Supporting Information

Caption of Figures

Fig. S1 Different non-covalent interactions in molecular structure of salt1. Color code: C, grey; O, red; N, blue.

Fig. S2 Six membered cyclic cavity of H_3BTC in 1 with included two molecules of $Pz^{Me2}H$. Color code: C, grey; H, white; O, red; N, blue.

Fig. S3 Cavities align to form channels in which Pz^{Me2}H molecules in salt 1. Color code: C, grey; H, white; O, red; N, blue. Hydrogen atoms have been omitted for clarity

Fig. S4 ORTEP drawing with 50% probability of salt 2.

Fig. S5 Six membered hexagonal cavity of H_3BTC with included two molecules of $Pz^{iPr2}H$ in salt 2. Color code: C, grey; H, white; N, blue; O, red

Fig. S6 Different non-covalent interactions in molecular structure of salt **2**. Color code: C, grey; O, red; N, blue.

Fig. S7 ORTEP drawing with 50% probability of salt 3.

Fig. S8 Six membered hexagonal cavity in salt **3** formed by H₃BTC having Pz^{tBu,iPr}H molecules. Color code: C, grey; H, white; N, blue; O, red.

Fig. S9 Different non-covalent interactions in molecular structure of salt **3**. Color code: C, grey; O, red; N, blue.

Fig. S10 C-H••• π interactions in salt **3**.

Fig. S11 ORTEP drawing with 50% probability of co-crystal 4.

Fig. S12 Four membered cyclic cavity of H_3BTC in 4, with hydrogen bonded $Pz^{Ph,Me}H$ molecules outside the cavity.

Fig. S13 Different non-covalent interactions in molecular structure of salt 4. Color code: C, grey; O, red; N, blue.

Fig. S14 Three dimensional packing of co-crystal 4.

Fig. S15 ORTEP drawing with 50% probability of salt 5.

Fig. S16 Hexagonal cavity formed by H₃BTC in **5** having 3-cumenyl-5-methylpyrazole. Color code: C, grey; H, white; N, blue; O, red.

Fig. S17 Different non-covalent interactions in molecular structure of salt **5**. Color code: C, grey; O, red; N, blue.

Fig. S18 π ••• π interactions in salt **5**.

Fig. S19 ORTEP drawing with 50% probability of salt 6.

Fig. S20 Different non-covalent interactions in molecular structure of salt **6**. Color code: C, grey; O, red; N, blue.

Fig. S21 Hydrogen bonded discrete hexameric unit in 6.

Fig. S22 ORTEP drawing with 50% probability of salt 7.

Fig. S23 Six membered rectangular cavity of H_3BTC in salt 7, filled with two BPz H_2 and one methanol molecule.

Fig. S24 Different non-covalent interactions in molecular structure of salt 7. Color code: C, grey; O, red; N, blue.

Fig. S25 Thermogravimetric curves for 1-7.

Fig. S26 Representative powder XRD patterns of salt 1 and co-crystal 4. The left and right patterns in each case correspond to experimental and simulated, respectively.









Fig. S3

























Fig. S11









































Fig. S26

(B)

Table S1 Details of the thermal analysis

Compound	Temperature range (°C) and weight loss (%)	Thermal
No.		stability (°C)
1	 (i) No weight loss upto 187 °C. (ii) Ist weight loss (~25 %) upto 250 °C. (iii) IInd weight loss (~45 %) upto 318 °C. (iv) Last continuous step of weight loss (100 %) upto 550 °C. 	187 °C
2	 (i) No loss upto 110 °C. (ii) Ist weight loss (~5 %) upto 124 °C and stable upto 177 °C. (iii) IInd weight loss (~26 %) upto 230 °C. (iv) Last continuous step of weight loss (100 %) upto 555 °C. 	177 °C
3	 (i) No loss upto 88 °C. (ii) Ist weight loss (~4.8 %) upto 115 °C and stable upto 180 °C. (iii) IInd weight loss (~38 %) upto 240 °C. (iv) Last continuous step of weight loss (100 %) upto 530 °C. 	180 °C
4	 (i) No loss upto 150 °C. (ii) Ist weight loss (~5 %) upto 202 °C. 	

	(iii) II nd weight loss (~32 %) upto 268 °C.	202 °C
	(iv) Last continuous step of weight loss (100 %) upto	
	526 °C.	
	(i) No loss upto 218° C.	
5	(ii) I st weight loss (~26 %) upto 270 °C.	218 °C
	(iv) Last continuous step of weight loss (100 %) upto	
	530 C.	
6	(i) No loss upto 236°C.	
	(ii) Last continuous step of weight loss (100 °C) upto	236 °C
	520 °C.	
	(i) No loss upto 168 °C.	
7	(ii) I st weight loss (\sim 3.8 %) upto 230 °C.	230 °C
	(iii) Last continuous step of weight loss (100 %) upto	
	596 °C.	