

SIT₁: Details of the parameters of Eq. 2 for various relaxation processes shown in SIF₂ (in CHXOL + CHPOL, $x_m = 0.25$).

Processes	Temp (K)	α_{HN}	β_{HN}	f_0 (Hz)	f_m (Hz)	$\Delta\epsilon$
α - process	152.8	0.066	0.733	1.28×10^{-2}	1.68×10^{-2}	19.71
	165.1	0.047	0.689	1.77×10^0	2.42×10^0	19.85
	180.8	0.074	0.718	1.02×10^2	1.39×10^2	20.23
	199.9	0.032	0.711	2.27×10^3	3.02×10^3	19.21
	224.8	0.021	0.737	4.93×10^4	6.31×10^4	18.35
	240.2	0.008	0.714	2.21×10^5	2.89×10^5	18.03
	260.5	0.000	0.714	1.33×10^6	1.73×10^6	17.19
β - process	152.8	0.555	1.00	3.31×10^1	3.31×10^1	0.582
	158.4	0.507	1.00	9.55×10^1	9.55×10^1	0.501
γ - process	100.7	0.840	1.00	1.39×10^1	1.39×10^1	0.2499
	105.9	0.828	1.00	1.27×10^2	1.27×10^2	0.2510
	110.6	0.817	1.00	5.68×10^2	5.68×10^2	0.2562
	115.0	0.797	1.00	2.65×10^3	2.65×10^3	0.2516
	120.3	0.764	1.00	1.59×10^4	1.59×10^4	0.2342
	125.2	0.752	1.00	5.18×10^4	5.18×10^4	0.2434
	130.5	0.728	1.00	1.35×10^5	1.35×10^5	0.2977

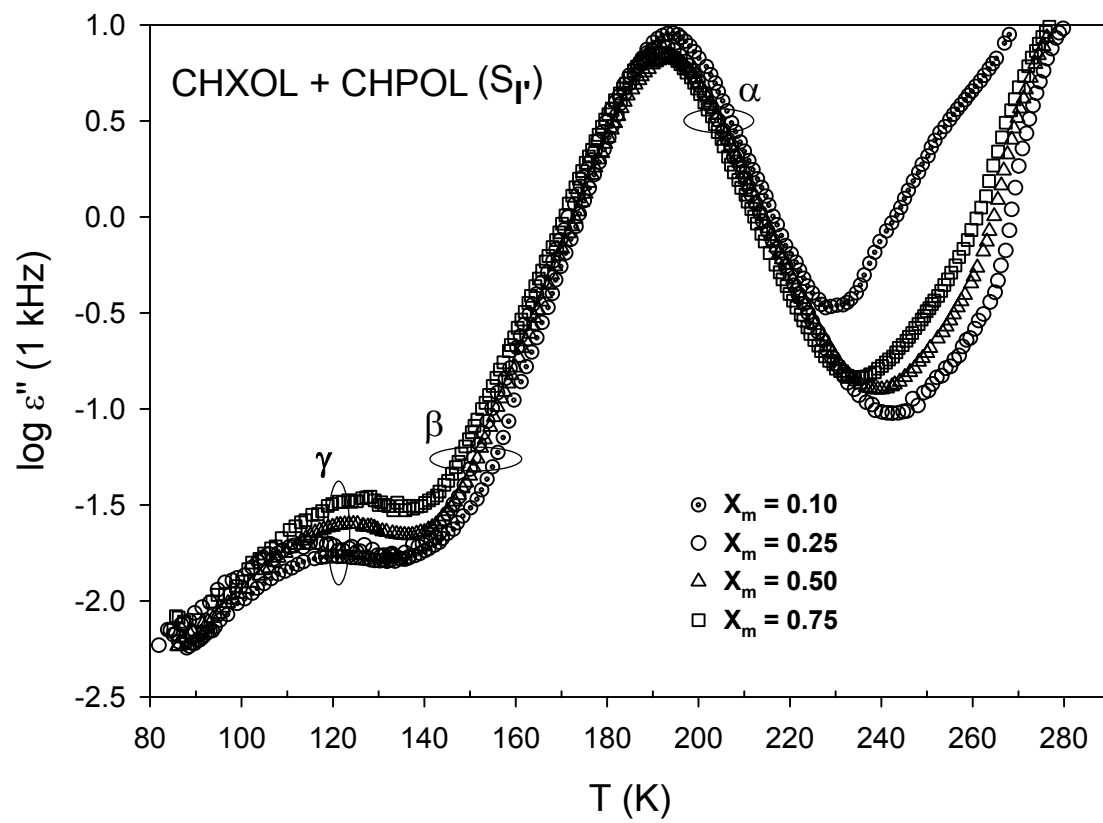
SIT₂: Details of the parameters of Eq. 2 for the fits shown in SIF₆ (in NPOL + NPGOL, $x_m = 0.13$).

