

## Supplementary Information

# Supramolecular Interactions Mediated Conformational Modulation of Flexible Linker Leading to Isolation of Metallo-Macrocycle in the Polyoxometalate Matrix: Hirshfeld surfaces and 2D finger print plots

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### Contents

**Section 1:** Additional diagrams of crystal structure

**Section 2:** Theoretical Calculations

- a. Optimized geometries and energies of cis and trans conformations of 1,4-bpimb linker
- b. Hirshfeld surfaces mapped with shape index (left) and curvedness (right) for the title compound

**Section 3:**

- a. Thermogravimetric curve
- b. PXRD patterns
- c. IR spectrum

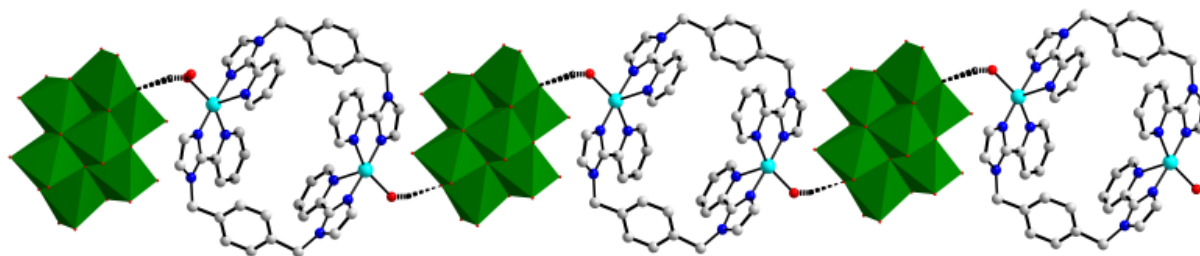
**Section 4:**

**Table 1** List of bond lengths in the octamolybdate anion and Cu(II) coordination sphere

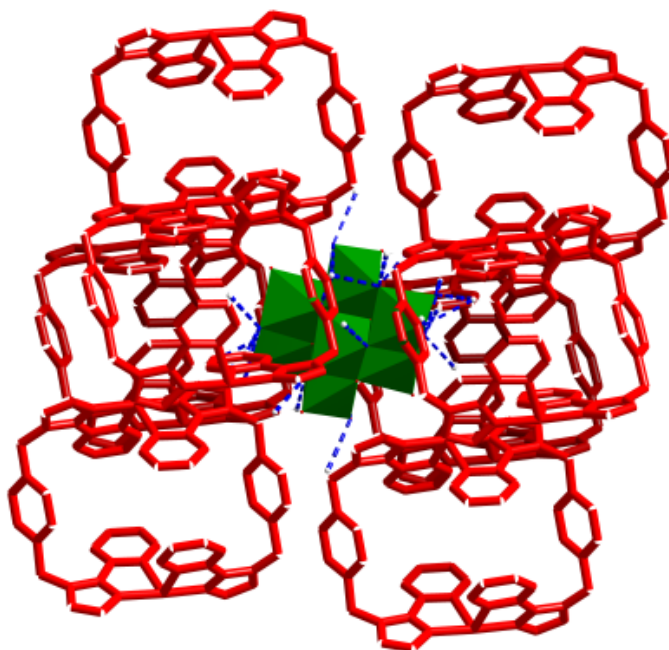
**Table 2** Complete list of bond lengths and bond angles

