

Supporting Information

Bilayer architecture based on heterometallic hexanuclear units

Wei-Hui Fang,^[a] and Guo-Yu Yang*^[a]

^[a] State Key Laboratory of Structural Chemistry, Fujian Institute of Research on the Structure of Matter, Chinese Academy of Sciences, Fuzhou, Fujian 350002, China

CONTENT

Supporting Tables:

[Table S1](#) Selected bond lengths (Å) for compound **1**.

[Table S2](#) BVS for Eu, Cu and hydroxyl oxygen atoms in **1**.

[Table S3](#) Hydrogen bond lengths (Å) and bond angles (°) for compounds **1**.

Supporting Figures:

[Fig. S1](#) The ball-stick view of the coordination environment of Eu ion.

[Fig. S2](#) The ball-stick view of the coordination environments of the Cu^{II} (up) and Cu^I (down) in compound **1**.

[Scheme S1](#) The coordination modes of HL ligands and related references.

[Fig. S3](#) Side view shows the thickness of the monolayers along the *b* axis.

[Fig. S4](#) The coordination environment of Eu^{III}₂Cu^{II}₄ core in **1**.

[Fig. S5](#) (a) Supramolecular structure of compound **1**. (b) Emphasized π - π interaction.

[Fig. S6](#) The bilayer architecture in **1**'.

[Fig. S7](#) Simulated and experimental PXRD patterns of **1**.

[Fig. S8](#) The IR spectrum of compound **1**.

[Fig. S9](#) TGA curves of compound **1**.

[Fig. S10](#) Solid-state emission spectra of HL and compound **1** at room temperature.

Supporting Tables:

Table S1 Selected bond lengths (Å) for compound **1^a**

Eu-O(1)	2.393(7)	Cu(1)-O(9)	1.950(7)
Eu-O(1W)	2.506(9)	Cu(1)-O(10)	1.932(6)
Eu-O(4)	2.398(7)	Cu(1)-O(14)	2.842(3)
Eu-O(5)	2.361(7)	Cu(2)-N(2)C	2.010(9)
Eu-O(6)A	2.483(7)	Cu(2)-O(2)	1.971(7)
Eu-O(7)	2.374(8)	Cu(2)-O(6)A	2.502(7)
Eu-O(8)A	2.468(7)	Cu(2)-O(9)	1.972(7)
Eu-O(10)	2.409(7)	Cu(2)-O(10)	1.936(6)
Cu(1)-N(1)B	1.982(8)	Cu(2)-O(13)	2.557(2)
Cu(1)-O(3)	1.976(7)	Cu(3)-N(3)	2.036(1)
Cu(1)-O(8)A	2.506(7)	Cu(3)-N(4)D	2.030(5)

^a Symmetry transformations used to generate equivalent atoms: A: $-x+2, -y+1, -z$; B: $x, y-1, z$; C: $x, y, z-1$; D: $x, y-1, z+1$.

Table S2 BVS for Eu, Cu and hydroxyl oxygen atoms in **1^a**

Atom	BVS
Eu	2.84
Cu1	1.95
Cu2	1.95
Cu3	0.64
O9	0.85
O10	1.37

^a $BVS = \exp[(R1-d)/0.37]$

Table S3 Hydrogen bond lengths (Å) and bond angles (°) for compounds **1^a**

D-H...A	d(D-H)	d(H...A)	d(D...A)	<DHA
O9-H(0B)...O2W	0.90	2.29	2.87(1)	122.4
O10-H(0A)...O3W	0.96	1.90	2.85(2)	168.3
O4W-H(42)...O5 ^A	0.83	2.37	3.14(1)	157.1

^a Symmetry code: A: $-x+2, -y+1, -z+3$.

