

Supplementary Information

Analytical data of SiPc oligomers

(a) SiPc dimer, HO(SiPcO)₂H:

Anal. Calcd for C₉₆H₉₈N₁₆O₃Si₂: C, 72.97; H, 6.25; N, 14.18. Found: C, 73.95; H, 6.75; N, 13.19.

MALDI-TOF-mass m/z = 1579.1 (M⁺), UV-vis (Toluene) λ/nm (logε) 639.5 (5.5), 342.0 (5.1)

(b) SiPc trimer, HO(SiPcO)₃H:

Anal. Calcd for C₁₄₄H₁₄₆N₂₄O₄Si₃: C, 73.25; H, 6.23; N, 14.24. Found: C, 73.32; H, 6.65; N, 13.24(4).

MALDI-TOF-mass m/z = 2359.6 (M⁺), UV-vis (Toluene) λ/nm (logε) 626.5 (5.5), 335.5 (5.4)

(c) SiPc tetramer, HO(SiPcO)₄H:

Anal. Calcd for C₁₉₂H₁₉₄N₃₂O₅Si₄: C, 73.39; H, 6.22; N, 14.26. Found: C, 74.06; H, 6.27; N, 13.32.

MALDI-TOF-mass m/z = 3139.1 (M⁺), UV-vis (Toluene) λ/nm (logε) 621.5 (5.6), 330.5 (5.3)

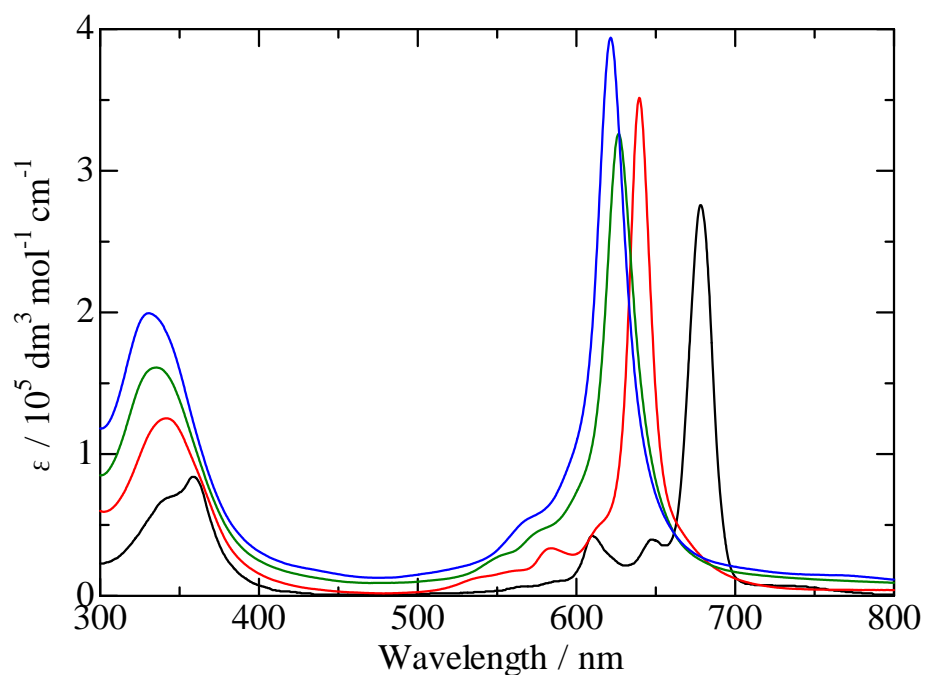


Fig. S1 Ground-state absorption spectra of SiPc oligomers (dimer: red line, trimer: green line, tetramer: blue line) in toluene in addition to a spectrum of the corresponding monomer (black line). The Q band peak (monomer: 678 nm, dimer: 639.5 nm, trimer: 626.5 nm, tetramer: 621.5 nm) shifts to the blue-side with increasing the number of SiPc units, which is explained by the exciton interaction. The ϵ values of the Soret and Q band increase due to oligomerization reflecting an increase in the number of SiPc units, whereas the ϵ value of the Q band is slightly smaller in the trimer than the dimer because of the broad bandwidth of the trimer.

