

TABLE 1S: Vibrational frequencies computed at the ^aB3LYP/6-31G(*d*) and ^bMP2/6-31G(*d,p*) levels of theory for reactants, transition structures and products involved in the H- and Cl-transfer reactions: F₃C + CX_{4-n}H_n (*n* = 1, 2 or 3; X = Cl or F) and F₃C + CCl_{4-n}H_n (*n* = 1, 2 or 3). Experimental values are given in italics.

Species	Frequencies (cm ⁻¹)																					
^c CF ₃ H	487	487	681	1127	1160	1160	1404	1404	3084													
	479	479	667	1111	1158	1158	1392	1392	3084													
	<i>508</i>	<i>508</i>	<i>700</i>	<i>1137</i>	<i>1152</i>	<i>1152</i>	<i>1376</i>	<i>1376</i>	<i>3035</i>													
F ₃ C- [*] HCF ₃	1708 <i>i</i>	18	31	37	186	187	212	495	497	509	511	682	747	1038	1205	1232	1233	1241	1241	1497	1498	
	2116 <i>i</i>	21	29	29	184	184	215	476	476	491	491	658	722	1007	1188	1205	1205	1207	1207	1495	1495	
^c F ₃ C	486	486	676	1071	1256	1256																
	477	477	669	1068	1249	1249																
	<i>500</i>	<i>500</i>	<i>701</i>	<i>1090</i>	<i>1090</i>	<i>1259</i>																
^c CF ₂ H ₂	509	1108	1109	1172	1251	1473	1538	3001	3062													
	497	1096	1115	1150	1236	1460	1546	2999	3060													
	<i>529</i>	<i>1090</i>	<i>1116</i>	<i>1176</i>	<i>1262</i>	<i>1435</i>	<i>1435</i>	<i>2949</i>	<i>3013</i>													
F ₃ C- [*] HCF ₂ H	1721 <i>i</i>	15	30	57	184	189	406	491	498	529	710	1054	1142	1152	1190	1198	1226	1317	1480	1492	3023	
	2154 <i>i</i>	23	33	57	186	191	443	483	485	514	707	1055	1123	1139	1185	1198	1224	1312	1511	1546	3032	
^d HF ₂ C	526	1000	1163	1181	1347	3059																
	514	1041	1143	1171	1338	3070																
	-	<i>946</i>	<i>1164</i>	<i>1173</i>	<i>1317</i>	-																
^c CFH ₃	1072	1182	1182	1494	1494	1502	2979	3051	3051													
	1057	1166	1166	1477	1485	1485	2976	3075	3075													
	<i>1049</i>	<i>1182</i>	<i>1182</i>	<i>1464</i>	<i>1467</i>	<i>1467</i>	<i>2965</i>	<i>3006</i>	<i>3006</i>													
F ₃ C- [*] HCFH ₂	1766 <i>i</i>	18	42	82	186	377	467	494	551	710	1073	1128	1184	1185	1214	1220	1433	1441	1497	3008	3102	
	2193 <i>i</i>	29	52	88	185	415	461	484	552	705	1073	1104	1169	1185	1203	1222	1451	1470	1534	3009	3116	
^d H ₂ FC	602	1162	1182	1478	3083	3225																
	733	1151	1156	1473	3078	3228																
	-	<i>1163</i>	<i>1170</i>	-	-	-																
^c CCl ₃ H	256	256	360	655	723	723	1240	1240	3142													
