

## Electronic Supporting Information

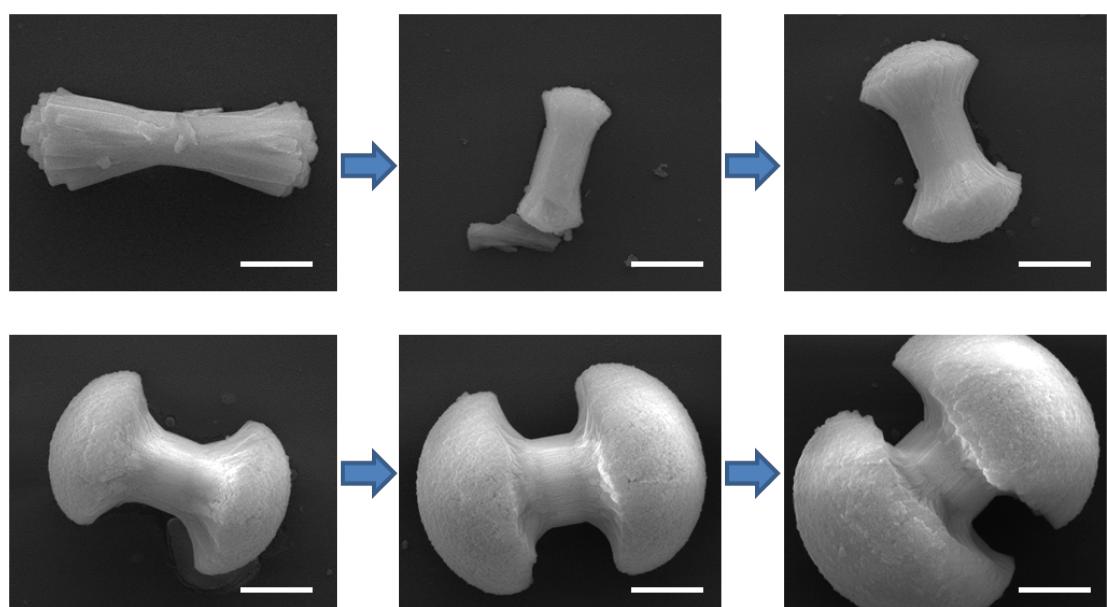
# Structural, Luminescent and Magnetic Properties of Yb<sup>3+</sup>-Er<sup>3+</sup> Doped Gd<sub>2</sub>O<sub>3</sub> Hierarchical Architectures