## Section 1

### Additional diagrams of crystal structure



**Fig. S1. 1D chain formed due to O-H...O interactions forming a 1D chain.**



**Fig. S2. Supramolecular interactions around the anion.**

## Section 2

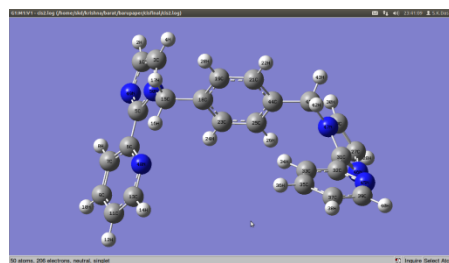
### Theoretical Calculations

#### a. Optimized geometries and energies of cis and trans conformations of 1,4-bpimb linker

```
cis2.log (-/krishna/barat/barupaper/cisfinal) - VIM
skd@skd: ~
cis2.log (-/krishna/barat/barupaper/cisfinal)
1640.66 1668.52 1687.54 1689.98 1692.63
1724.16 1735.53 1737.30 1746.59 1765.86
1830.04 1840.03 1844.83 1866.70 1870.01
1874.29 1884.19 1897.32 1903.63 1933.29
1988.09 1995.64 2010.90 2029.39 2061.37
2067.87 2085.34 2089.40 2107.66 2118.22
2129.24 2145.81 2159.76 2172.01 2176.87
2223.94 2232.68 2235.24 2306.04 2315.51
2323.40 2339.35 2344.13 2379.77 4375.84
4399.57 4445.55 4511.86 4526.98 4528.80
4531.26 4553.61 4562.82 4567.16 4571.93
4588.83 4592.55 4597.60 4598.36 4629.85
4654.75 4656.92 4693.84 4696.78

Zero-point correction= 0.397187 (Hartree/Particle)
Thermal correction to Energy= 0.420990
Thermal correction to Enthalpy= 0.421935
Thermal correction to Gibbs Free Energy= 0.336960
Sum of electronic and Zero-point Energies= -1255.017264
Sum of electronic and thermal Energies= -1254.993461
Sum of electronic and thermal Enthalpies= -1254.992516
Sum of electronic and thermal Free Energies= -1255.077491

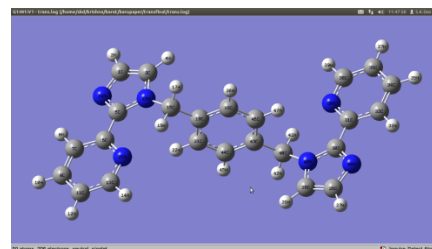
E (Thermal) CV S
KCal/Mol Cal/Mol-Kelvin Cal/Mol-Kelvin
Total 264.175 93.120 178.044
Electronic 0.000 0.000 0.000
Translational 0.889 2.981 43.791
Rotational 0.889 2.981 30.831
Vibrational 262.398 87.158 98.221
Vibration 1 0.593 1.987 8.767
Vibration 2 0.593 1.987 7.775