Processes	Temp (K)	α_{HN}	β_{HN}	f_0 (Hz)	f_m (Hz)	$\Delta\epsilon$
α - process	156.5	0.152	0.830	4.81×10^{-2}	5.80×10^{-1}	6.32
	168.8	0.163	0.855	2.50×10^2	2.92×10^0	8.33
	187.7	0.132	0.799	9.10×10^1	1.12×10^2	10.47
	200.4	0.104	0.783	3.18×10^3	3.99×10^3	10.50
	224.9	0.049	0.714	9.38×10^4	1.24×10^5	10.05
	235.4	0.021	0.604	2.92×10^5	4.39×10^5	10.27
β - process	156.5	0.631	1.00	2.65×10^1	2.65×10^1	0.220
γ - process	105.4	0.658	1.00	1.59×10^4	1.59×10^4	0.102
	110.5	0.621	1.00	5.20×10^4	5.19×10^4	0.101
	115.9	0.586	1.00	1.54×10^5	1.54×10^5	0.099
	120.2	0.563	1.00	3.19×10^5	3.18×10^5	0.099
	125.8	0.577	1.00	1.41×10^6	1.40×10^6	0.116

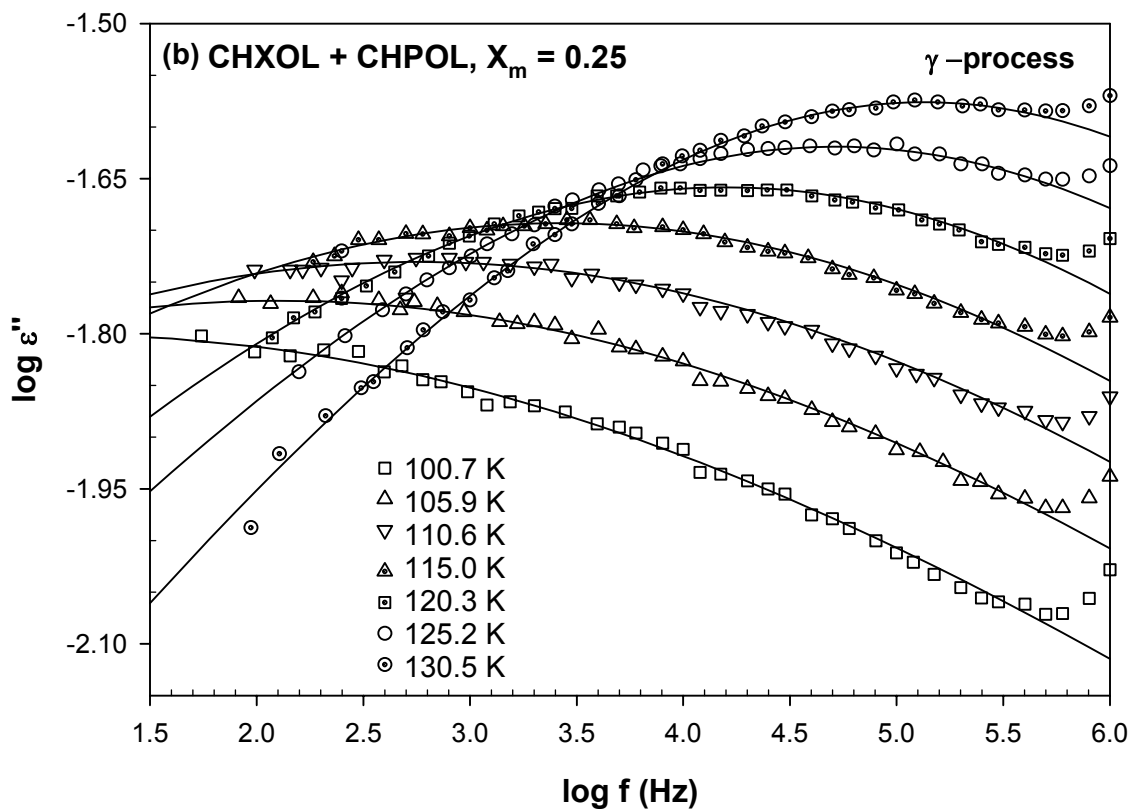
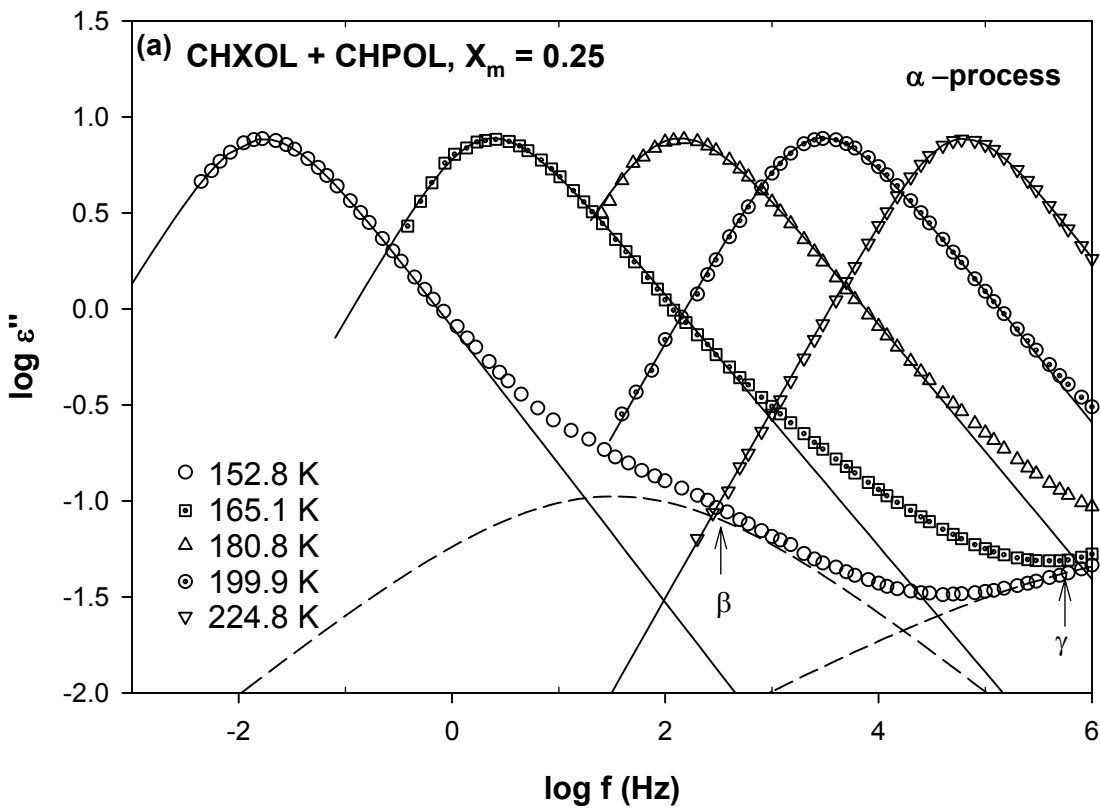
SIT₃. Details of parameters of Eq. 2 for the fits shown in SIF₇ (in CNCH + CHC, X_m = 0.125).