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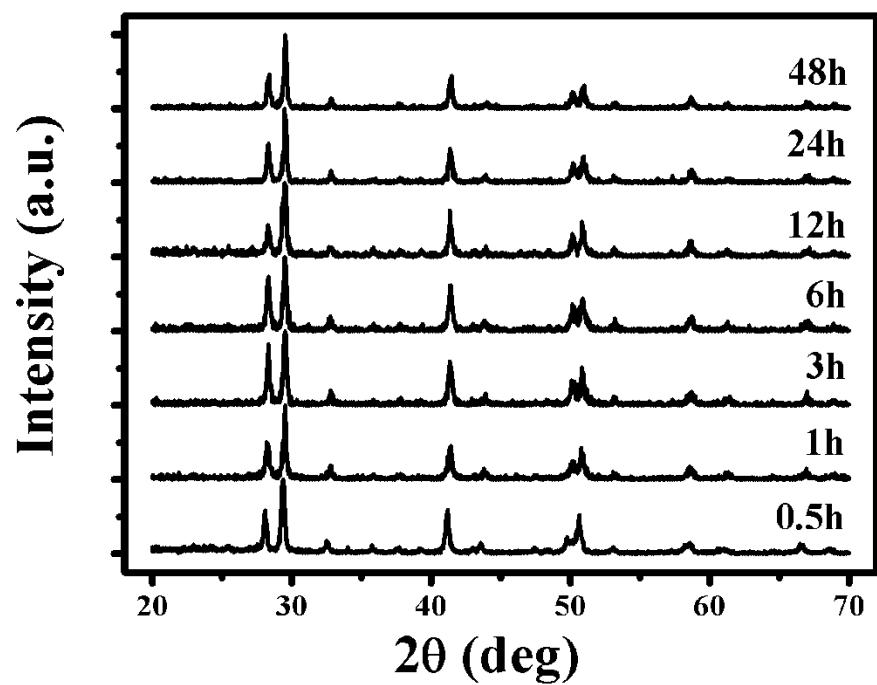
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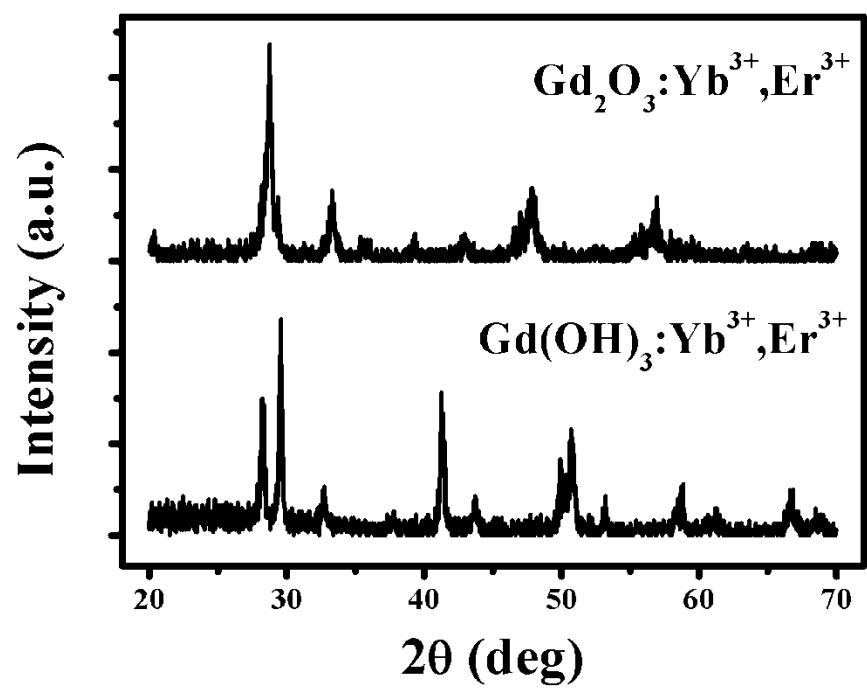
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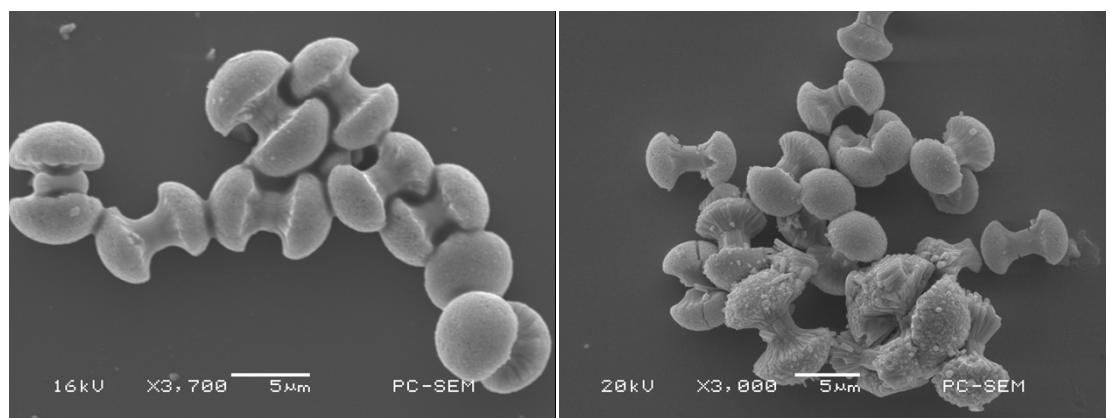
**Fig.S1** SEM images (performed on SEM #1) of  $\text{Gd}(\text{OH})_3$  micro-crystals prepared in 12 h (bar = 2  $\mu\text{m}$ ). The evolutionary morphology of particles can be found in the same sample, which indicates the growth process of dumbbell-shape  $\text{Gd}(\text{OH})_3$  microcrystals should be time-dependent.