Vibration 3 0.593 1.986 7.229
Vibration 4 0.593 1.985 6.298
Vibration 5 0.594 1.983 5.539
Vibration 6 0.594 1.982 5.487
Vibration 7 0.595 1.980 5.068
Vibration 8 0.598 1.970 4.251
```



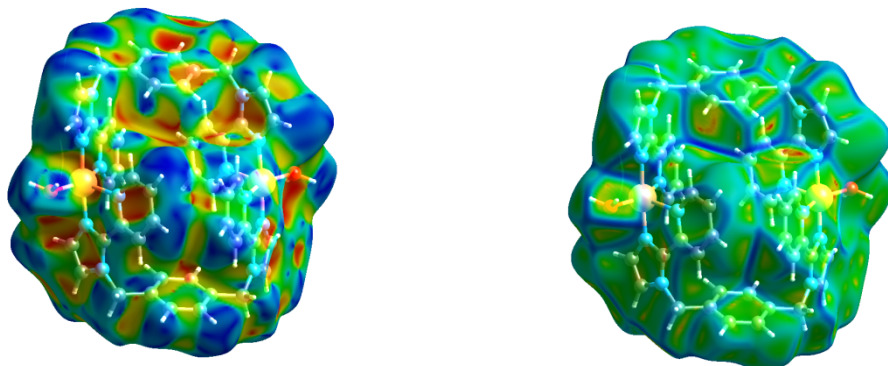
```
trans.log (-/krishna/barat/barupaper/transfinal) - VIM
skd@skd: ~
trans.log (-/krishna/barat/barupaper/transfinal)
1638.75 1687.48 1687.49 1692.31 1692.67
1732.25 1736.53 1741.46 1751.99 1769.23
1832.92 1834.14 1866.26 1867.68 1870.51
1876.03 1883.16 1888.91 1918.80 1930.25
1988.42 1998.38 2027.79 2030.21 2060.02
2061.81 2088.46 2105.84 2105.89 2118.80
2120.32 2146.57 2146.96 2177.30 2177.88
2224.25 2232.02 2232.44 2305.45 2305.54
2323.89 2343.98 2344.00 2380.19 4396.70
4396.71 4498.36 4499.09 4530.34 4530.35
4539.60 4541.82 4565.67 4565.68 4579.80
4582.18 4596.05 4596.67 4630.82 4630.83
4656.05 4656.06 4697.69 4697.70

Zero-point correction= 0.397540 (Hartree/Particle)
Thermal correction to Energy= 0.421327
Thermal correction to Enthalpy= 0.422271
Thermal correction to Gibbs Free Energy= 0.335464
Sum of electronic and Zero-point Energies= -1255.029234
Sum of electronic and thermal Energies= -1255.005447
Sum of electronic and thermal Enthalpies= -1255.004503
Sum of electronic and thermal Free Energies= -1255.091310

E (Thermal) CV S
KCal/Mol Cal/Mol-Kelvin Cal/Mol-Kelvin
Total 264.387 92.988 182.700
Electronic 0.000 0.000 0.000
Translational 0.889 2.981 43.791
Rotational 0.889 2.981 36.895
Vibrational 262.609 87.027 102.013
Vibration 1 0.592 1.987 10.402
Vibration 2 0.593 1.987 8.374
Vibration 3 0.593 1.986 7.345
Vibration 4 0.593 1.986 6.853
Vibration 5 0.593 1.985 6.487
Vibration 6 0.593 1.985 6.003
Vibration 7 0.595 1.979 4.937
Vibration 8 0.597 1.971 4.328
```