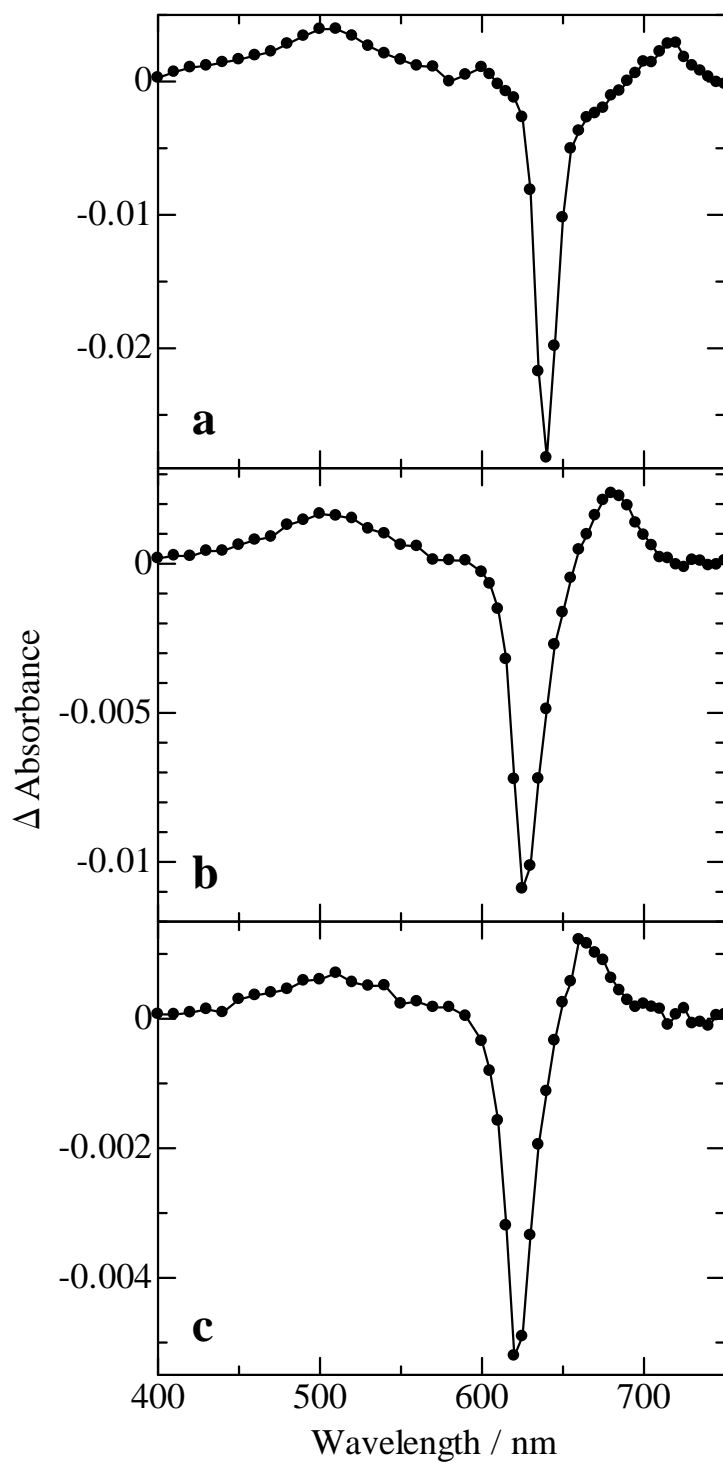


Fig. S2 Transient difference absorption spectra of SiPc oligomers (**a**: dimer, **b**: trimer, **c**: tetramer) in toluene.