	263	263	367	668	783	783	1245	1245	3094													
	<i>261</i>	<i>261</i>	<i>363</i>	<i>680</i>	<i>774</i>	<i>774</i>	<i>1220</i>	<i>1220</i>	<i>3034</i>													
F ₃ C- [*] HCCl ₃	1650 <i>i</i>	5	47	49	156	156	172	266	266	397	493	495	630	749	756	756	1108	1223	1223	1435	1435	
	1973 <i>i</i>	10	54	54	157	157	176	273	273	407	482	482	643	752	817	818	1115	1211	1212	1451	1453	
F ₃ C- [*] ClCCl ₂ H	407 <i>i</i>	7	45	55	139	165	197	229	302	385	505	507	658	697	783	998	1063	1246	1278	1279	3152	
^d HCl ₂ C	299	470	741	861	1243	3185																
	305	539	761	899	1258	3162																
	-	-	-	<i>902</i>	<i>1226</i>	-																
^c CCl ₂ H ₂	292	694	702	895	1138	1285	1474	3083	3148													
	299	719	763	899	1149	1291	1457	3051	3123													
	<i>282</i>	<i>717</i>	<i>758</i>	<i>898</i>	<i>1153</i>	<i>1268</i>	<i>1467</i>	<i>2999</i>	<i>3040</i>													
F ₃ C- [*] HCCl ₂ H	1700 <i>i</i>	5	47	62	160	164	296	394	493	495	686	735	758	977	1111	1213	1217	1231	1441	1454	3109	
	2048 <i>i</i>	4	38	55	160	164	302	402	482	486	692	748	816	985	1118	1202	1215	1236	1450	1452	3077	
F ₃ C- [*] ClCClH ₂	431 <i>i</i>	8	51	76	153	218	218	306	506	508	674	713	756	999	1069	1101	1267	1269	1428	3101	3213	
	729 <i>i</i>	11	52	80	171	245	247	335	498	499	669	743	782	983	1081	1138	1239	1241	1426	3063	3175	
^d H ₂ ClC	136	820	993	1411	3140	3285																
	335	841	1005	1429	3125	3271																
	<i>402</i>	<i>827</i>	-	<i>1356</i>	-	-																
^c CClH ₃	707	1025	1025	1386	1481	1481	3036	3134	3134													
	747	1027	1027	1389	1466	1466	3010	3121	3121													
	<i>731</i>	<i>1017</i>	<i>1017</i>	<i>1355</i>	<i>1455</i>	<i>1455</i>	<i>2968</i>	<i>3054</i>	<i>3054</i>													
F ₃ C- [*] HCClH ₂	1715 <i>i</i>	6	50	88	166	371	469	494	550	709	755	1047	1059	1131	1208	1214	1421	1440	1449	3061	3160	
	2080 <i>i</i>	10	43	79	166	383	460	483	549	702	785	1048	1062	1139	1202	1207	1418	1435	1452	3039	3142	
F ₃ C- [*] ClCH ₃	445 <i>i</i>	9	76	78	233	234	249	509	511	679	721	722	1000	1108	1258	1258	1434	1434	3065	3213	3214	
	727 <i>i</i>	15	81	81	250	250	283	499	499	670	777	777	983	1169	1232	1232	1431	1431	3032	3188	3188	
^c H ₃ C	449	1404	1404	3078	3249	3249																
	374	1415	1415	3076	3263	3263																
	<i>580</i>	<i>1383</i>	<i>1383</i>	<i>3002</i>	<i>3184</i>	<i>3184</i>																