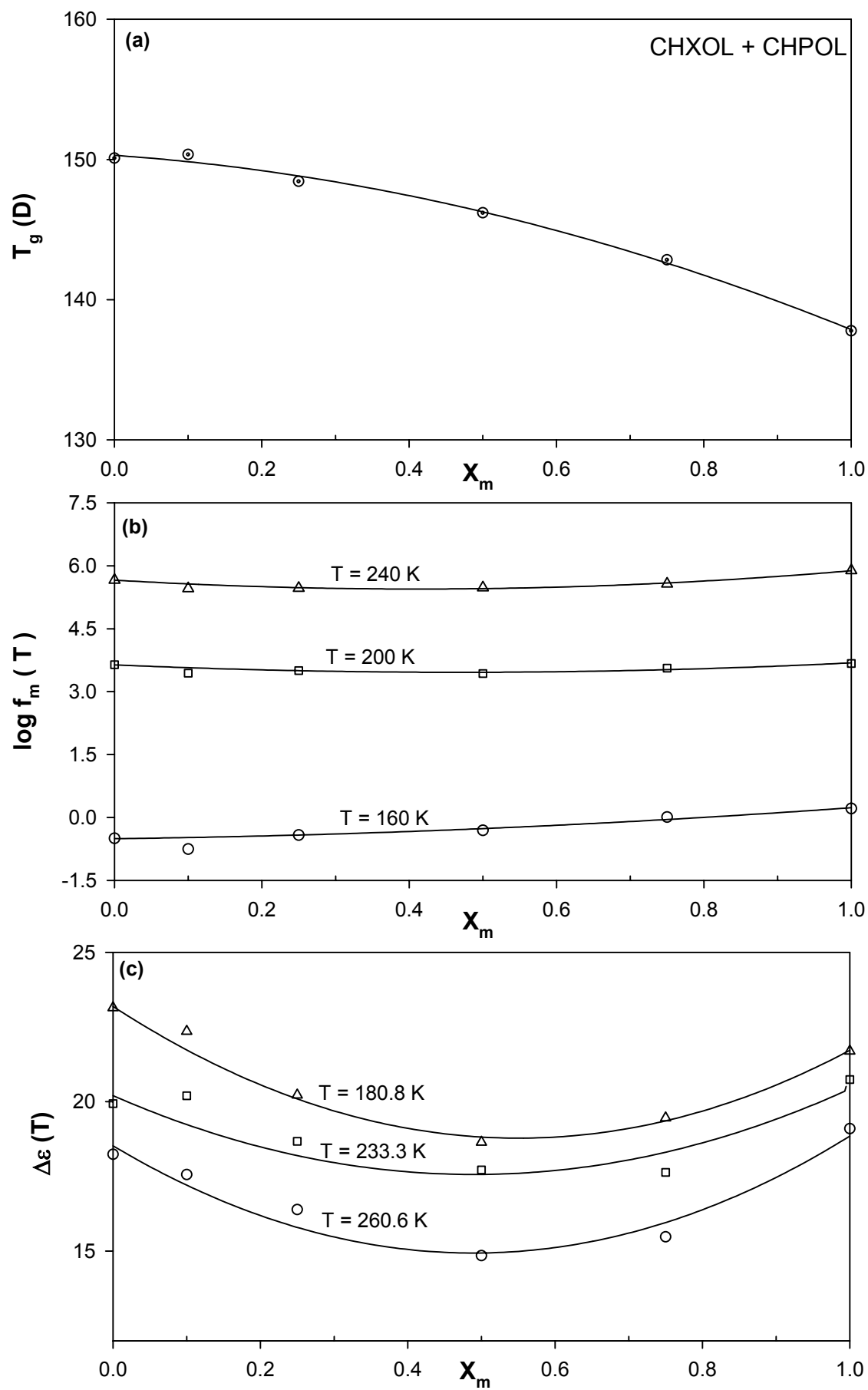
Process	Temp (K)	α_{HN}	β_{HN}	f_0 (Hz)	f_m (Hz)	$\Delta\epsilon$
α'	185.6	0.406	0.943	5.55×10^1	6.07×10^1	1.565
	190.1	0.405	1.000	1.99×10^2	1.99×10^2	1.498
	195.2	0.355	1.000	5.90×10^2	5.90×10^2	1.355
	200.8	0.322	1.000	1.77×10^3	1.77×10^3	1.239
α	135.4	0.180	0.657	2.25×10^{-2}	3.49×10^{-2}	11.03
	150.5	0.140	0.586	2.76×10^1	4.66×10^1	11.51
	165.4	0.146	0.599	3.98×10^3	6.62×10^3	11.96
	185.6	0.045	0.388	1.75×10^5	3.92×10^5	12.56
	195.2	0.011	0.299	5.54×10^5	1.49×10^6	13.82
β	100.8	0.505	0.232	1.93×10^1	3.24×10^2	0.284
	105.5	0.634	0.350	5.17×10^1	8.48×10^2	0.336
	110.1	0.680	0.421	2.31×10^2	3.26×10^3	0.389
	115.1	0.705	0.501	1.14×10^3	1.14×10^4	0.422
	120.4	0.735	0.738	1.59×10^4	4.99×10^4	0.440
	125.0	0.759	1.000	1.67×10^5	1.67×10^5	0.481