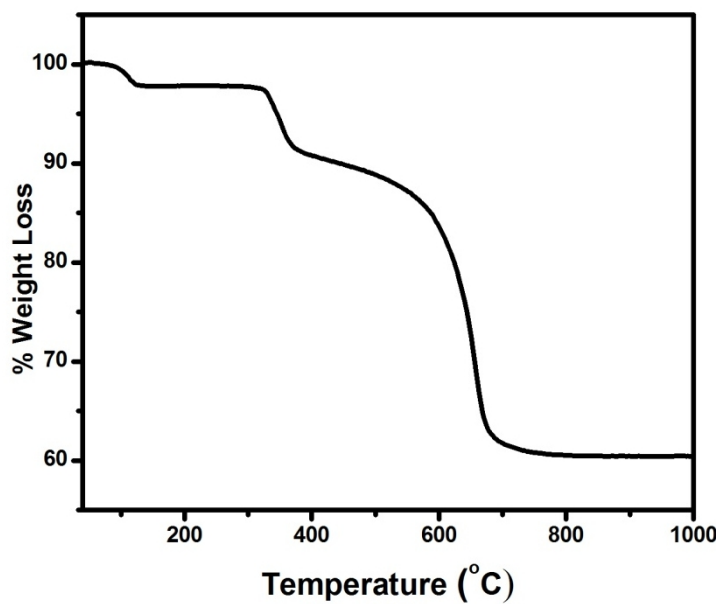


**b. Hirshfeld surfaces mapped with shape index (left) and curvedness (right) for the title compound**

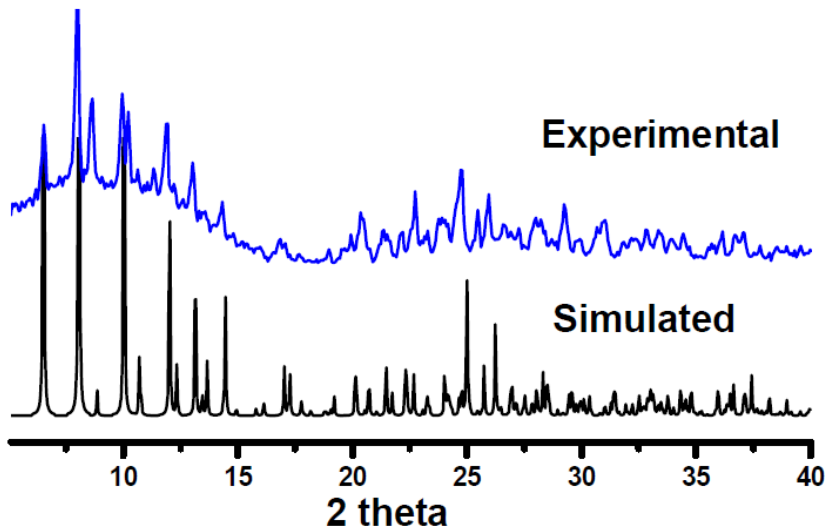


### **Section 3**

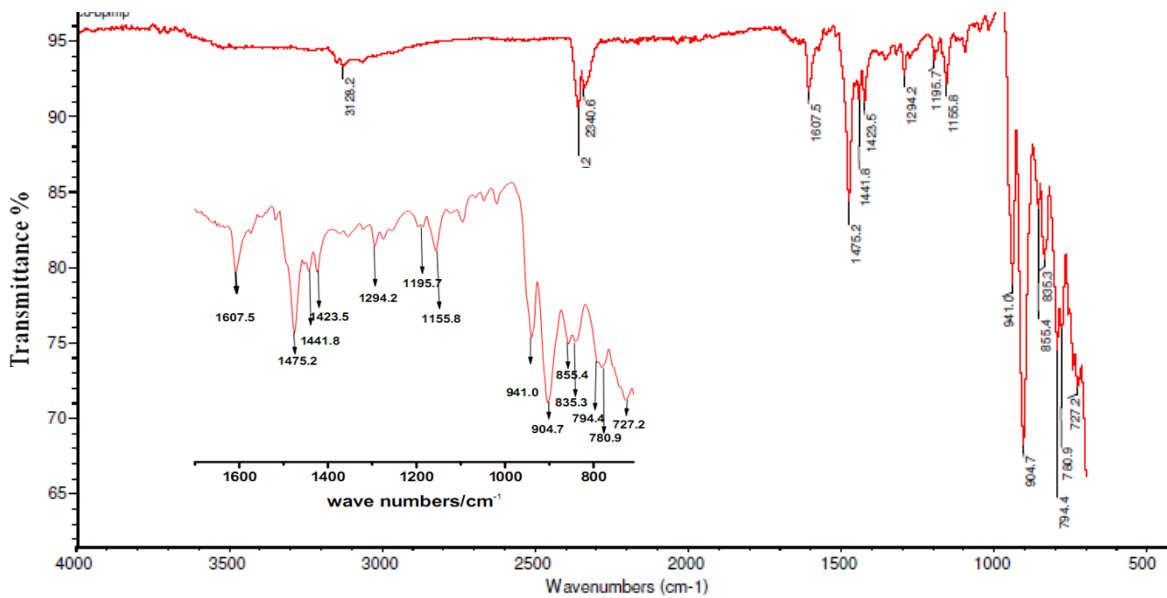
**a. Thermogravimetric curve of  $[\text{Cu}(\text{1,4-bpimb})(\text{H}_2\text{O})_2][\text{Mo}_8\text{O}_{26}] \cdot 4\text{H}_2\text{O}$**



**b. PXRD of  $[\text{Cu}(1,4\text{-bpimb})(\text{H}_2\text{O})_2][\text{Mo}_8\text{O}_{26}] \cdot 4\text{H}_2\text{O}$**



**c. IR spectrum of  $[\text{Cu}(1,4\text{-bpimb})(\text{H}_2\text{O})_2][\text{Mo}_8\text{O}_{26}] \cdot 4\text{H}_2\text{O}$**



