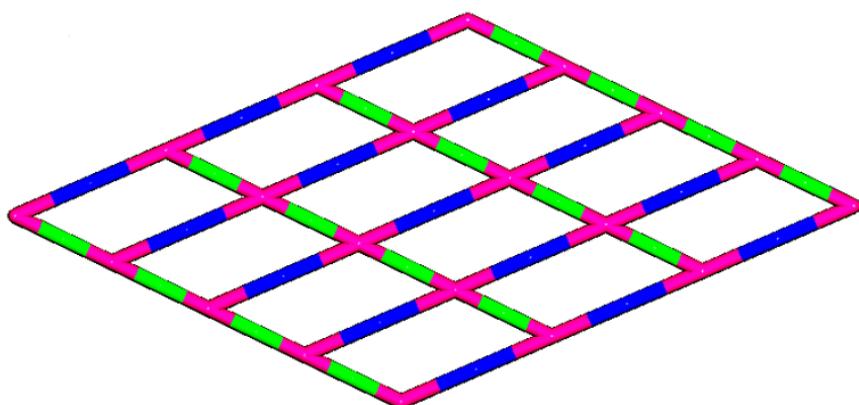


Fig. S14. Schematic representation of 2D topology of 6 (the pink sticks represent the Co center, green sticks represent the L3 ligand and the blue sticks represent *p*-bdc²⁻).

