-Electronic Supporting Information-Ultrafast Dynamics in Iron Tetracarbonyl Olefin Complexes investigated with Two-Dimensional Vibrational Spectroscopy

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1 Determination of the T_1 relaxation time of the vibrational modes



Figure S1 Delay-dependent intensity of the $B_2/A_1^{(1)}$ modes of Fe(CO)₄(cinnamic acid) when pumped at 1980, 2034, and 2086 cm⁻¹. The intensity is plotted on a log scale. The curves represent a least-squares fit of a single exponential decay from 14–150 ps. The T_1 lifetime of the overall vibrational relaxation of the carbonyl stretch modes of Fe(CO)₄(cinnamic acid) is 59±5 ps.



Figure S2 Delay-dependent intensity of the $B_2/A_1^{(1)}$ modes of Fe(CO)₄(dimethyl fumarate) when pumped at 2007, 2041, and 2105 cm⁻¹. The intensity is plotted on a log scale. The curves represent a least-squares fit of a single exponential decay from 14–150 ps. The T_1 lifetime of the overall vibrational relaxation of the carbonyl stretch modes of Fe(CO)₄(dimethyl fumarate) is 66 ± 2 ps.





Figure S3 Top: Weights of the principal components obtained from a Singular Value Decomposition (SVD) of the delay-dependence of the cross peaks of $Fe(CO)_4$ (cinnamic acid) (left) and $Fe(CO)_4$ (dimethyl fumarate) (right) 2D-IR spectra. The magnitude of the weights are plotted on a log scale. Bottom: First four target vectors of the SVD of $Fe(CO)_4$ (cinnamic acid) (left) and $Fe(CO)_4$ (dimethyl fumarate) (right). In both complexes, the first three components are the most significant. The delay dependent behavior of the fourth component suggests it is an uncorrelated noise contribution to the data. The target vectors of both complexes are very similar in relative magnitude and delay dependence which indicates their dynamics follow the same mechanism. The $Fe(CO)_4$ (dimethyl fumarate) measurement is noisier compared to that of $Fe(CO)_4$ (cinnamic acid). This is evident from the higher magnitude of the fourth singular value.