^aUpper values correspond to the optimised frequencies at B3LYP/6-31G(*d*) level of theory. ^bLower values correspond to the optimised frequencies at MP2/6-31G(*d,p*) level of theory. ^cRef. 39. ^dRef. 41.

TABLE 2S: Vibrational frequencies computed at the B3LYP/6-31G(*d*) level of theory for reactants, transition structures and products involved in the Cl-transfer reactions: $F_3C + CCl_{4-n}F_n$ ($n = 0, 1$ or 2). Experimental values are given in italics.

Species	Frequencies (cm ⁻¹)																					
^a CCl ₂ F ₂	259	307	409	419	431	641	846	1115	1214													
	<i>261</i>	<i>322</i>	<i>436</i>	<i>436</i>	<i>457</i>	<i>667</i>	<i>922</i>	<i>1099</i>	<i>1162</i>													
F ₃ C- [*] CICClF ₂	393 <i>i</i>	14	46	53	145	179	195	239	258	376	415	507	509	574	665	743	1007	1133	1236	1279	1281	
^b F ₂ CIC	349	398	578	734	1149	1236																
	-	-	<i>599</i>	<i>761</i>	<i>1148</i>	-																
^a CCl ₃ F	239	239	340	385	385	511	793	793	1117													
	<i>241</i>	<i>241</i>	<i>349.5</i>	<i>398</i>	<i>398</i>	<i>535</i>	<i>847</i>	<i>847</i>	<i>1085</i>													
F ₃ C- [*] CICCl ₂ F	388 <i>i</i>	9	42	48	142	162	179	224	231	308	379	458	505	507	585	696	827	1007	1153	1284	1285	
^b FCl ₂ C	275	368	451	583	865	1177																
	-	-	-	<i>747</i>	<i>919</i>	<i>1143</i>																
^a CCl ₄	214	214	309	309	309	442	730	730	730													
	<i>218</i>	<i>218</i>	<i>314</i>	<i>314</i>	<i>314</i>	<i>458</i>	<i>776</i>	<i>776</i>	<i>776</i>													
F ₃ C- [*] CICCl ₃	370 <i>i</i>	1	41	43	138	154	156	209	210	291	291	379	503	505	518	685	770	772	1008	1288	1289	
^a Cl ₃ C	269	270	326	472	857	858																
	<i>240</i>	<i>240</i>	<i>450</i>	<i>460</i>	<i>898</i>	<i>898</i>																
^a CCIF ₃	340	340	457	551	551	773	1099	1264	1264													
	<i>350</i>	<i>350</i>	<i>474</i>	<i>560</i>	<i>560</i>	<i>782</i>	<i>1106</i>	<i>1217</i>	<i>1217</i>													

^aRef. 40. ^bRef.41.

TABLE 3S: Partition functions ($Q_T(T)$ translational, $Q_R(T)$ rotational, $Q_V(T)$ vibrational, $Q_E(T)$ electronic, $Q_{V'}(T)$ vibrational corrected for hindered rotation, and $Q_{HR}(T)$ hindered rotor, $T = 298.15$ K), hindered rotational barriers (V_o , kcal mol⁻¹ $T = 298.15$ K) and imaginary frequencies ($\tilde{\nu}$ cm⁻¹), optimised at the ^aB3LYP/6-31G(d) and ^bMP2/6-31G(d,p) levels of theory, for the species involved in H- and Cl-transfer the reactions: F₃C + CX_{4-n}H_n ($n = 1, 2$ or 3 ; X = F or Cl) and F₃C + CCl_{4-n}F_n ($n = 0, 1$ or 2).

Species	$Q_T(T), 10^{-7}$	$Q_R(T), 10^{-3}$	$Q_V(T)$	$Q_E(T)$	$Q_{V'}(T)$	$Q_{HR}(T)$	V_o	$\tilde{\nu}$
F ₃ C ^a	2.25	11.50	1.28	2.00				
	2.25	11.50	1.30	2.00				
CF ₃ H	2.30	11.90	1.29	1.00				
	2.30	12.00	1.30	1.00				
F ₃ C-*HCF ₃	6.43	512.00		2.00	335.00	41.43	2.38	1708i
	6.43	502.00		2.00	436.00	53.00	3.50	2116i
CF ₂ H ₂	1.47	6.29	1.11	1.00				
	1.47	6.32	1.12	1.00				
F ₃ C-*HCF ₂ H	5.23	380.00		2.00	142.00	36.68	1.39	1721i
	5.23	354.00		2.00	126.00	48.00	3.11	2154i
CFH ₃	0.78	0.90	1.02	1.00				
	0.78	0.90	1.02	1.00				
F ₃ C-*HCFH ₂	4.10	249.00		2.00	46.50	23.54	0.81	1766i
	4.10	223.00		2.00	35.70	32.60	2.15	2193i
CCl ₃ H	5.03	70.20	2.69	1.00				
	5.03	67.60	2.55	1.00				
F ₃ C-*HCCl ₃	10.00	1110.00		2.00	462.00	51.80	0.26	1650i
	10.00	1070.00		2.00	351.00	51.70	1.22	1973i
F ₃ C-*ClCCl ₂ H	10.00	1360.00		4.03	415.00	47.16	0.52	407i
CCl ₂ H ₂	3.02	24.90	1.45	1.00				
	3.02	23.90	1.44	1.00				
F ₃ C-*HCCl ₂ H	7.43	736.00		2.00	137.00	47.76	0.25	1700i
	7.43	708.00		2.00	178.00	47.60	0.25	2048i
F ₃ C-*ClCClH ₂	7.43	706.00		4.03	119.00	29.74	1.25	431i
	7.43	661.00		4.03	89.70	29.90	0.49	729i
CClH ₃	1.39	1.77	1.05	1.00				
	1.39	1.71	1.05	1.00				
F ₃ C-*HCClH ₂	5.10	366.00		2.00	41.80	30.67	0.15	1715i
	5.10	355.00		2.00	52.80	30.50	0.09	2080i
F ₃ C-*ClCH ₃	5.10	275.00		4.03	437.00	11.32	0.05	445i
	5.10	263.00		4.03	363.00	11.20	0.13	727i
CCl ₂ F ₂	5.16	91.10	2.96	1.00				
F ₃ C-*ClCClF ₂	10.20	1250.00		4.03	495.00	45.84	1.90	393i
CCl ₃ F	6.22	93.00	4.26	1.00				
F ₃ C-*ClCCl ₂ F	11.50	1610.00		4.03	899.00	48.92	0.79	388i
CCl ₄	7.35	33.00	6.40	1.00				
F ₃ C-*ClCCl ₃	12.90	2000.00		4.03	1620.00	52.13	0.02	370i