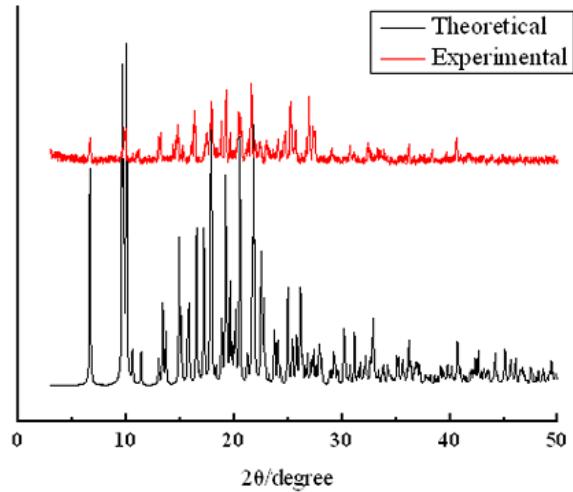


Fig. S15. The theoretical and experimental PXRD pattern of **1**

Fig. S16. The theoretical and experimental PXRD pattern of **2**

Fig. S17. The theoretical and experimental PXRD pattern of **3**

Fig. S18. The theoretical and experimental PXRD pattern of **4**

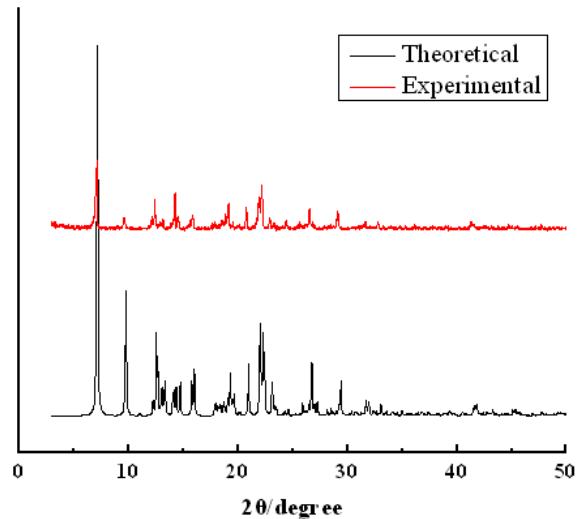


Fig. S19. The theoretical and experimental PXRD pattern of **5**

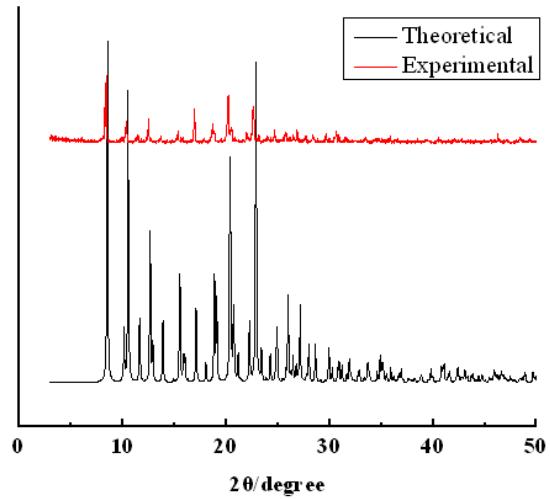


Fig. S20. The theoretical and experimental PXRD pattern of 6

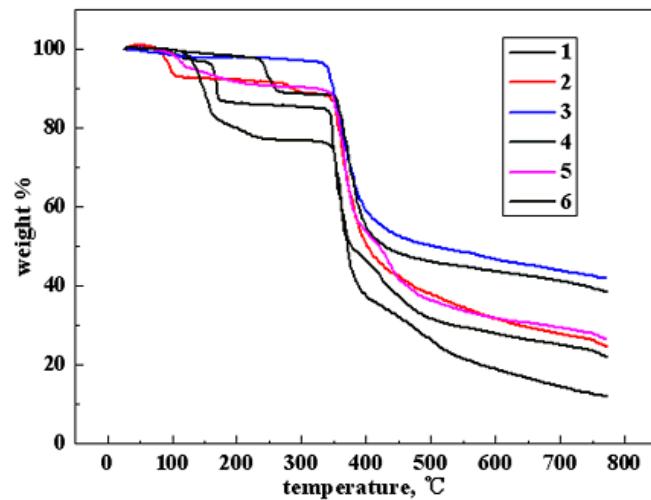


Fig. S21. TGA curves of compounds 1-6

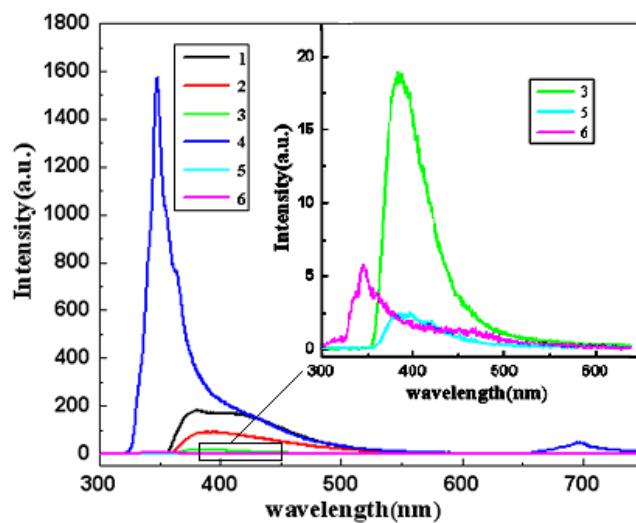


Fig. S22. Solid-state photoluminescence spectra for compounds 1-6

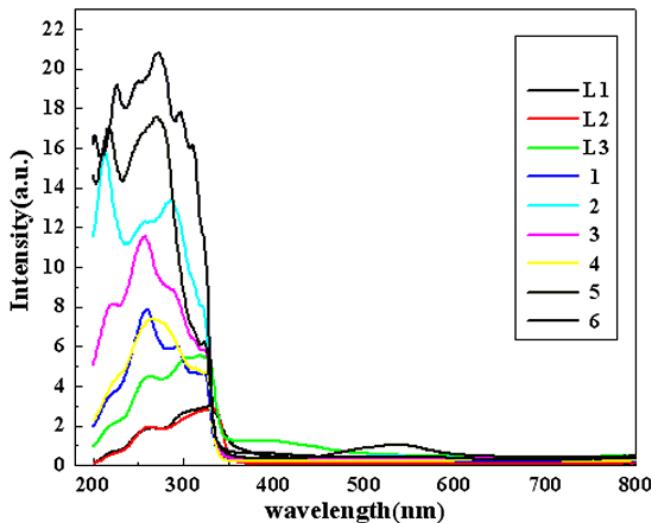


Fig. S23. Solid-state UV-vis diffuse reflectance spectra for compounds **1-6**

Table S1. Selected bond distances (\AA) and angles ($^{\circ}$) for **1**.