Supporting Figures:

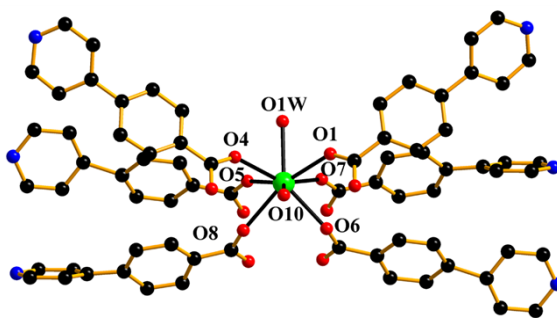


Fig. S1 The ball-stick view of the coordination environment of Eu ion. Atoms having “A” in their labels are symmetry-generated. Symmetry code A: $-x+2, -y+1, -z$.

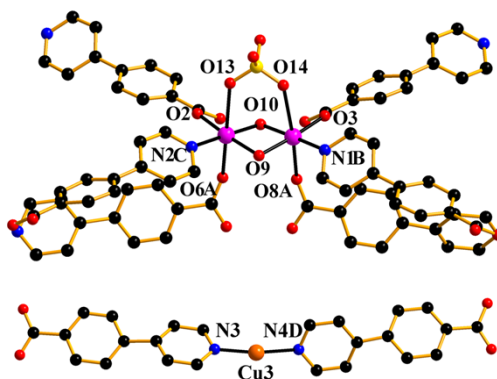
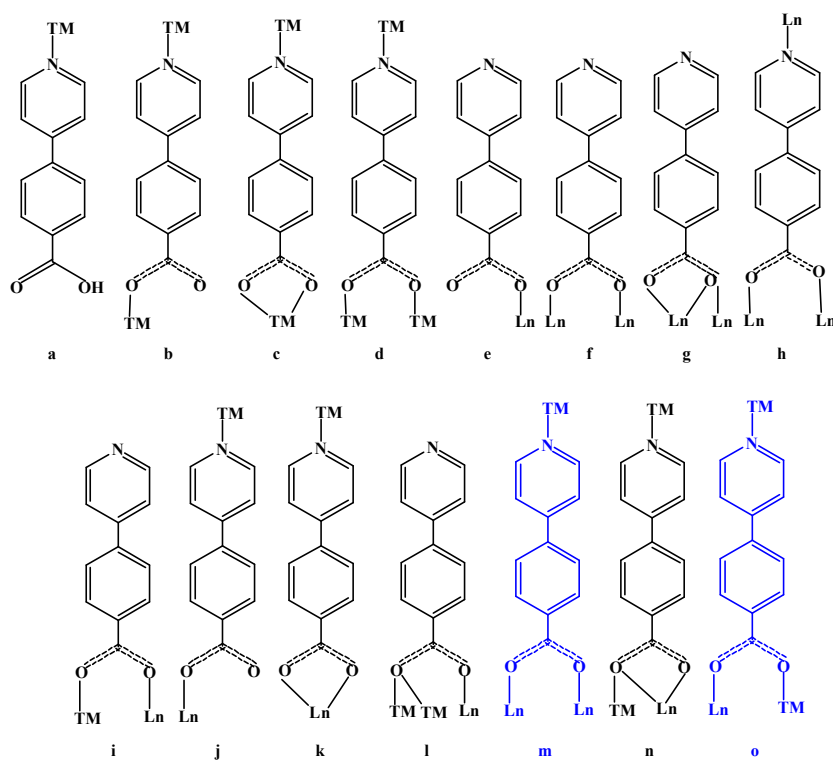


Fig. S2 The ball-stick view of the coordination environments of the Cu^{II} (up) and Cu^I (down) in compound **1**. Atoms having B-D in their labels are symmetry-generated. Symmetry codes B: $x, y-1, z$; C: $x, y, z-1$; D: $x, y-1, z+1$.



Scheme S1 The coordination modes of HL ligands and related references.

The mode (a) is seen in the reference 1; The mode (b) is given in the reference 2; The mode (c) is found in the reference

2a,c,3; The mode (d) is seen in the reference 2b,4; The modes (e,i,l,n) are seen in the reference 5; The modes (f-h) are seen in the reference 6; The mode (j) is seen in the reference 7; The mode (k) is seen in the reference 7,8; The mode (m) is seen in the reference 7,9 and herein; The mode (o) is seen in the reference herein.