SIF₁

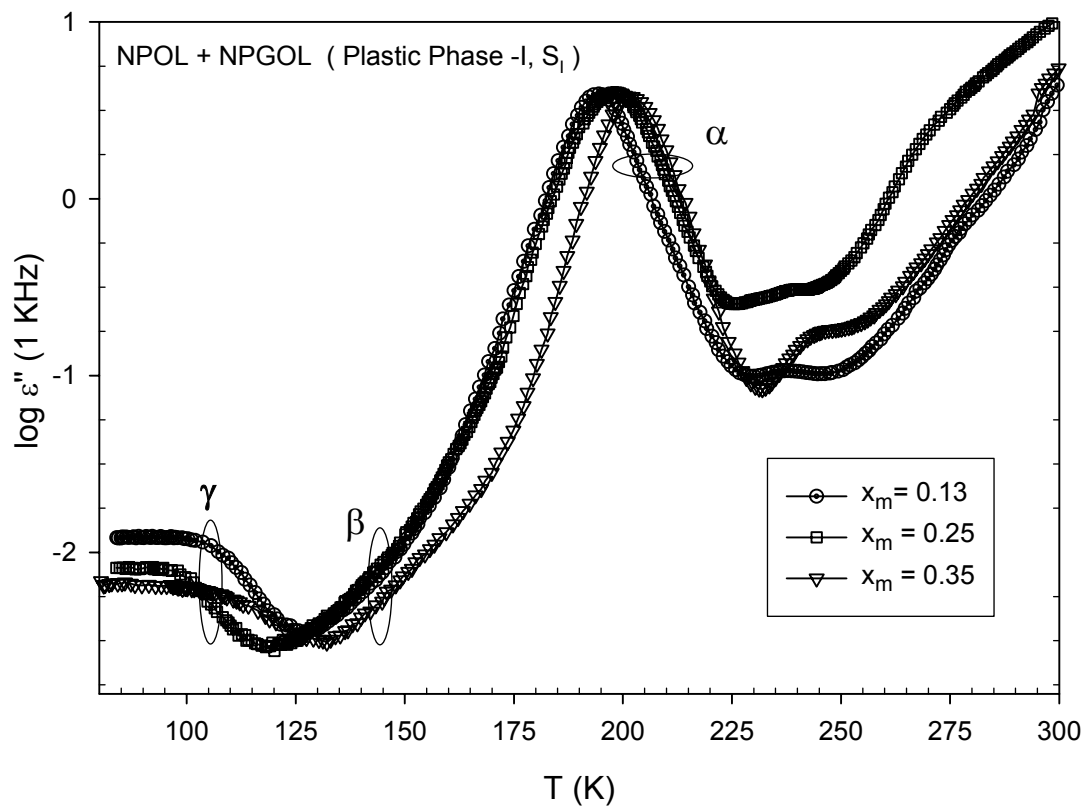


SIF₂