^aUpper values correspond to those optimised at the B3LYP/6-31G(d) level of theory. ^bLower values correspond to those optimised at MP2/6-31G(d,p) level of theory.

TABLE 4S: Electronic charge distribution on reactants and transition structures, polarity variation, Δp_{C-X} (X = H or Cl), L parameter and reaction enthalpies, $\Delta_r H_{298.15\text{ K}}$ (kcal mol⁻¹) for the reactions: $F_3C + CX_{4-n}H_n$ ($n = 1, 2$ or 3 ; X = F or Cl) and $F_3C + CCl_{4-n}F_n$ ($n = 0, 1$ or 2).

Species		NPA		L	$\Delta_r H_{298.15\text{ K}}$		
		q_C	q_H		q_{Cl}	B3LYP/6-31G(d)	MP2/6-31G(d,p)
CF ₃ H		0.97	0.16				
F ₃ C- [*] HCF ₃		1.02	0.00				
Δp_{C-H}	0.21			1.000	0.00	0.00	0.00
CF ₂ H ₂		0.45	0.16				
F ₃ C- [*] HCF ₂ H		0.52	0.10				
Δp_{C-H}	0.13			0.755	-4.73	-5.04	-3.59
CFH ₃		-0.17	0.18				
F ₃ C- [*] HCFH ₂		-0.28	0.11				
Δp_{C-H}	0.03			0.742	-5.07	-5.18	-3.65
CCl ₃ H		-0.34	0.28	0.02			
F ₃ C- [*] HCCl ₃		-0.01	0.06	0.04			
Δp_{C-H}	0.55			0.595	-11.45	-11.44	-10.54
F ₃ C- [*] CICCl ₂ H		-0.03	0.14	-0.09			
Δp_{C-Cl}	0.42			0.621	-14.45	-	-8.44
CCl ₂ H ₂		-0.46	0.26	-0.03			
F ₃ C- [*] HCCl ₂ H		-0.02	0.05	0.01			
Δp_{C-H}	0.66			0.682	-8.46	-9.21	-7.27
F ₃ C- [*] CICClH ₂		-0.01	0.12	-0.11			
Δp_{C-Cl}	0.54			0.816	-9.17	-	-6.07
CClH ₃		-0.65	0.24	-0.08			
F ₃ C- [*] HCClH ₂		-0.03	0.04	-0.02			
Δp_{C-H}	0.82			0.819	-5.90	-6.61	-5.90
F ₃ C- [*] CICH ₃		0.01	0.23	-0.11			
Δp_{C-Cl}	0.69			1.088	-3.24	-	-2.38
CCl ₂ F ₂		0.67		0.00			
F ₃ C- [*] CICClF ₂		0.51		-0.10			
Δp_{C-Cl}	0.07			0.846	-6.73	-	-3.60
CCl ₃ F		0.22		0.04			
F ₃ C- [*] CICCl ₂ F		0.24		-0.08			
Δp_{C-Cl}	0.14			0.594	-13.49	-	-13.00
CCl ₄		-0.26		0.07			
F ₃ C- [*] CICCl ₃		-0.05		-0.05			
Δp_{C-Cl}	0.32			0.477	-21.16	-	-15.30

^aRef. 39.