- (a) R. Sekiya, S. Nishikiori and K. Ogura, *Inorg. Chem.*, 2006, **45**, 9233; (b) R. Sekiya and S. Nishikiori, *Cryst. Growth Des.*, 2011, **11**, 5574; (c) G. C. Ou, J. Z. Gu, T. B. Lu and R. L. Luck, *J. Mol. Struct.*, 2005, **740**, 143; (d) P. Teo, L. L. Koh and T. S. A. Hor, *Inorg. Chem.*, 2008, **47**, 6464; (e) X. L. Jia, J. Zhou, S. T. Zheng and G. Y. Yang, *J. Cluster Sci.*, **2009**, **20**, 555; (f) R. Sekiya and S. Nishikiori, K. Ogura, *Chem. Lett.*, 2006, **35**, 614.
- (a) O. R. Evans and W. B. Lin, *Chem. Mater.*, 2001, **13**, 2705; (b) X. M. Zhang, Y. Q. Wang, Y. Song and E. Q. Gao, *Inorg. Chem.*, 2011, **50**, 7284; (c) T. B. Lu and R. L. Luck, *Inorg. Chim. Acta*, 2003, **351**, 345.
- (a) X. J. Zhou, B. Y. Li, G. H. Li, Q. Zhou, Z. Shi and S. H. Feng, *CrystEngComm*, 2012, **14**, 4664; (b) G. Mehlana, S. A. Bourne and G. Ramon, *Dalton Trans.*, 2012, **41**, 4224; (c) Y. P. He, Y. X. Tan and J. Zhang, *CrystEngComm*, 2012, **14**, 6359.
- (a) Y. B. Zhang, H. L. Zhou, R. B. Lin, C. Zhang, J. B. Lin, J. P. Zhang and X. M. Chen, *Nat. Commun.*, 2012, **3**, 642; (b) M. H. Zeng, Q. X. Wang, Y. X. Tan, S. Hu, H. X. Zhao, L. S. Long and M. Kurmoo, *J. Am. Chem. Soc.*, 2010, **132**, 2561; (c) X. M. Zhang, K. Wang, Y. Q. Wang and E. Q. Gao, *Dalton Trans.*, 2011, **40**, 12742; (d) M. C. Das, H. Xu, S. C. Xiang, Z. J. Zhang, H. D. Arman, G. D. Qian and B. L. Chen, *Chem.-Eur. J.*, 2011, **17**, 7817; (e) Z. Yin, Q. X. Wang and M. H. Zeng, *J. Am. Chem. Soc.*, 2012, **134**, 4857.
- M. B. Zhang, H. M. Chen, R. X. Hu and Z. L. Chen, *CrystEngComm*, 2011, **13**, 7019.
- (a) Z. L. Wang, W. H. Fang and G. Y. Yang, *Chinese. J. Struct. Chem.*, 2009, **28**, 1453; (b) Z. L. Wang, W. H. Fang and G. Y. Yang, *J. Cluster Sci.*, 2009, **20**, 725; (c) W. H. Fang, Z. L. Wang and G. Y. Yang, *J. Cluster Sci.*, 2010, **21**, 187; (d) W. H. Fang, L. Cheng, L. Huang and G. Y. Yang, *Inorg. Chem.*, 2013, **52**, 6.
- W. H. Fang and G. Y. Yang, *CrystEngComm*, 2013, **15**, 9504.
- Z. L. Wang, W. H. Fang and G. Y. Yang, *Chem. Commun.*, 2010, **46**, 8216.
- H. M. Chen, R. X. Hu and M. B. Zhang, *Inorg. Chim. Acta*, 2011, **379**, 34.

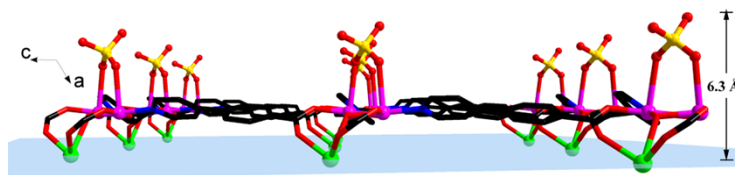


Fig. S3 Side view shows the thickness of the monolayers along the *b* axis.