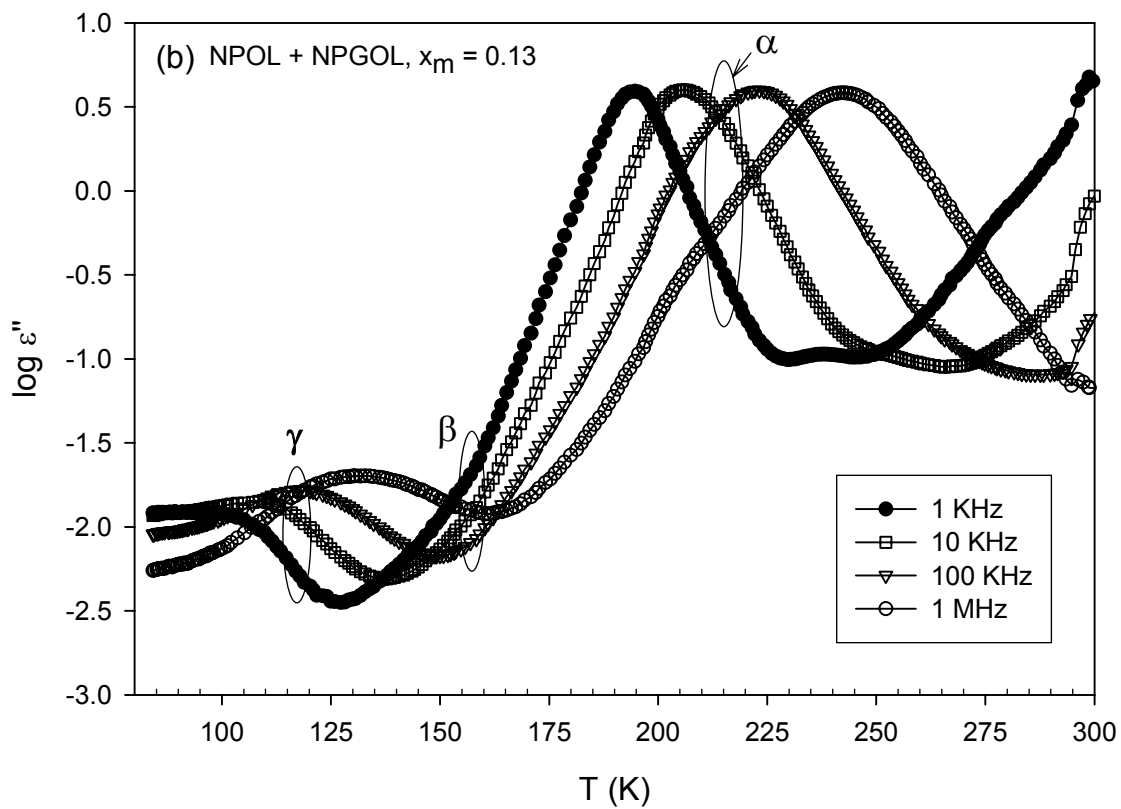
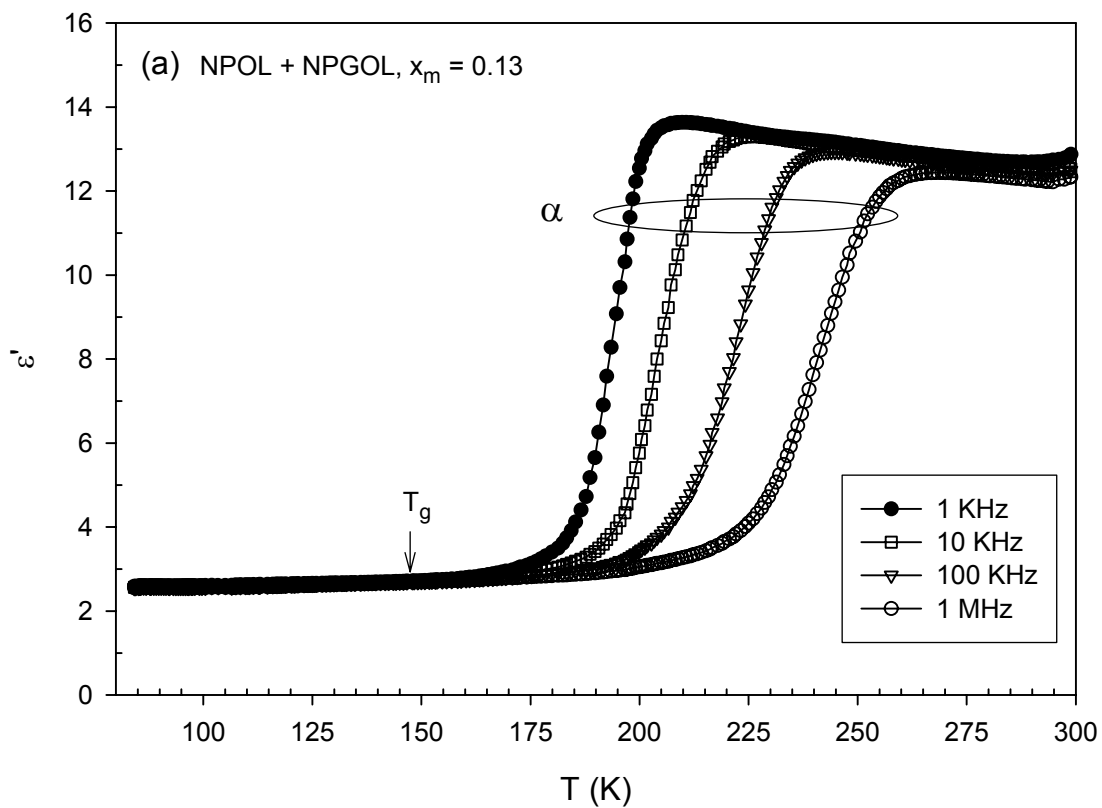