## Section 4

**Table 1.** List of bond lengths in the octamolybdate anion and Cu(II) coordination sphere

Mo-O <sub>t</sub>	Mo-O(μ <sub>2</sub> )	Mo-O(μ <sub>3</sub> )	Mo-O(μ <sub>4</sub> )	Cu-N/ Cu-O
Mo(1)-O(1) 1.685(6)	Mo(1)-O(2) 1.740(6)	Mo(1)-O(3) 1.946(5)	Mo(1)-O(13) 2.131(5)	Cu(5)-N(1) 2.051(7)
Mo(2)-O(5) 1.698(6)	Mo(2)-O(6) 1.901(5)	Mo(1)-O(12) 1.946(6)	Mo(1)-O(13) <sup>#1</sup> 2.377(5)	Cu(5)-N(2) 1.932(7)
Mo(2)-O(4) 1.711(6)	Mo(3)-O(9) 1.920(5)	Mo(2)-O(3) 1.973(6)	Mo(2)-O(13) 2.318(5)	Cu(5)-N(4) 2.197(7)
Mo(3)-O(10) 1.697(6)	Mo(4)-O(6) 1.898(5)	Mo(2)-O(12) <sup>#1</sup> 2.348(6)	Mo(3)-O(13) 2.299(5)	Cu(5)-N(5) 1.963(8)
Mo(3)-O(11) 1.699(6)	Mo(4)-O(9) 1.925(6)	Mo(3)-O(12) 1.972(6)	Mo(4)-O(13) 2.493(5)	Cu(5)-O(16) 2.037(7)
Mo(4)-O(8) 1.687(6)	O(2)-Mo(4) <sup>#1</sup> 2.274(6)	Mo(3)-O(3) <sup>#1</sup> 2.357(6)		
Mo(4)-O(7) 1.701(6)				
#1 -x+1, -y+2, -z				

**Table 2**

### Complete list of bond lengths and bond angles

C(1)-C(2)	1.328(13)	C(20)-C(21)	1.354(13)
C(1)-N(5)	1.356(11)	C(21)-C(22)	1.387(13)
C(2)-N(6)	1.377(12)	C(22)-N(1)	1.319(11)
C(3)-N(5)	1.313(11)	C(23)-N(3)	1.456(11)
C(3)-N(6)	1.338(10)	C(23)-C(24)	1.489(12)
C(3)-C(4)	1.498(12)	C(24)-C(14) <sup>#1</sup>	1.374(12)
C(4)-N(4)	1.340(11)	C(24)-C(12) <sup>#1</sup>	1.403(12)
C(4)-C(5)	1.377(12)	Cu(5)-N(2)	1.932(7)
C(5)-C(6)	1.401(12)	Cu(5)-N(5)	1.963(8)
C(6)-C(7)	1.378(12)	Cu(5)-O(16)	2.037(7)
C(7)-C(8)	1.391(12)	Cu(5)-N(1)	2.051(7)
C(8)-N(4)	1.319(11)	Cu(5)-N(4)	2.197(7)
C(9)-N(6)	1.463(11)	Mo(1)-O(1)	1.685(6)
C(9)-C(10)	1.512(13)	Mo(1)-O(2)	1.740(6)
C(10)-C(11)	1.368(12)	Mo(1)-O(3)	1.946(5)
C(10)-C(13)	1.402(12)	Mo(1)-O(12)	1.946(6)
C(11)-C(12)	1.379(12)	Mo(1)-O(13)	2.131(5)
C(12)-C(24) <sup>#1</sup>	1.403(12)	Mo(1)-O(13) <sup>#2</sup>	2.377(5)
C(13)-C(14)	1.390(12)	Mo(2)-O(5)	1.698(6)
C(14)-C(24) <sup>#1</sup>	1.374(12)	Mo(2)-O(4)	1.711(6)
C(15)-C(16)	1.345(12)	Mo(2)-O(6)	1.901(5)
C(15)-N(2)	1.365(10)	Mo(2)-O(3)	1.973(6)
C(16)-N(3)	1.365(10)	Mo(2)-O(13)	2.318(5)
C(17)-N(2)	1.321(11)	Mo(2)-O(12) <sup>#2</sup>	2.348(6)
C(17)-N(3)	1.357(11)	Mo(3)-O(10)	1.697(6)
C(17)-C(18)	1.462(12)	Mo(3)-O(11)	1.699(6)
C(18)-N(1)	1.367(11)	Mo(3)-O(9)	1.920(5)
C(18)-C(19)	1.375(12)	Mo(3)-O(12)	1.972(6)
C(19)-C(20)	1.376(12)		

