

Support Information

A formation mechanism of oxygen vacancies in a MnO_2 monolayer: A DFT+U study

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¹⁵ S1: Side view of MnO_2 monolayer ($4 \times 4 \times 1$ supercell)

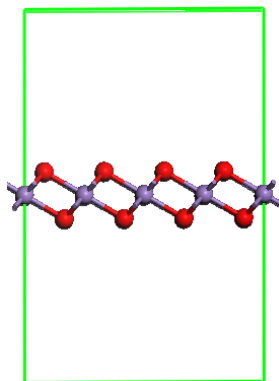
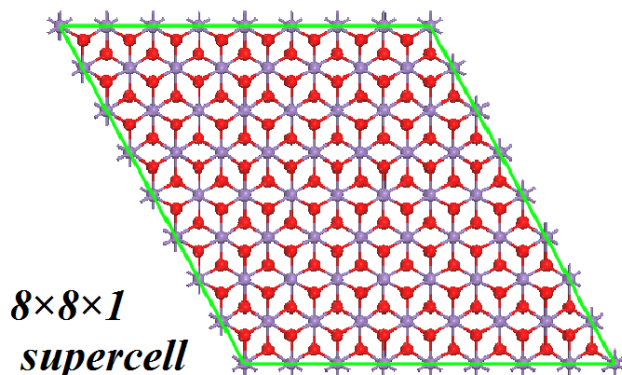


Fig. S1 Side view of perfect MnO_2 nanosheets modelled by $4 \times 4 \times 1$ supercell. Mn and O are indicated

²⁰ by purple and red spheres.

S2: Models of $8 \times 8 \times 1$ supercell (top view)



²⁵ **Fig. S2** Perfect MnO_2 nanosheets modelled by $8 \times 8 \times 1$ supercell. Mn and O are indicated by purple and red spheres..

S3: Calculated reaction energies from $8 \times 8 \times 1$ supercell

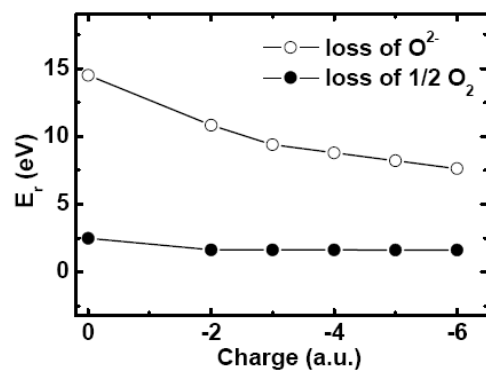


Fig. S3 Reaction energies (E_r) for the release of neutral oxygen molecules and oxygen ions (O^{2-})
s calculated from the $8 \times 8 \times 1$ supercell with $U = 4.0$ eV.