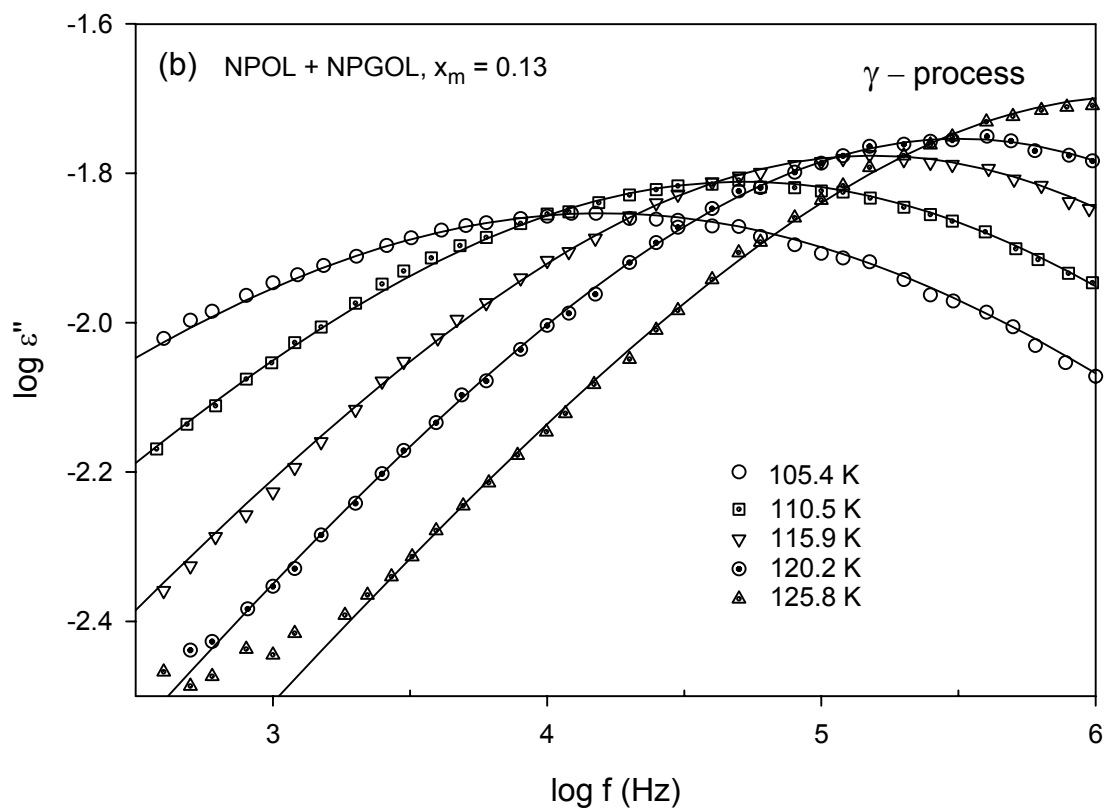
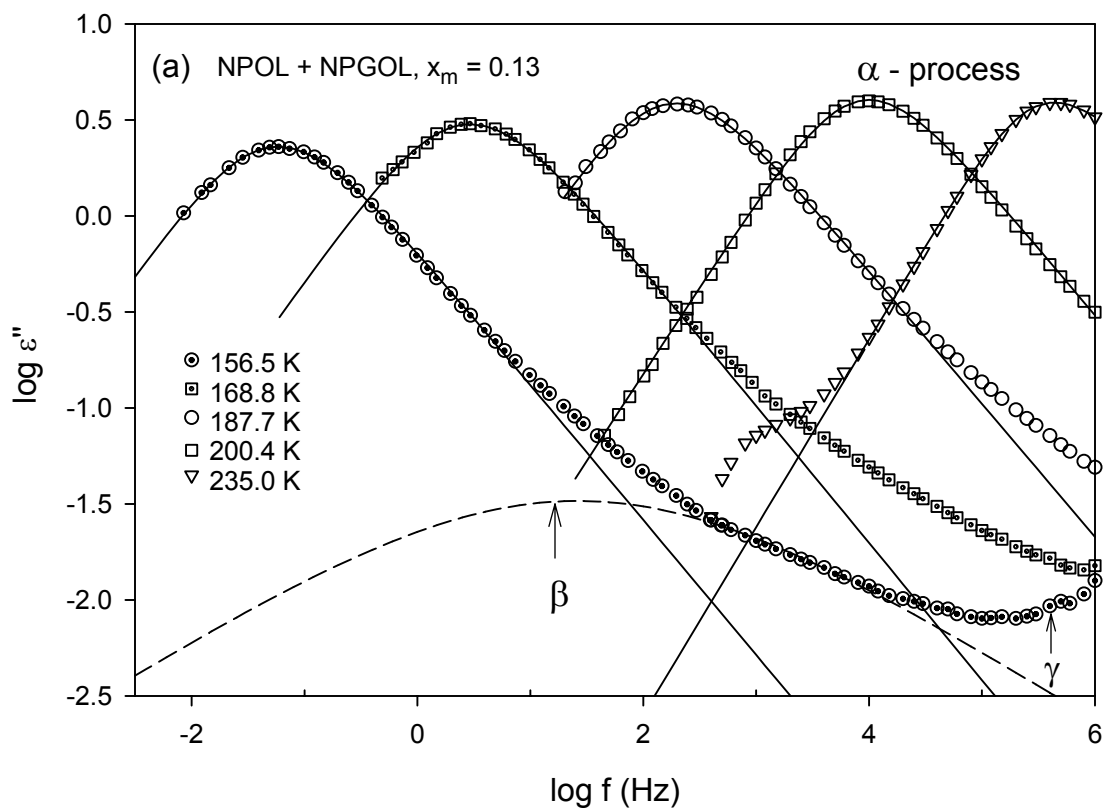
SIF₄