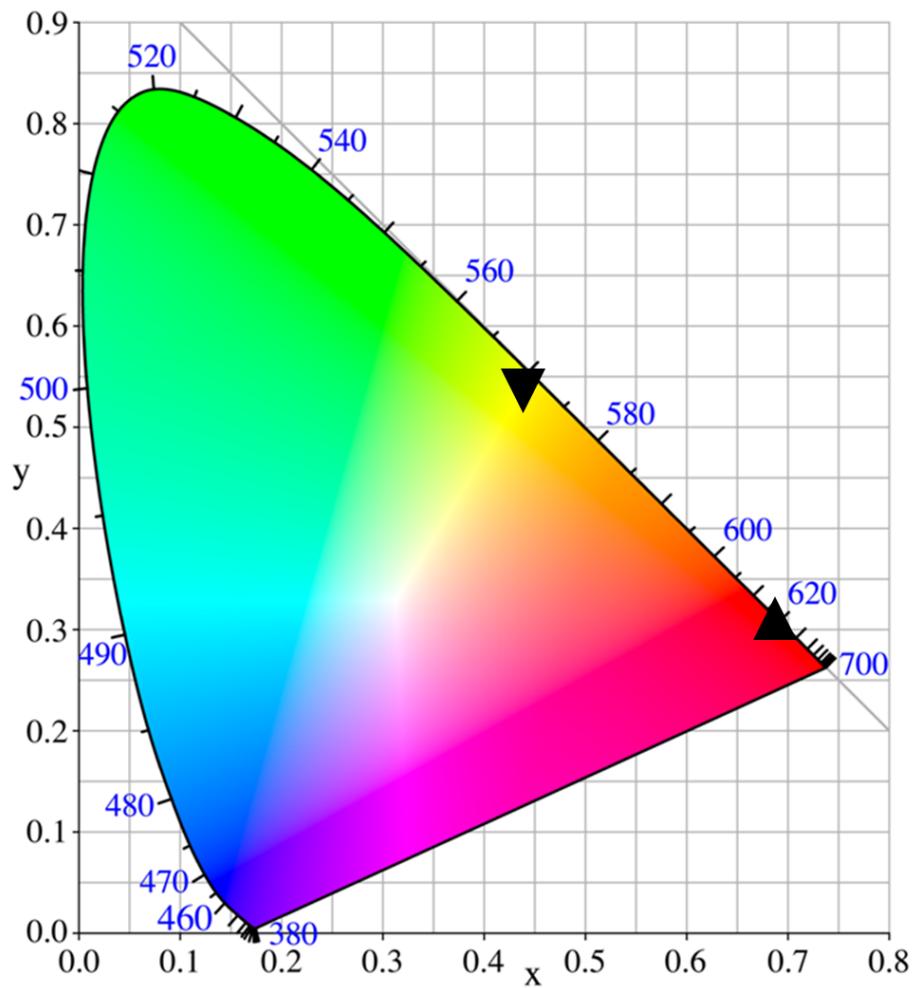
**Fig.S2** XRD patterns of  $\text{Gd}(\text{OH})_3$  at different reaction times. All samples are showing hexagonal phase.