Mo(3)-O(13)	2.299(5)	N(3)-C(23)-C(24)	113.6(7)
Mo(3)-O(3)#2	2.357(6)	C(14)#1-C(24)-C(12)#1	118.2(8)
Mo(4)-O(8)	1.687(6)	C(14)#1-C(24)-C(23)	123.5(8)
Mo(4)-O(7)	1.701(6)	C(12)#1-C(24)-C(23)	118.3(8)
Mo(4)-O(6)	1.898(5)	N(2)-Cu(5)-N(5)	177.3(3)
Mo(4)-O(9)	1.925(6)	N(2)-Cu(5)-O(16)	93.4(3)
Mo(4)-O(2)#2	2.274(6)	N(5)-Cu(5)-O(16)	88.6(3)
Mo(4)-O(13)	2.493(5)	N(2)-Cu(5)-N(1)	81.2(3)
O(2)-Mo(4)#2	2.274(6)	N(5)-Cu(5)-N(1)	97.7(3)
O(3)-Mo(3)#2	2.357(6)	O(16)-Cu(5)-N(1)	155.4(3)
O(12)-Mo(2)#2	2.348(6)	N(2)-Cu(5)-N(4)	99.9(3)
O(13)-Mo(1)#2	2.377(5)	N(5)-Cu(5)-N(4)	78.1(3)
		O(16)-Cu(5)-N(4)	94.0(3)
C(2)-C(1)-N(5)	110.2(8)	N(1)-Cu(5)-N(4)	110.6(3)
C(1)-C(2)-N(6)	106.4(8)	O(1)-Mo(1)-O(2)	104.3(3)
N(5)-C(3)-N(6)	111.0(7)	O(1)-Mo(1)-O(3)	101.8(3)
N(5)-C(3)-C(4)	119.2(8)	O(2)-Mo(1)-O(3)	96.3(2)
N(6)-C(3)-C(4)	129.8(7)	O(1)-Mo(1)-O(12)	101.2(3)
N(4)-C(4)-C(5)	123.2(8)	O(2)-Mo(1)-O(12)	97.1(2)
N(4)-C(4)-C(3)	111.3(7)	O(3)-Mo(1)-O(12)	149.5(2)
C(5)-C(4)-C(3)	125.4(8)	O(1)-Mo(1)-O(13)	99.3(2)
C(4)-C(5)-C(6)	117.8(8)	O(2)-Mo(1)-O(13)	156.4(2)
C(7)-C(6)-C(5)	119.1(8)	O(3)-Mo(1)-O(13)	78.7(2)
C(6)-C(7)-C(8)	118.6(8)	O(12)-Mo(1)-O(13)	78.1(2)
N(4)-C(8)-C(7)	122.8(9)	O(1)-Mo(1)-O(13)#2	174.8(2)
N(6)-C(9)-C(10)	113.1(7)	O(2)-Mo(1)-O(13)#2	80.9(2)
C(11)-C(10)-C(13)	119.3(9)	O(3)-Mo(1)-O(13)#2	77.8(2)
C(11)-C(10)-C(9)	118.0(8)	O(12)-Mo(1)-O(13)#2	77.4(2)
C(13)-C(10)-C(9)	122.7(8)	O(13)-Mo(1)-O(13)#2	75.5(2)
C(10)-C(11)-C(12)	121.5(8)	O(5)-Mo(2)-O(4)	105.1(3)
C(11)-C(12)-C(24)#1	120.1(8)	O(5)-Mo(2)-O(6)	100.4(3)
C(14)-C(13)-C(10)	119.0(8)	O(4)-Mo(2)-O(6)	101.7(3)
C(24)#1-C(14)-C(13)	121.9(8)	O(5)-Mo(2)-O(3)	100.3(3)
C(16)-C(15)-N(2)	108.8(8)	O(4)-Mo(2)-O(3)	98.3(3)
C(15)-C(16)-N(3)	107.4(7)	O(6)-Mo(2)-O(3)	146.2(2)
N(2)-C(17)-N(3)	110.0(8)	O(5)-Mo(2)-O(13)	160.0(2)
N(2)-C(17)-C(18)	118.7(8)	O(4)-Mo(2)-O(13)	94.8(2)
N(3)-C(17)-C(18)	131.2(8)	O(6)-Mo(2)-O(13)	77.7(2)
N(1)-C(18)-C(19)	121.3(8)	O(3)-Mo(2)-O(13)	73.7(2)
N(1)-C(18)-C(17)	111.2(8)	O(5)-Mo(2)-O(12)#2	88.5(2)
C(19)-C(18)-C(17)	127.5(8)	O(4)-Mo(2)-O(12)#2	164.5(2)
C(18)-C(19)-C(20)	118.4(9)	O(6)-Mo(2)-O(12)#2	82.7(2)
C(21)-C(20)-C(19)	120.1(9)	O(3)-Mo(2)-O(12)#2	71.5(2)
C(20)-C(21)-C(22)	119.3(8)	O(13)-Mo(2)-O(12)#2	71.44(19)
N(1)-C(22)-C(21)	121.6(9)	O(10)-Mo(3)-O(11)	104.2(3)
O(10)-Mo(3)-O(9)	101.8(3)	O(9)-Mo(3)-O(3)#2	84.1(2)
O(11)-Mo(3)-O(9)	99.9(3)	O(12)-Mo(3)-O(3)#2	71.3(2)
O(10)-Mo(3)-O(12)	97.1(3)	O(13)-Mo(3)-O(3)#2	71.96(19)
O(11)-Mo(3)-O(12)	101.8(3)	O(8)-Mo(4)-O(7)	104.3(3)
O(9)-Mo(3)-O(12)	146.6(2)	O(8)-Mo(4)-O(6)	104.4(3)
O(10)-Mo(3)-O(13)	94.9(2)	O(7)-Mo(4)-O(6)	98.0(3)
O(11)-Mo(3)-O(13)	160.9(2)	O(8)-Mo(4)-O(9)	102.3(3)
O(9)-Mo(3)-O(13)	77.5(2)	O(7)-Mo(4)-O(9)	98.8(3)
O(12)-Mo(3)-O(13)	73.6(2)	O(6)-Mo(4)-O(9)	143.7(2)
O(10)-Mo(3)-O(3)#2	164.3(2)	O(8)-Mo(4)-O(2)#2	89.4(2)
O(11)-Mo(3)-O(3)#2	88.9(2)	O(7)-Mo(4)-O(2)#2	166.3(2)