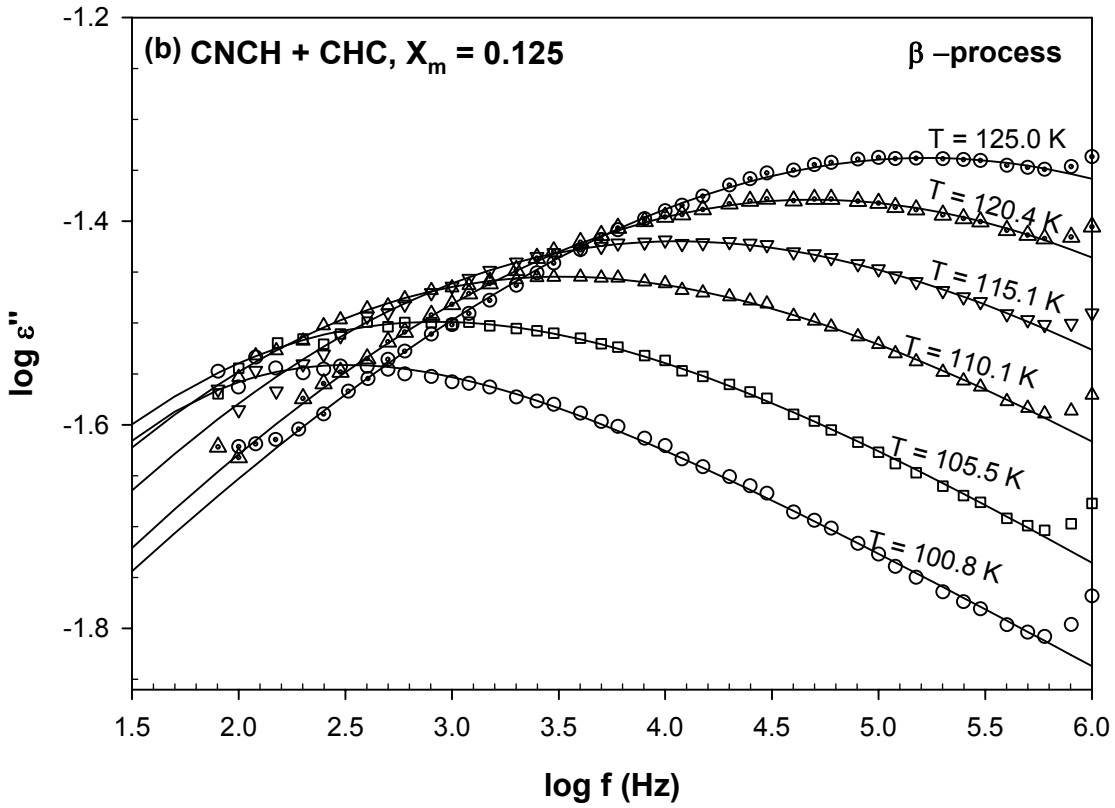
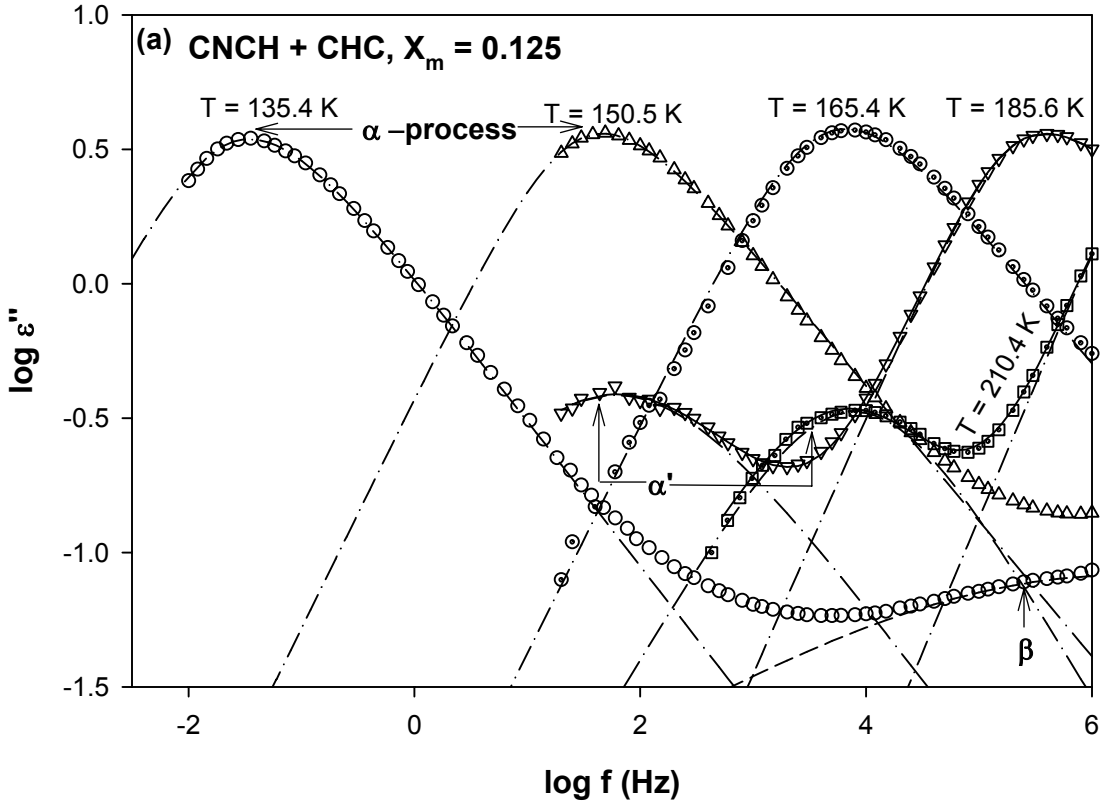
SIF₅