**Fig.S3** XRD patterns of  $\text{Yb}^{3+}$ ,  $\text{Er}^{3+}$  co-doped  $\text{Gd}(\text{OH})_3$  and  $\text{Gd}_2\text{O}_3$ . The lanthanide doping precursor has the same hexagonal structure with the non-doped precursor. And



**Fig.S4** SEM images of  $\text{Yb}^{3+}/\text{Er}^{3+}$  co-doped  $\text{Gd}(\text{OH})_3$  and  $\text{Gd}_2\text{O}_3$ .



**Fig.S5** CIE chromaticity coordinate diagram in UC and QS processes. The upward triangle and the downward triangle represent pure red UC emission and yellow QS emission, respectively.

**Table S1** Particle sizes and corresponding magnetic mass susceptibilities of reported Gd<sup>3+</sup>-based complexes.

	Size (nm)	Magnetic mass susceptibility (emu g <sup>-1</sup> Oe <sup>-1</sup> )	Ref
KGdF <sub>4</sub> :Yb,Er	11.6	5.27×10 <sup>-5</sup>	1
KGdF <sub>4</sub>	49	1.03×10 <sup>-4</sup>	2
KGdF <sub>4</sub>	25	9.85×10 <sup>-5</sup>	3
BaGdF <sub>5</sub>	37	7.0×10 <sup>-5</sup>	4
BaGdF <sub>5</sub>	18.2	6.91×10 <sup>-5</sup>	5
NaGdF <sub>4</sub> :Yb,Er	845	9.82×10 <sup>-5</sup>	6
NaYb <sub>0.15</sub> Gd <sub>0.85</sub> F <sub>4</sub>	10	7.74×10 <sup>-5</sup>	7
Gd <sub>2</sub> O <sub>3</sub>	250(D)/2000(L)	2.8×10 <sup>-5</sup>	8
Gd <sub>2</sub> O <sub>3</sub>	7000(D)/9000(L)	1.132×10 <sup>-4</sup>	this work

- [1] P.K. Chu et al., Magnetic and upconverted luminescent properties of multifunctional lanthanide doped cubic KGdF<sub>4</sub> nanocrystals. *Nanoscale*. 2010, 2, 2805-2810
- [2] J. Lin et al., Room temperature synthesis of hydrophilic Ln<sup>3+</sup>-doped KGdF<sub>4</sub> nanoparticles with controllable size: energy-transfer, size-dependent and color-tunable luminescence properties. *Nanoscale*. 2012, 4, 3450-3459
- [3] X.Y. Chen et al., Amine-functionalized lanthanide-doped KGdF<sub>4</sub> nanocrystals as potential optical/magnetic multimodal bioprobes. *J. Am. Chem. Soc.* 2012, 134, 1323-1330
- [4] Y.H. Huang et al., Crystal structure and up- and down-conversion properties of Yb<sup>3+</sup>,Ho<sup>3+</sup> codoped BaGdF<sub>5</sub> solid solution with different morphologies. *CrystEngComm*. 2012, 14, 3131-3141
- [5] Q. Yang et al., Upconversion luminescence and magnetic properties of ligand-free monodisperse lanthanide doped BaGdF<sub>5</sub> nanocrystals. *J. Lumin.* 2011, 131, 2544-2549
- [6] J.H. Hao et al., Down- and up-conversion photoluminescence, cathodoluminescence and paramagnetic properties of NaGdF<sub>4</sub>:Yb<sup>3+</sup>,Er<sup>3+</sup> submicron disks assembled from primary nanocrystals. *J. Mater. Chem.* 2010, 20, 3178-3185
- [7] Q. Yang et al., Modifying crystal phase, shape, size, optical and magnetic properties of monodispersed multifunctional NaYbF<sub>4</sub> nanocrystals through lanthanide doping. *CrystEngComm*. 2011, 13, 4276
- [8] C. Tan et al., Pure red upconversion photoluminescence and paramagnetic properties of Gd<sub>2</sub>O<sub>3</sub>:Yb<sup>3+</sup>,Er<sup>3+</sup> nanotubes prepared via a facile hydrothermal process. *Mater. Lett.* 2012, 73, 147