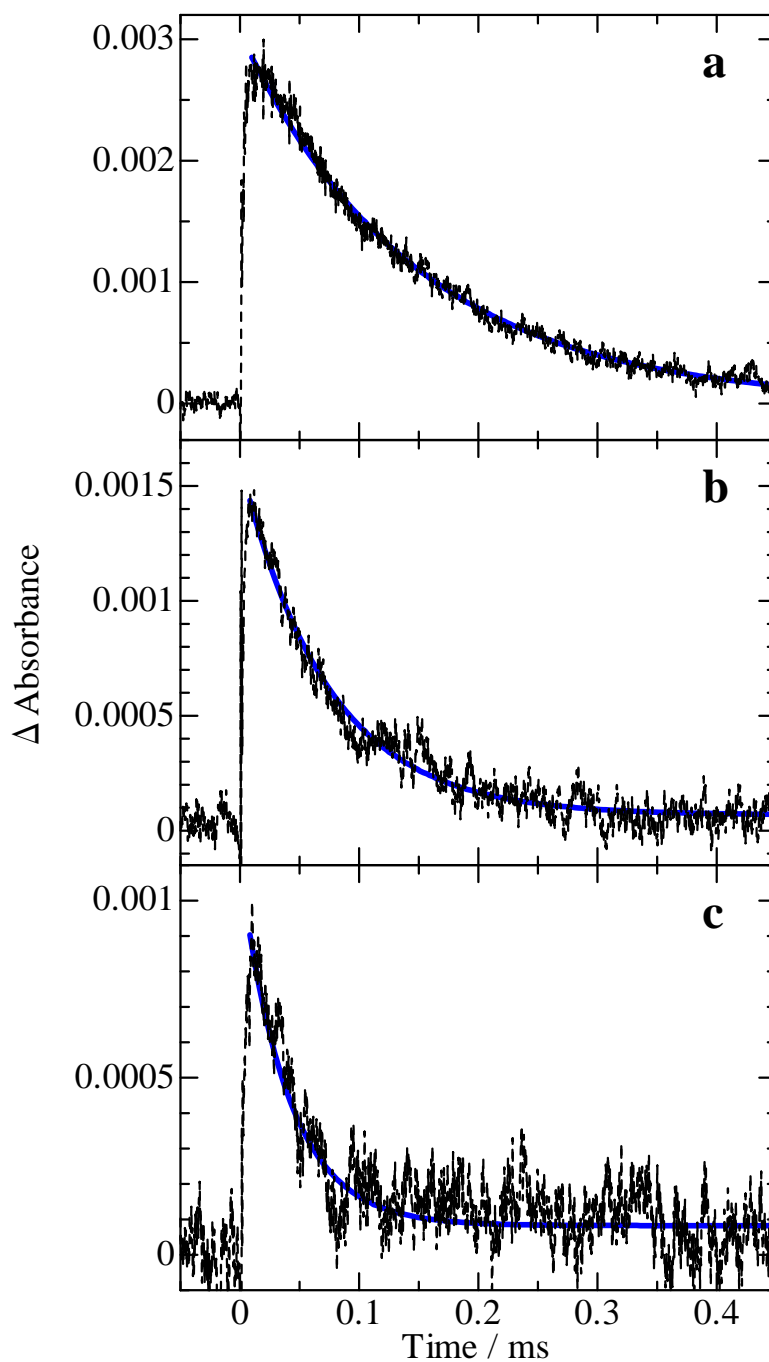


Fig. S3 Time-profiles of transient absorption signals of the dimer (**a**), trimer (**c**), and tetramer (**d**), respectively. The T_1 lifetime decreases in the order monomer ($500 \mu\text{s}$) > dimer ($150 \mu\text{s}$) > trimer ($73 \mu\text{s}$) > tetramer ($40 \mu\text{s}$).

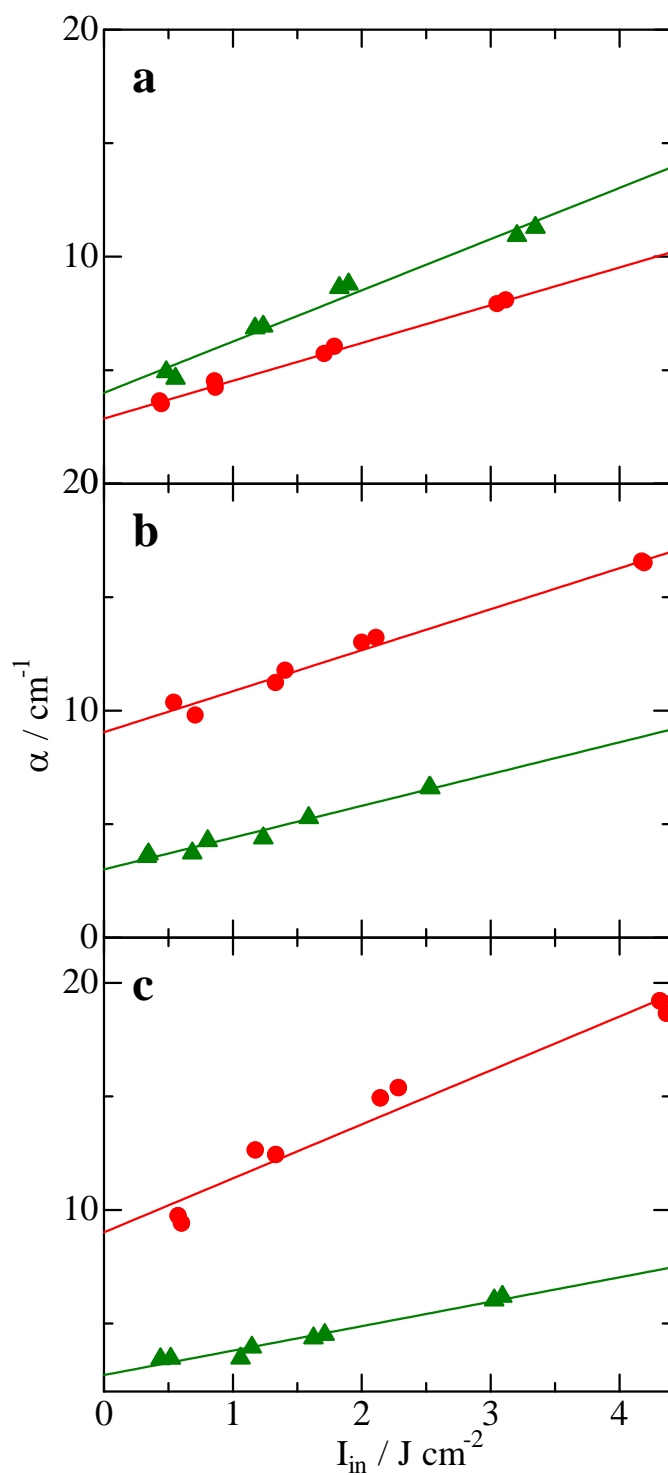


Fig. S4 Relationships between the incident light (I_{in}) and α ($= \alpha_0 + \beta I_{in}$) at 532 nm (▲) and for the excited-state Q band (●: 720nm, 670nm, and 660nm for the dimer, trimer, and tetramer, respectively). The linear (α_0) and non-linear absorption (β) coefficients were evaluated from the gradients and y-intercepts in these figures, respectively.