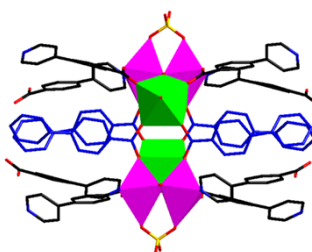


Fig. S4 The coordination environment of Eu^{III}₂Cu^{II}₄ core in **1**. Twelve ligands: four in mode I (blue), eight in mode II (black).

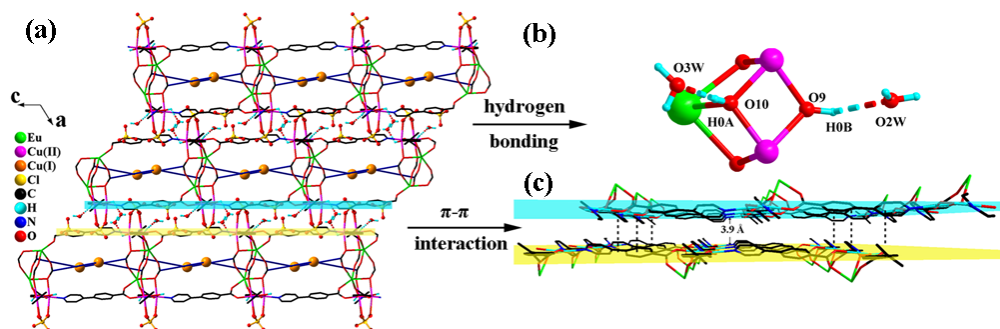


Fig. S5 (a) Supermolecular structure of compound **1**. (b-c) Emphasized hydrogen bonding and π-π interaction. Perchlorate groups are omitted for clarity.

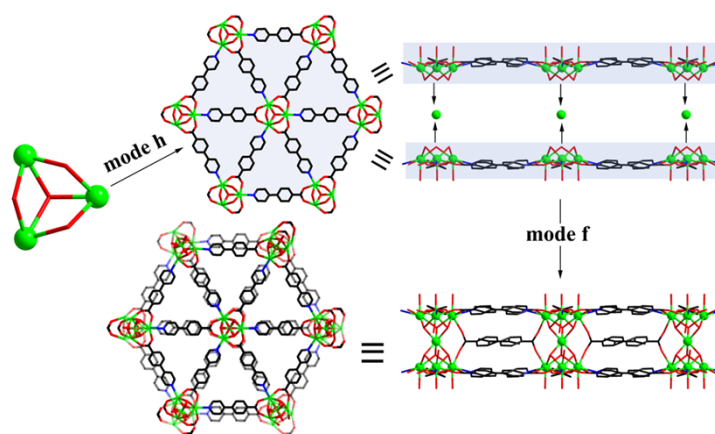


Fig. S6 The bilayer architecture in **1**' (ref. 16).

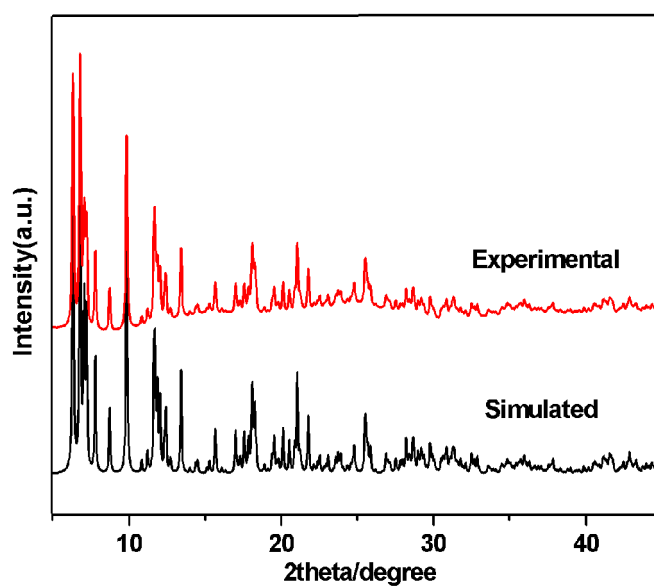


Fig. S7 Simulated and experimental PXRD patterns of **1**.