Cd(1)-O(5)	2.230(3)	Cd(1)-N(1)	2.250(3)
Cd(1)-O(3)	2.307(2)	Cd(1)-N(4)	2.264(3)
Cd(1)-O(4)	2.372(2)		
O(5)-Cd(1)-N(1)	120.38(9)	O(5)-Cd(1)-N(4)	98.48(10)
N(1)-Cd(1)-N(4)	92.32(10)	O(5)-Cd(1)-O(3)	116.06(10)
N(1)-Cd(1)-O(3)	96.99(9)	N(4)-Cd(1)-O(3)	132.15(9)
O(5)-Cd(1)-O(4)	103.63(9)	N(1)-Cd(1)-O(4)	135.64(9)
N(4)-Cd(1)-O(4)	86.20(9)	O(3)-Cd(1)-O(4)	55.33(8)

Symmetry transformations used to generate equivalent atoms: #1 1+x, y, z; #2 x, 1.5-y, 0.5+z.

Table S2. Selected bond distances (\AA) and angles ($^{\circ}$) for **2**.

Co(1)-O(3)	2.083(6)	Co(1)-N(1)	2.089(8)
Co(1)-N(4)	2.095(8)	Co(1)-O(6)	2.116(6)
Co(1)-O(2W)	2.156(7)	Co(1)-O(1W)	2.158(7)
O(3)-Co(1)-N(1)	90.8(3)	O(3)-Co(1)-N(4)	90.6(3)
N(1)-Co(1)-N(4)	178.5(3)	O(3)-Co(1)-O(6)	176.6(3)
N(1)-Co(1)-O(6)	90.7(3)	N(4)-Co(1)-O(6)	88.0(3)

O(3)-Co(1)-O(2W)	92.0(2)	N(1)-Co(1)-O(2W)	89.8(3)
N(4)-Co(1)-O(2W)	90.7(3)	O(6)-Co(1)-O(2W)	84.9(3)
O(3)-Co(1)-O(1W)	92.9(3)	N(1)-Co(1)-O(1W)	89.8(3)
N(4)-Co(1)-O(1W)	89.5(3)	O(6)-Co(1)-O(1W)	90.2(3)
O(2W)-Co(1)-O(1W)	175.1(2)		

Symmetry transformations used to generate equivalent atoms: ${}^{\#1}x-2, y-1, z; {}^{\#2}1-x, -y, 1-x, 3+z.$

Table S3. Selected bond distances (\AA) and angles ($^\circ$) for **3**.

Zn(1)-O5	1.942(3)	Zn(1)-N5	1.978(4)
Zn(1)-N1	1.982(4)	Zn(1)-O3	1.987(3)
Zn(2)-O9 ${}^{\#1}$	1.920(4)	Zn(2)-O8	1.929(4)
Zn(2)-N4 ${}^{\#1}$	1.957(5)	Zn(2)-N8 ${}^{\#3}$	1.968(4)
O(5)- Zn(1)-N(5)	103. 47(16)	O(5)- Zn(1)-N(1)	108.85(16)
N(5)- Zn(1)-N(1)	120.20(17)	O(5)- Zn(1)-O(3)	96. 70(15)
N(5)- Zn(1)-O(3)	109.43(15)	N(1)- Zn(1)-O(3)	114.89(16)
O(9) ${}^{\#1}$ - Zn(2)-O(8)	96.81(16)	O(9)- Zn(2)-N(4) ${}^{\#1}$	110.20(17)
O(8)- Zn(2)-N(4) ${}^{\#1}$	107.76(18)	O(9) ${}^{\#1}$ - Zn(2)-N(8) ${}^{\#3}$	112.77(18)
O(8)- Zn(2)-N(8) ${}^{\#3}$	116.32(17)	N(4) ${}^{\#1}$ - Zn(2)-N(8) ${}^{\#3}$	111.95(19)

Symmetry transformations used to generate equivalent atoms: ${}^{\#1}3-x, 1-y, 1-z; {}^{\#2}x-2, y-1, z; {}^{\#3}x+2, y+1, z.$

Table S4. Selected bond distances (\AA) and angles ($^\circ$) for **4**.

