# Hybrid QM/QM Simulations of Photochemical Reactions in the Molecular Crystal N-salicylidene-2-chloroaniline

4 Michał A. Kochman,<sup>a</sup> Andrzej Bil,<sup>b,c</sup> Carole A. Morrison<sup>a,\*</sup>

5 <sup>a</sup> School of Chemistry and EaSTCHEM Research School, The University of

6 Edinburgh, King's Buildings, West Mains Road, Edinburgh, EH9 3JJ, UK

<sup>7</sup> <sup>b</sup> Institute of Physical Chemistry, University of Zürich, Winterthurerstrasse
8 190, CH-8057 Zürich, Switzerland

9 <sup>c</sup> Faculty of Chemistry, University of Wrocław, F. Joliot-Curie 14, 50-383
10 Wrocław, Poland

11

12 Celebrating 300 years of Chemistry at Edinburgh.

13

- 14 \* Corresponding author: C.Morrison@ed.ac.uk
- 15

## 16 1 Molecular Packing in the Photochromic 17 Polymorph of SCA

18 In order to further assess the crystal packing of SCA, we have carried 19 out a Voronoi-Dirichlet tesselation<sup>1</sup> of the crystal structure using the 20 computer program TOPOS 4.0.<sup>2</sup> This process assigns to each atom (which 21 in this context is considered to be a point positioned at the nucleus) a 22 Voronoi-Dirichlet polyhedron (VDP) that contains all points whose distance 23 to that atom is not greater than the distance to any other atom. In this 24 manner, space is partitioned completely and disjointly into VDPs associated 25 with individual atoms. (The partitioning is not disjoint in the rigorous sense, 26 because the surface of each polyhedron is shared with the neighbouring

polyhedra, but this fine point can be ignored here.) The VDP of an atom 27 may be roughly interpreted as the region of space taken up by the atom 28 29 and, by extension, the union of VPDs of atoms comprising a molecule may 30 be regarded as the region of space belonging to that molecule. We must bear in mind, however, that although the Voronoi-Dirichlet tesselation 31 32 provides a convenient means to analyze the molecular packing, the VDPs 33 are purely geometric constructs that do not reflect the electronic structure 34 of a crystal, and it would therefore be meaningless to use them to analyze 35 system properties other than the spatial arrangement of atoms. 36

In Figure 1 we show two views of the union of atomic VDPs belonging 37 to a single SCA molecule in the crystal lattice. It can be seen that the 38 39 union of VDPs follows the shape of the molecule quite closely, consisting 40 of two disk-like parts, which correspond to the two aromatic rings, joined at the edge. In particular, no significant bulge or protuberance is visible 41 42 above or below the aromatic rings, indicating that the crystal structure 43 contains no large voids on either side of the rings. This, in turn, suggests that a photoisomerisation mechanism involving a simple rotation around 44 the C3-C4 bond is obstructed by intermolecular steric interactions, whereas 45 46 the pedal-motion mechanism of Harada *et al.*, in which the two aromatic rings are less strongly displaced from their original positions, seems likely to 47 be able to proceed with less steric hindrance. 48

49

50

Figure 1: Atomic VDPs of an SCA molecule in the crystal lattice. Above each view of the VDPs, the SCA molecule is displayed to scale and in the same orientation.



Note that in addition to the photochromic polymorph of SCA whose crystal structure is discussed above, a non-photochromic SCA polymorph has recently been reported<sup>3</sup> in which the molecules are planar and  $\pi$ -stacked. The density calculated from the crystal cell and contents of the non-photochromic polymorph at room temperature is slightly higher than that of the photochromic polymorph, at 1.405 g/cm<sup>3</sup> and 1.359 g/cm<sup>3</sup>, respectively.

## 57 2 Construction of the 1×1×2 Supercell 58 Containing a *Trans*-keto Molecule

In the experimentally determined structure of the *trans*-keto tautomer of N-3,5-di-tert-butylsalicylidene-3-nitroaniline embedded in a lattice consisting predominantly of *cis*-enol molecules, the *trans*-keto molecule is positioned very similarly to the original *cis*-enol molecule.<sup>4</sup> Hence, in order to generate the starting geometry of the supercell containing a single *trans*-keto molecule, one molecule in the experimental crystal structure was converted from the *cis*-enol into the *trans*-keto form by manually moving the atoms H1, H2, C3 and N1 while preserving the position of the two aromatic rings. The structure obtained in this way was subsequently optimised at the periodic DFT level of theory (see Section 3.4.2 of the main body of the present 69 paper for a detailed description of the parameters of the DFT calculation).

70~ The default BFGS optimisation algorithm was applied, with the following

71 convergence criteria: energy tolerance  $2.0 \times 10^{-5}$  eV/atom, maximum force 72 tolerance  $5.0 \times 10^{-2}$  eV/Å, displacement tolerance  $1.0 \times 10^{-3}$  Å. The resulting

72 tolerance  $5.0 \times 10^{-2} \text{ eV/A}$ , displacement tolerance  $1.0 \times 10^{-3}$  A. The resulting 73 optimised geometry of the supercell containing one *trans*-keto molecule and

seven *cis*-enol molecules is shown in Figure 6(b) in the main body of the

75 present paper.

#### 76 **References**

- 77 [1] V. A. Blatov, *Cryst. Rev.*, 2004, **10**, 249.
- 78 [2] TOPOS website: http://www.topos.ssu.samara.ru/
- [3] K. Johmoto, T. Ishida, A. Sekine, H. Uekusa, Y. Ohashi, *Acta Cryst.*,
   2012, B68, 297.
- [4] J. Harada, H. Uekusa, Y. Ohashi, J. Am. Chem. Soc., 1999, 121,
  5809.