O(6)-Mo(4)-O(2)#2	78.5(2)	Mo(1)-O(3)-Mo(3)#2	109.7(3)
O(9)-Mo(4)-O(2)#2	77.7(2)	Mo(2)-O(3)-Mo(3)#2	104.2(2)
O(8)-Mo(4)-O(13)	158.6(2)	Mo(4)-O(6)-Mo(2)	118.1(3)
O(7)-Mo(4)-O(13)	97.0(2)	Mo(3)-O(9)-Mo(4)	118.3(3)
O(6)-Mo(4)-O(13)	73.4(2)	Mo(1)-O(12)-Mo(3)	108.9(3)
O(9)-Mo(4)-O(13)	72.7(2)	Mo(1)-O(12)-Mo(2)#2	110.6(2)
O(2)#2-Mo(4)-O(13)	69.29(18)	Mo(3)-O(12)-Mo(2)#2	104.5(2)
C(22)-N(1)-C(18)	119.2(8)	Mo(1)-O(13)-Mo(3)	92.0(2)
C(22)-N(1)-Cu(5)	127.1(6)	Mo(1)-O(13)-Mo(2)	91.55(19)
C(18)-N(1)-Cu(5)	113.5(5)	Mo(3)-O(13)-Mo(2)	162.6(3)
C(17)-N(2)-C(15)	107.0(7)	Mo(1)-O(13)-Mo(1)#2	104.5(2)
C(17)-N(2)-Cu(5)	115.2(6)	Mo(3)-O(13)-Mo(1)#2	97.83(19)
C(15)-N(2)-Cu(5)	137.8(6)	Mo(2)-O(13)-Mo(1)#2	97.8(2)
C(17)-N(3)-C(16)	106.8(7)	Mo(1)-O(13)-Mo(4)	164.7(3)
C(17)-N(3)-C(23)	128.2(8)	Mo(3)-O(13)-Mo(4)	86.95(17)
C(16)-N(3)-C(23)	124.9(8)	Mo(2)-O(13)-Mo(4)	85.18(17)
Mo(1)-O(2)-Mo(4)#2	118.9(3)	Mo(1)#2-O(13)-Mo(4)	90.78(17)
Mo(1)-O(3)-Mo(2)	109.0(3)		

\*\*\*\*\*End\*\*\*\*\*