Zn(1)-O4	1.930(3)	Zn(1)-N1	2.005(3)
Zn(1)-O6 ${}^{\#1}$	2.007(4)	Zn(1)- N4 ${}^{\#2}$	2.033(3)
Zn(1)-O5	2.499(6)		
O(4)- Zn(1)-N(1)	115. 58(13)	O(4)- Zn(1)- O(6)	108.50(16)
N(1)- Zn(1)- O(6) ${}^{\#1}$	104.17(14)	O(4)- Zn(1)- N(4)	95. 35(13)
N(1)- Zn(1)- N(4)	102.83(13)	O(6) ${}^{\#1}$ - Zn(1)- N(4) ${}^{\#2}$	133.68(18)
O(6)- Zn(1)- N(5)	55.49(17)	O(5)- Zn(1)- N(4)	82.86(16)

O(5)- Zn(1)- N(1)	101.04(16)	O(5)- Zn(1)- O(4)	146.81(16)
O(5)- Zn(1)- O(6)	55.49(17)		

Symmetry transformations used to generate equivalent atoms: $\#^1 1-x, -0.5+y, 1.5-z;$
 $\#^2 1.5-x, -0.5+y, z.$

Table S5. Selected bond distances (\AA) and angles ($^\circ$) for **5**.

Co(1)-O5	2.041(6)	Co(1)-N1	2.087(7)
Co(1)- N4 $^{\#1}$	2.113(9)	Co(1)-O3	2.169(5)
Co(1)-O4	2.246(6)	Co(1)- O1W	2.136(6)
O(5)-Co(1)-N(1)	101. 6(3)	O(5)-Co(1)- N(4) $^{\#1}$	92.3(3)
N(1)-Co(1)- N(4) $^{\#1}$	89.4(3)	O(5)-Co(1)- O(1W)	89.8(2)
N(1)-Co(1)- O(1W)	91.7(3)	N(4) $^{\#1}$ -Co(1)- O(1W)	177.4(3)
O(5)-Co(1)-O(3)	103.6(2)	N(1)-Co(1)-O(3)	154.7(3)
N(4)-Co(1)-O(3)	90.8(3)	O(1W)-Co(1)-O(3)	87.4(2)
O(5)-Co(1)-O(4)	162.2(2)	N(1)-Co(1)-O(4)	95.4(2)
N(4) $^{\#1}$ -Co(1)-O(4)	92.9(3)	O(1W)-Co(1)-O(4)	84.6(2)
O(3)-Co(1)-O(4)	59.3(2)		

Symmetry transformations used to generate equivalent atoms: $\#^1-x, 1-y, -z.$

Table S6. Selected bond distances (\AA) and angles ($^\circ$) for **6**.

Co(1)-O2 $^{\#1}$	2.097(3)	Co(1)- O1W $^{\#1}$	2.098(4)
Co(1)- N1 $^{\#1}$	2.128(3)		
O(2) $^{\#1}$ -Co(1)- O(2)	180. 00(1)	O(2)-Co(1)- O(1W)	88.19(12)
O(2) $^{\#1}$ -Co(1)- O(1W)	91.81(12)	O(1W)-Co(1)-O(1W)	180. 00(1)
O(2) $^{\#1}$ -Co(1)- N(1)	94.37(11)	O(2)-Co(1)- N(1) $^{\#1}$	85.63(11)
O(1W) $^{\#1}$ -Co(1)- N(1) $^{\#1}$	88.63(14)	O(1W)-Co(1)- N(1) $^{\#1}$	91.37(14)
N(1) $^{\#1}$ -Co(1)- N(1)	180. 00(2)		

Symmetry transformations used to generate equivalent atoms: $\#^1-x, -2-y, 2+z.$