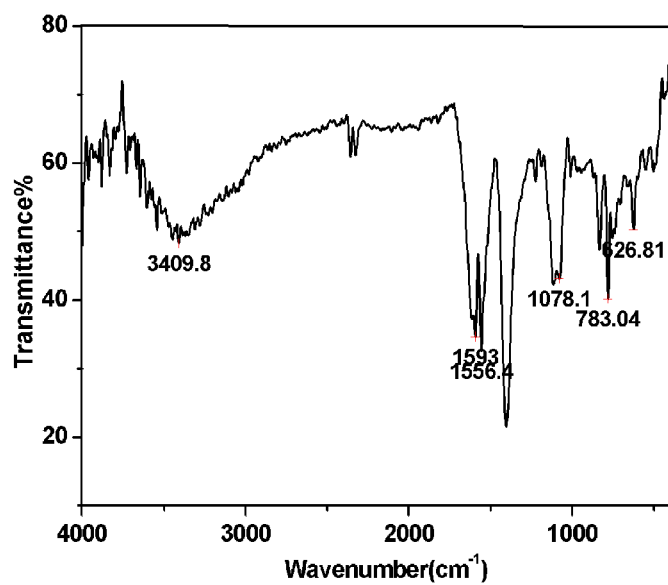


Fig. S8 The IR spectrum of compound **1**.

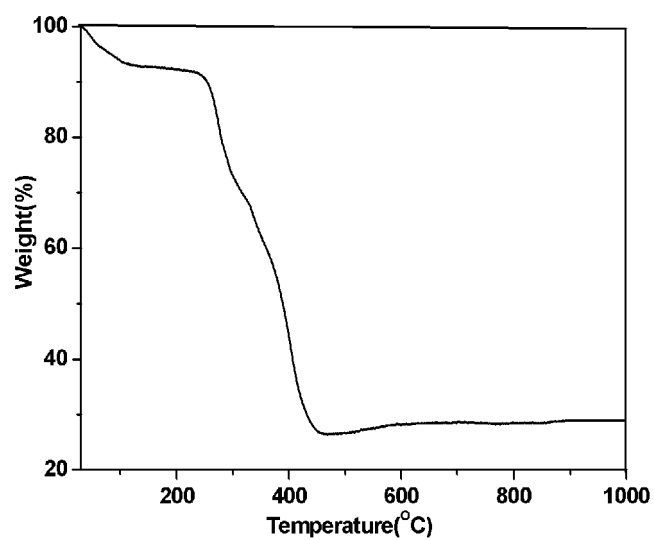


Fig. S9 TGA curve of compound 1.

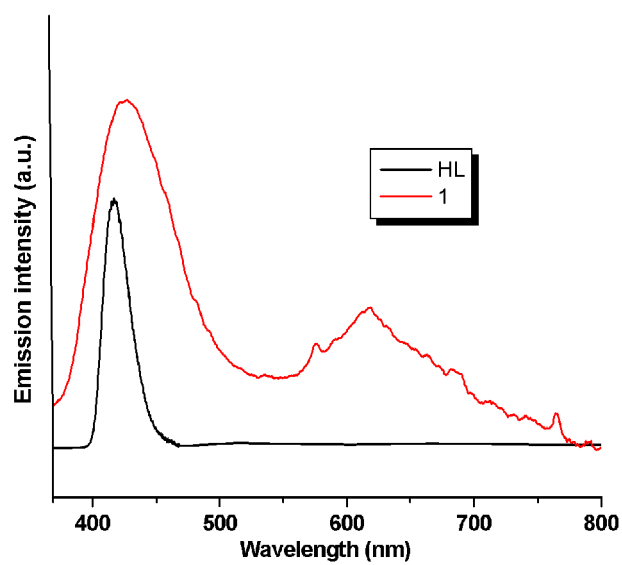


Fig. S10 Solid-state emission spectra of HL and compound 1 at room temperature.