#### Snapshots from Trajectory 1

t = 0 fs  $E(S_1) - E(S_0) = 3.636 \text{ eV}$ 



$$t = 5 \text{ fs}$$
  $E(S_1) - E(S_0) = 3.316 \text{ eV}$ 



$$t = 10 \text{ fs}$$
  $E(S_1) - E(S_0) = 2.923 \text{ eV}$ 



$$t = 15 \text{ fs}$$
  $E(S_1) - E(S_0) = 3.139 \text{ eV}$ 



$$t = 20 \text{ fs}$$
  $E(S_1) - E(S_0) = 3.356 \text{ eV}$ 



$$t = 25 \text{ fs}$$
  $E(S_1) - E(S_0) = 2.751 \text{ eV}$ 



$$t = 30 \text{ fs}$$
  $E(S_1) - E(S_0) = 2.297 \text{ eV}$ 



$$t = 35 \text{ fs}$$
  $E(S_1) - E(S_0) = 2.251 \text{ eV}$ 



$$t = 40 \text{ fs}$$
  $E(S_1) - E(S_0) = 2.291 \text{ eV}$ 



$$t = 240 \text{ fs}$$
  $E(S_1) - E(S_0) = 2.429 \text{ eV}$ 



$$t = 250 \text{ fs}$$
  $E(S_1) - E(S_0) = 2.120 \text{ eV}$ 



$$t = 260 \text{ fs}$$
  $E(S_1) - E(S_0) = 2.185 \text{ eV}$ 



$$t = 270 \text{ fs}$$
  $E(S_1) - E(S_0) = 1.496 \text{ eV}$ 



$$t = 280 \text{ fs}$$
  $E(S_1) - E(S_0) = 1.752 \text{ eV}$ 



$$t = 290 \text{ fs}$$
  $E(S_1) - E(S_0) = 1.036 \text{ eV}$ 



$$t = 300 \text{ fs}$$
  $E(S_1) - E(S_0) = 0.768 \text{ eV}$ 



$$t = 310 \text{ fs}$$
  $E(S_1) - E(S_0) = 0.264 \text{ eV}$ 



#### Snapshots from Trajectory 3

t = 0 fs  $E(S_1) - E(S_0) = 3.699 \text{ eV}$ 



$$t = 10 \text{ fs}$$
  $E(S_1) - E(S_0) = 2.986 \text{ eV}$ 



$$t = 20 \text{ fs}$$
  $E(S_1) - E(S_0) = 3.096 \text{ eV}$ 



$$t = 30 \text{ fs}$$
  $E(S_1) - E(S_0) = 2.960 \text{ eV}$ 



$$t = 40 \text{ fs}$$
  $E(S_1) - E(S_0) = 2.880 \text{ eV}$ 



$$t = 50 \text{ fs}$$
  $E(S_1) - E(S_0) = 2.854 \text{ eV}$ 



$$t = 60 \text{ fs}$$
  $E(S_1) - E(S_0) = 1.675 \text{ eV}$ 



$$t = 70 \text{ fs}$$
  $E(S_1) - E(S_0) = 1.586 \text{ eV}$ 



$$t = 80 \text{ fs}$$
  $E(S_1) - E(S_0) = 0.399 \text{ eV}$ 



$$t = 90 \text{ fs}$$
  $E(S_1) - E(S_0) = 0.502 \text{ eV}$ 



$$t = 100 \text{ fs}$$
  $E(S_1) - E(S_0) = 0.325 \text{ eV}$ 



$$t = 110 \text{ fs}$$
  $E(S_1) - E(S_0) = 0.896 \text{ eV}$ 



$$t = 120 \text{ fs}$$
  $E(S_1) - E(S_0) = 0.073 \text{ eV}$ 



#### Snapshots from Trajectory 3'

t = 440 fs  $E(S_1) - E(S_0) = 1.987 \text{ eV}$ 



### t = 450 fs $E(S_1) - E(S_0) = 1.516 \text{ eV}$



$$t = 460 \text{ fs}$$
  $E(S_1) - E(S_0) = 1.617 \text{ eV}$ 



$$t = 470 \text{ fs}$$
  $E(S_1) - E(S_0) = 1.421 \text{ eV}$ 



$$t = 480 \text{ fs}$$
  $E(S_1) - E(S_0) = 1.335 \text{ eV}$ 



$$t = 490 \text{ fs}$$
  $E(S_1) - E(S_0) = 1.720 \text{ eV}$ 



$$t = 500 \text{ fs}$$
  $E(S_1) - E(S_0) = 1.254 \text{ eV}$ 



#### t = 510 fs $E(S_1) - E(S_0) = 1.865 \text{ eV}$



#### t = 520 fs $E(S_1) - E(S_0) = 1.437 \text{ eV}$



$$t = 530 \text{ fs}$$
  $E(S_1) - E(S_0) = 1.388 \text{ eV}$ 



#### t = 540 fs $E(S_1) - E(S_0) = 1.060 \text{ eV}$



#### t = 550 fs $E(S_1) - E(S_0) = 0.927 \text{ eV}$



#### t = 560 fs $E(S_1) - E(S_0) = 0.691 \text{ eV}$



#### t = 570 fs $E(S_1) - E(S_0) = 0.310 \text{ eV}$



#### t = 580 fs $E(S_1) - E(S_0) = 0.509 \text{ eV}$



$$t = 590 \text{ fs}$$
  $E(S_1) - E(S_0) = 0.253 \text{ eV}$ 