SIF₆



SIF₇



Supplementary Figure Captions

SIF₁. CHXOL + CHPOL binary system: Variation of $\log \varepsilon''$ with temperature at a test frequency of 1 kHz, for different concentrations of CHPOL. Note that the presence of a smaller process designated as β - process along with the α - and γ - processes. It may be noted that the α' -process present³⁸ in pure CHXOL is not resolvable in this binary system.

SIF₂. Double logarithmic plot of ε'' vs. frequency at different temperatures for various relaxation processes of CHXOL + CHPOL binary system with $x_m = 0.25$. (a) α -, & β - process (b) γ -process. The solid line corresponds to the HN- equation (2) for the α - process, dashed line corresponds to the CC-fit for the resolved β -process in panel (a) and solid line corresponds to the CC-fit for β -process in panel (b). The rise in the loss at frequencies above 10 kHz in (a) is due to the β -process. The parameters of eq. 2 for the α - & γ -processes are given in supplementary Table (SIT₁).

SIF₃. Variation of various physical parameters of CHXOL + CHPOL binary system with mole fraction (x_m) of second component i .e. CHPOL: (a) $T_g(D)$, [where $T_g(D)$ is the temperature at which the f_m value is 10^{-3} Hz]; (b) $\log f_m$ & (c) dielectric strength ($\Delta\varepsilon$) at three fixed temperatures. The thick lines are fits to eq. 5.

SIF₄ . NPOL + NPGOL binary system: Variation of $\log \varepsilon''$ with temperature at a test frequency of 1 kHz, for different concentrations of NPGOL. Note that the presence of a smaller process designated as β - process along with the α - and γ - processes.

SIF₅. Behavior of NPOL + NPGOL binary system for $x_m = 0.13$. Temperature variation of the (a) real and (b) imaginary parts of the complex permittivity at various test frequencies. The phase designated as S₁ is the solid solution.

SIF₆. Double logarithmic plot of ε'' vs. frequency at different temperatures for various relaxation processes of NPOL + NPGOL binary system with $x_m = 0.13$. (a) α -, & β -process (b) γ -process. The solid line corresponds to the HN- equation (2) for the α -process, dashed line corresponds to the CC-fit for the resolved β -process in panel (a) and solid line corresponds to the CC-fit for γ -process in panel (b). The rise in the loss at frequencies above 10 kHz in (a) is due to the γ -process. The parameters of eq. 2 for the α - & γ -processes are given in SIT₂.

SIF₇. Double logarithmic plot of ε'' vs. frequency of CNCH + CHC binary system for different temperatures with $x_m = 0.125$: (a) α - process and (b) β - process. The dashed dotted line in panel (a) and solid line in panel (b) corresponds to the HN-parameters shown in SIT₃. For $T = 185.6$ K, the solid line represents the typical ansatz (HN + CC) fit to resolve the α' - and α - processes.